



ChemApp Applications in Multiphysics and Discrete Event Simulations

T. Marin-Alvarado, M4Dynamics
GTT-User's Meeting 2018





Overview

- Background
 - M4Dynamics
 - Multiphysics Simulation
 - Discrete Event Simulation
- M4Dynamics™ COMSOL® ChemApp® Interface
 - Multiphysics and thermochemistry

M4D-CCI™
- M4Dynamics™ Arena® ChemApp® Interface
 - Discrete Event Simulation and thermochemistry

M4D-ACI™



Background – M4Dynamics

- Based in Toronto, since 2015:
 - Consulting in Multiphysics and Thermodynamic based Process Simulation
 - Software development
 - Training
 - Custom physics models, COMSOL Application Development and Hosting



Discrete Event Simulation

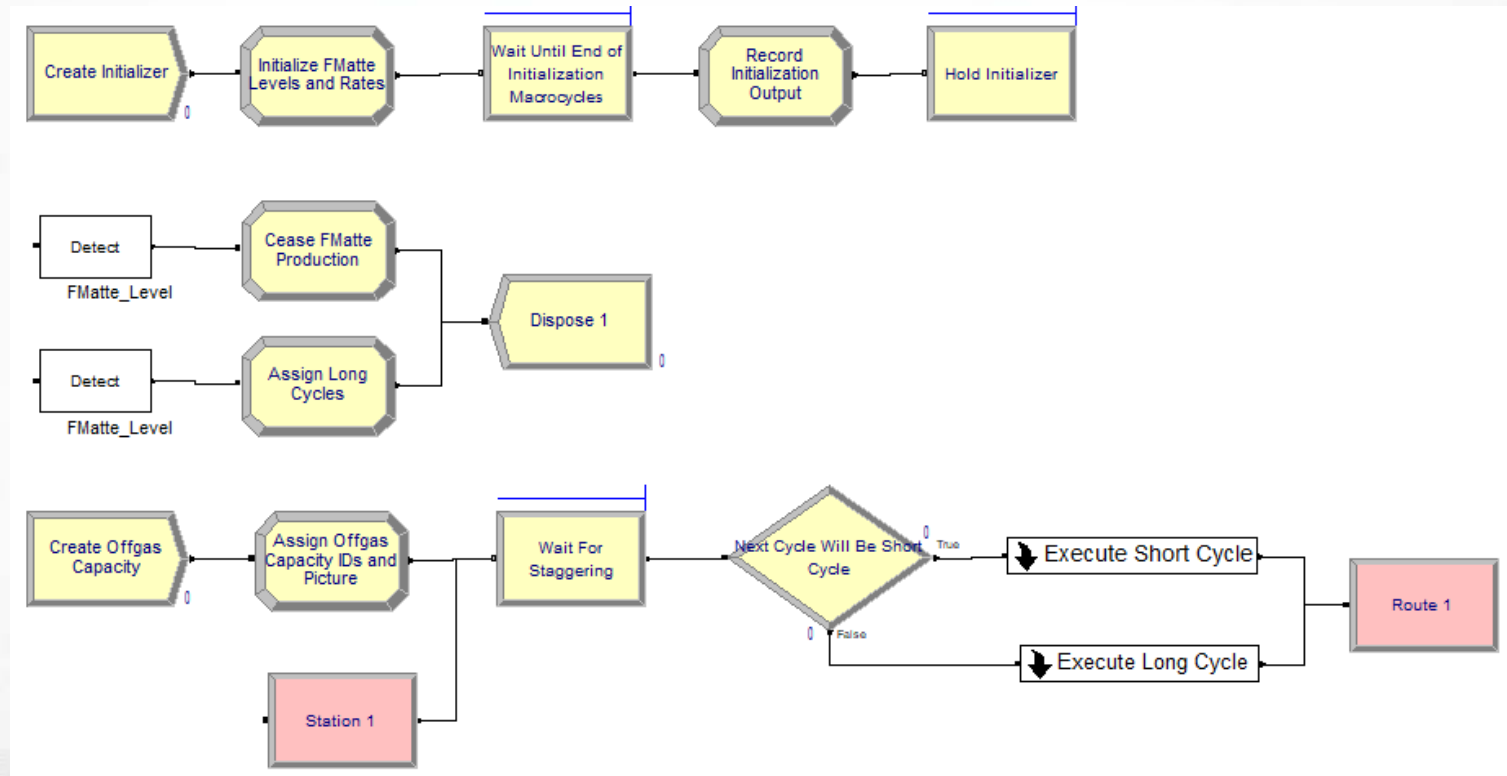
- Used in Extractive Metallurgy to:
 - Scheduling simulation of metallurgical plants → several unit operation connected in a flowsheet, a series of events trigger actions between the unit operations
 - Identify operational bottlenecks and find opportunities to optimize resources
 - Evaluate impact, for example, of equipment capacity and utilization on overall plant performance
 - Often, these type of simulation consider time spans of several days or years to obtain statistically significant results



Discrete Event Simulation

- Motivation:
 - This type of simulation can be viewed as a very high level heat and mass balance flowsheet
 - However, it lacks of any thermochemistry component or tool
 - It relies on “recipes” of operation:
 - How many inches of metal level / hour are generated on a furnace before a tapping event is triggered
 - Duration of blow in a converter
 - This events might be set by constant values but they don't depend on composition, temperature, heat balance, etc. → thermochemistry component

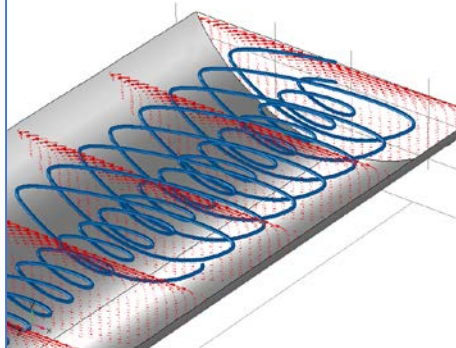
- Arena model for Smelter/Converter



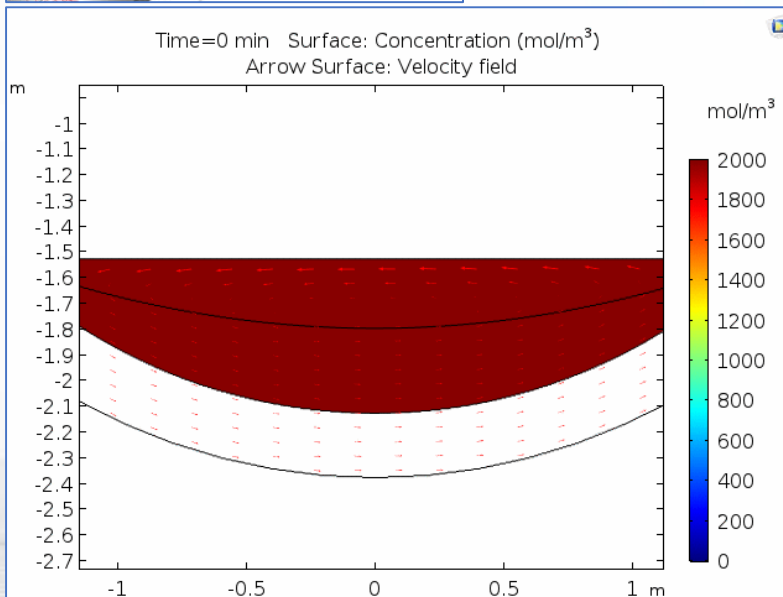
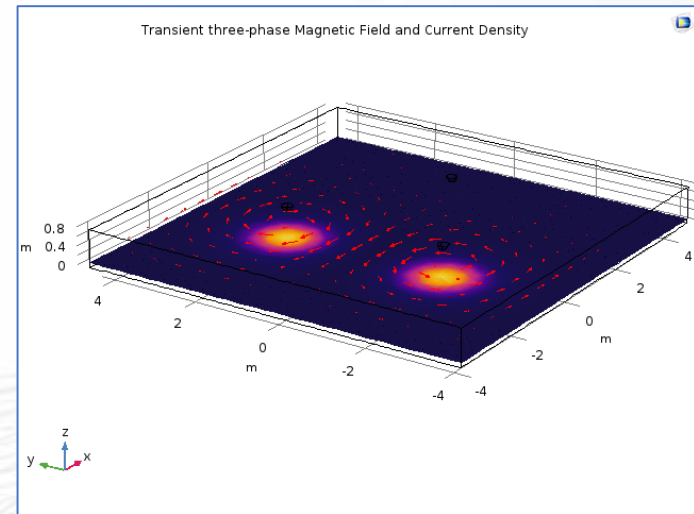
Cu Smelter model: Smelter + Converting simulation based on discrete event simulation
 Courtesy of Prof. Alessandro Navarra

Multiphysics Simulation

3D Granular flow inside rotary kiln

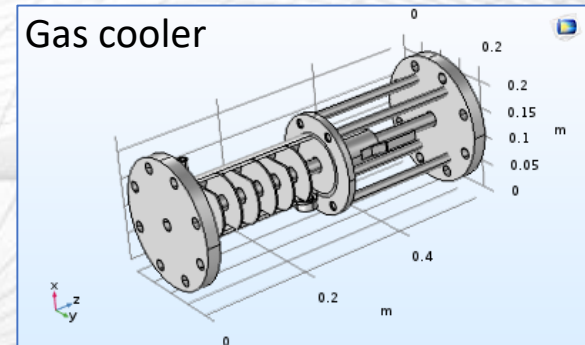


Magnetic field and current density on 3-electrode furnace



Concentration of reactants inside rotary kiln

Gas cooler





M4D-CCI™

M4Dynamics™ COMSOL® ChemApp© Interface

Motivation:

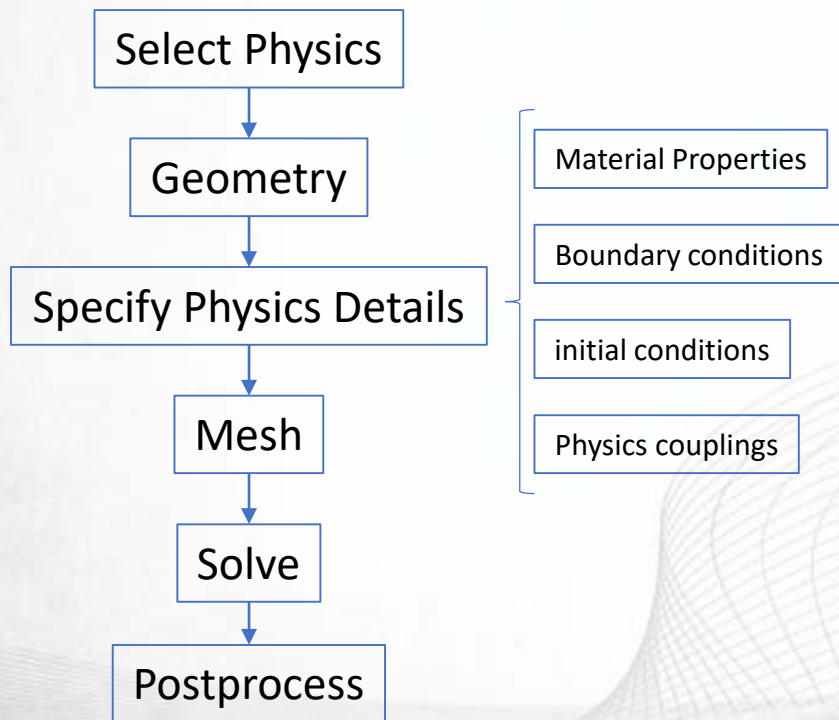
- We use COMSOL as Multiphysics simulation software. It is based on FEA:
 - Vast range of physics: Fluid flow, mechanical, thermal, structural, electrical,... All sharing common PDE formulation
 - Based on space discretization: 0D, 1D, 2D, 3D
 - Steady state, time-dependant, and other analysis
 - Allows detailed simulation of metallurgical reactors combining heat and mass transfer, fluid flow, etc
 - Does not have a thermochemistry component



M4D-CCI™

Typical COMSOL Workflow

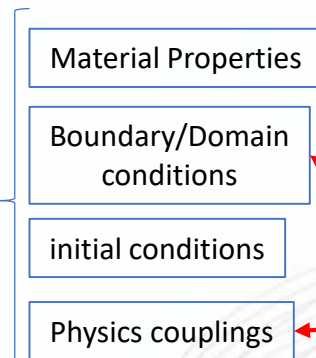
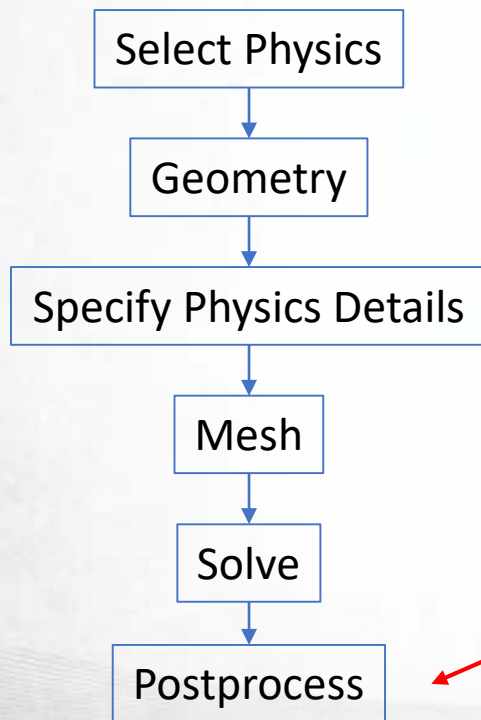
COMSOL workflow



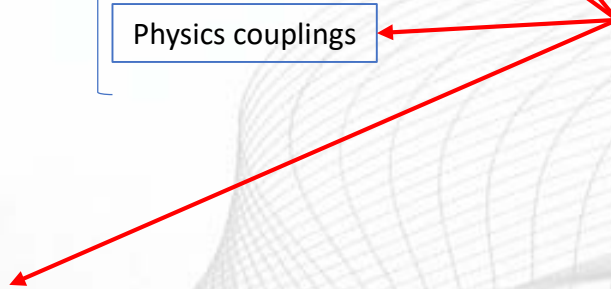
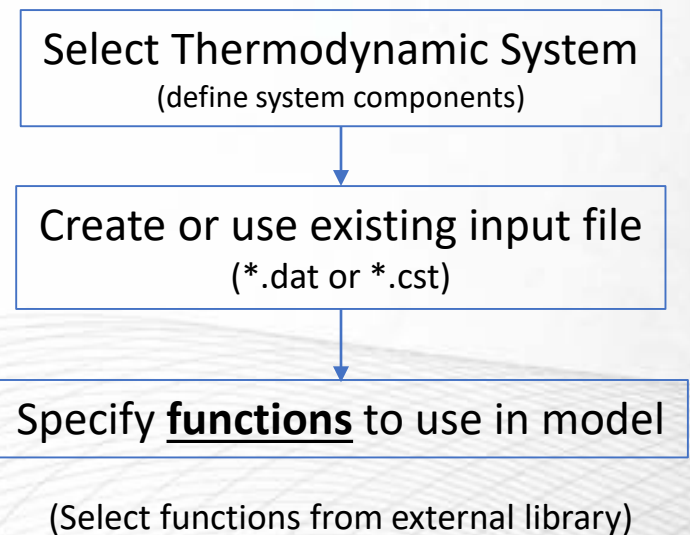


Adding thermochemistry library to the workflow

COMSOL workflow

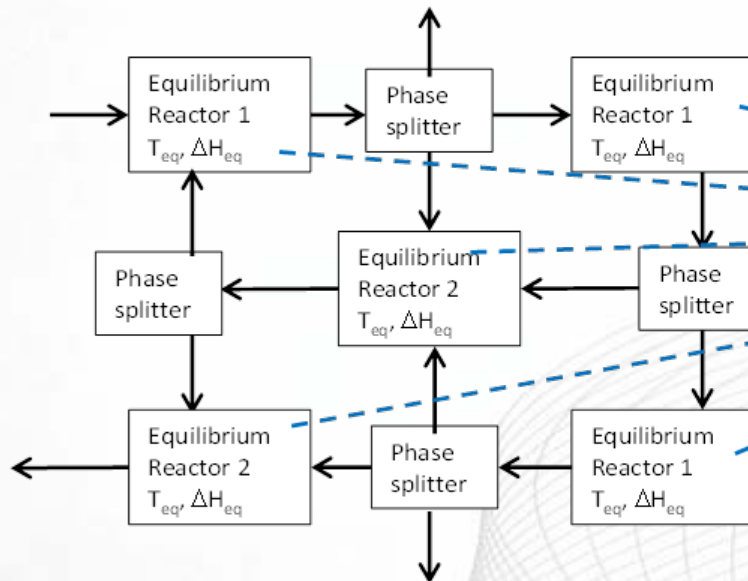


M4D-CCI™

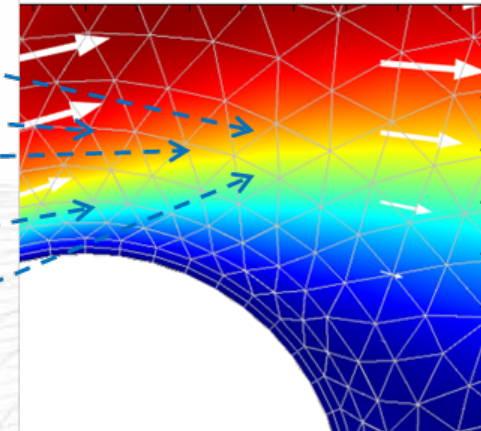


Concept

Analogy to Local Equilibrium Concept



Finite Element Numerical Mesh





M4D-CCI™

How to use

- M4D-CCI is built as an “external function” in COMSOL → like an add-in
- It allows to:
 - Evaluate thermodynamic properties → it can help define complex material properties.
 - Perform equilibrium calculations based on COMSOL variables → one-way coupling
 - Affect COMSOL variables based on equilibrium results → two-way coupling



M4D-CCI™

How to use – Defining external function in COMSOL

External function decl.

The screenshot shows the COMSOL Model Builder interface. On the left, the 'Model Builder' tree shows the project structure: Ternary_NiFeS_3DLiquidus_Projection.mph (root) > Global Definitions > Parameters > M4D-CCI. A red box highlights 'M4D-CCI' and a red arrow points to it from the 'External function decl.' text. On the right, the 'Settings' window for the 'M4D-CCI' external function is shown. The 'Library' field contains the path 'mics\M4DCCI\v10\bin\win64\M4D-CCI.dll', highlighted with a red box and a red arrow pointing to it from the text 'M4D-CCI.dll'. Below this, a table lists functions defined in the DLL, highlighted with a red box and a red arrow pointing to it from the text 'These functions are defined in M4D-CCI'. The table has two columns: 'Function name' and 'Arguments'. The 'Advanced' section at the bottom has a red box around the 'Initialization data' field, which contains 'NiFe_Matte.cst', with a red arrow pointing to it from the text 'Thermodynamic input file'.

Model Builder

Ternary_NiFeS_3DLiquidus_Projection.mph (root)

- Global Definitions
 - Parameters
 - M4D-CCI
- Materials
- Component 1 (comp 1)
- Study 1
- Results
 - Data Sets
 - Views
 - Derived Values
 - Tables
 - 2D Phase Targets
 - 2D Phase Fraction
 - Export
 - Reports

Settings

External

Plot Create Plot

Label: M4D-CCI

Functions

Library: mics\M4DCCI\v10\bin\win64\M4D-CCI.dll Browse...

Function name	Arguments
M4DCCI_VERSION	1
M4DCCI_BUILD	1
ChemApp_VERSION	1
SC_PTarget	T,Pidx,X1,X2,X3
SC_FTarget	T,Pidx,X1,X2,X3
SC_PhaseAmount	T,Pidx,X1,X2,X3

Function name:

Arguments:

Derivatives

Advanced

Initialization data: NiFe_Matte.cst

Thread safe

M4D-CCI.dll

These functions are defined in M4D-CCI

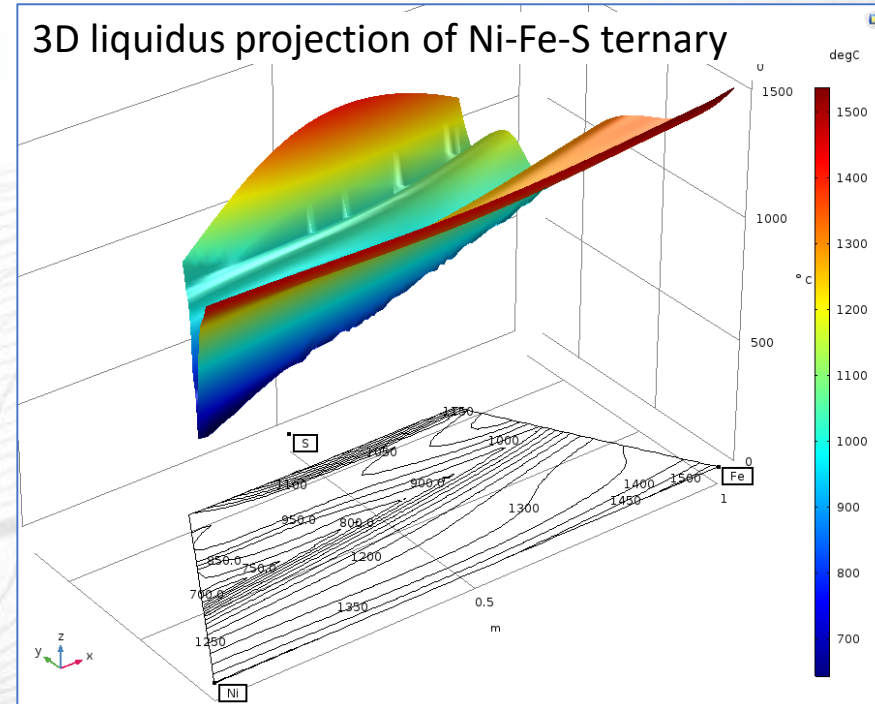
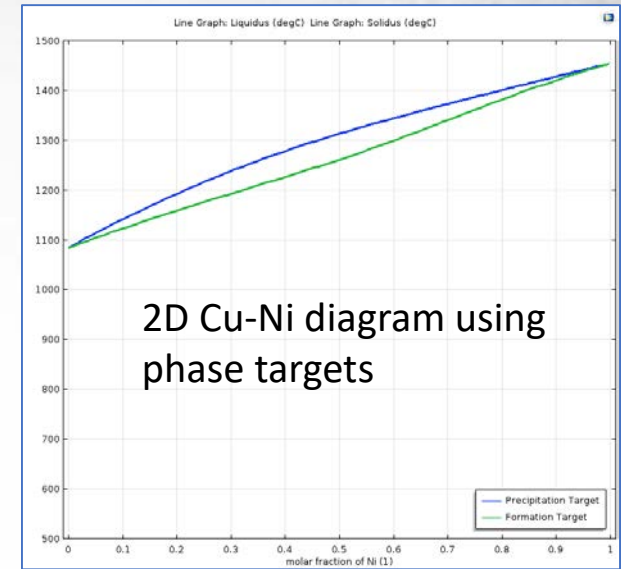
Thermodynamic input file



M4D-CCI™

Examples of use

- Equilibrium Calculations - General
 - Thermodynamic functions: G, H, S, Cp
 - Amount, fractions, activities, etc. of phases, system components, phase constituents,
 - fraction,
 - Phase Target: precipitation and formation targets
 - Fixed chemical potential calculation of amount and fraction of phases, phase constituent and system components



Examples of use

- Equilibrium Calculations – Chemical Reaction Kinetics

$$\frac{dc_i}{dt} = R_i$$

For a set of ODE's with c_i as the variable to solve for

The source term depends on the rate of reactions

$$R_i = \sum_j R_j = \sum_j v_{ij} r_j$$

$$r_j = k_j^f \prod_{i=1}^{Q_r} c_i^{v_{ij}} - k_j^r \prod_{i=1}^{Q_p} c_i^{v_{ij}}$$

Which depends on the rate constants for each reaction

$$k_j^f = A_j^f e^{\left(\frac{-Ea_j^f}{R_g T}\right)}$$

At equilibrium, reverse and forward rate constants are related by the equilibrium constant

$$K_{e,j} = e^{\left(\frac{-\Delta G_j^0}{R \cdot T}\right)}$$

Equilib constant is calculated by M4D-CCI for each reaction




M4D-CCI™

Examples of use

- Equilibrium Calculations – Reaction Kinetics + Energy Balance

All of the previous +:

$$V \sum_i c_i \cdot C_{p,i} \cdot \frac{dT}{dt} = Q$$

$$Q = -V \sum_j H_j \cdot r_j$$


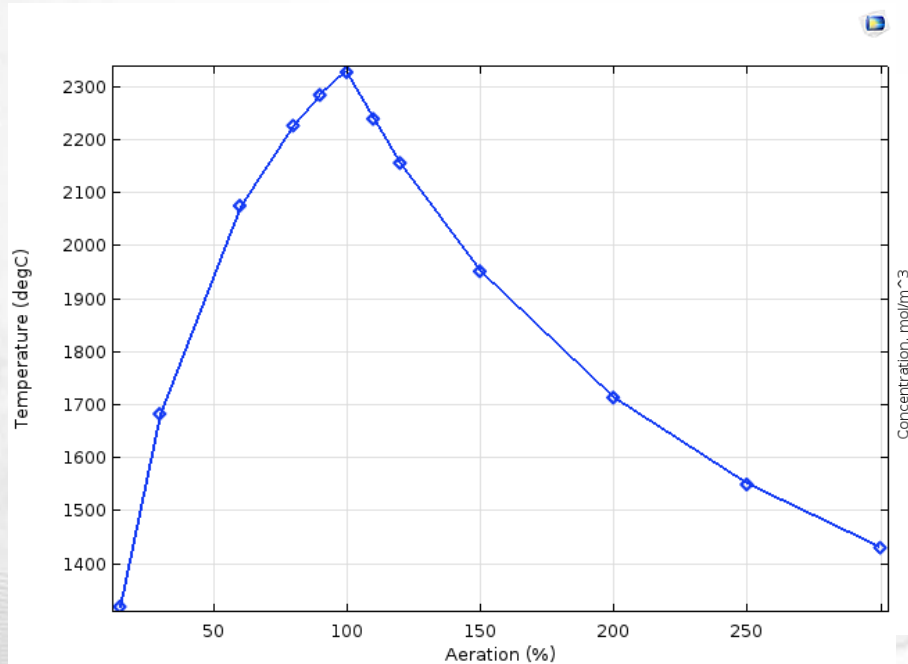
The system's temperature depends on the heat source from chemical reactions

The volumetric heat source, depends on Enthalpy of reactions and rate of reactions

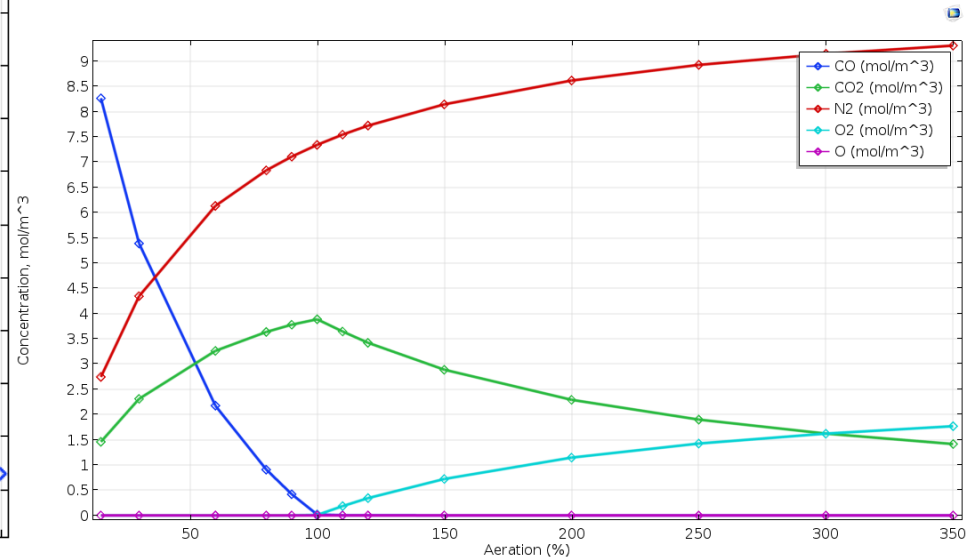
Examples of use

- Equilibrium Calculations – Reaction Kinetics + Energy Balance

Example: 0D combustion of preheated CO + Air mixture



Effect of aeration on adiabatic flame T
Initial gas T = 650°C



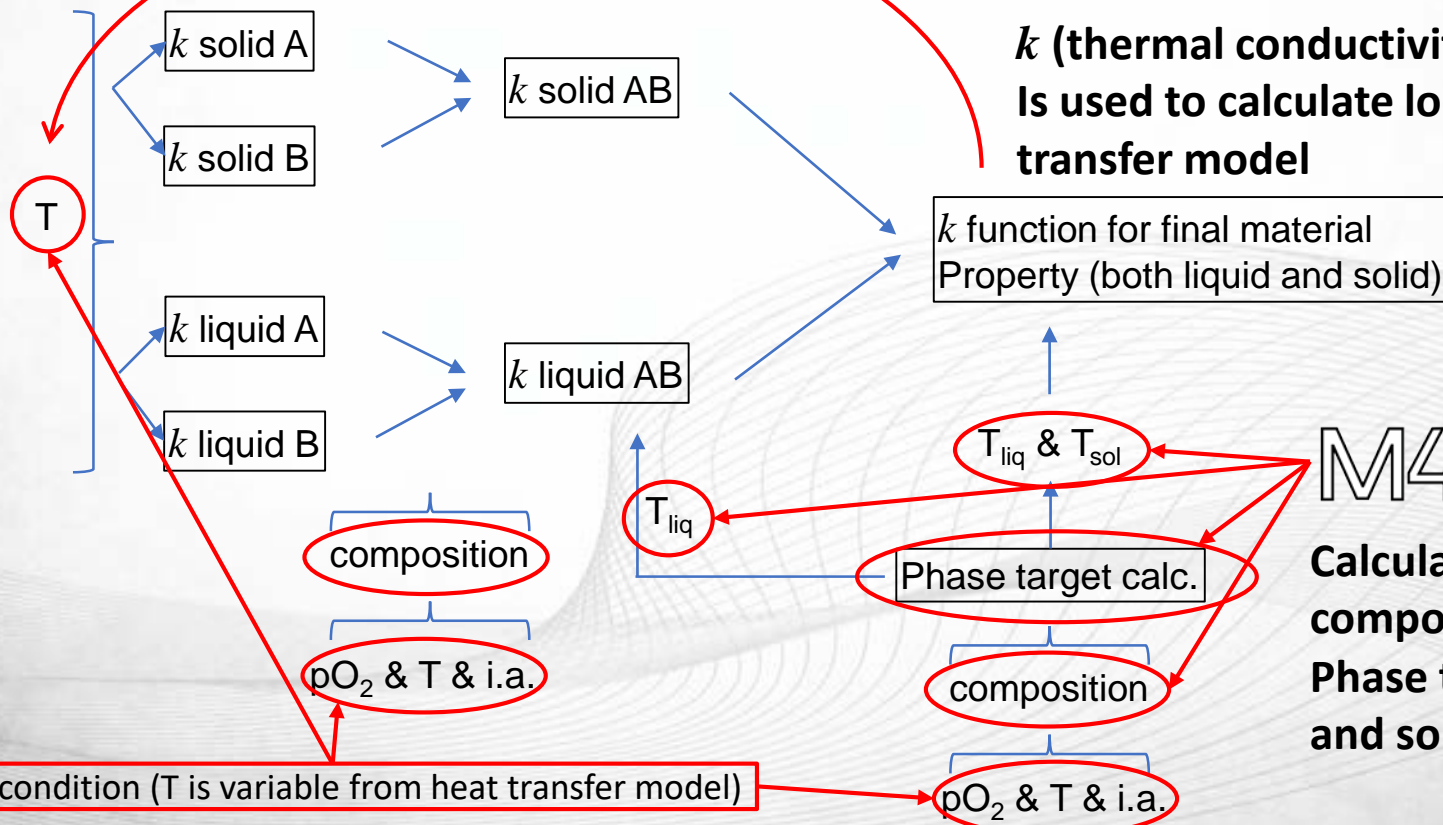
Equilibrium concentration vs aeration

Examples of use

- Complex material properties:

- Consider a material domain that can be either liquid or solid, depending on the local concentration and temperature. The material could be an alloy AB (solid or liquid)
- The purpose is to defined a single material property within that space domain

k (thermal conductivity)
Is used to calculate local T on the heat transfer model

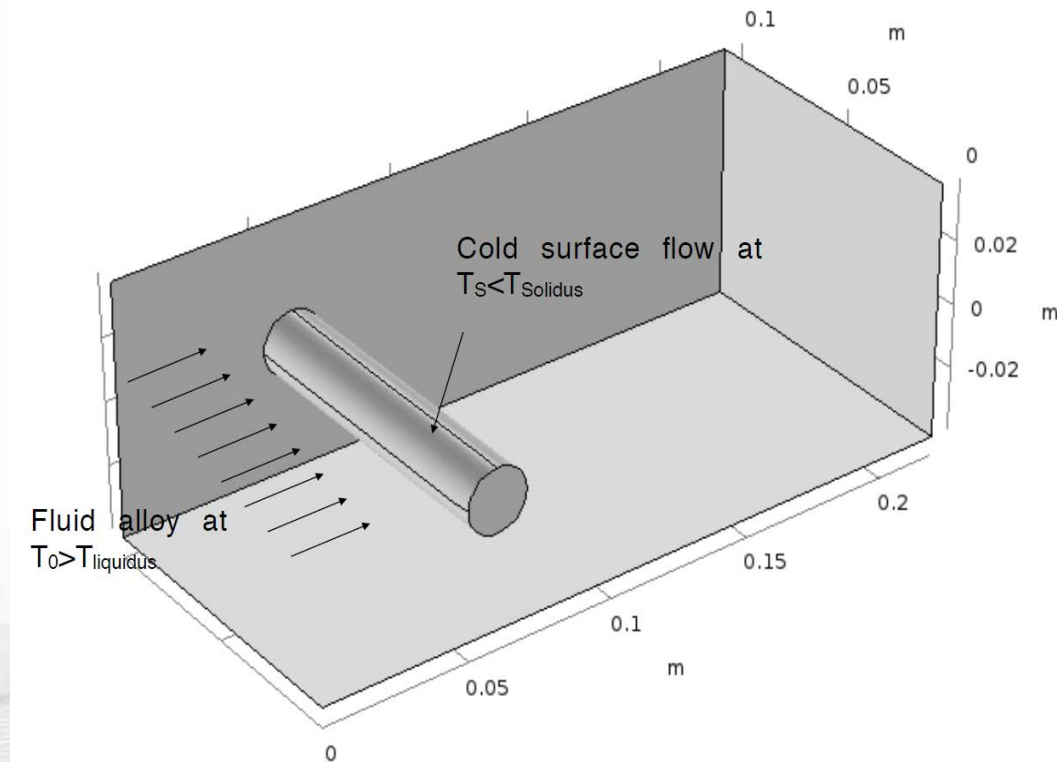


M4D-CCI™

Calculate local eq. compositions, Phase targets → T liq and sol.

Examples of use

- Phase Change - Solidification of non-ideal liquid as function of space and time and composition

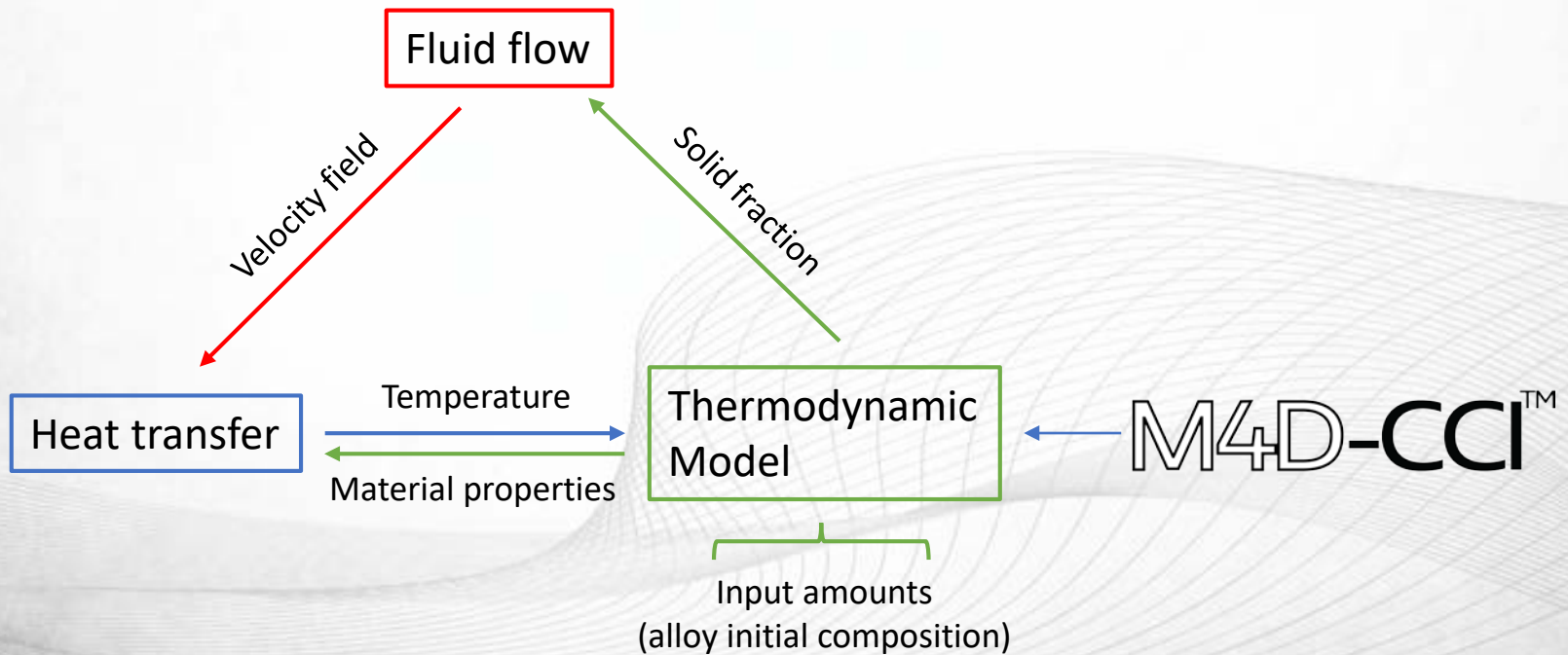




M4D-CCI™

Examples of use

- Solidification model uses a fixed numerical grid to track liquid/solid interface
- Model approach:

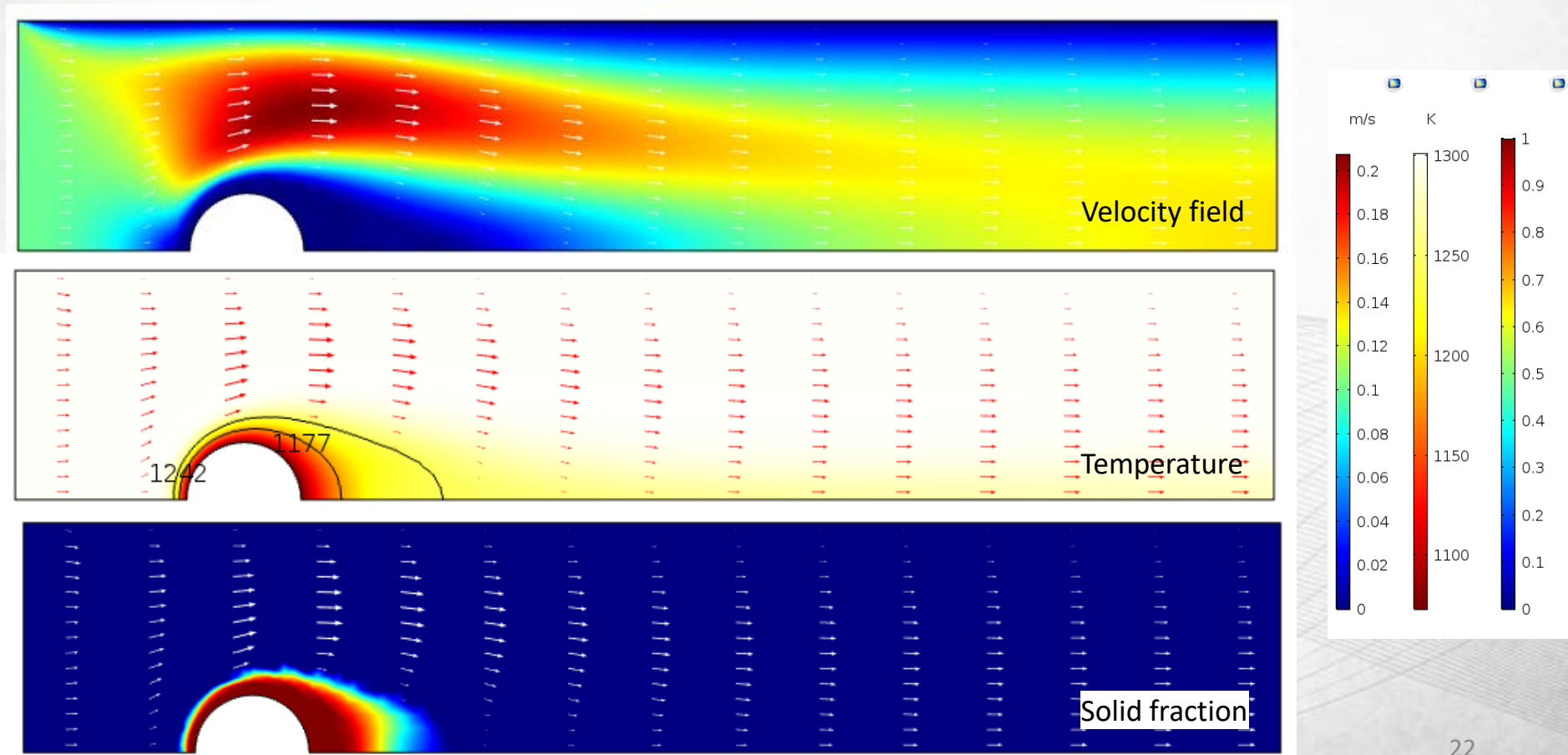




M4D-CCI™

Examples of use

- Phase Change - Results

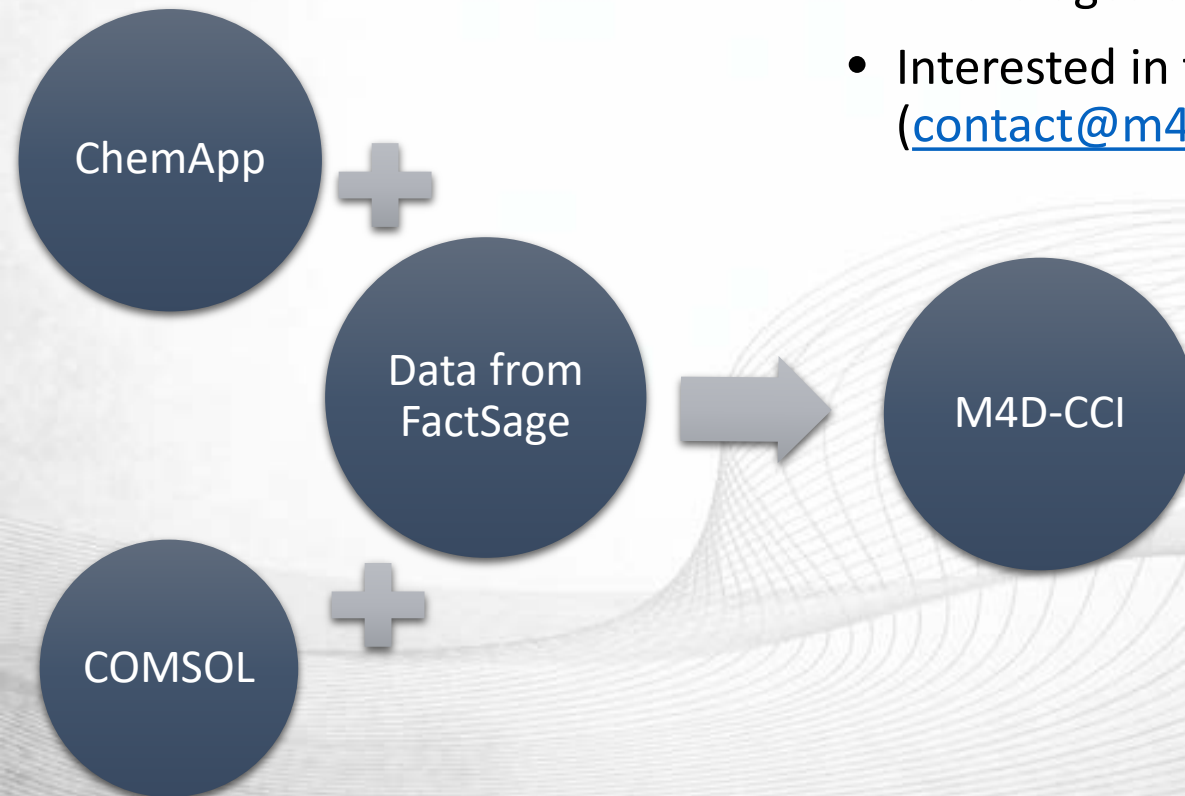




M4D-CCI™

Current State of Development

- Compatible with COMSOL versions 5.1 up to 5.3a
- 64-bit ChemApp version
- Final stages of documentation
- Interested in testing it?
(contact@m4dynamics.com)





Thanks!