

## Application of thermochemistry for the prediction of slagging in coal fired boilers



S. Guhl, M. Klinger, R. Schimpke, C. Hommel, D. Vogt, P. Gehre, D. Kirschenmann, D. Safronov,  
B. Meyer, T. Brunne\*

\*Lausitz Energie Kraftwerke AG

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# R&D Project KORRISTENT

## KORRISTENT

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- **time:** 01/01/2016 - 31/12/2019
- **project ID:** 03ET7066A

## Cooperation partner

- TU Bergakademie Freiberg
  - Institute of Energy Process Engineering and Chemical Engineering (IEC)
  - Institute of Analytical Chemistry (IAC)
  - Institute of Ceramic, Glass and Construction Materials (IKGB)
- Lausitz Energie Kraftwerke AG

## Objective

Development of solutions for reduced slagging of coal fired boilers especially for critical fuels

## Introduction

# Power station Boxberg

### Issue

- 900 MW el., pulverised fuel firing systems, tower type boiler, **blend of RW/NO lignite**
- **Slagging of boiler wall**, sintered deposits at “Schott” (first horizontal tube bundle heat exchanger)
- Automated cleaning system (water cannons) keeps slagging manageable
- **decreasing coal quality in future** requires R&D

### Work plan

- Analysing context between coal/ash properties, operation parameters and slagging based on process samples
  - Chemical-mineralogical characterization of process samples (XRF, REM-EDX/WDX + EBSD, XRD, CT)
  - Development and application of ICP-OES with electro-thermal evaporation
  - experimental investigations (impedance spectroscopy, simulation of slagging in drop tube reactor)
- Modelling via thermochemistry and CFD
- providing applicable proposal for solution

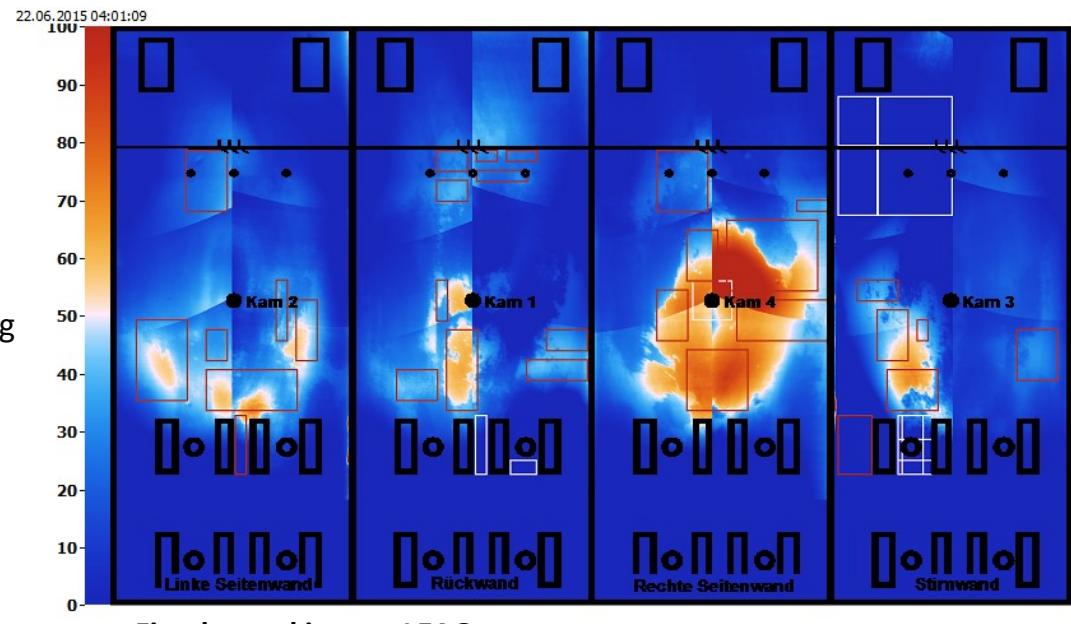


Fig: thermal image, LEAG

### Sample pool

Sample period	blend RW/NO	coal	fly ash	wet ash	deposits
May/June 2015	50/50...70/30...100/0	-	78	-	1 aluminosilicatic slagged
October 2015	70/30	<b>42 shift samples</b>	<b>42 shift samples</b>	<b>42 shift samples</b>	-
December 2015	60/40	-	-	-	1 Fe-dominated
July 2016	100/0	1 mixed sample	48	103	1 aluminosilicatic slagged (large slagged lumps) + 34 Fe-dominated
November 2016	0/100	1 mixed sample	-	-	-
August 2017	50/50	-	-	-	17 Fe-dominated
December 2016	100	coals and sediments from opencast mine RW			

## Coal (set of 42 shift samples RW/NO 70/30)

Oxid	Average Ma.-%	Min. Ma.-%	Max. Ma.-%
<b>SiO<sub>2</sub></b>	<b>39,8</b>	<b>30,0</b>	<b>50,2</b>
<b>Al<sub>2</sub>O<sub>3</sub></b>	<b>14,7</b>	<b>7,7</b>	<b>20,1</b>
<b>Fe<sub>2</sub>O<sub>3</sub></b>	<b>12,2</b>	<b>7,6</b>	<b>15,3</b>
<b>CaO</b>	<b>12,2</b>	<b>7,7</b>	<b>15,8</b>
MgO	2,5	1,8	3,2
SO <sub>3</sub>	16,2	10,4	22,5
Na <sub>2</sub> O	0,2	0,2	0,2
K <sub>2</sub> O	1,1	0,6	1,8
TiO <sub>2</sub>	1,1	0,7	1,5
Summe	100		

Mineral	Formel	Ma.-% coal wf
<b>Quartz</b>	<b>SiO<sub>2</sub></b>	<b>10-13</b>
<b>Kaolinite</b>	<b>Al<sub>2</sub>(Si<sub>2</sub>O<sub>5</sub>)(OH)<sub>4</sub></b>	<b>5-11</b>
Muscovite	KAl <sub>2</sub> [(OH) <sub>2</sub>   AlSi <sub>3</sub> O <sub>10</sub> ]	0-9
<b>Pyrite</b>	<b>FeS<sub>2</sub></b>	<b>2-5</b>
Marcasite	FeS <sub>2</sub>	max. 0,5
Anatase	TiO <sub>2</sub>	1-2
Gypsum	CaSO <sub>4</sub> * 2H <sub>2</sub> O	0-3

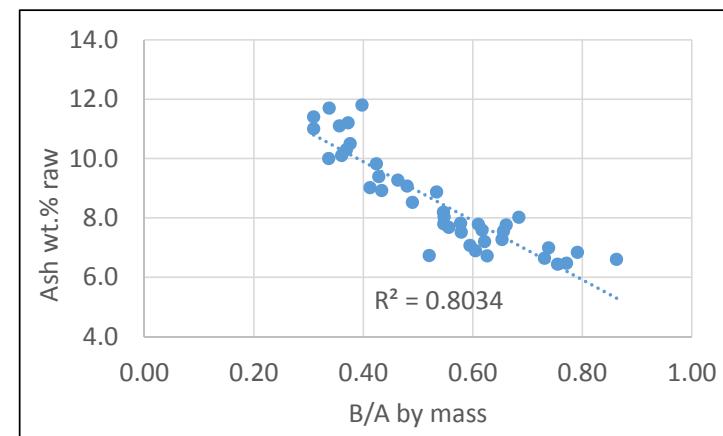
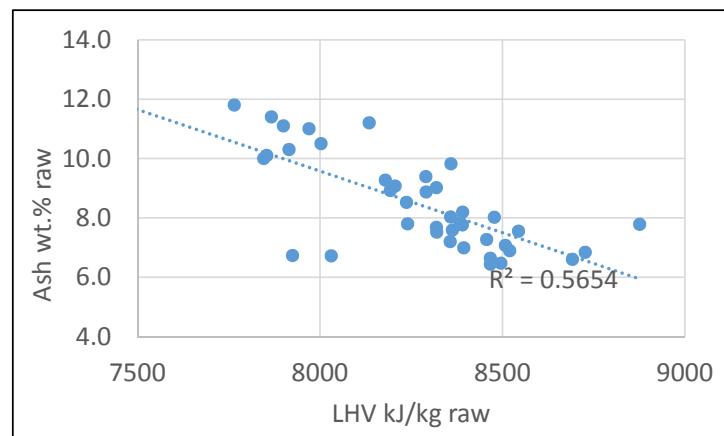
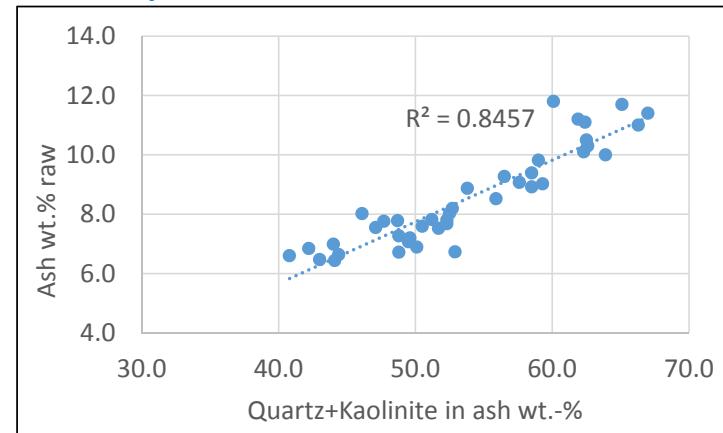
- Ash content 6–12 Ma.-% raw, B/A = 0,3–0,9
- Quartz, clay minerals and pyrite dominate
- Al, Si, 50% Fe, K, Ti → extraneous ash (70%), sediments
- Ca, Mg, 50% Fe → inherent ash (30%), humates
- Flow temperature in red. atmosphere in average 1330°C

→ highly fluctuating coal properties

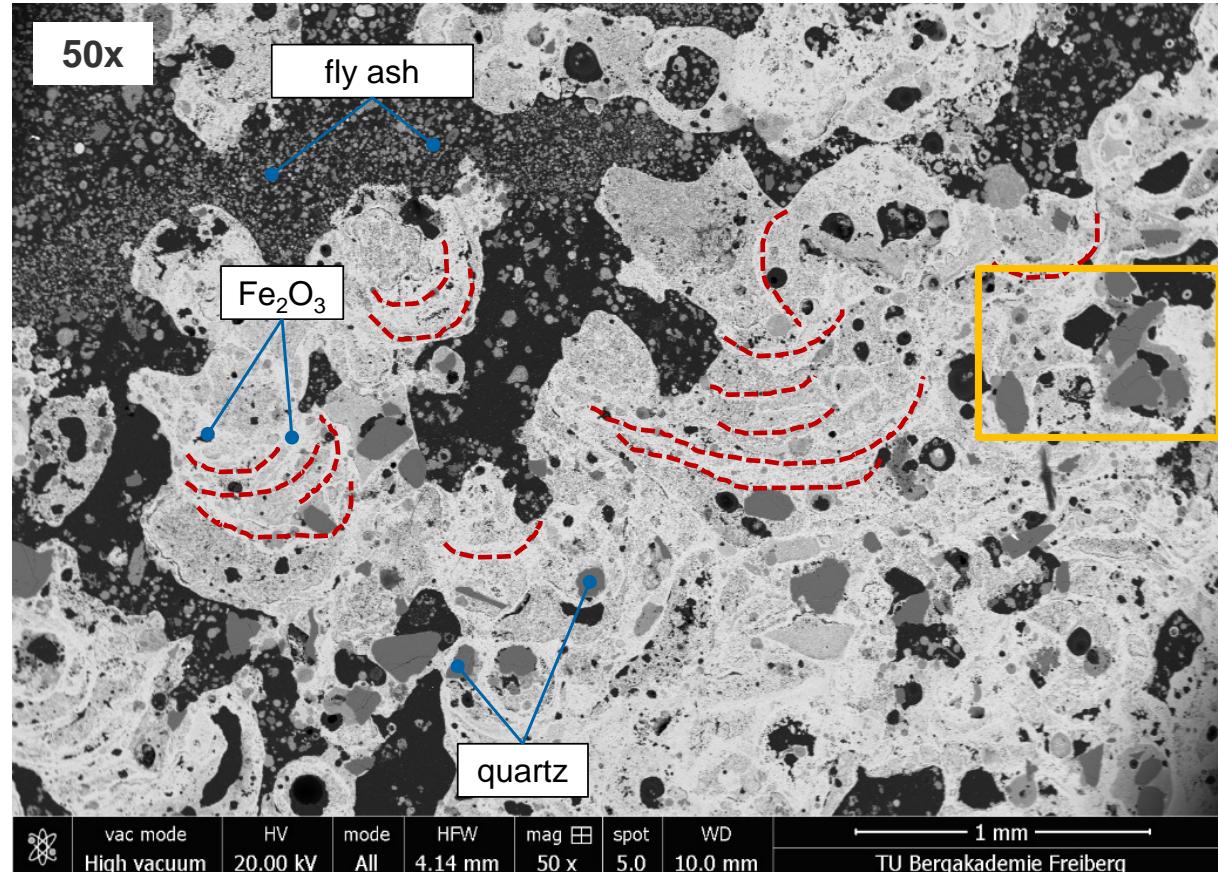
$$\frac{\text{Base}}{\text{Acid}} = \frac{\text{Na}_2\text{O} + \text{K}_2\text{O} + \text{CaO} + \text{MgO} + \text{Fe}_2\text{O}_3}{\text{SiO}_2 + \text{Al}_2\text{O}_3 + \text{TiO}_2}$$

### Coal (set of 42 shift samples RW/NO 70/30)

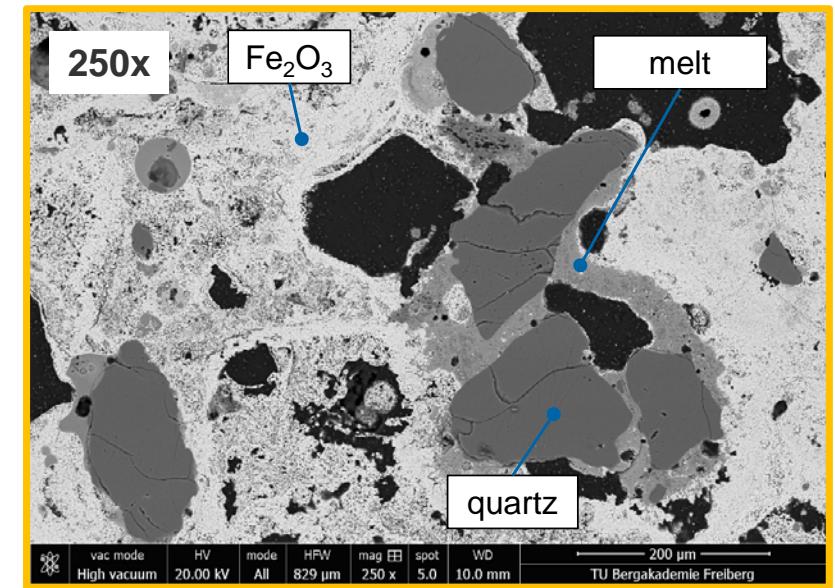
- Sediments quartz and clay minerals (kaolinite, illite, muscovite) determine ash content, LHV and B/A
- quartz and kaolinite can be calculated via XRF-Data and stoichiometry (proven by direct measurement via XRD)



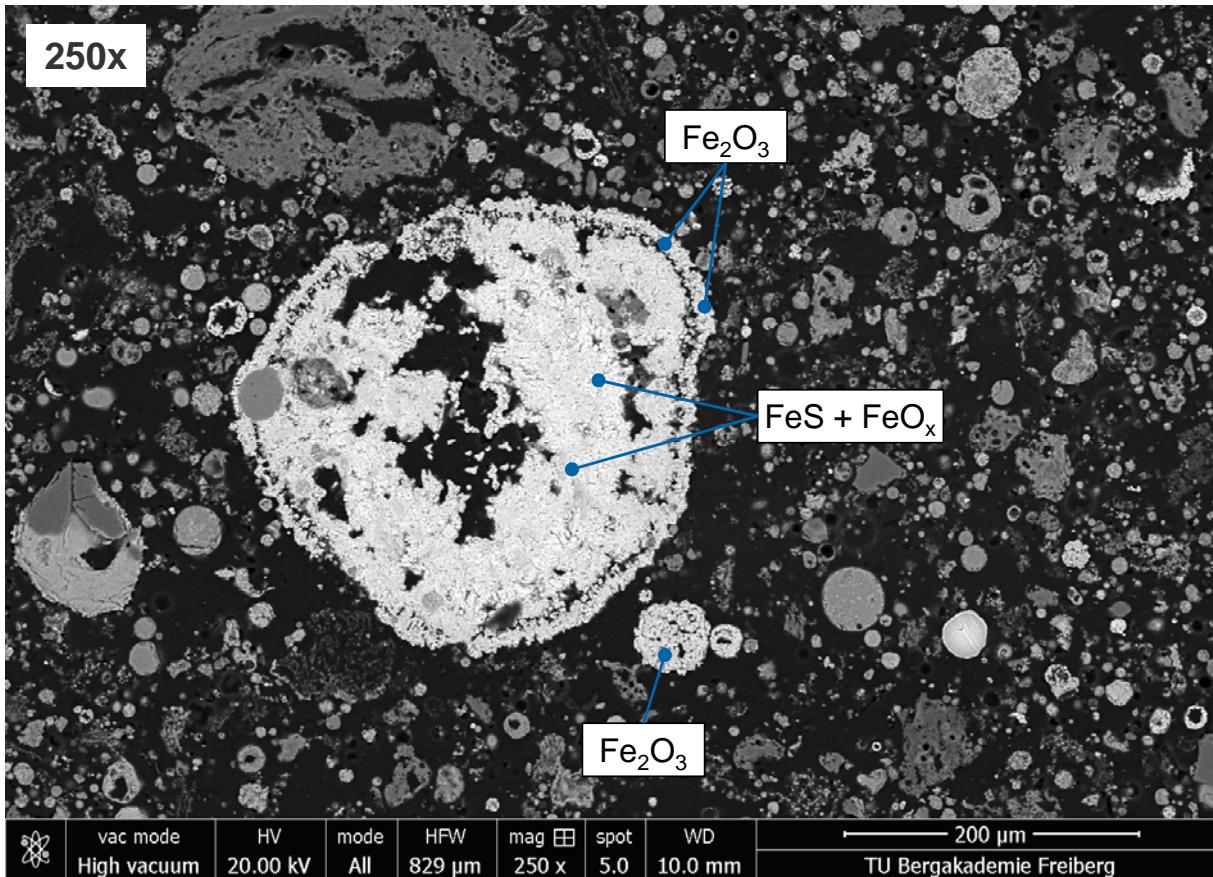
## Fe-dominated deposit from boiler wall – SEM



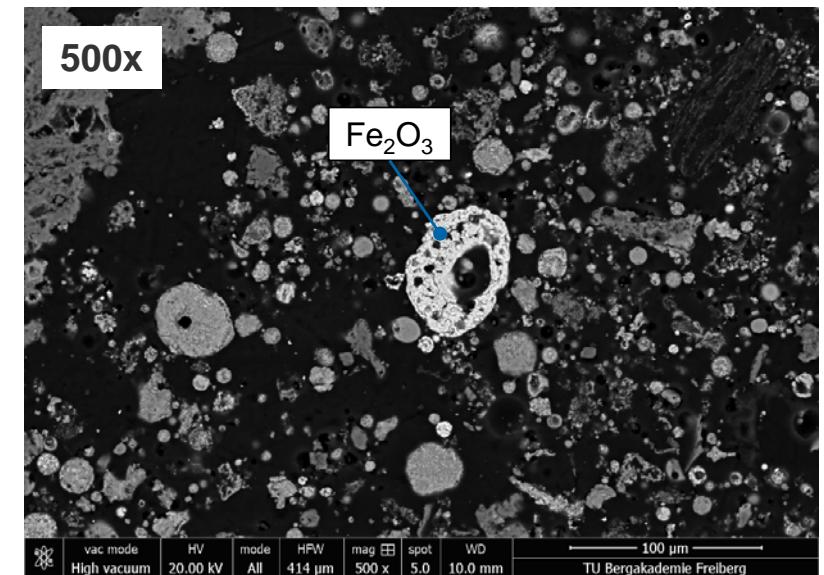
- **$\text{Fe}_2\text{O}_3$ -Matrix** (layers, formerly melt) with inclusions of aluminosilicate slag and quartz grains
- intermediate layers from fly ash (unspecific composition)



## Fly ash



- $\text{Fe}_2\text{O}_3$  particles
- in large Fe-rich particles FeS-rich centre
- due to particle form formerly liquid



### Wet ash and „Schlackestürze“ (large slagged lumps)

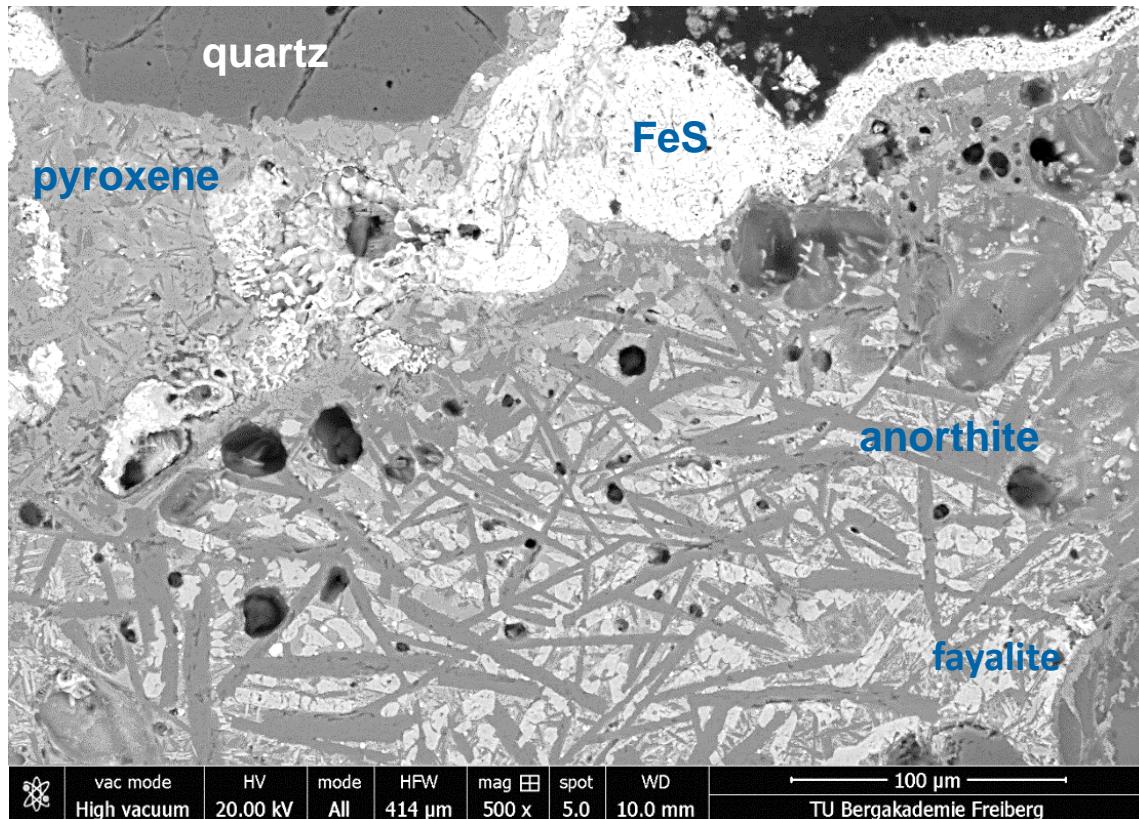


Fig.: coarser wet ash

- relicts of quartz and clay minerals in **recrystallized matrix** ( $\text{CaFeSi}_2\text{O}_6$  and  $\text{CaMgSi}_2\text{O}_6$ ), anorthite ( $\text{CaAl}_2\text{Si}_2\text{O}_8$ ), fayalite ( $\text{Fe}_2[\text{SiO}_4]$ )),
- FeS-inclusions
- same composition for aluminosilicate slagged lumps

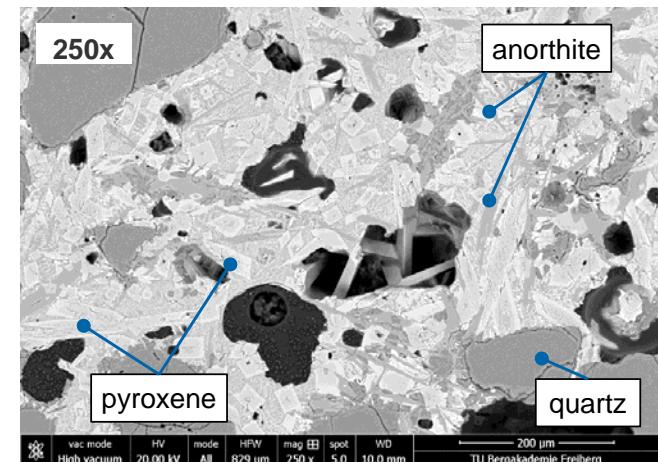
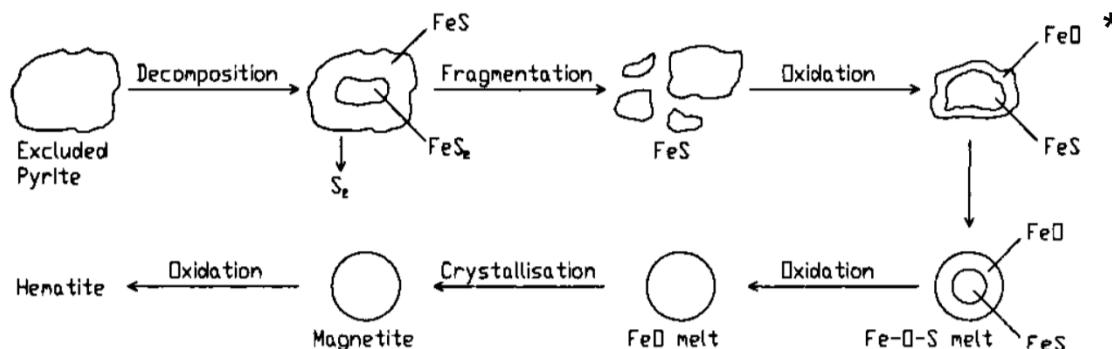


Fig.: Schlackestürz

## Mechanism of deposit/slag formation

- formation of FeO-FeS-melt from pyrite
- formation of initial layer on boiler wall, collection of fly ash particles
- mechanism 1: oxidizing of FeO-FeS-melt forms  $\text{Fe}_2\text{O}_3$  (see SEM results of Fe-dominated deposits)
- mechanism 2: secondary reaction of FeO-FeS-melt with fly ash particles results in aluminosilicate melt with relicts of quartz and decomposed clay minerals (see SEM results of coarser wet ash and *Schlackestürze*)

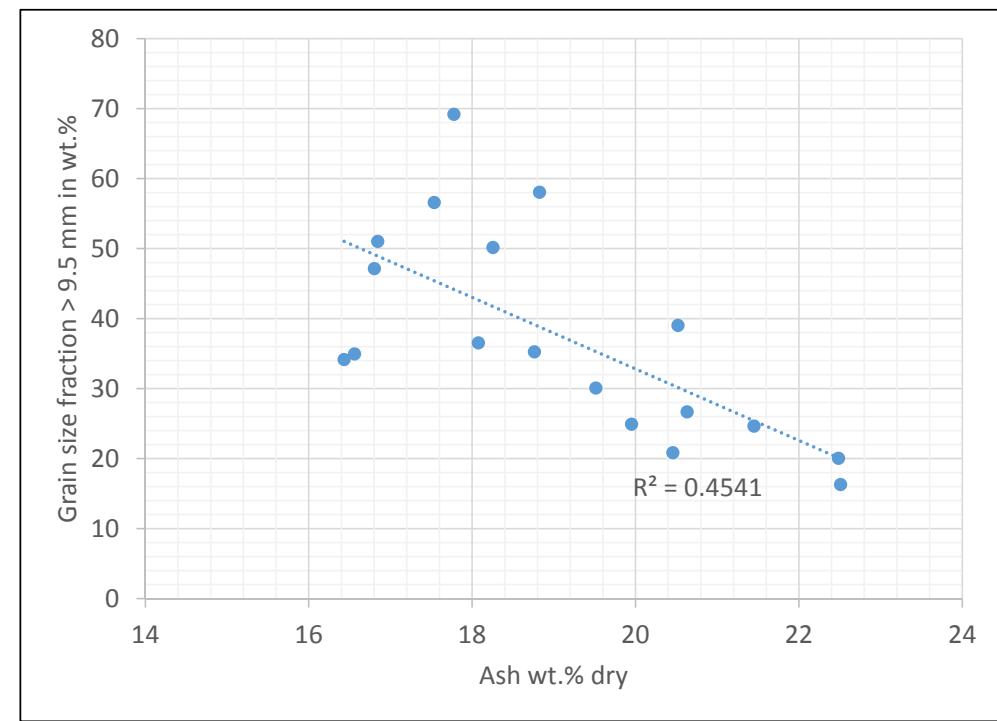
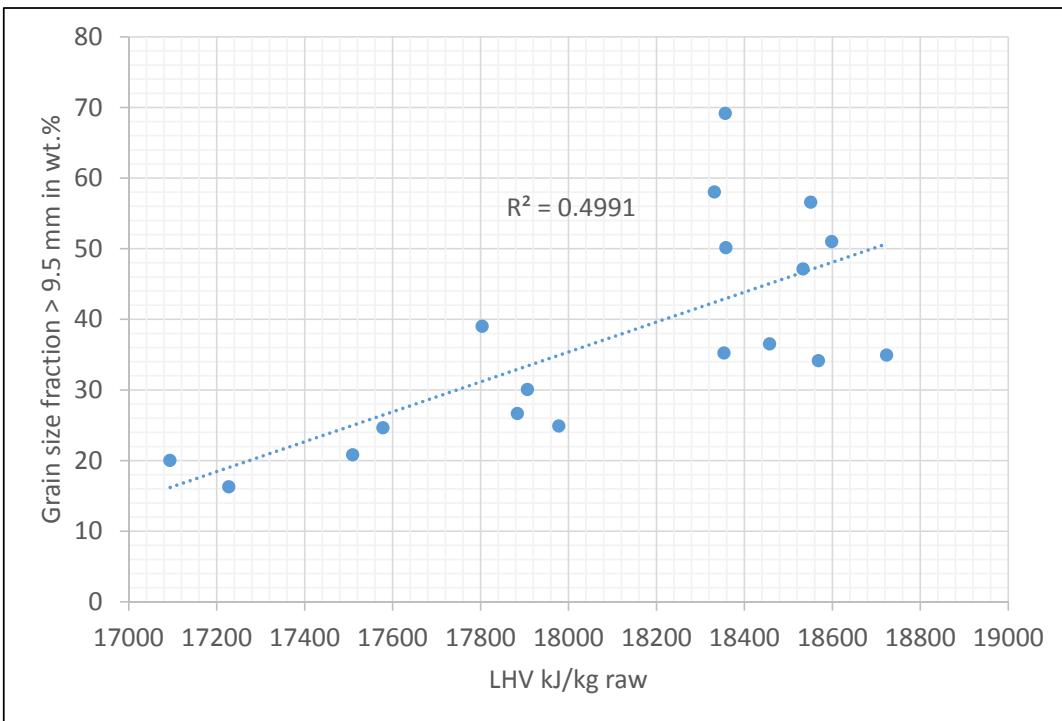


- coarser wet ash fraction corresponds to cleaned slag/deposit from boiler wall  
 → definition of slag index: relative amount of wet ash fraction > 9.5 mm

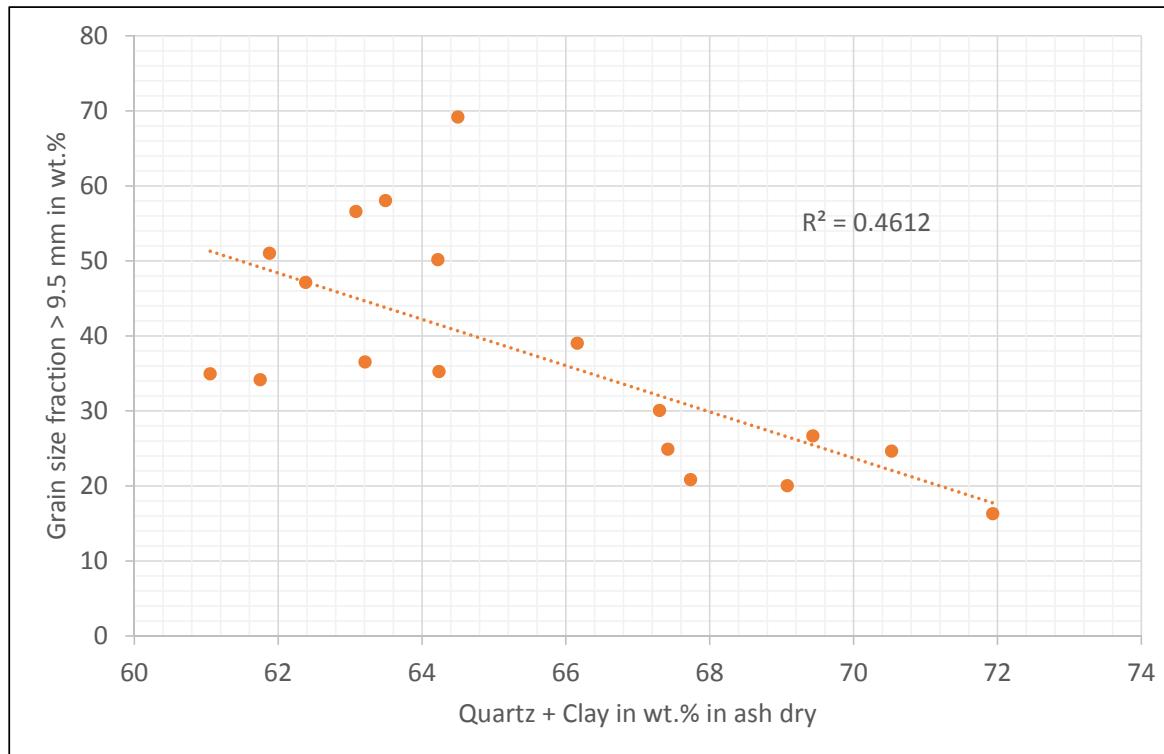
\*McLennan et al.: Ash formation mechanisms during pf combustion in reducing conditions. Energy Fuels 2000, 14 (1), 150-159

## Correlations between slagging and coal properties

- in general a mass balances (coal ash = wet ash + fly ash) is difficult to achieve (Representative samples? Frequency of sampling vs frequency of coal property change? Sampling point before coal distributor!) → moving average over two shifts

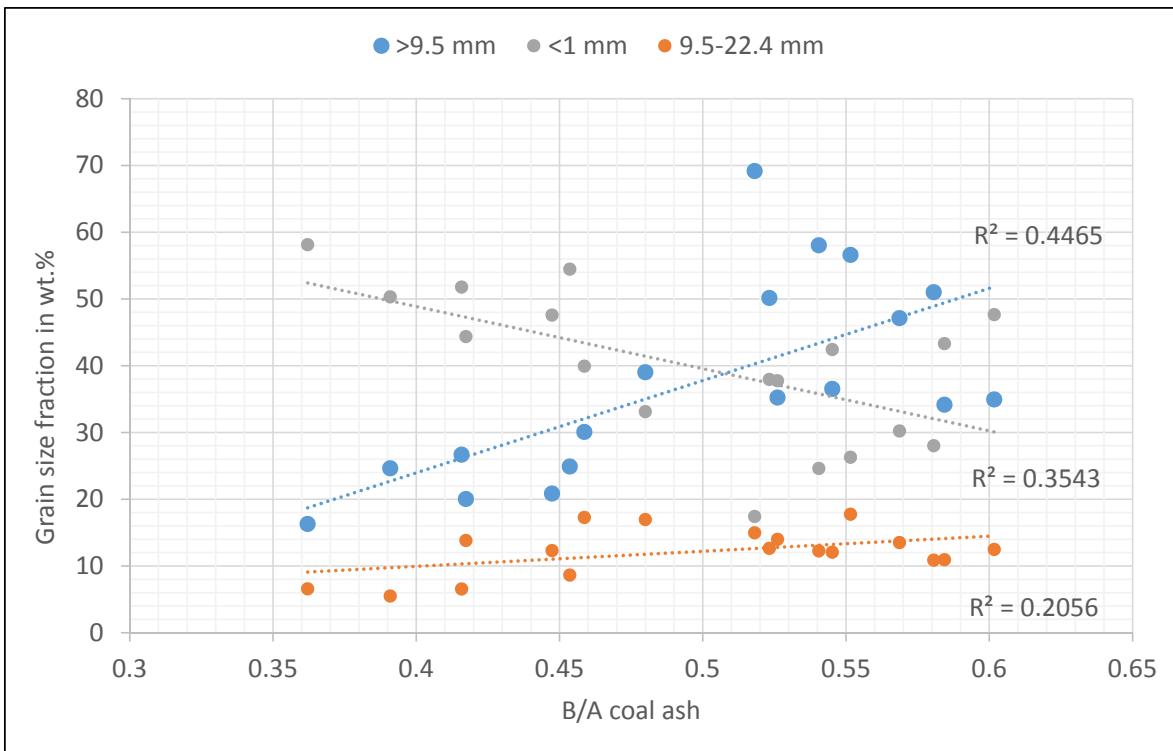


## Correlations between slagging and coal properties



- Salgging ~ heating value
- Salgging ~ 1/ash content
- Salgging ~ 1/(quartz + clay minerals)

## Correlations between slagging and coal properties

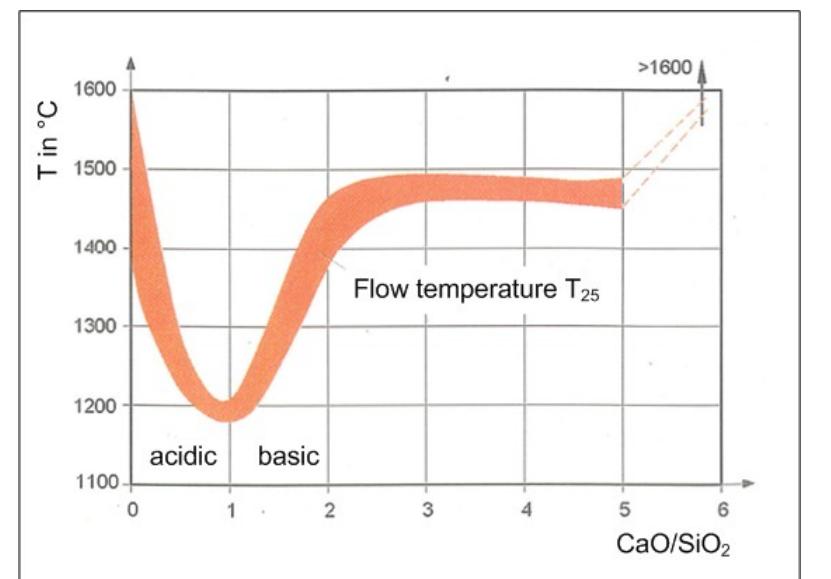


- **Slagging ~ B/A**
- B/A represents indirectly also other slagging sensitive parameters (ash, quartz+clay, LHV, pyrite)
- B/A connects ash composition with temperature of slag formation ( $T_{\text{liq}} = f(\text{B/A})$ )

$$\frac{\text{Base}}{\text{Acid}} = \frac{\text{Na}_2\text{O} + \text{K}_2\text{O} + \text{CaO} + \text{MgO} + \text{Fe}_2\text{O}_3}{\text{SiO}_2 + \text{Al}_2\text{O}_3 + \text{TiO}_2}$$

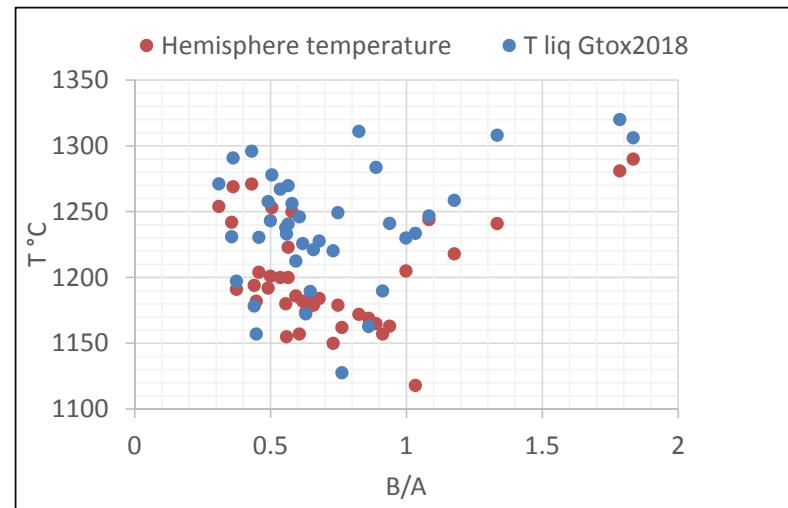
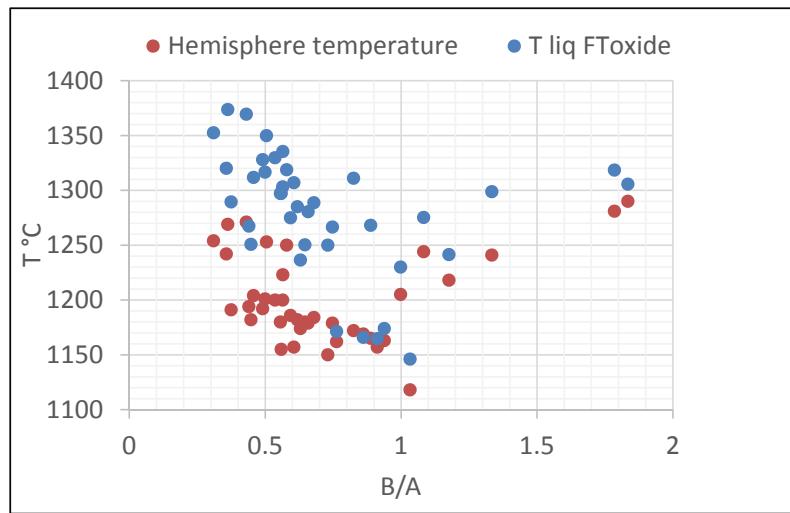
# Basic calculations

- Database: Factsage 7.3 (FToxic; Gtox v2018)
- Calculations in SimuSage (adjustment of T for 100% slag)
- Input:
  - raw coal composition
  - air–fuel ratio = 0.9
  - no flue gas recirculation considered
- validation against AFT (hemisphere temperature at CO/CO<sub>2</sub> 40/60 by vol.) and well established correlation of  $T_{liq}=f(B/A)$



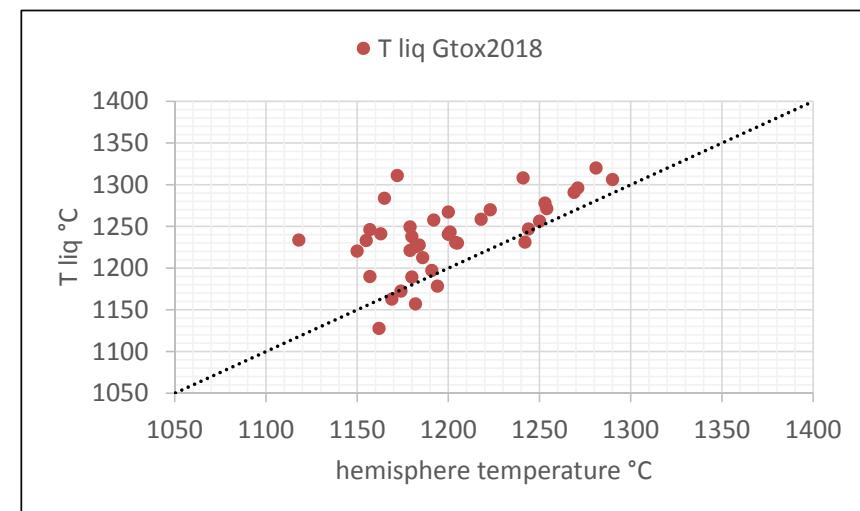
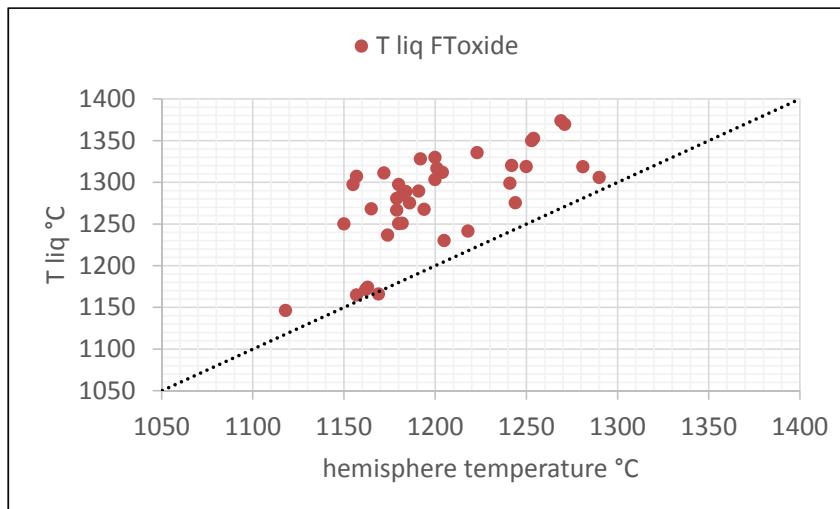
\* modified from: J. Schmalfeld: Die Veredelung und Umwandlung von Kohle - Technologien und Projekte 1970-2000 in Deutschland. Hamburg: Deutsche Wissenschaftliche Gesellschaft für Erdöl, Erdgas und Kohle e.V.. 2008

## Basic calculations



- correlation of  $T_{liq} = f(B/A)$  in general visible for both databases

## Basic calculations



- Reproducibility of AFT-temperatures for reducing atmosphere 80 K (according to ISO 540: 1995E)
- robust (simple) prediction of slagging as function of temperature and coal ash composition possible
- not considered: special role of pyrite, inert amounts of quartz and clay (mullite)

## Extended Model

- FactSage 7.2, FactPS, Ftoxic, FTmisc
- Input data:
  - fuel analysis of 42 coal samples (LEAG-shift data: ultimate analysis, disulphide, XRF of ash)
  - XRF of wet ash fraction > 9.5 mm (average)
- high degree of freedom (air-fuel ratio, inert components, T – range):
  - modeling FeO-FeS-melt
  - reduce number of free parameters (T, reactive SiO<sub>2</sub>-content)
  - apply model on 42 coal samples

### 1. calculation: pyrite induced melts

Input streams:

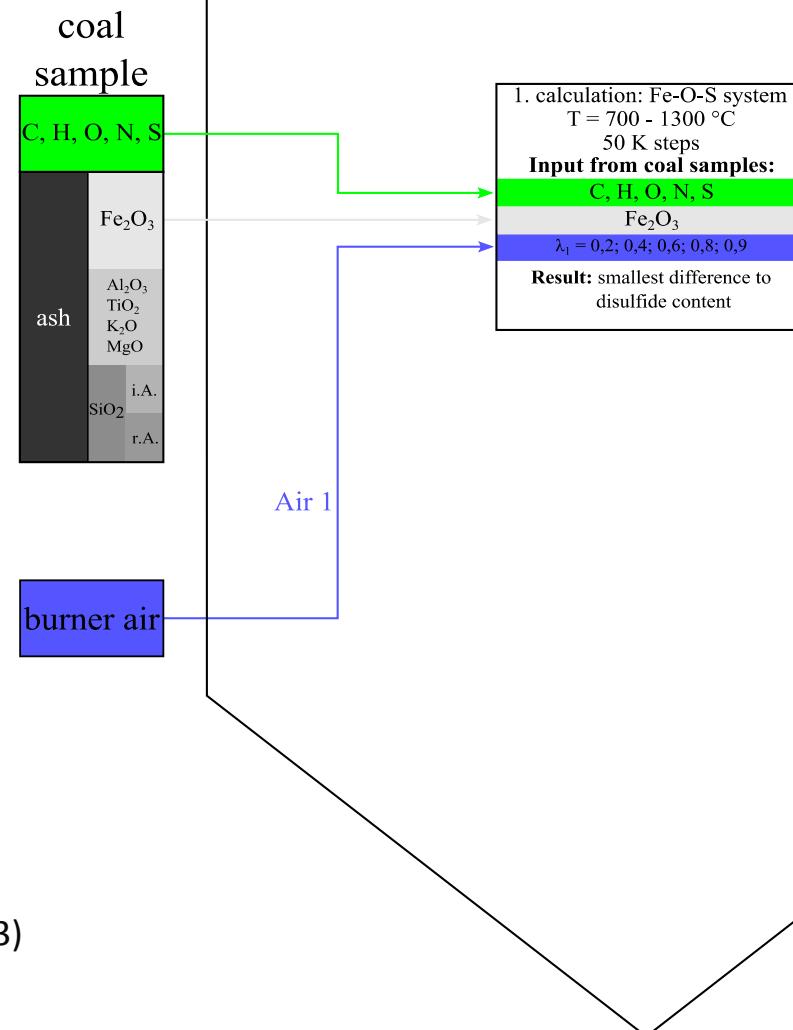
- C, H, O, N, S and water of respective coal sample
- Fe content of the ash ( $\text{Fe}_2\text{O}_3$ )
- Combustion air

Parameter variation:

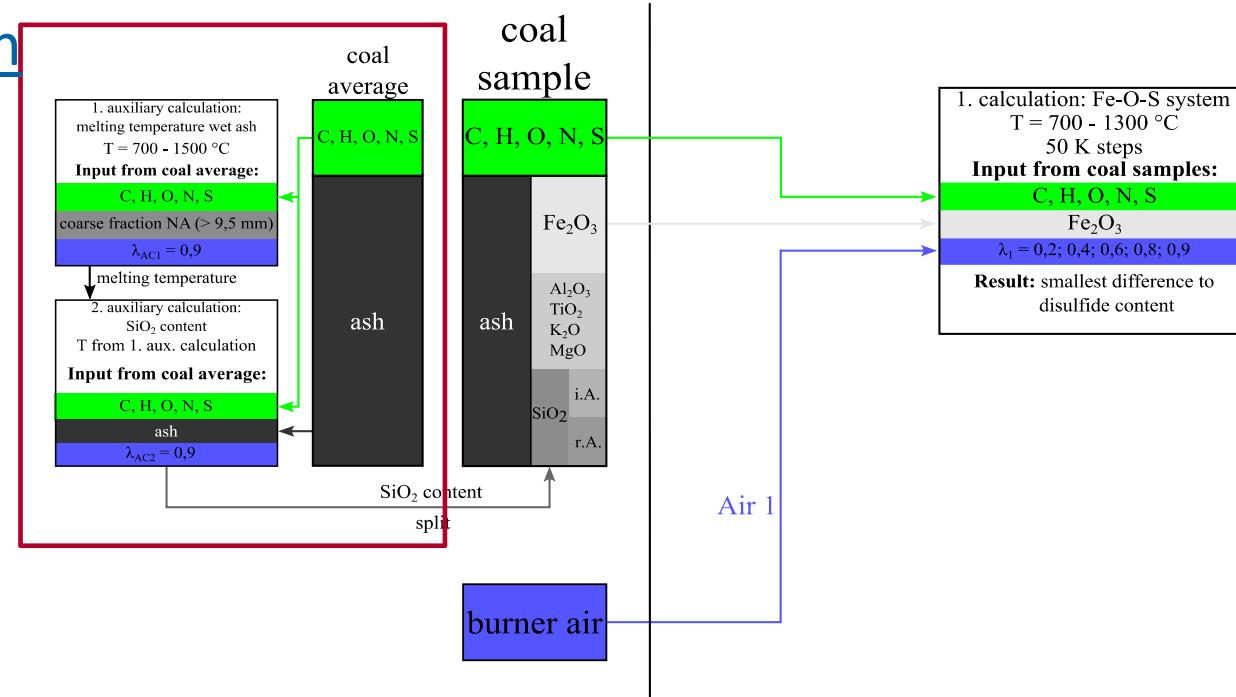
- Air ratio  $\lambda_1 = 0,2; 0,4; 0,6; 0,8; 0,9$
- Temperature:  $T = 700 - 1300 \text{ }^\circ\text{C}$
- variation of T and  $\lambda_1$  to minimize the difference between the amounts of
  - disulphide in the coal sample and
  - sulphur in the resulting liquid phase

Result:

$T$  and  $\lambda$  of  $\text{FeO}-\text{FeS}$ -melt formation (ca.  $950-1050 \text{ }^\circ\text{C}$ ,  $\lambda = 0,2-0,3$ )



## Auxiliary calculation

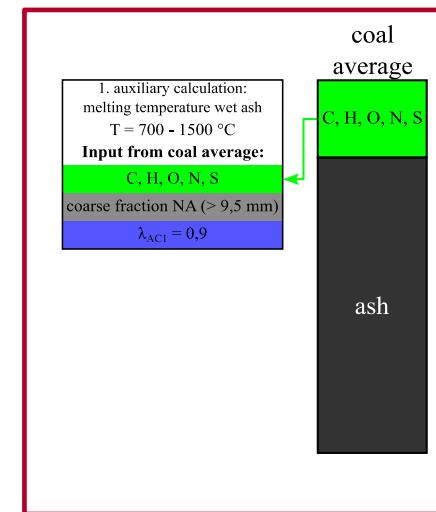


## Auxiliary calculation

### 1. melting temperature of the average slagged wet ash (>9.5mm)

Input:

- C, H, O, N, S and water of the average coal sample
- Ash composition and content of the coarse wet ash fraction > 9,5 mm
- Combustion air  $\lambda_{AC1} = 0,9$  (burner belt section)
- Variation of temperature  $T = 700 - 1500 \text{ }^{\circ}\text{C}$  for 100% slag



Result: melting temperature of the average slagged wet ash  
 $1153 \text{ }^{\circ}\text{C}$

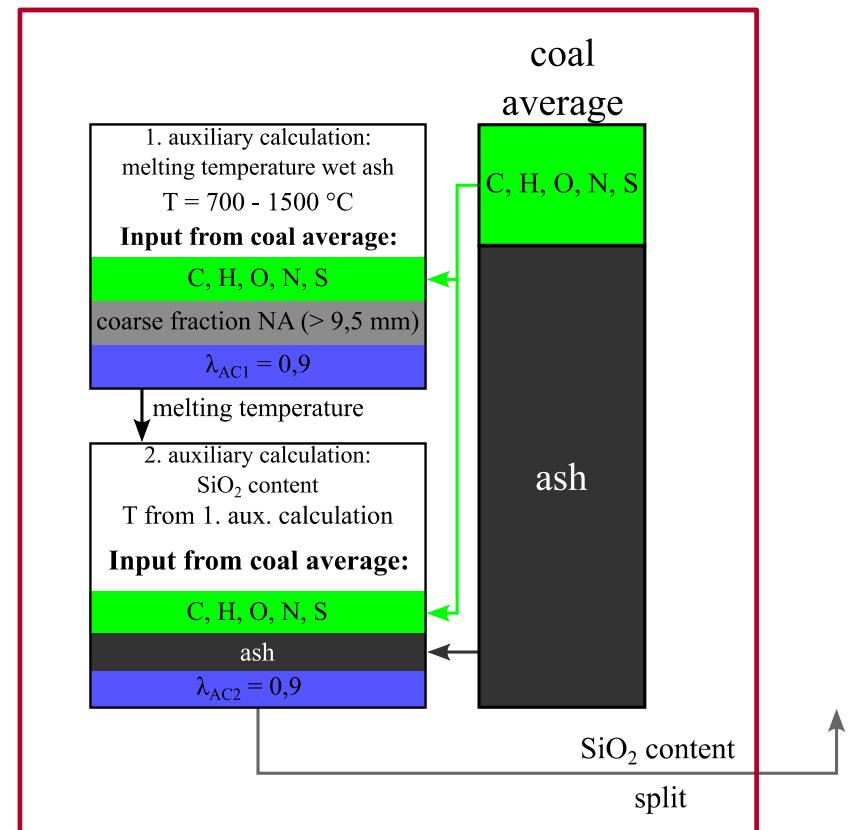
## Auxiliary calculation

### 2. reactive SiO<sub>2</sub> content

Input:

- C, H, O, N, S and water of the average coal sample
- ash composition of average coal sample
- Combustion air  $\lambda_{AC2} = 0,9$
- Temperature: melting temperature from 1. auxiliary calculation
- Variation of the reactive SiO<sub>2</sub> amount to minimize the difference between the composition of the coarse wet ash fraction and the liquid phase

Result: reactive SiO<sub>2</sub> content of 43 wt. %

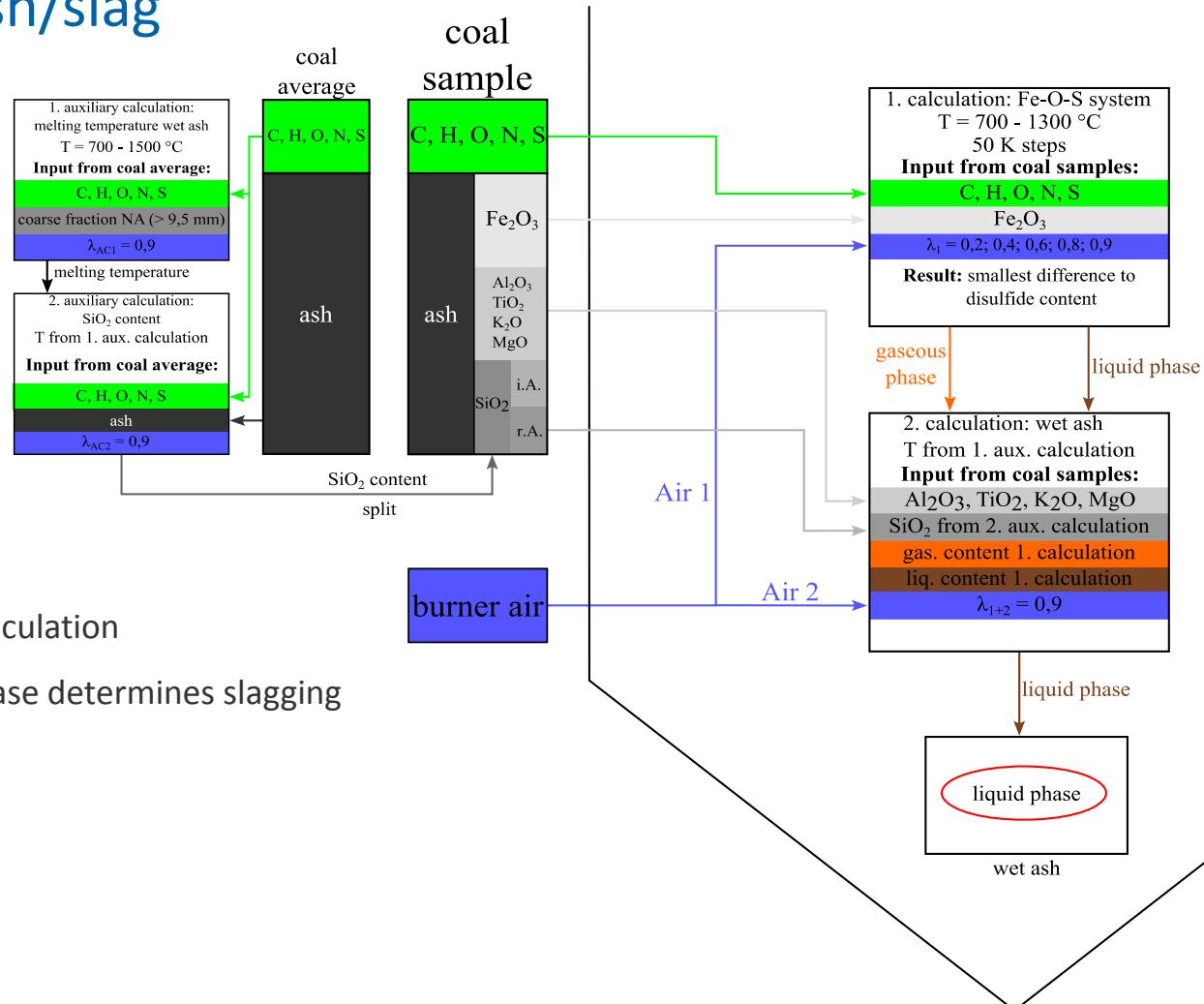


## 2. calculation: wet ash/slag

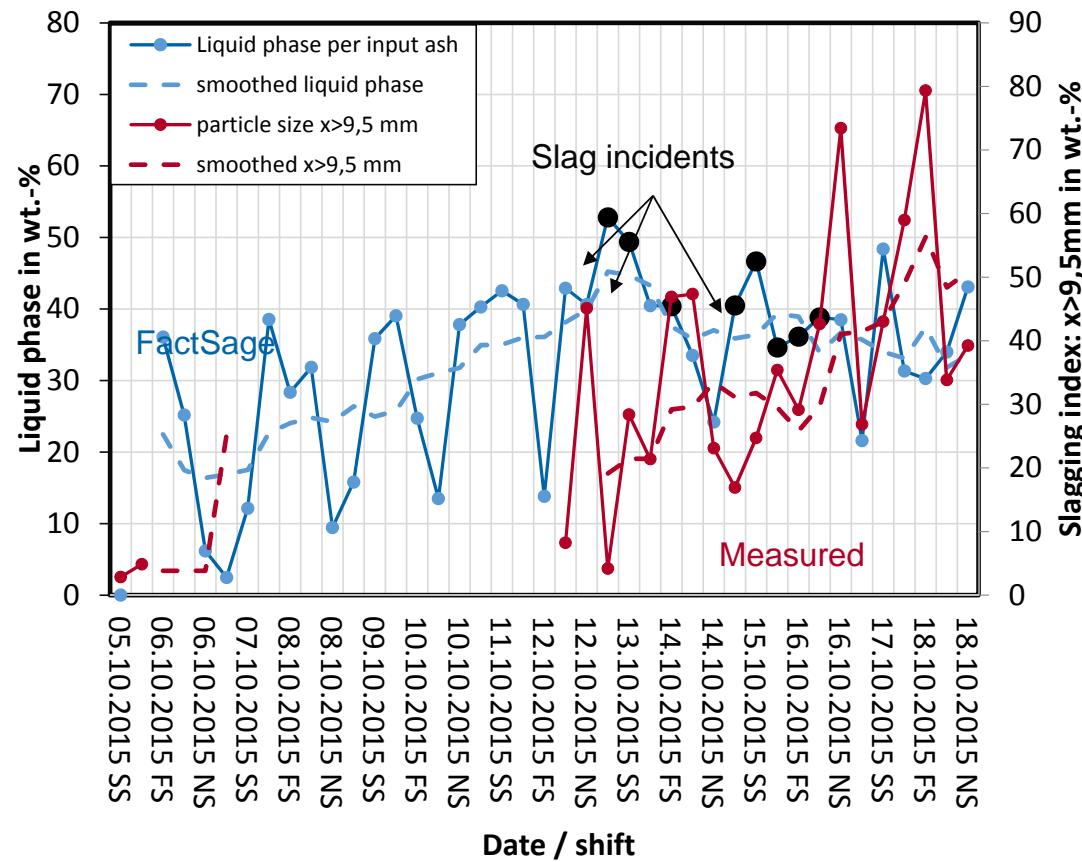
Input:

- 42 shift coal samples
- gaseous and liquid phases from 1. calculation
- Remaining ash components of coal samples (excl. Fe)
- reactive amount  $\text{SiO}_2$  content based on 2. auxiliary calculation
- Combustion air  $\lambda_{1+2} = 0.9$
- Temperature from 1. auxiliary calculation

Result: amount of slag phase determines slagging tendency



## Calculated slag phase vs Slagging index and Slag incidents



- relations between coal properties and slagging extracted ( $\text{slagging} = f(\text{LHV, quartz, clay, pyrite})$ )
- B/A allows for robust estimation of slagging propensity, whereas  $\text{slagging} \sim \text{B/A}$  within investigated range of composition
- application thermochemistry allows prediction of T-dependent slag formation
- further improvement of thermochemical model by
  - analysis of inert components (quartz, clay in wet ash and fly ash via XRD)
  - improved estimation of temperature (as function of LHV and boiler load)
- practical application of model (or key figure):
  - simplification of model (have to be based on measurable coal and operational parameters)
  - implementation of ETV-ICP-OES as online measurement (multielement analysis, indirectly detection of quartz, clay minerals and pyrite)

## Gefördert durch:



aufgrund eines Beschlusses  
des Deutschen Bundestages

### Contact:

TU Bergakademie Freiberg

Dr.-Ing. Stefan Guhl

Institute of Energy Process Engineering and Chemical Engineering  
Fuchsmühlenweg 9  
D-09599 Freiberg

Tel: +49 3731 39 4485

E-Mail: stefan.guhl@iec.tu-freiberg.de