

Combining the power of **computational thermochemistry** with the convenience of **Python** programming:

My experience with **ChemAppPy**

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1. My background

- I am metallurgical engineer specialised in modeling, control and optimization of metallurgical furnaces (steelmaking, non-ferrous metallurgy)
- 10 years of experience using **thermochemical** application
 - **FactSage**™ regularly
 - **SimuSage** for dynamic process modeling, in a close cooperation with GTT (oxygen converter, lead smelting in the TSL process)
- limited experience with coding:
 - applications developed mainly by co-workers (programmers)
 - moderate experience using the software **Matlab**:
 - university days
 - mathematical process modeling + solving numerical problems (with assistance)

2. Why ChemAppPy?

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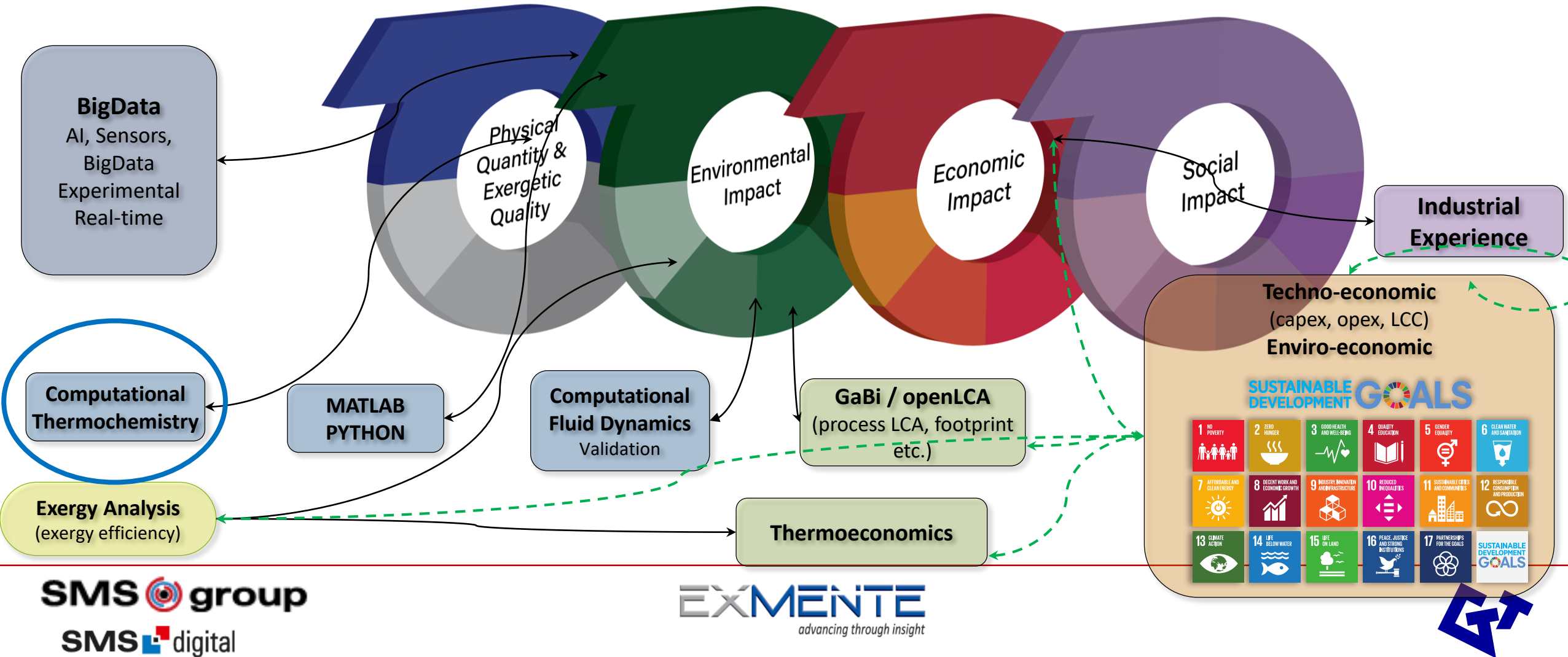
➤ FactSage

- a **powerful** tool that I use on a daily basis for evaluating the thermodynamic equilibrium state relevant to our processes
- there has always been a need to use FactSage calculation options as an integrated module into a **programming environment** so that we can
 - **automate** thermochemical calculation tasks
 - **inputs** for the calculations
 - the **post-processing** of the results
 - integrate computational thermochemistry into existing **digital platforms** in SMS group

➡ those possibilities are provided by: **ChemApp** + **ChemAppPy**

2. Why ChemAppPy?

Digital TWINNING resource systems of SMS: Integrating our expertise, in-depth understanding of technology, and theoretical knowledge into **digital platforms**



2. Why ChemAppPy?

- **ChemApp** provides the flexibility that we are seeking
 - is an API (application programmer's interface)
 - However, there was a **language** issue!
 - It has an interface to **C** and **Fortran**
 - For someone with a little programming experience, learning **those** coding languages is challenging
 - this was the case for me during learning **DELPHI** and **C#**
 - made me question the **effort-benefit** since I will use those programs only „occasionally“
 - **Python** was recommended to me as a much easier to learn coding language

2. Why ChemAppPy?

- **ChemAppPy**: became commercially available in 2019
 - made **ChemApp** available in a **Python** environment
 - we considered this as a chance to achieve our goals
 - The original functions in **ChemApp** are now grouped into few **classes**, which are easier to remember and work with:

Info, Units, ThermochemicalSystem, EquilibriumCalculation, StreamCalculation, PhaseMapCalculation
 - Further additional functions are included in order to
 - make the **calculations** easier and quicker to do
 - **post-processing** and **visualizing** the calculation results.

3. What I gained so far

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➤ Programming with Python:

- using thermochemistry in Python made me part of a large online community from which there is a lot to learn and exchange!
- The effort to learn Python was considerably less than for other coding languages:
 - Python-[Training](#) is one part of ChemAppPy-[Training](#): The fundamental concepts + the basic functions needed for ChemAppPy are introduced.
 - It took me about [2 weeks](#) with an effort of [2-4 h/day](#) to get the basics of Python and be able to work with it by myself
 - once getting over the „[hurdle](#)“ of learning how to code
 - I can now use many powerful python packages such as [Pandas](#) (data preparation and analysis) and [Matplotlib](#) (visualization)

3. What I gained so far

- Enjoying not only the power but also the **friendliness** of ChemAppPy
 - The calculation functions are grouped into few **Classes**
 - when the class is called, a list of the corresponding functions appears:

- The indexing adopted is consistent throughout the program (**A** for amount, **ph** for phase, **pc** for phase const,.....)

➡ focus more on **what** I want to do rather on **how** to do it

➡ less learning effort Vs. more **comfort** and **fun** working with the program!

EquilibriumCalculation.

```
m set_IA_pc(cls, cls_1, ph, p... AbsoluteEquilibriumCalculationBase
m get_result_object(cls, cls_... AbsoluteEquilibriumCalculationBase
m get_eq_A_ph(cls, cls_1, ph) EquilibriumCalculationBase
m get_eq_A(cls, cls_1) EquilibriumCalculationBase
m calculate_eq(cls, cls_1, print_results,... EquilibriumCalculation
m set_eq_P(cls, cls_1, double_value) EquilibriumCalculationBase
m set_eq_T(cls, cls_1, double_value) EquilibriumCalculationBase
m set_eq_AC_pc(cls, cls_1, ph... AbsoluteEquilibriumCalculationBase
m calculate_eq_IA(cls, cls_1, list_IA_ran... EquilibriumCalculation
m calculate_eq_P(cls, cls_1, P_guess, pri... EquilibriumCalculation
m calculate_eq_T(cls, cls_1, T_guess, pri... EquilibriumCalculation
m calculate_eq_V(cls, cls_1, V_guess, pri... EquilibriumCalculation
Press Strg+. to choose the selected (or first) suggestion and insert a dot afterwards Next Tip
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Conclusion

- The need to flexibly use thermochemistry in SMS has been there for a while
- The development and availability of ChemAppPy brought us closer to achieving our goals
 - we were now able to automate the thermochemical calculations and thus work more efficiently
 - we are heading into our goal of integrating ChemAppPy into our **digital platforms** in the scope a close cooperation between:
 - SMS group: R&D + Non-Ferrous Process and Technology
 - SMS digital
 - GTT and Ex Mente

Thank you for your attention!

Questions?  please contact me under:

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