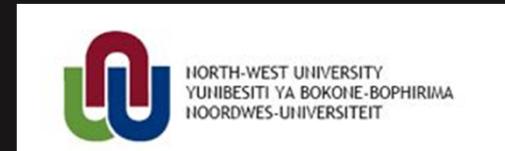


# THERMO CHEMICAL SIMULATION OF A COAL GASIFICATION PROCESS

(APPLYING WASTE CAPTURED CO<sub>2</sub> EITHER FROM COMBUSTION OR GASIFICATION)



**AFRICARY**  
**African Carbon Energy**

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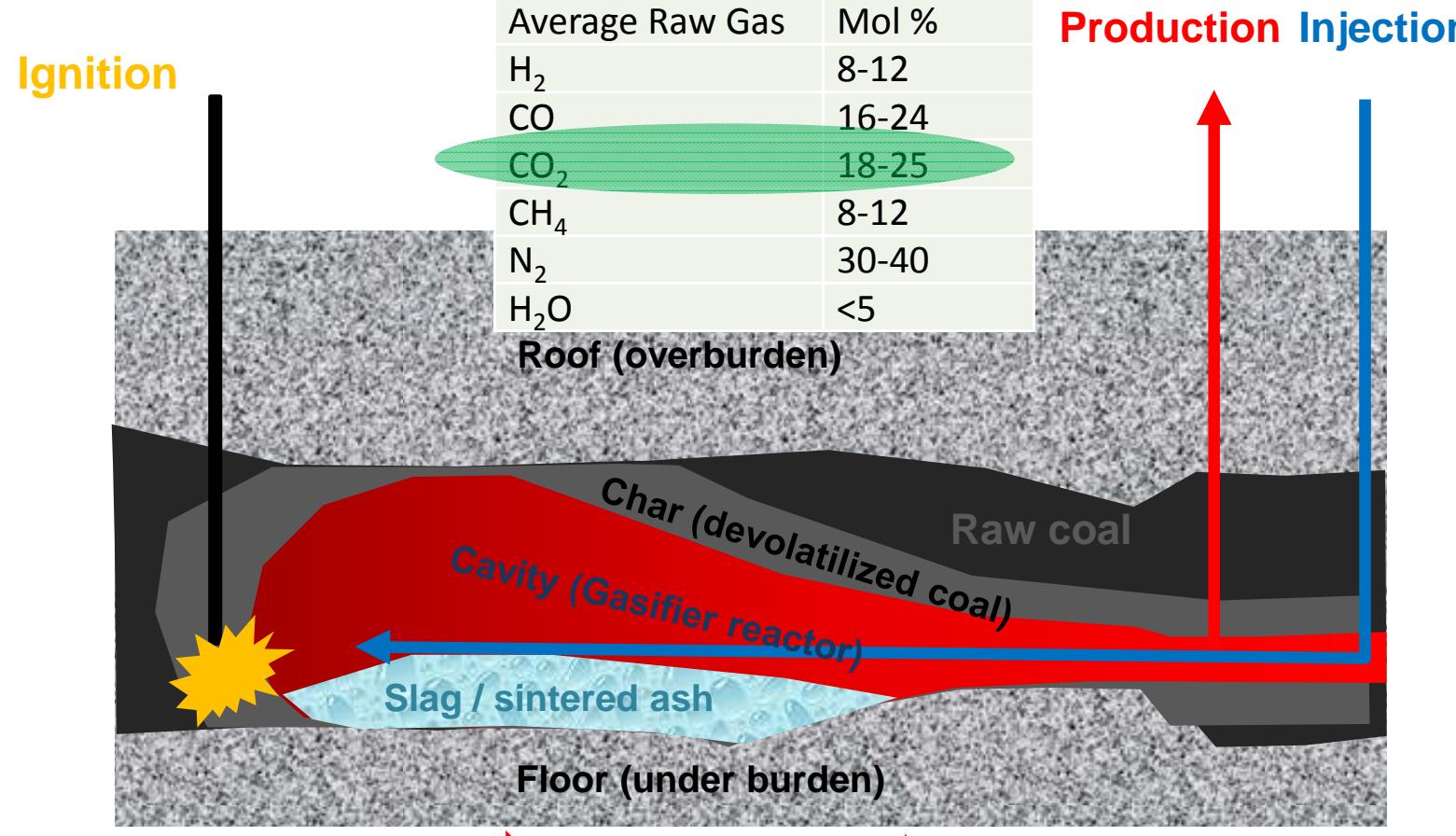




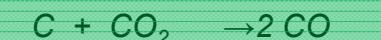
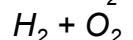
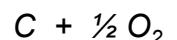
# ROADMAP FOR THE PRESENTATION

- UNDERGROUND COAL GASIFICATION – EFFICIENT IN-SITU CO<sub>2</sub> CAPTURE AND CONVERSION
- Entrained flow Gasification
  - E- Gas™ (Reliance Industries) – AC Collins

# UCG (OPERATION AND GAS COMPOSITION)



**COMBUSTION** → **GASIFICATION** → **PYROLYSIS** → **DRYING**

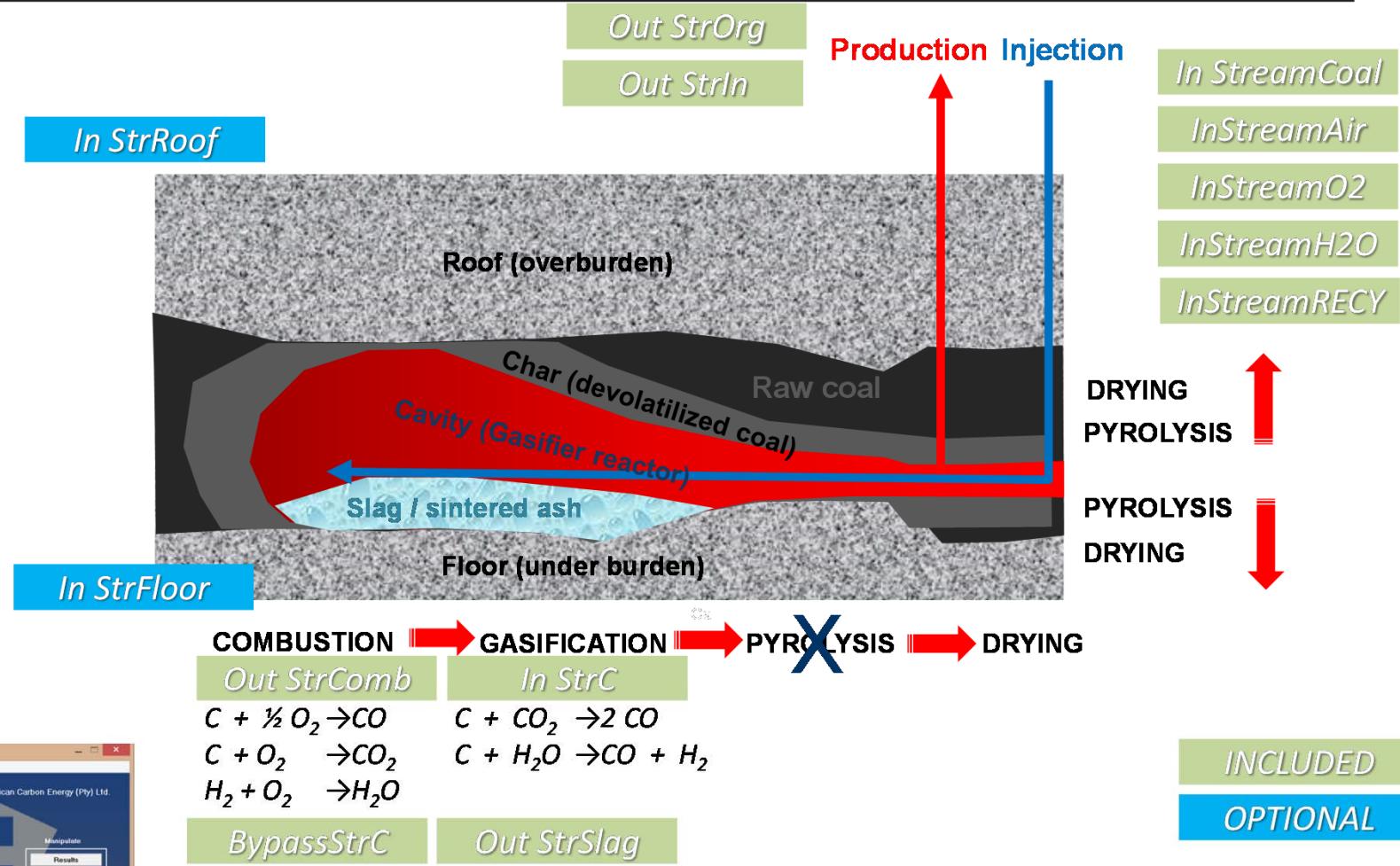


# PURPOSE OF THIS STUDY

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1. **CO<sub>2</sub>, as gasification agent for UCG**, has received little attention and only studied by a few researchers to date - will be investigated on a thermo-equilibrium basis in this study.
2. Eftekhari, et. al., (2012) and others concluded further that Carbon Capture and Storage (CCS) in a geological form as sequestration within UCG is not practical and cannot result to a near zero-emission process - and **capture (utilizing) of CO<sub>2</sub> coupled with UCG** has to be studied in novel manners and approaches.
3. Evaluate under thermo-equilibrium conditions the positive influence which **CO<sub>2</sub> utilization as UCG gasification agent in parallel gasifiers** will positively influence the carbon emissions reduction and environmental footprint, gas quality, as well as reduction of CO<sub>2</sub> per ton coal when recycled to a second or third UCG cavity.

# FACTSAGE™ THERMO EQUILIBRIUM APPROACH (SYNGAS)



# SIMULATION PROCESS

## (Coal and operational inputs)



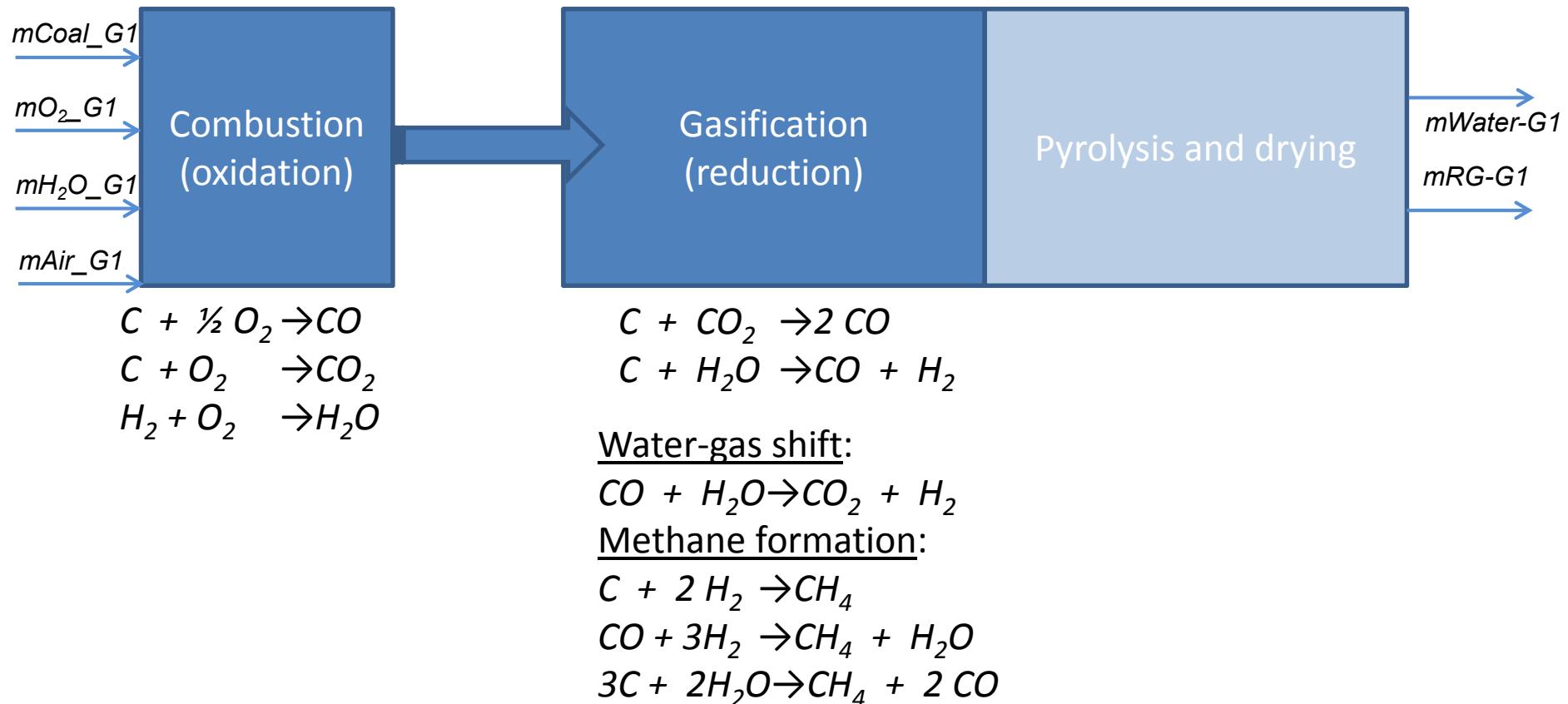
PROXIMATE (AR)	Mass %
Moisture	4.3
Fixed carbon	45.7
Volatile matter	19.2
Ash	30.8
ULTIMATE (DAF)	
C	78.9
H	4.3
N	2.1
S	0.9
O	13.9
ASH OXIDES (%)	
SiO <sub>2</sub>	52.8
Al <sub>2</sub> O <sub>3</sub>	27.2
Fe <sub>2</sub> O <sub>3</sub>	4.9
P <sub>2</sub> O <sub>5</sub>	0.1
TiO <sub>2</sub>	1.3
CaO	6.4
MgO	1.0
K <sub>2</sub> O	0.5
Na <sub>2</sub> O	0.4
SO <sub>3</sub>	4.9
Trace Elements	0.5

Not included

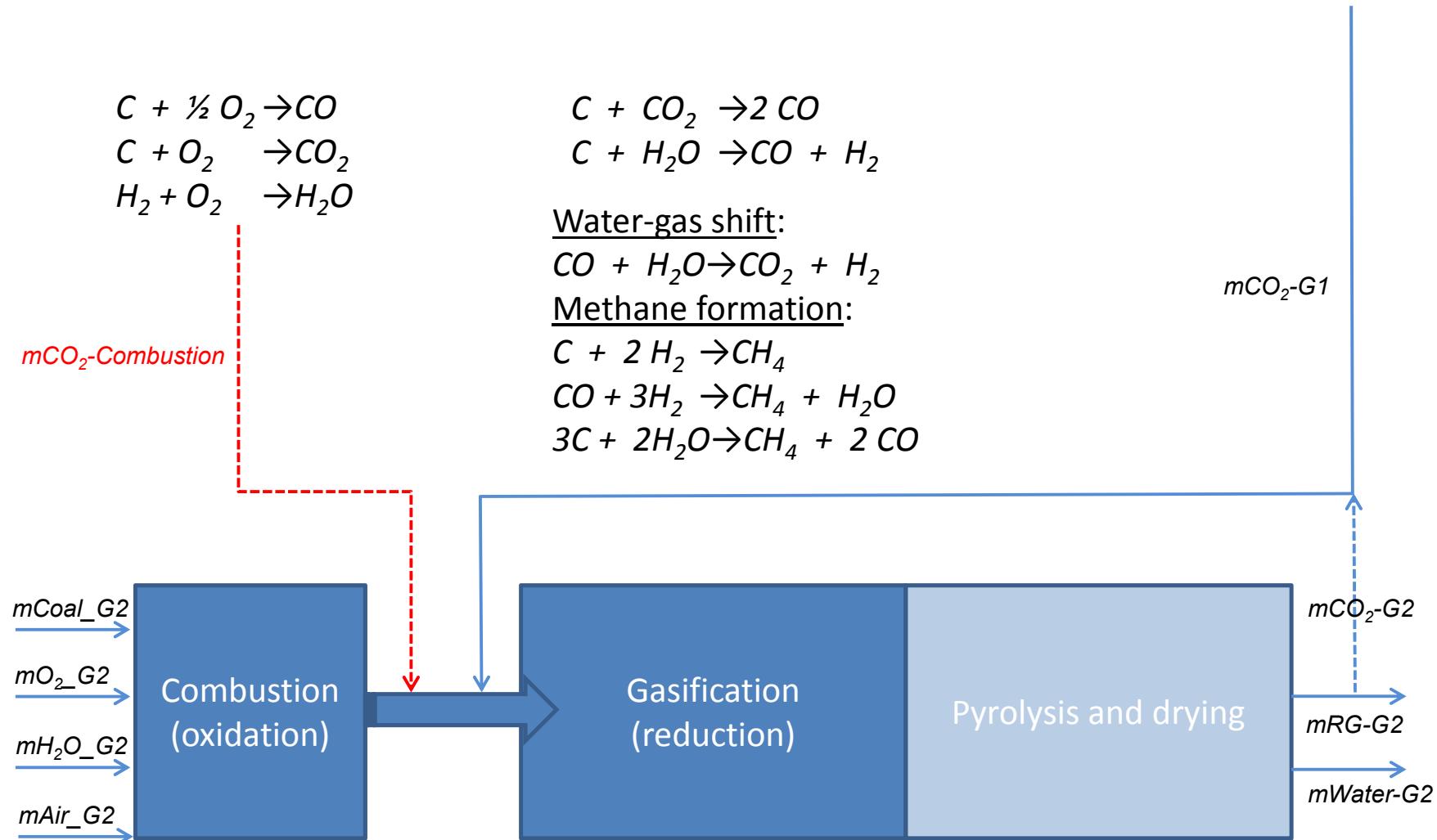
Property	Mass %
H <sub>2</sub> O	2.9
H <sub>2</sub>	0.15
CH <sub>4</sub>	4.01
CO	0.98
CO <sub>2</sub>	7.2
N <sub>2</sub>	2.1
Tar and oils	5.6

- In situ coal = 50 000kg/hr
- Total agent (O<sub>2</sub> + Air) = 45 000kg/hr
- C + O<sub>2</sub> → CO<sub>2</sub> (**Combustion zone**) – all O<sub>2</sub> molecules converted and remaining carbon to **Gasification zone**
- Water is the sum of all ground water and recycled water and will be kept at a constant for both G1 and G2.
- No contribution from roof and floor in this study – only individual coal seam
- 18 trace elements included

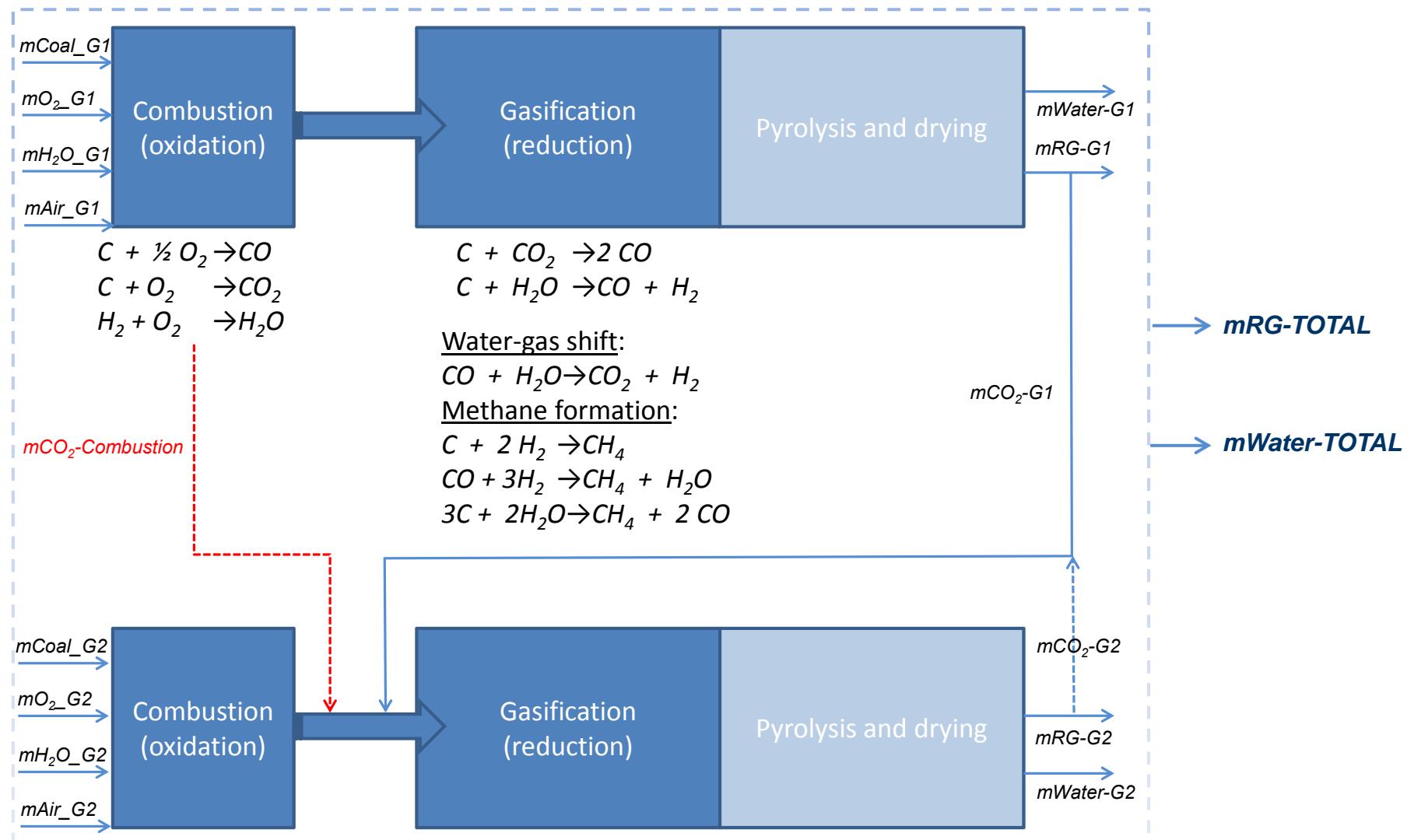
# SIMULATION PROCESS (Gasifier 1)



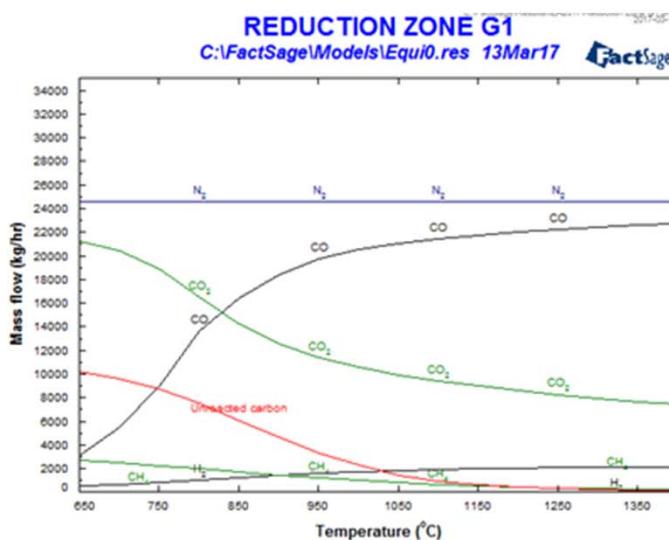
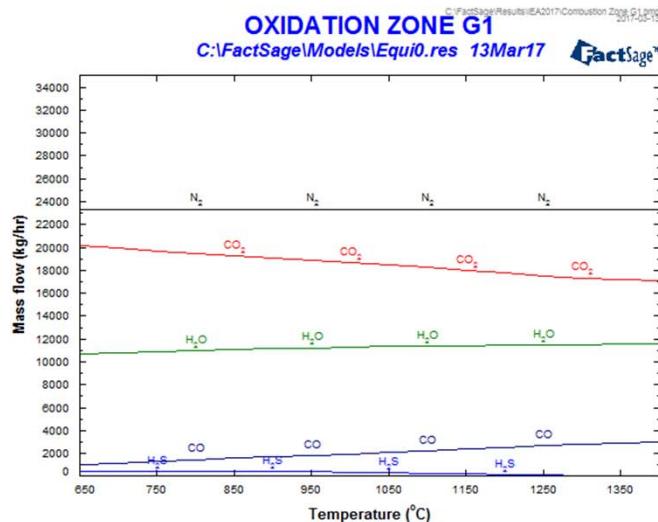
# SIMULATION PROCESS (Gasifier 2)



# SIMULATION PROCESS (G1+G2)



# RESULTS (GASIFIER 1)

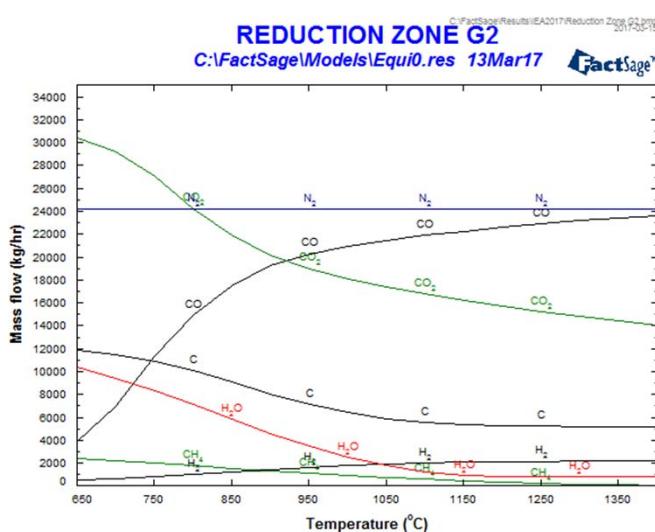
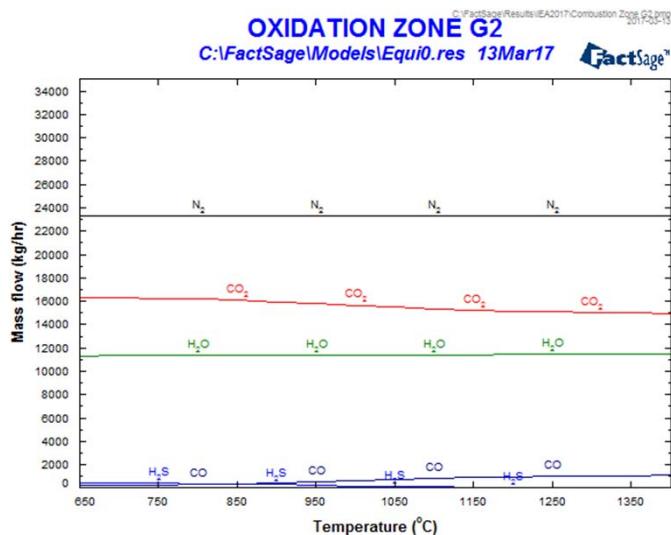


	kg/hr	mol %
CO	2 070	3.5
CO <sub>2</sub>	17 563	18.9
N <sub>2</sub>	24 080	40.3
H <sub>2</sub> O	11 909	31.3
H <sub>2</sub> S+COS	576	0.8
Reacted COAL	±15 000	

$C + O_2 \rightarrow CO_2$  (**Combustion zone**) – all O<sub>2</sub> molecules converted and remaining carbon to **Gasification zone**

	kg/hr	mol %
H <sub>2</sub>	341	7.2
CO	20 153	30.6
CO <sub>2</sub> <i>(mCO<sub>2</sub>_G1)</i>	<b>10 556</b>	<b>10.2</b>
CH <sub>4</sub>	3 394	9.0
N <sub>2</sub>	25 975	39.5
H <sub>2</sub> O	1 144	2.7
H <sub>2</sub> S	577	0.7
Raw gas (mRG_G1)	62 140	
Unconverted Coal	101	

# RESULTS (GASIFIER 2)

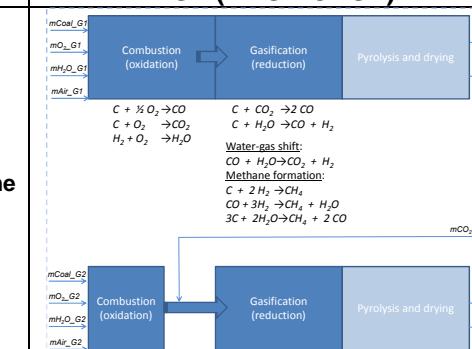
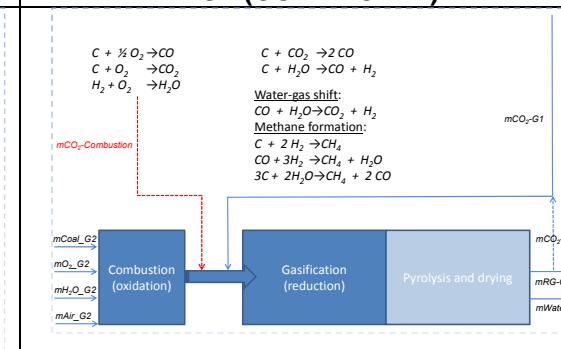
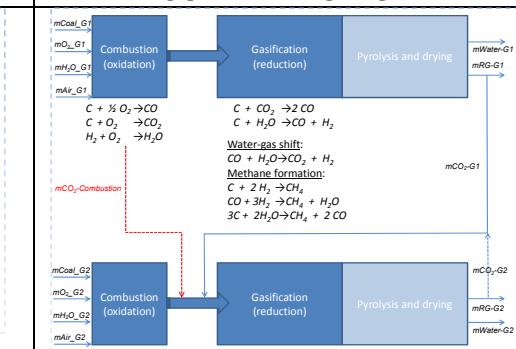


1.  $mCO_2\_G1 = 10\ 556\ kg/hr$
2.  $mAir\_G2 = mAir\_G1 - mCO_2\_G1$

	kg/hr	mol %
CO	799	1.5
CO <sub>2</sub>	15 493	18.6
N <sub>2</sub>	23 727	44.7
H <sub>2</sub> O	11 795	34.3
H <sub>2</sub> S+COS	454	0.7
Reacted COAL	17 280	

	kg/hr	mol %
H <sub>2</sub>	356	7.3
CO	22 238	32.8
CO <sub>2</sub> (mCO <sub>2</sub> _G2)	13 590	12.8
CH <sub>4</sub>	3 537	9.1
N <sub>2</sub>	23 485	34.6
H <sub>2</sub> O	1 192	2.7
H <sub>2</sub> S	601	0.7
Raw gas (mRG_G2)	64 999	
Unreacted COAL	5000	

# RESULTS (COMBINED G1 + G2)

INDEX	G1 (BASE CASE)	G2 (CO2 RECYCLE)	COMBINED G1+G2
<b>Flow scheme</b>	 <p>Reactions for G1:</p> <ul style="list-style-type: none"> <li><math>C + \frac{1}{2} O_2 \rightarrow CO</math></li> <li><math>C + O_2 \rightarrow CO_2</math></li> <li><math>H_2 + O_2 \rightarrow H_2O</math></li> <li><math>C + CO_2 \rightarrow 2 CO</math></li> <li><math>C + H_2O \rightarrow CO + H_2</math></li> <li><math>CO + H_2O \rightarrow CO_2 + H_2</math></li> <li><math>Methane formation:</math></li> <li><math>C + 2 H_2 \rightarrow CH_4</math></li> <li><math>CO + 3H_2 \rightarrow CH_4 + H_2O</math></li> <li><math>3C + 2H_2O \rightarrow CH_4 + 2 CO</math></li> </ul>	 <p>Reactions for G2:</p> <ul style="list-style-type: none"> <li><math>C + \frac{1}{2} O_2 \rightarrow CO</math></li> <li><math>C + O_2 \rightarrow CO_2</math></li> <li><math>H_2 + O_2 \rightarrow H_2O</math></li> <li><math>CO + H_2O \rightarrow CO_2 + H_2</math></li> <li><math>Methane formation:</math></li> <li><math>C + 2 H_2 \rightarrow CH_4</math></li> <li><math>CO + 3H_2 \rightarrow CH_4 + H_2O</math></li> <li><math>3C + 2H_2O \rightarrow CH_4 + 2 CO</math></li> </ul>	 <p>Reactions for Combined G1+G2:</p> <ul style="list-style-type: none"> <li><math>C + \frac{1}{2} O_2 \rightarrow CO</math></li> <li><math>C + O_2 \rightarrow CO_2</math></li> <li><math>H_2 + O_2 \rightarrow H_2O</math></li> <li><math>CO + H_2O \rightarrow CO_2 + H_2</math></li> <li><math>Methane formation:</math></li> <li><math>C + 2 H_2 \rightarrow CH_4</math></li> <li><math>CO + 3H_2 \rightarrow CH_4 + H_2O</math></li> <li><math>3C + 2H_2O \rightarrow CH_4 + 2 CO</math></li> </ul>
<b>Coal usage (kg/hr)*</b>	49 800		
<b>CO<sub>2</sub> produce (kg/hr)</b>	10 556 (recycle)		
<b>Oxygen (kg/hr)**</b>	14 512		
<b>Syngas (kg/hr)***</b>	62 140		
<b>CO<sub>2</sub>:Syngas</b>	0.17		
<b>CO<sub>2</sub>:Coal</b>	0.21		
<b>Nitrogen (kg/hr)</b>	25 975		

\* Total coal for equilibrium

\*\* Total oxygen molecules air + pure oxygen

\*\*\* Syngas including CO<sub>2</sub>

\*\*\*\* Excluding CO<sub>2</sub> from G1 recycled

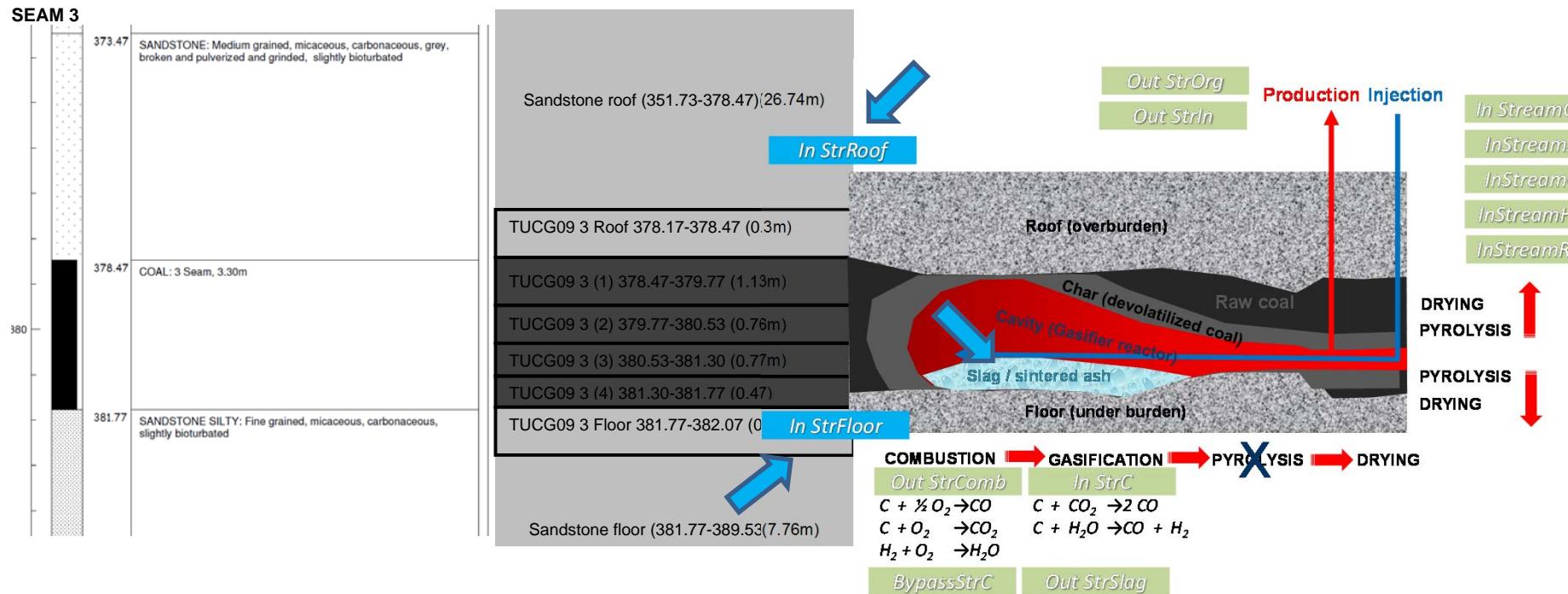
# CONCLUSIONS

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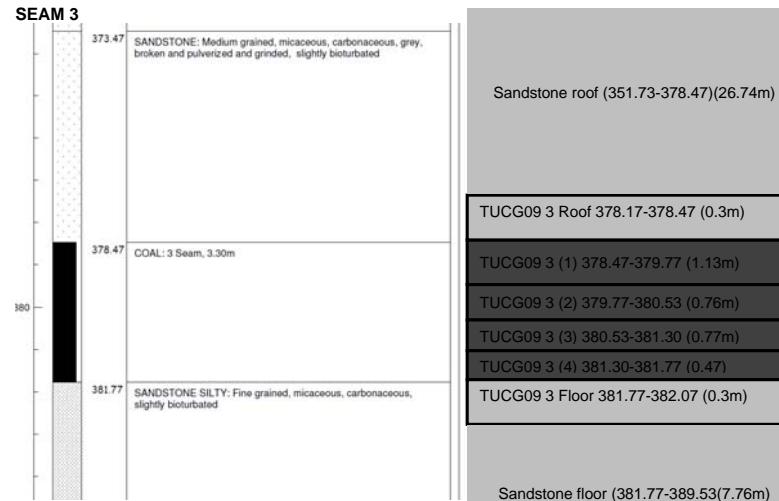
- ❑ Thermo-equilibrium add value in
  - ❑ determining the optimized process conditions for a specific feed and operating scenario
  - ❑ coal ash fusibility characteristics & slag-profile behaviour prediction to prevent operational shut-downs
  - ❑ new flow schemes during the concept and design phase
- ❑ Coal composition from either core drillings or actual mine samples can be used as input for the complex equilibrium calculations using FACTSAGE™.

# NEXT STEPS IN PROGRESS

(Influence of roof, floor and slag on CO<sub>2</sub> capture and flow temperature with recycling)



# NEXT STEPS IN PROGRESS



Sample naming	Anatase	Dolomite	Halite (Sodium Chloride)	Kaolinite	Micro-cline	Muscovite	Plagioclase	Pyrite	Quartz	Rutile	Siderite
	(TiO <sub>2</sub> )	(CaMg(CO <sub>3</sub> ) <sub>2</sub> )	(NaCl)	(Al <sub>2</sub> (Si <sub>2</sub> O <sub>5</sub> )(OH) <sub>4</sub> )	(KAlSi <sub>3</sub> O <sub>8</sub> )	(KAl <sub>3</sub> Si <sub>3</sub> O <sub>10</sub> (OH)2)	((Na,Ca)(Si,Al)4O <sub>8</sub> )	(FeS <sub>2</sub> )	(SiO <sub>2</sub> )	(TiO <sub>2</sub> )	(FeCO <sub>3</sub> )
<b>TUCG09/3 ROOF/1</b>	0.7	0.7	0.61	17.48	12.47	5.20	16.12	0.47	45.17	1.12	0.00
<b>TUCG09/3 /1</b>											
<b>TUCG09/3 /2</b>											
<b>TUCG09/3 /3</b>	4.3	0.0	0.00	52.50	5.96	5.17	3.39	2.33	24.78	1.58	0.00
<b>TUCG09/3 /4</b>											
<b>TUCG09/3 LOOR/1</b>	1.3	0.9	0.07	37.74	9.08	8.75	1.96	0.39	38.60	1.14	0.11
<b>TUCG09/1 /1</b>	2.2	0.0	0.34	23.20	6.18	11.25	5.20	1.50	49.24	0.93	0.00



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