

***EFFECT OF GASIFICATION OPERATING TEMPERATURE ON EITHER LEACHABILITY OF ASH AND MIGRATION OF ELEMENTS DURING AN UCG PROCESS OR RECOVERY OF SPECIFIC ELEMENTS SUCH AS K, AL AND TI FROM A FIXED BED GASIFICATION PROCESS***



***GTT WORKSHOP, AACHEN, 1-3 JULY 2015***

Prof JC van Dyk  
**AC Collins\* (author)**  
Prof CA Strydom  
JF Brand



NORTH-WEST UNIVERSITY  
YUNIBESITI YA BOKONE-BOPHIRIMA  
NOORDWES-UNIVERSITEIT  
POTCHEFSTROOM CAMPUS

# Motivation

Results - Equilib 850 C (page 1/20)

Output Edit Show Pages

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

1050 C	1075 C	1100 C	1125 C	1150 C	1175 C	1200 C	1225 C	1250 C	1275 C	1300 C	1325 C	
725 C	750 C	775 C	800 C	825 C	- 850 C -	875 C	900 C	925 C	950 C	975 C	1000 C	1025 C

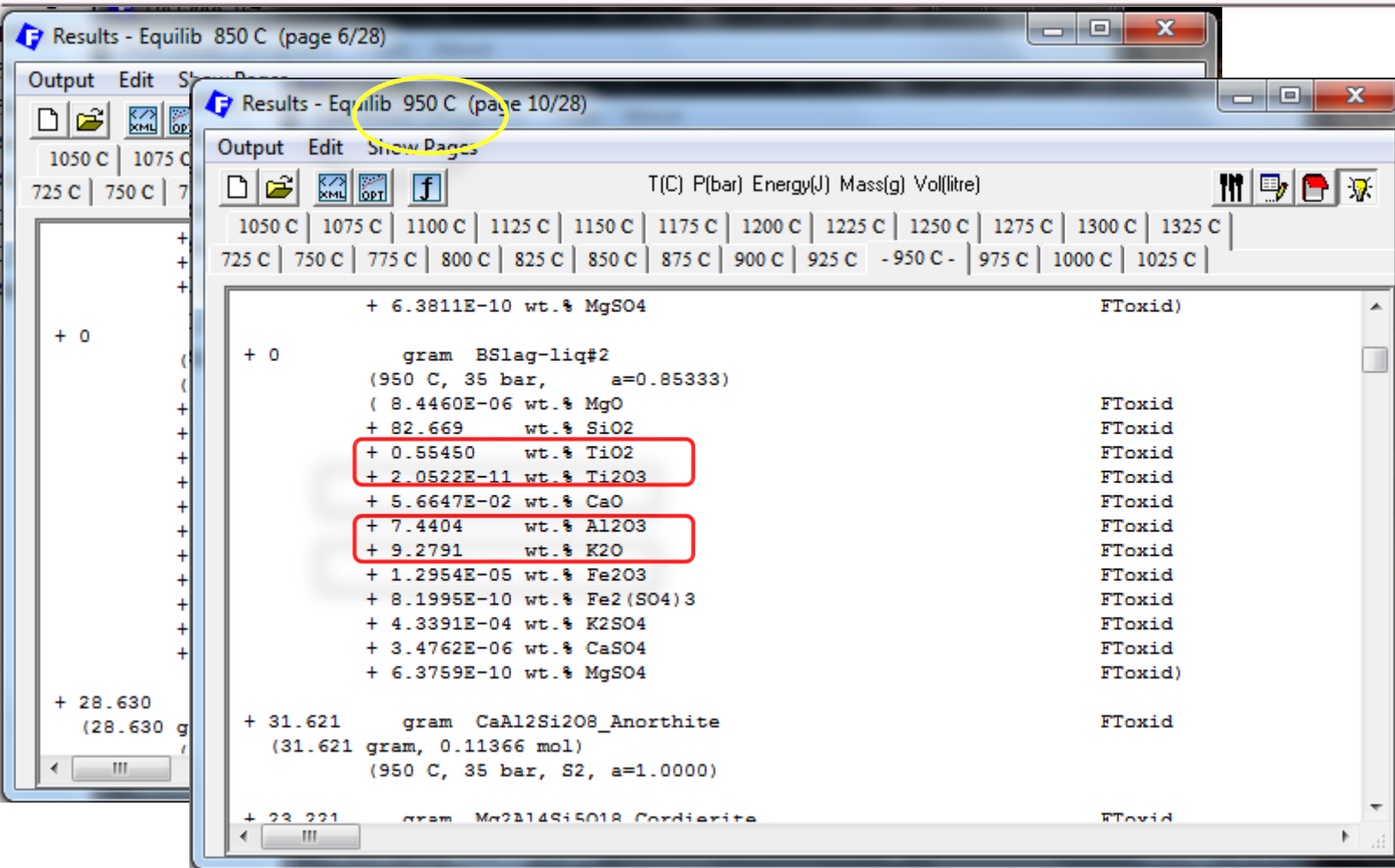
```
+ 2.9254E-04 wt.% K2SO4          FToxid
+ 1.9686E-06 wt.% CaSO4          FToxid
+ 2.5313E-11 wt.% MgSO4          FToxid)

+ 0      gram BSlag-liq#2
(850 C, 35 bar, a=0.77612)
( 5.2014E-07 wt.% MgO             FToxid
+ 82.741 wt.% SiO2                FToxid
+ 0.28498 wt.% TiO2               FToxid
+ 4.9280E-13 wt.% Ti2O3           FToxid
+ 4.7868E-02 wt.% CaO             FToxid
+ 7.2650 wt.% Al2O3                FToxid
+ 9.6604 wt.% K2O                 FToxid
+ 1.5724E-06 wt.% Fe2O3           FToxid
+ 6.4207E-11 wt.% Fe2(SO4)3       FToxid
+ 2.9141E-04 wt.% K2SO4          FToxid
+ 1.8949E-06 wt.% CaSO4          FToxid
+ 2.5330E-11 wt.% MgSO4          FToxid)

+ 28.630 gram CaAl2Si2O8_Anorthite FToxid
(28.630 gram, 0.10291 mol)
(850 C, 35 bar, S2 a=1.0000)
```

**Ideal operating conditions for the recovery of minerals from the coal ash**

# Motivation



Results - Equilib 850 C (page 6/28)

Results - Equilib 950 C (page 10/28)

Output Edit Show Pages

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

1050 C	1075 C	1100 C	1125 C	1150 C	1175 C	1200 C	1225 C	1250 C	1275 C	1300 C	1325 C	
725 C	750 C	775 C	800 C	825 C	850 C	875 C	900 C	925 C	- 950 C -	975 C	1000 C	1025 C

+ 6.3811E-10 wt.% MgSO4 FToxid)

+ 0 gram BSlag-liq#2 (950 C, 35 bar, a=0.85333)

( 8.4460E-06 wt.% MgO FToxid

+ 82.669 wt.% SiO2 FToxid

+ 0.55450 wt.% TiO2 FToxid

+ 2.0522E-11 wt.% Ti2O3 FToxid

+ 5.6647E-02 wt.% CaO FToxid

+ 7.4404 wt.% Al2O3 FToxid

+ 9.2791 wt.% K2O FToxid

+ 1.2954E-05 wt.% Fe2O3 FToxid

+ 8.1995E-10 wt.% Fe2(SO4)3 FToxid

+ 4.3391E-04 wt.% K2SO4 FToxid

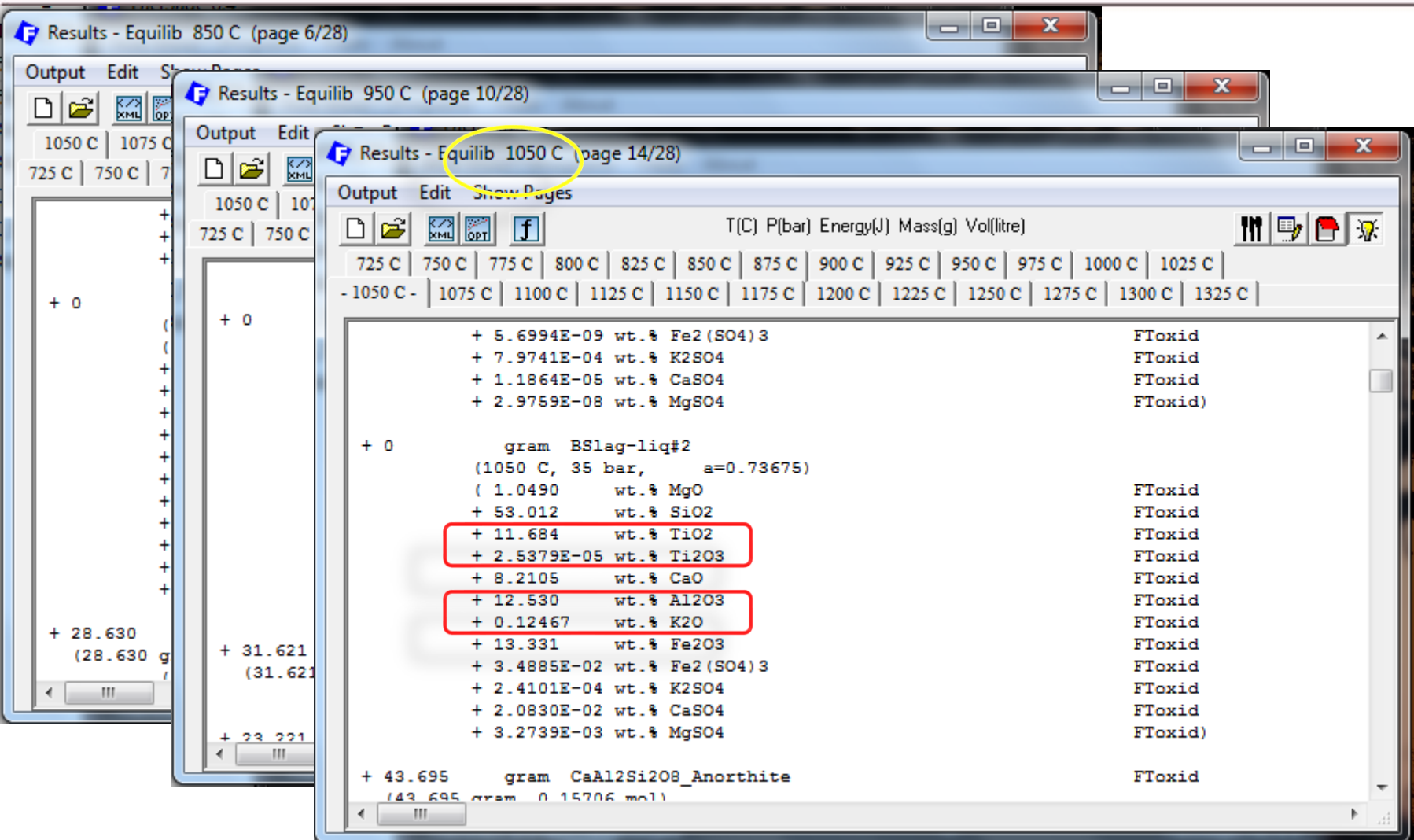
+ 3.4762E-06 wt.% CaSO4 FToxid

+ 6.3759E-10 wt.% MgSO4 FToxid)

+ 31.621 gram CaAl2Si2O8\_Anorthite FToxid (31.621 gram, 0.11366 mol) (950 C, 35 bar, S2, a=1.0000)

+ 23.221 gram Mg2Al4Si5O18\_Cordierite FToxid

# Motivation



Results - Equilib 850 C (page 6/28)

Results - Equilib 950 C (page 10/28)

Results - Equilib 1050 C (page 14/28)

Output Edit Show Pages

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

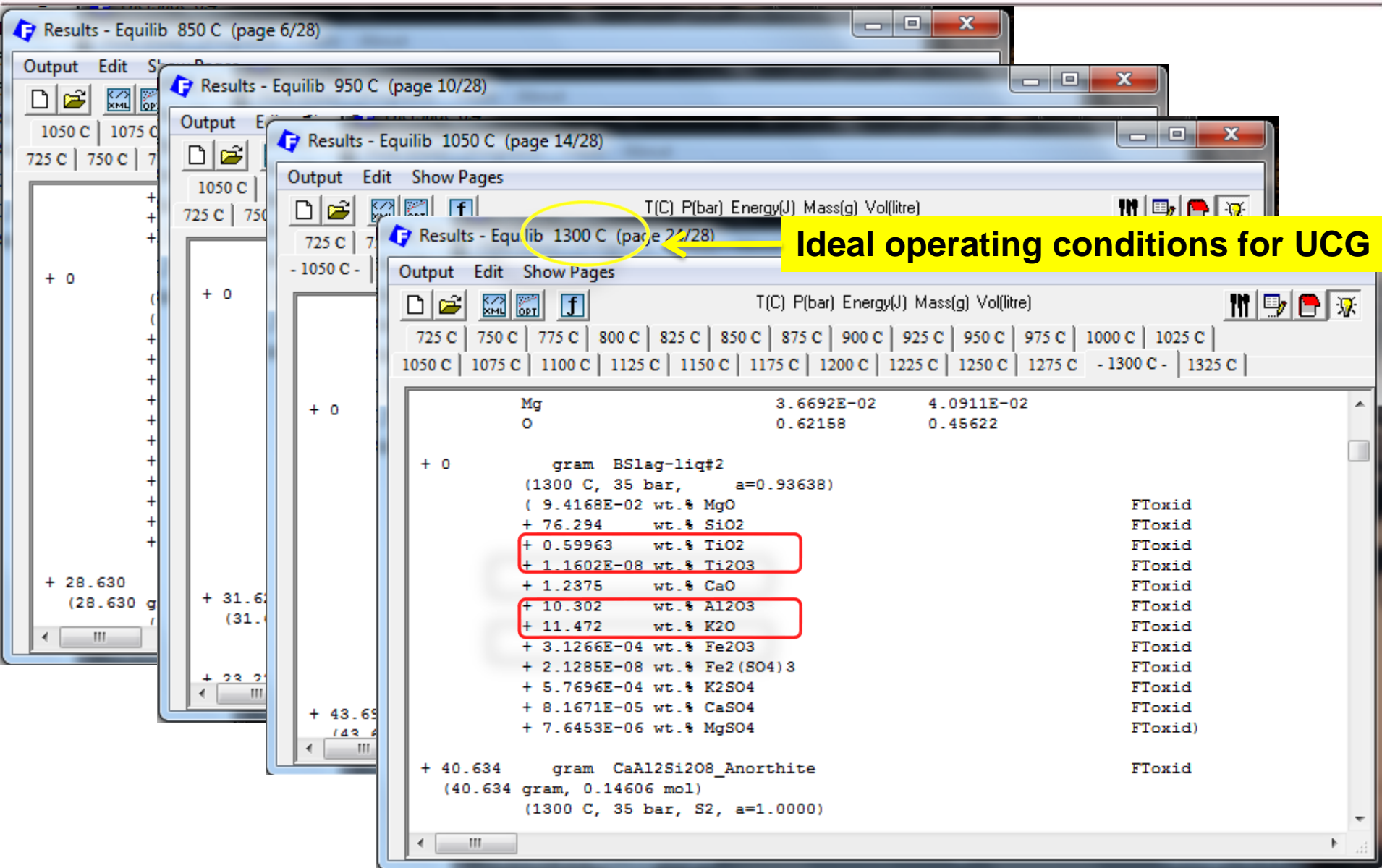
725 C | 750 C | 775 C | 800 C | 825 C | 850 C | 875 C | 900 C | 925 C | 950 C | 975 C | 1000 C | 1025 C |  
- 1050 C - | 1075 C | 1100 C | 1125 C | 1150 C | 1175 C | 1200 C | 1225 C | 1250 C | 1275 C | 1300 C | 1325 C |

```
+ 5.6994E-09 wt.% Fe2(SO4)3 FToxid
+ 7.9741E-04 wt.% K2SO4 FToxid
+ 1.1864E-05 wt.% CaSO4 FToxid
+ 2.9759E-08 wt.% MgSO4 FToxid

+ 0 gram BSlag-liq#2
(1050 C, 35 bar, a=0.73675)
( 1.0490 wt.% MgO FToxid
+ 53.012 wt.% SiO2 FToxid
+ 11.684 wt.% TiO2 FToxid
+ 2.5379E-05 wt.% Ti2O3 FToxid
+ 8.2105 wt.% CaO FToxid
+ 12.530 wt.% Al2O3 FToxid
+ 0.12467 wt.% K2O FToxid
+ 13.331 wt.% Fe2O3 FToxid
+ 3.4885E-02 wt.% Fe2(SO4)3 FToxid
+ 2.4101E-04 wt.% K2SO4 FToxid
+ 2.0830E-02 wt.% CaSO4 FToxid
+ 3.2739E-03 wt.% MgSO4 FToxid

+ 43.695 gram CaAl2Si2O8_Anorthite FToxid
(43.695 gram, 0.15706 mol)
```

# Motivation



Results - Equilib 850 C (page 6/28)

Results - Equilib 950 C (page 10/28)

Results - Equilib 1050 C (page 14/28)

Results - Equilib 1300 C (page 21/28) ← **Ideal operating conditions for UCG**

Output Edit Show Pages

T(C)	P(bar)	Energy(J)	Mass(g)	Vol(litre)																				
725 C	750 C	775 C	800 C	825 C	850 C	875 C	900 C	925 C	950 C	975 C	1000 C	1025 C	1050 C	1075 C	1100 C	1125 C	1150 C	1175 C	1200 C	1225 C	1250 C	1275 C	- 1300 C -	1325 C

+ 0

gram BSlag-liq#2  
(1300 C, 35 bar, a=0.93638)

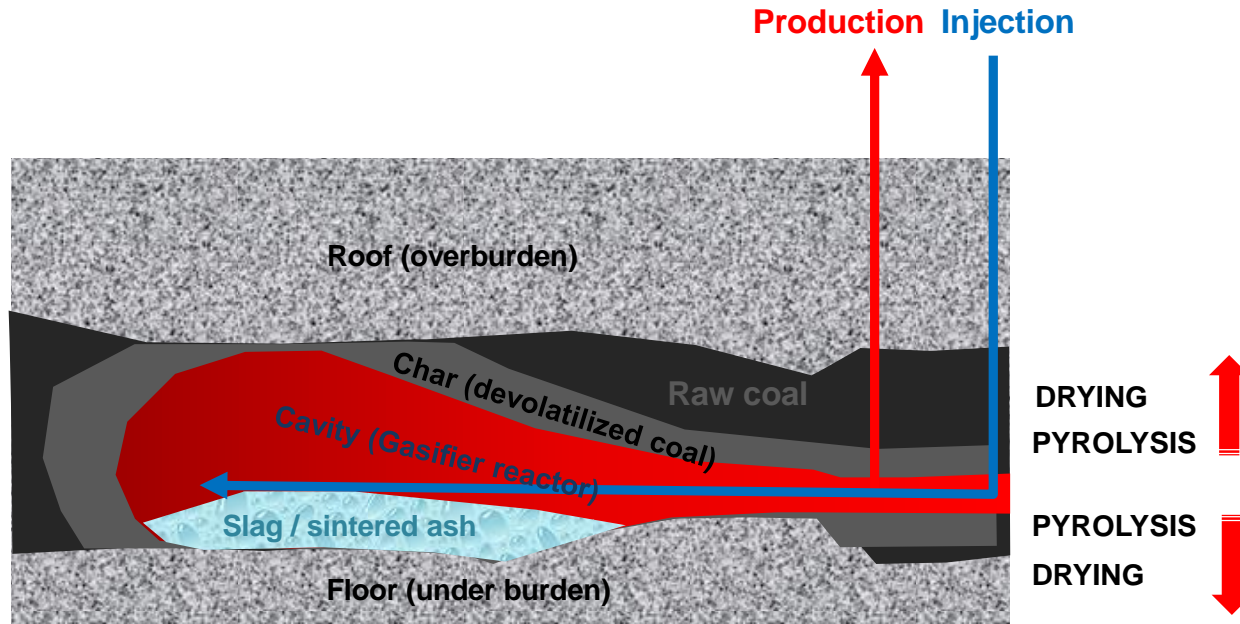
Component	wt. %	FToxid
MgO	9.4168E-02	FToxid
SiO2	76.294	FToxid
TiO2	0.59963	FToxid
Ti2O3	1.1602E-08	FToxid
CaO	1.2375	FToxid
Al2O3	10.302	FToxid
K2O	11.472	FToxid
Fe2O3	3.1266E-04	FToxid
Fe2(SO4)3	2.1285E-08	FToxid
K2SO4	5.7696E-04	FToxid
CaSO4	8.1671E-05	FToxid
MgSO4	7.6453E-06	FToxid

+ 40.634 gram CaAl2Si2O8\_Anorthite  
(40.634 gram, 0.14606 mol)  
(1300 C, 35 bar, S2, a=1.0000)

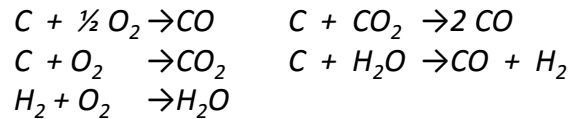
- **The influence of operating temperature on mineral species;**
- **Determination of whether these species are recoverable/leachable from the ash or lost/captured in the slag;**
- **The influence of gasification temperature on the slagging behaviour of coal samples with different characteristics;**
- **The influence of potassium on slagging behaviour when present in high concentrations within the coal;**
- **The influence of a specific potassium compound on the slagging behaviour when added to the coal in different percentages.**



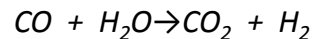
# FactSage Model of UCG Gasifier



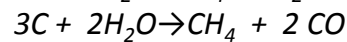
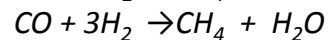
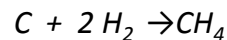
COMBUSTION → GASIFICATION → PYROLYSIS → DRYING



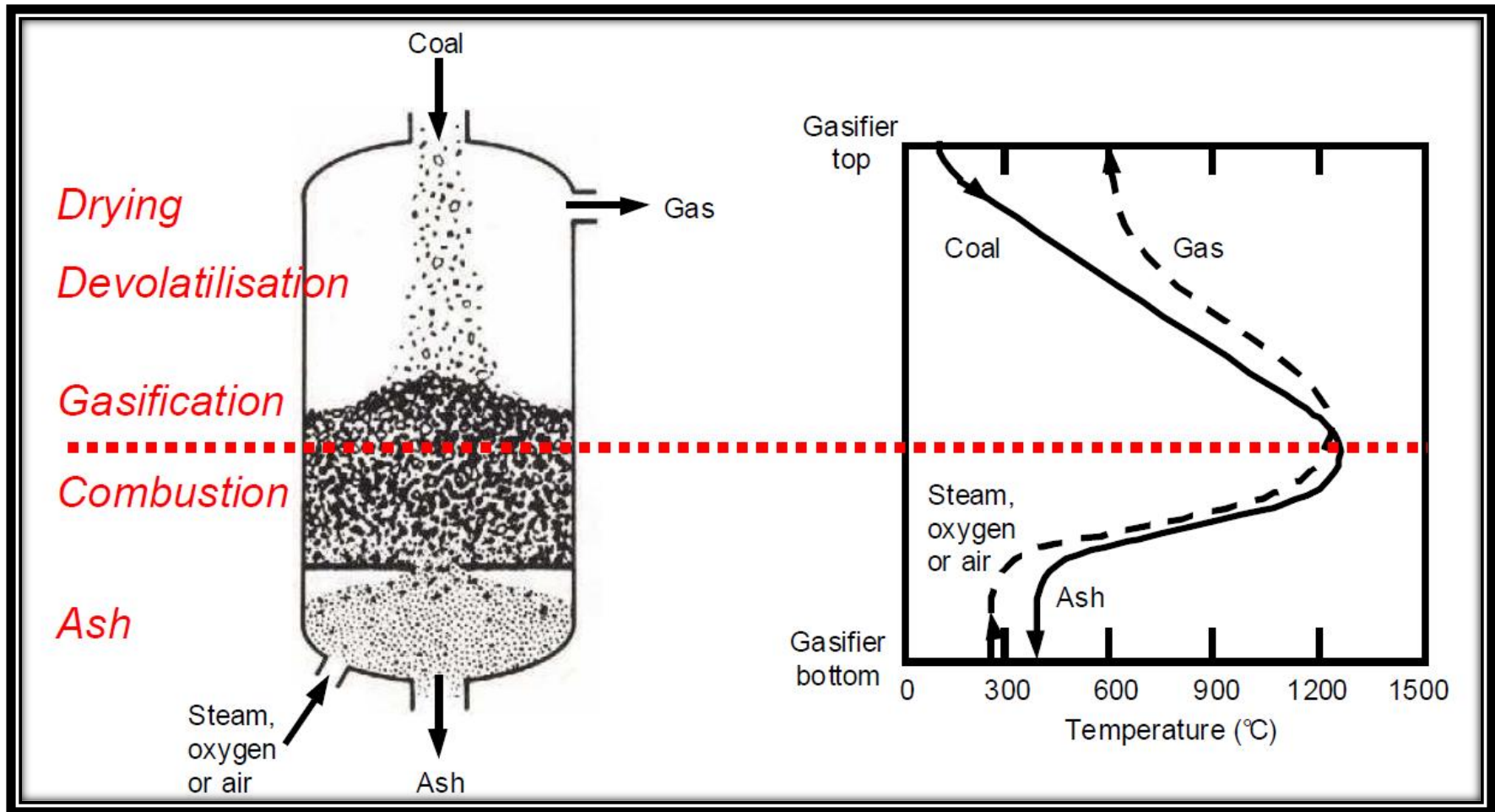
Water-gas shift:



Methane formation:

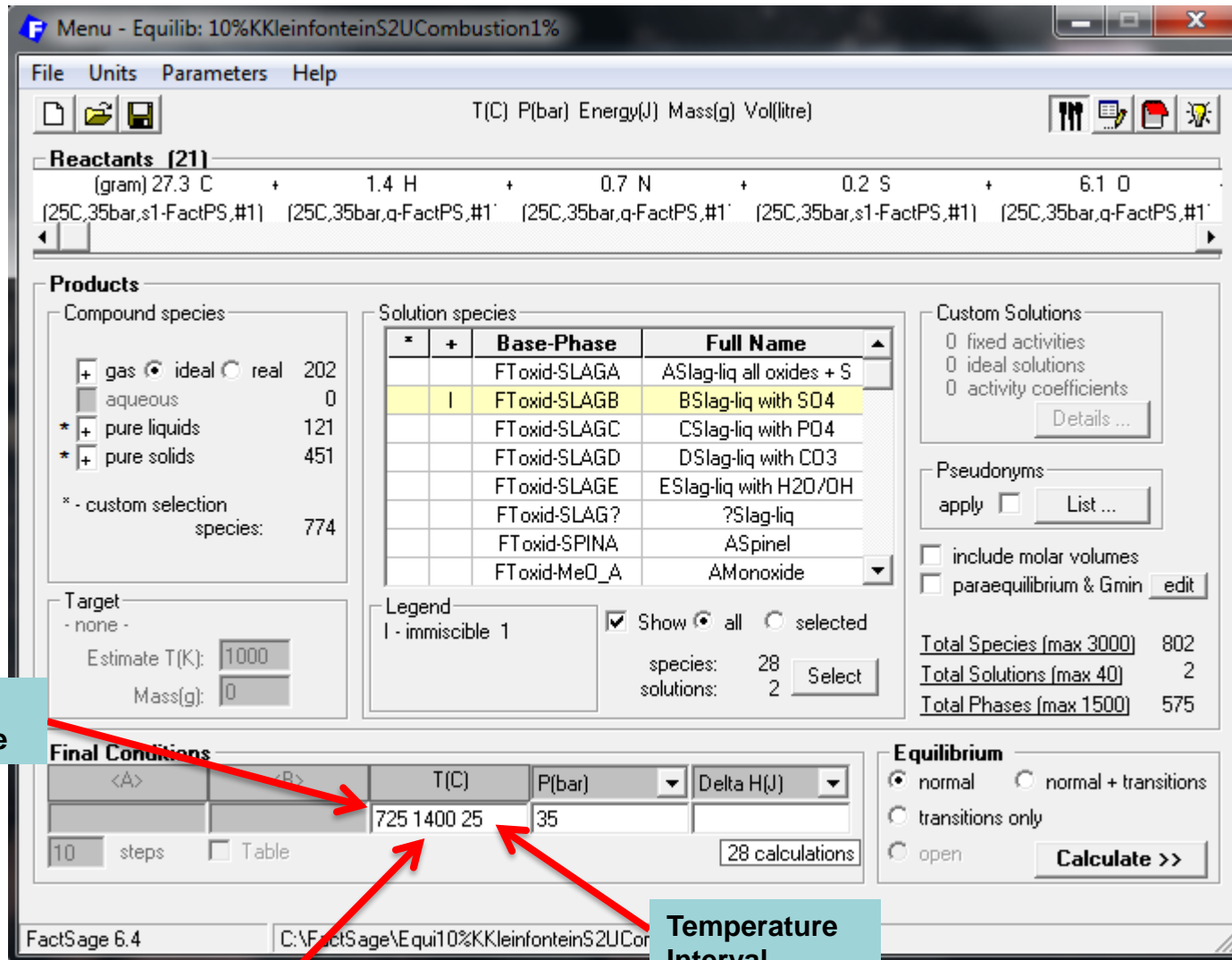


# FactSage Model of FBDB Gasifier





# FactSage Model



Menu - Equilib: 10%KKleinfonteinS2UCombustion1%

File Units Parameters Help

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

**Reactants [21]**

(gram) 27.3 C + 1.4 H + 0.7 N + 0.2 S + 6.1 O

(25C,35bar,s1-FactPS,#1) (25C,35bar,q-FactPS,#1) (25C,35bar,q-FactPS,#1) (25C,35bar,s1-FactPS,#1) (25C,35bar,q-FactPS,#1)

**Products**

Compound species

- gas ideal real 202
- aqueous 0
- pure liquids 121
- pure solids 451
- custom selection species: 774

Solution species

*	+	Base-Phase	Full Name
		FToxid-SLAGA	ASlag-liq all oxides + S
	I	FToxid-SLAGB	BSlag-liq with SO4
		FToxid-SLAGC	CSlag-liq with PO4
		FToxid-SLAGD	DSlag-liq with CO3
		FToxid-SLAGE	ESlag-liq with H2O/OH
		FToxid-SLAG?	?Slag-liq
		FToxid-SPINA	ASpinel
		FToxid-MeO_A	AMonoxide

Legend: I - immiscible 1

show all selected

species: 28 solutions: 2

Custom Solutions

- fixed activities
- ideal solutions
- activity coefficients

Pseudonyms

- include molar volumes
- paraequilibrium & Gmin

Total Species (max 3000) 802  
Total Solutions (max 40) 2  
Total Phases (max 1500) 575

Target: none

Estimate T(K): 1000  
Mass(g): 0

**Final Conditions**

<A>	<B>	T(C)	P(bar)	Delta H(J)
		725	1400	25

10 steps Table 28 calculations

**Equilibrium**

- normal normal + transitions
- transitions only
- open

Calculate >>

Starting Temperature

End Temperature

Temperature Interval

1 - 10 | 11 - 20 | 21 - 21

Mass(g)	Species	Phase	T(C)	P(total)**	Stream#	Data
42.4	C	solid-1-FactPS Graphi	25	35	1	FactPS
+ 3.2	H	gas-FactPS	25	35	1	FactPS
+ 0.7	N	gas-FactPS	25	35	1	FactPS
+ 1.0	S	solid-1-FactPS Orthor	25	35	1	FactPS
+ 13.0	O	gas-FactPS	25	35	1	FactPS
+ 52.7	SiO2	solid-1-FToxid Quartz	25	35	1	FToxid
+ 15.2	Al2O3	solid-1-FToxid gamma	25	35	1	FToxid
+ 5.2	Fe2O3	solid-1-FToxid hematit	25	35	1	FToxid
+ 0.1	(P2O5)2	solid-FactPS	25	35	1	FactPS
+ 0.8	TiO2	solid-1-FToxid Rutile	25	35	1	FToxid

Initial Conditions

Next >>

Reactants - Equilib

File Edit Table Units Data Search Help

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

1 - 10 | 11 - 20

42.4

+ 3.2

+ 0.7

+ 1.0

+ 13.0

+ 52.7

+ 15.2

+ 5.2

+ 0.1

+ 0.8

FactSage 6.4

Reactants - Equilib

File Edit Table Units Data Search Help

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

1 - 10 | 11 - 20 | 21 - 21

Mass(g)	Species	Phase	T(C)	P(total)**	Stream#	Data
+ 9.2	CaO	solid-FToxid Lime	25	35	1	FToxid
+ 3.6	MgO	solid-FToxid periclase	25	35	1	FToxid
+ 2.1	K2O	solid-FToxid	25	35	1	FToxid
+ 3.0	Na2O	solid-1-FToxid Solid-A	25	35	1	FToxid
+ 8.2	SO3	solid-FactPS	25	35	1	FactPS
+ 588.5	N2	gas-FactPS	100	35	2	FactPS
+ 320.7	O2	gas-FactPS	100	35	2	FactPS
+ 0.2	CO2	gas-FactPS	100	35	2	FactPS
+ 4.7	Ar	gas-FactPS	100	35	2	FactPS
+ 23.7	H2O	gas-FactPS Steam	100	35	2	FactPS

Initial Conditions

Next >>

FactSage 6.4 Compound: 3/12 databases Solution: 2/13 databases

Reactants - Equilib

File Edit Table Units Data Search Help

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

1 - 10 | 11 - 20

42.4

+ 3.2

+ 0.7

+ 1.0

+ 13.0

+ 52.7

+ 15.2

+ 5.2

+ 0.1

+ 0.8

FactSage 6.4

Reactants - Equilib

File Edit Table Units Data Search Help

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

1 - 10 | 11 - 20

9.2

3.6

2.1

3.0

8.2

588.5

320.7

0.2

4.7

23.7

FactSage 6.4

Reactants - Equilib

File Edit Table Units Data Search Help

T(C) P(bar) Energy(J) Mass(g) Vol(litre)

1 - 10 | 11 - 20 | 21 - 21

	Mass(g)	Species	Phase	T(C)	P(total)**	Stream#	Data
+	88.2	H2O	liquid-FactPS	25	35	1	FactPS

Initial Conditions

**Next >>**

FactSage 6.4 Compound: 3/12 databases Solution: 2/13 databases

# Coal samples

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Coal sample	Coal Rank	Base/Acid Ratio	Acidity
KFS2U	Medium Volatile Bituminous Coal	0.05	18.4
L#2	Lignite Coal	0.34	2.95
MD	High Volatile Bituminous Coal	0.33	2.91
WFS1	Medium Volatile Bituminous Coal	0.16	6.31

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# Characterization Results

	KFS2U	KFS2U + 10%K	L#2	L#2 + 10%K	MD	MD + 10%K	WFS1	WFS1 + 10%K
<b><u>Proximate analysis</u></b>								
Moisture content	3.3	2.7	18.0	26.2	3.7	2.9	4.0	3.8
Ash content	28.3	35.0	20.5	26.1	28.5	36.0	22.4	30.7
Volatile content	18.2	20.5	30.5	24.3	21.2	23.9	21.9	23.0
Fixed carbon	50.2	41.8	31.0	23.4	46.5	37.2	51.7	42.5
<b><u>Ultimate analysis</u></b>								
% C	56.4	47.8	43.3	32.8	53.7	45.6	59.1	50.6
% H	2.9	2.4	3.2	2.5	2.6	2.3	3.1	2.7
% N	1.2	1.2	0.7	0.6	1.3	1.2	1.4	1.2
% O	7.4	10.6	13.3	11.2	8.9	11.3	8.9	10.4
% S	0.5	0.3	1.0	0.5	1.3	0.8	1.1	0.8



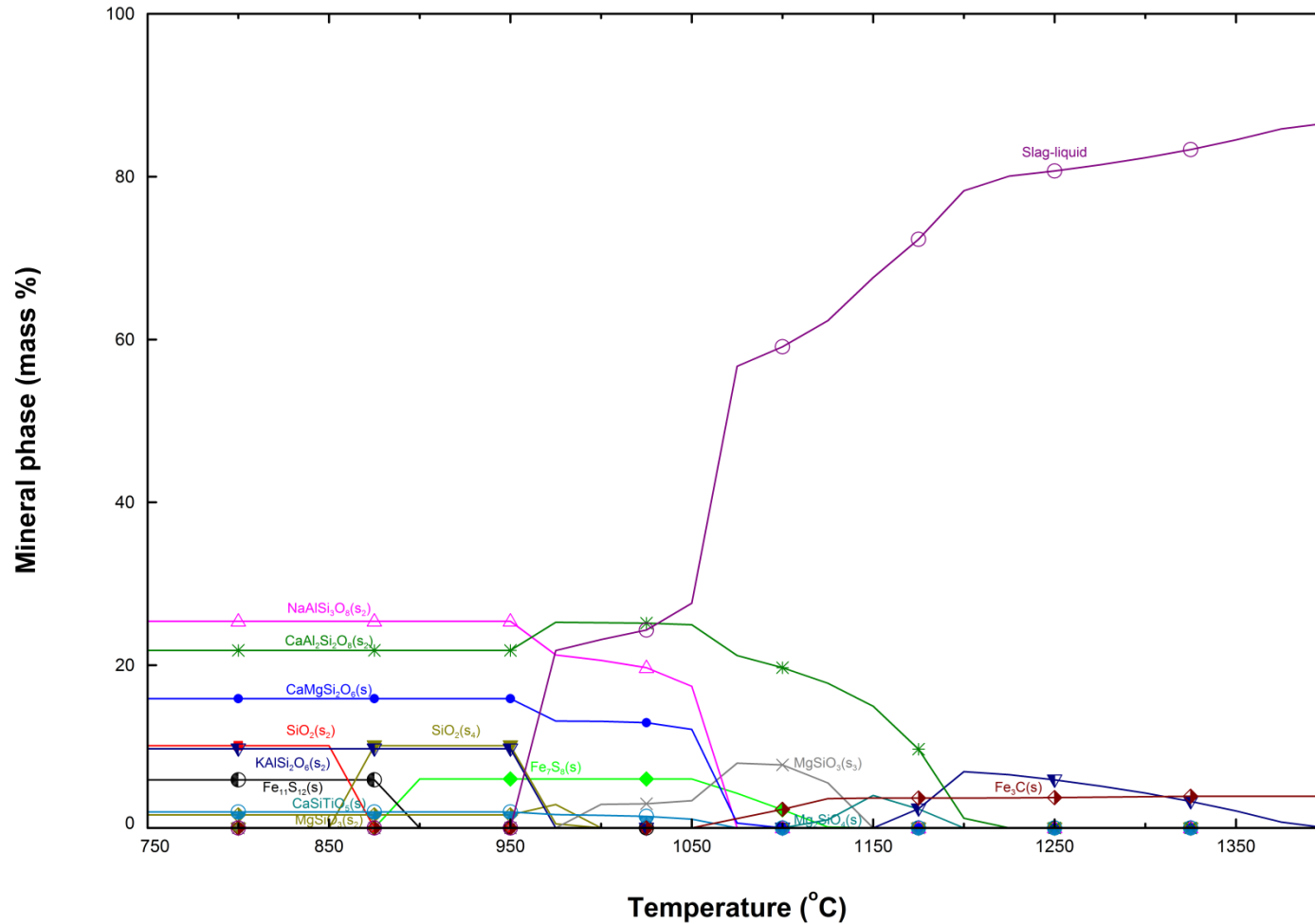
	KFS2U	KFS2U + 10%K	L#2	L#2 + 10%K	MD	MD + 10%K	WFS1	WFS1 + 10%K
<b><u>XRF</u></b>								
Al <sub>2</sub> O <sub>3</sub>	28.46	19.42	15.07	9.22	25.18	17.82	25.89	16.50
CaO	1.51	1.00	9.13	5.05	13.02	8.94	5.07	3.37
Cr <sub>2</sub> O <sub>3</sub>	0.12	0.02	0.03	nd	0.03	nd	0.03	0.02
Fe <sub>2</sub> O <sub>3</sub>	2.00	1.18	5.16	3.06	5.64	3.73	5.54	3.53
K <sub>2</sub> O	0.69	29.79	2.06	38.78	1.11	27.61	0.43	31.82
MgO	0.78	0.52	3.52	2.02	3.15	2.18	1.81	1.20
MnO	0.03	0.03	0.05	0.03	0.07	0.05	0.05	0.04
Na <sub>2</sub> O	Nd	0.17	2.94	1.88	nd	0.15	nd	0.27
P <sub>2</sub> O <sub>5</sub>	0.07	0.04	0.10	0.07	0.26	0.19	0.10	0.14
SiO <sub>2</sub>	63.17	43.17	52.16	31.66	41.60	29.70	55.15	34.17
TiO <sub>2</sub>	1.52	1.05	0.80	0.47	1.79	1.27	1.34	0.84
V <sub>2</sub> O <sub>5</sub>	0.02	0.02	0.02	0.02	0.02	0.02	0.03	0.03
ZrO <sub>2</sub>	0.09	0.06	0.03	0.02	0.08	0.05	0.06	0.04
BaO	0.02	nd	0.56	0.33	0.15	0.11	0.08	0.07
SrO	0.02	0.01	0.25	0.14	0.27	0.19	0.07	0.08
SO <sub>3</sub>	1.48	3.53	8.10	7.24	7.62	7.99	4.33	7.89

# Drying, Devolatilization and Gasification Zone 1

# Results

## Mineral transformation in Zone 1 - L#2

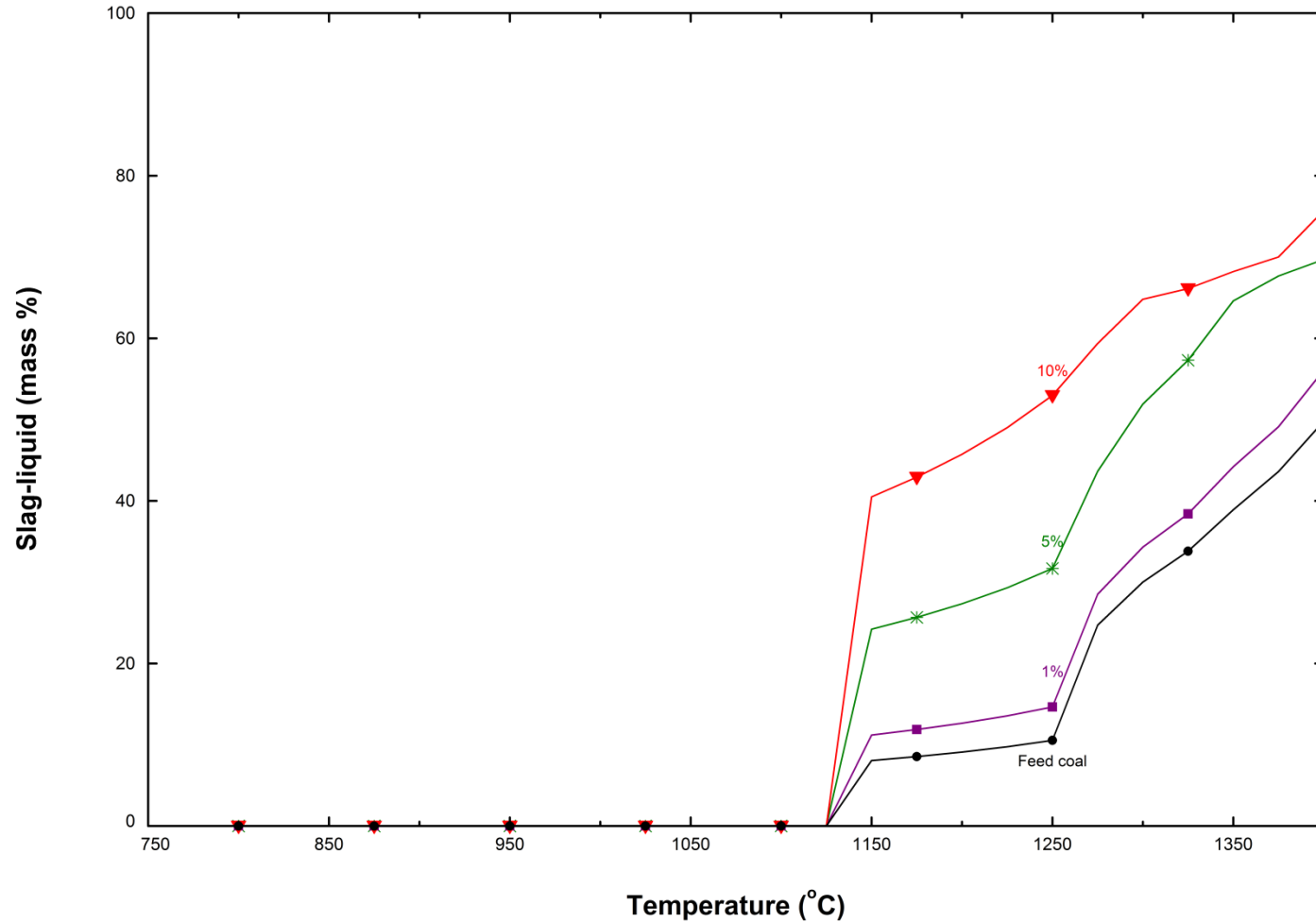
C:\FactSage\Equi0.res - AC Collins



# Results

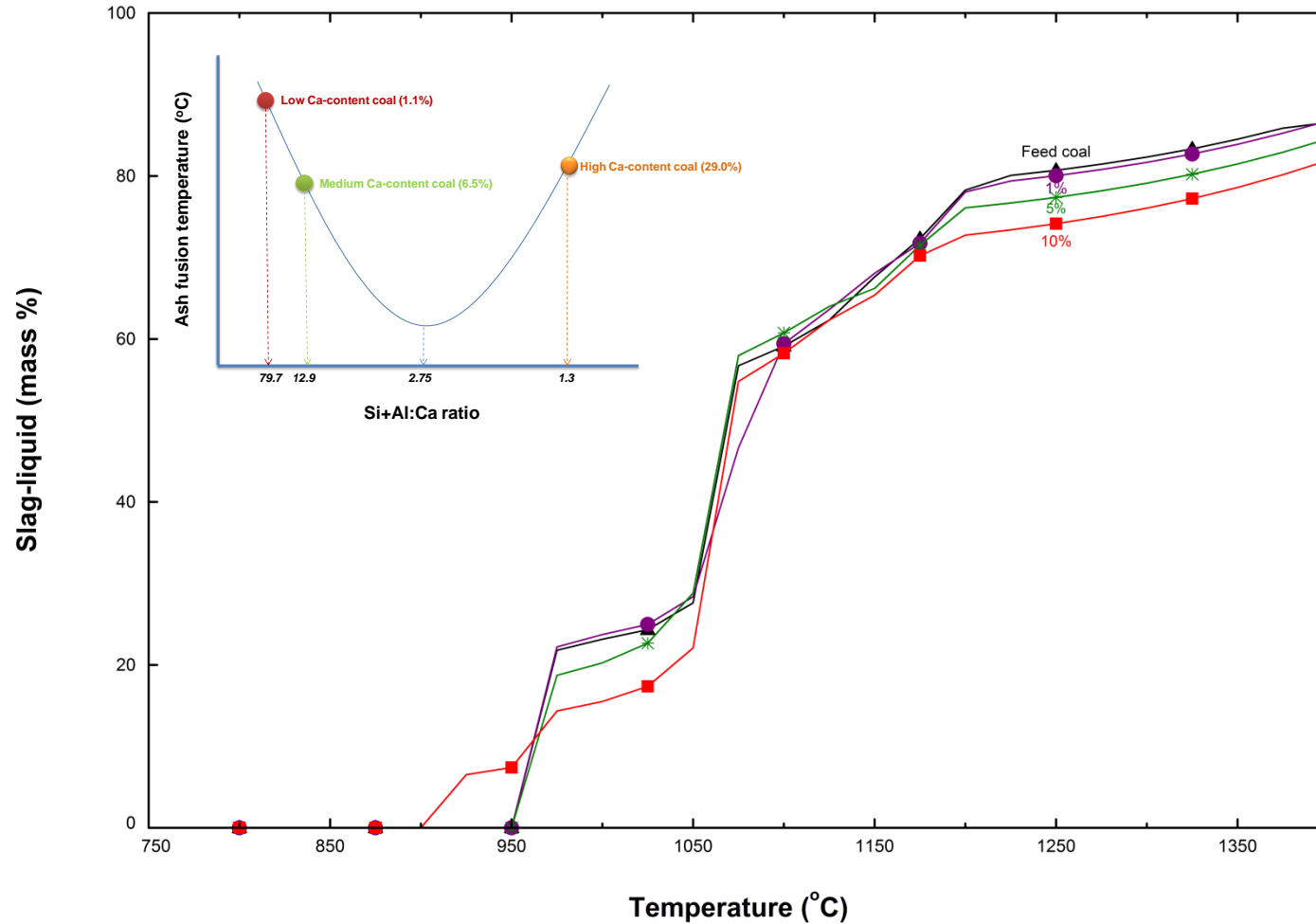
## Influence of Potassium addition on slag formation in Zone 1 - KFS2U

Addition of potassium according to ash percentage  
C:\FactSage\Equi0.res - AC Collins



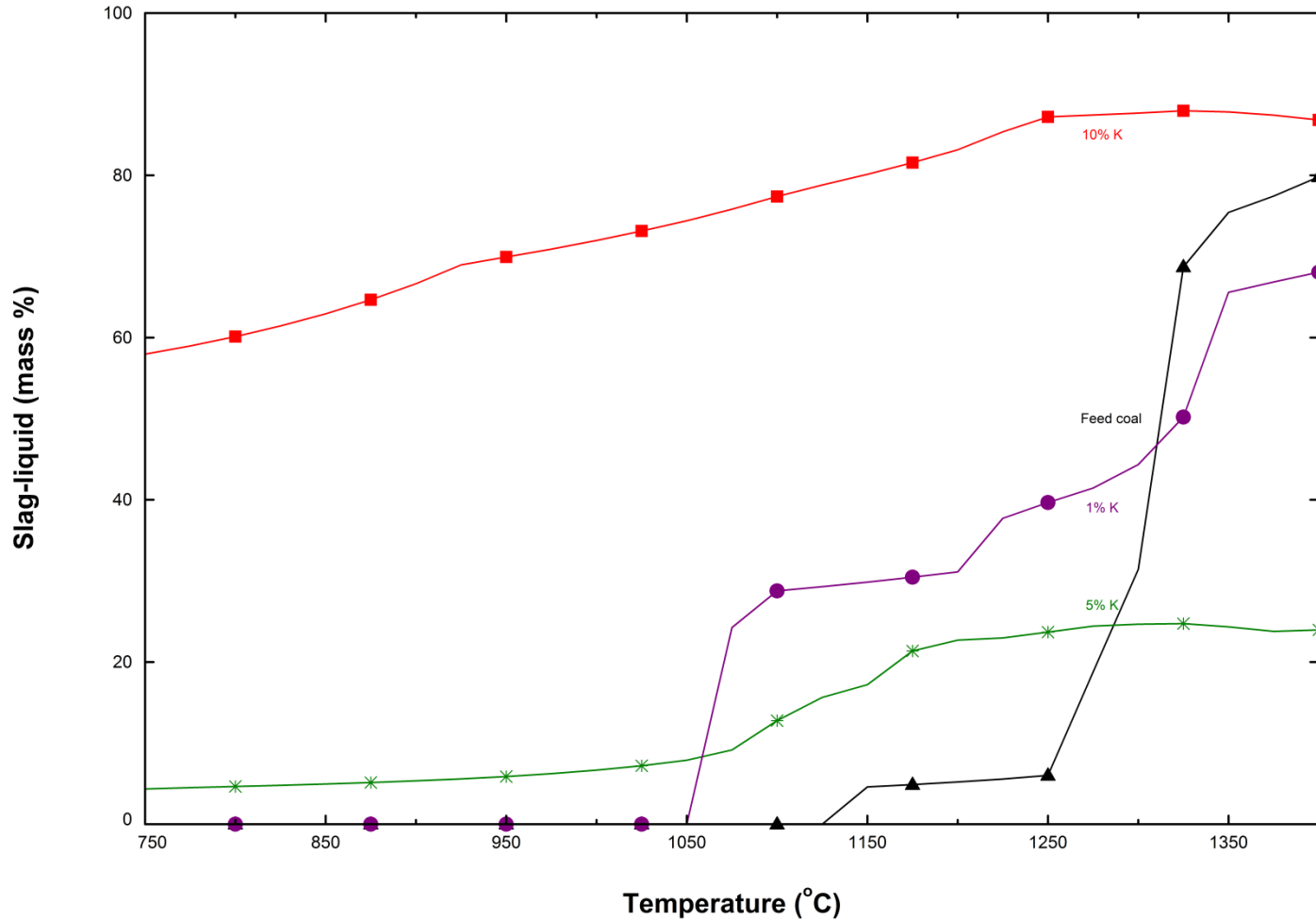
## Influence of Potassium addition of slag formation in Zone 1 - L#2

Addition of potassium according to ash percentage  
 C:\FactSage\Equi0.res - AC Collins



## Influence of Potassium addition on slag formation in Zone 1 - WFS1

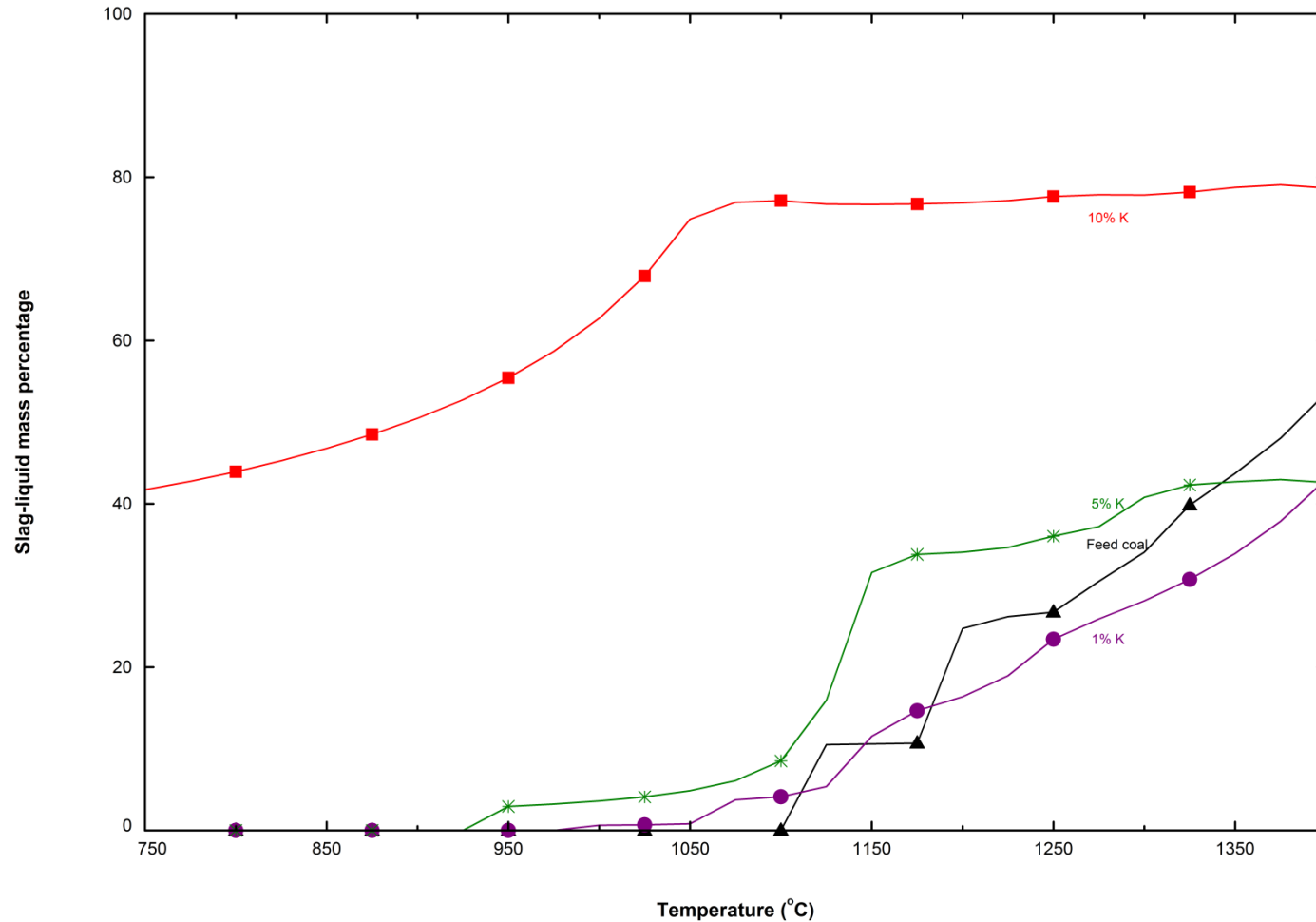
Addition of potassium according to coal mass  
C:\FactSage\Equi0.res - AC Collins





## Influence of Potassium addition on slag formation in Zone 1 - MD

Addition of potassium according to coal mass  
C:\FactSage\Equi0.res - AC Collins

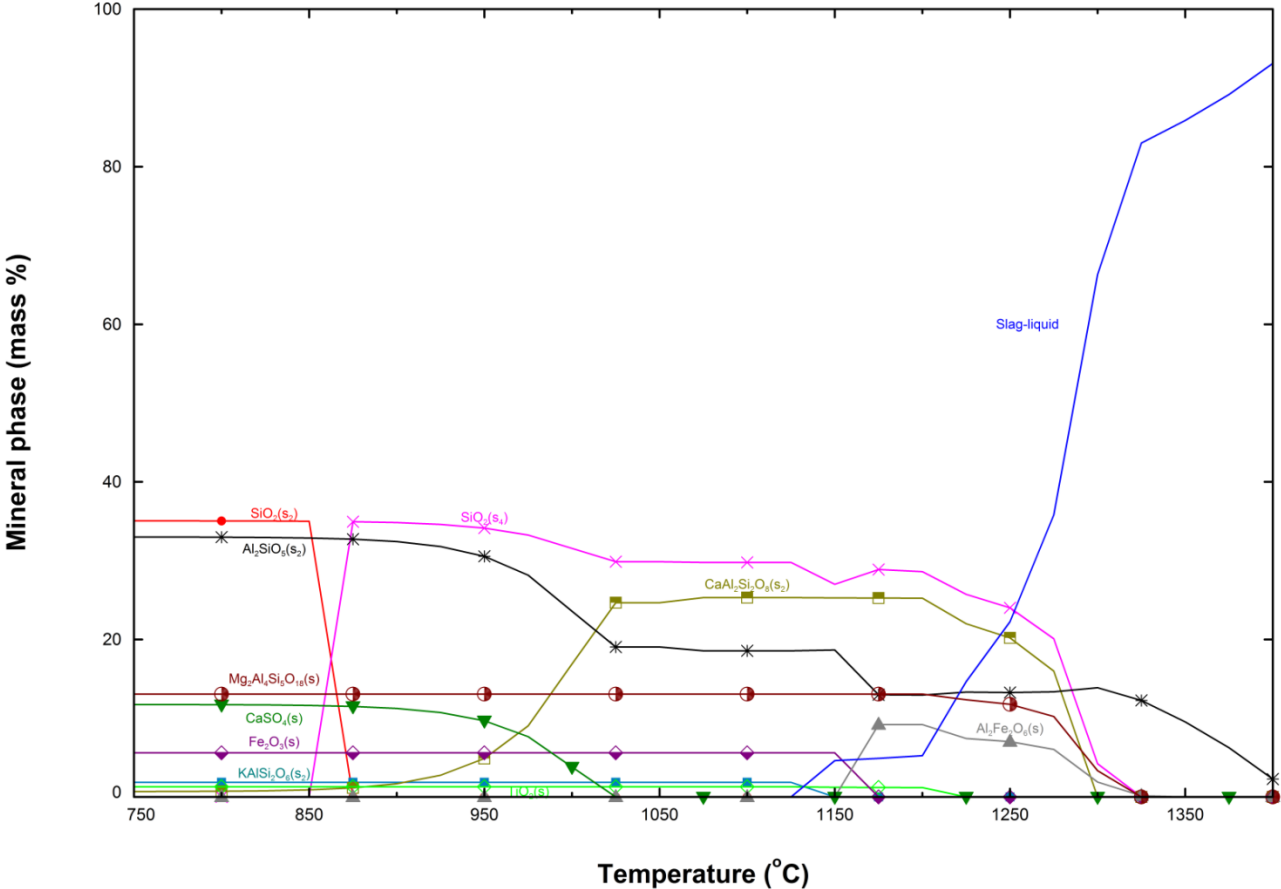


# Combustion and Ash Zone 2

# Results

## Mineral transformation in Zone 2 - WFS1

C:\FactSage\Equi0.res - AC Collins

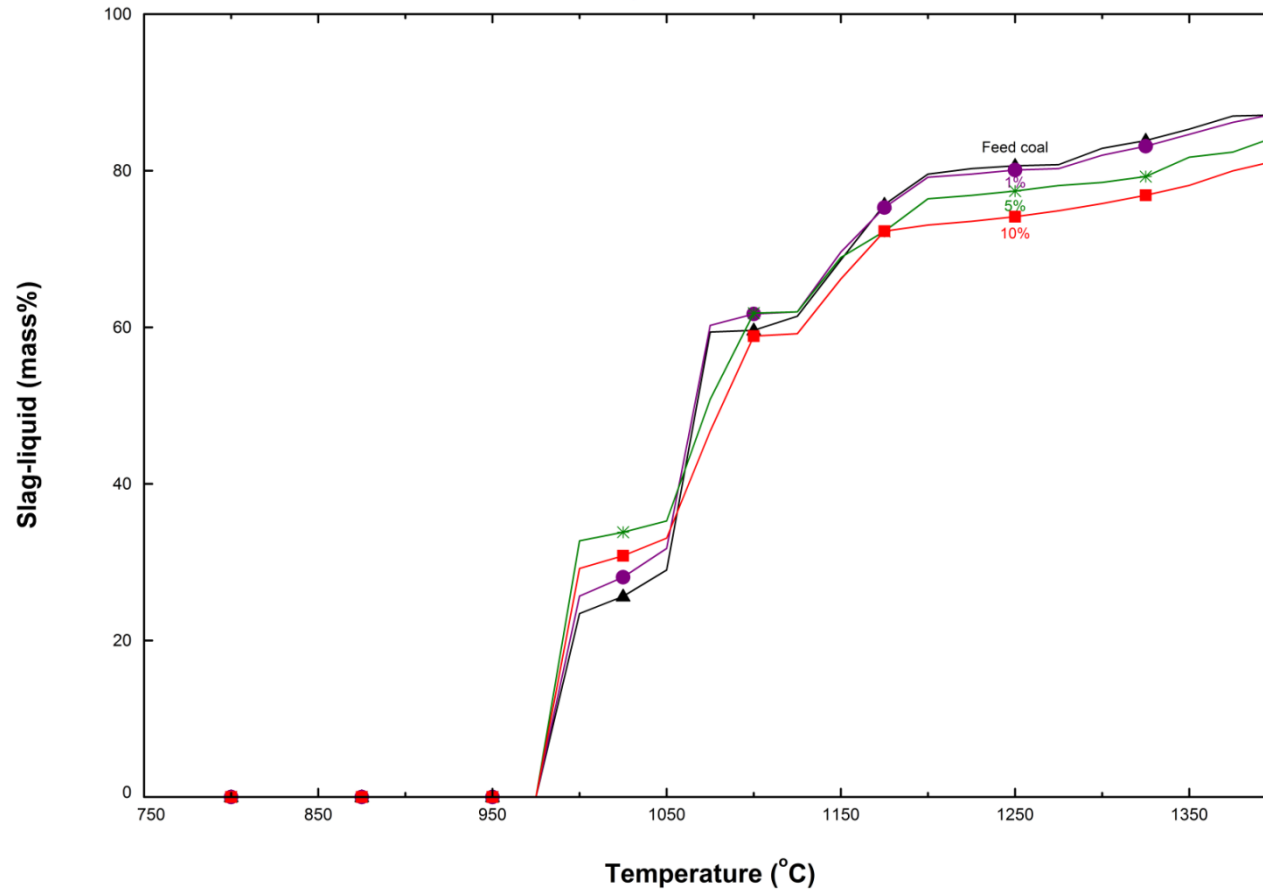


Cooling process in combustion zone

# Results

## Influence of Potassium addition on slag formation in Zone 2 - L#2

Addition of potassium according to ash percentage  
C:\FactSage\Equi0.res - AC Collins

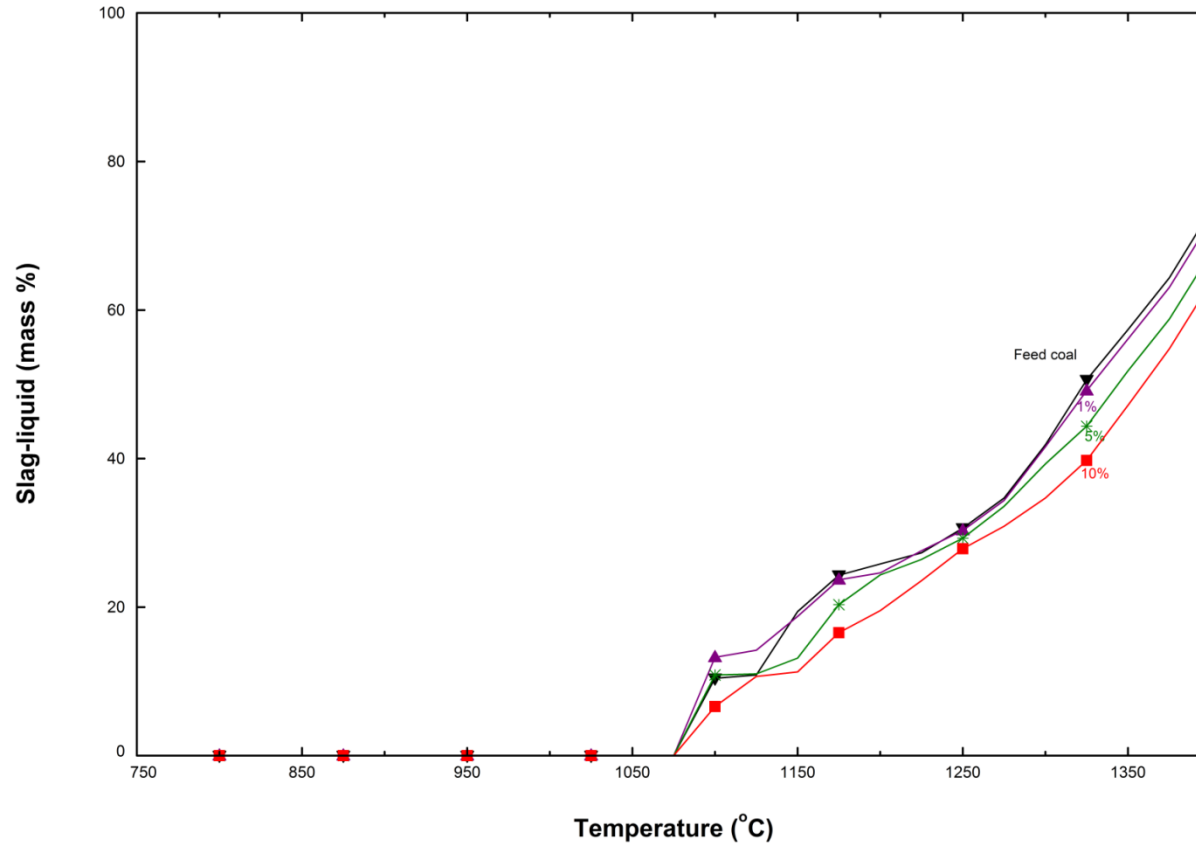


Cooling process in combustion zone

# Results

## Influence of Potassium addition on slag formation in Zone 2 - MD

Addition of potassium according to ash percentage  
C:\FactSage\Equi0.res - AC Collins

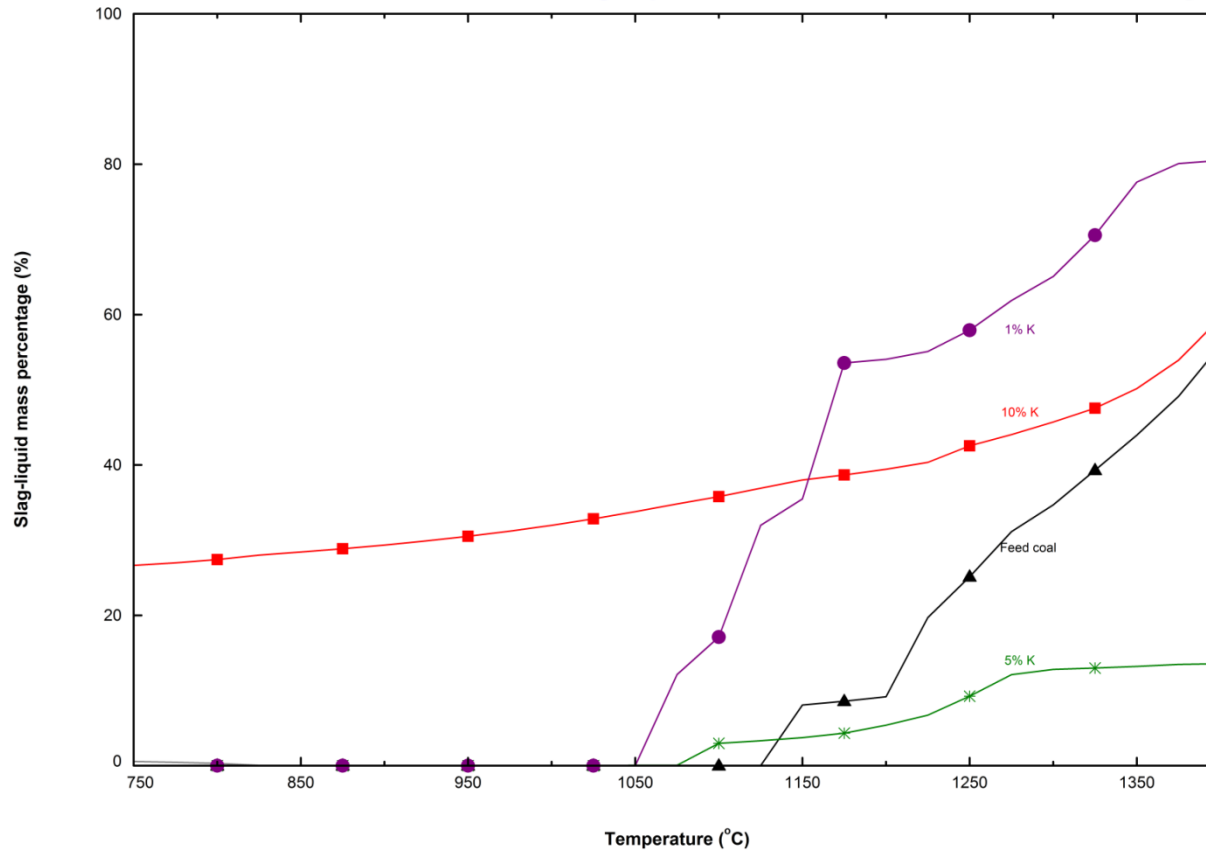


Cooling process in combustion zone

# Results

## Influence of Potassium addition on slag formation in Zone 2 - KFS2U

Addition of potassium according to coal mass  
C:\FactSage\Equi0.res - AC Collins

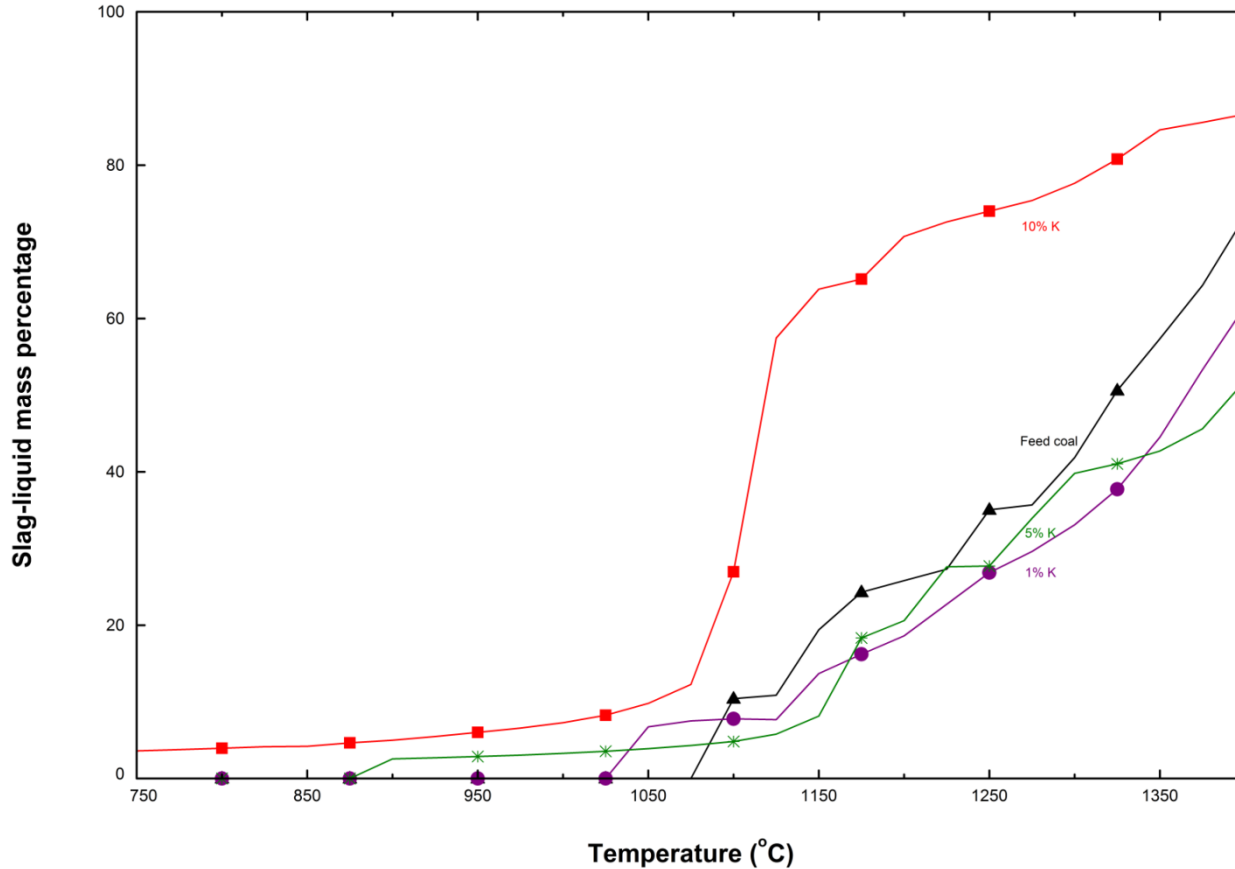


Cooling process in combustion zone



## Influence of Potassium addition on slag formation in Zone 2 - MD

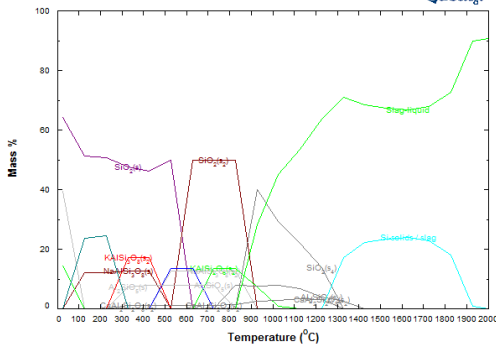
Addition of potassium according to coal mass  
C:\FactSage\Equi0.res - AC Collins



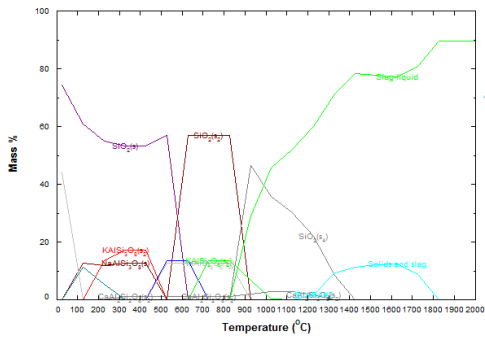
← Cooling process in combustion zone

# Gasification science – mineral transformation and ash behaviour

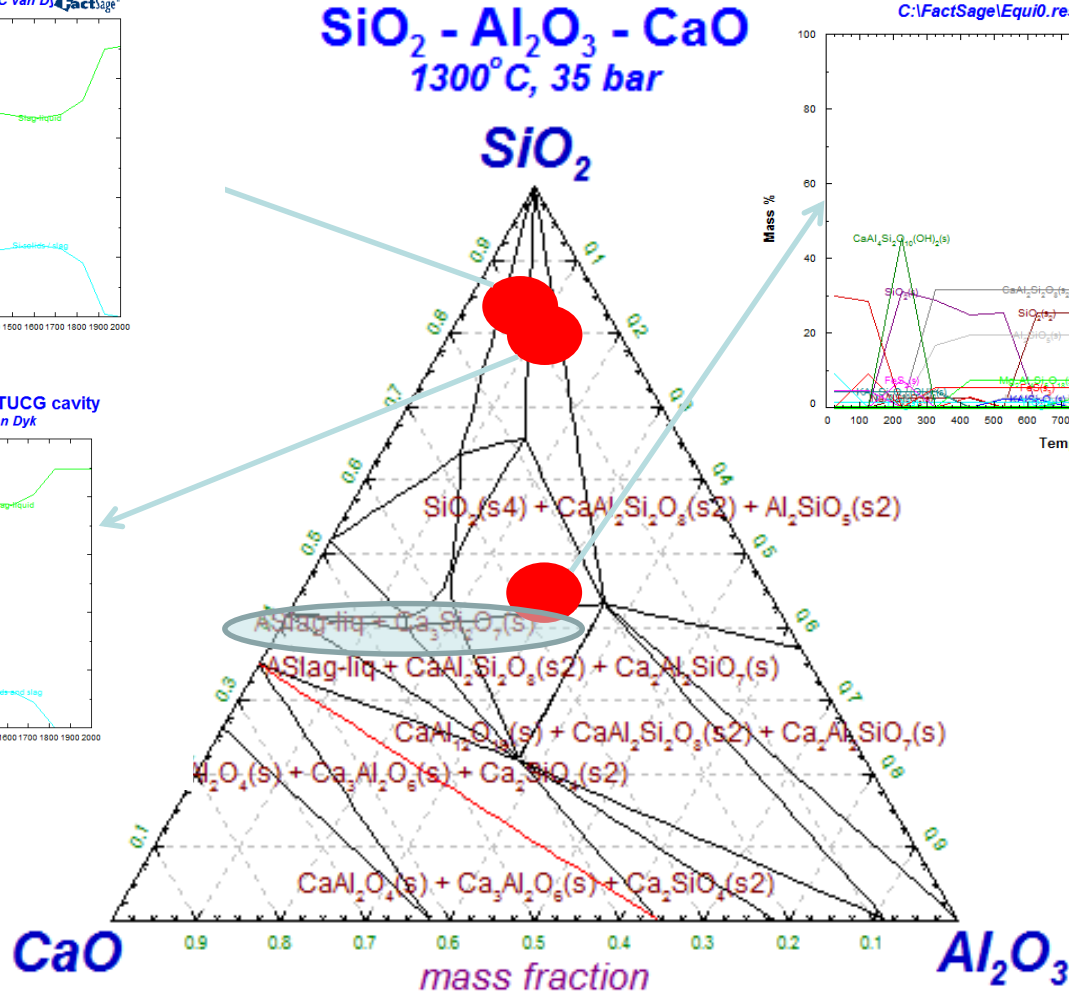
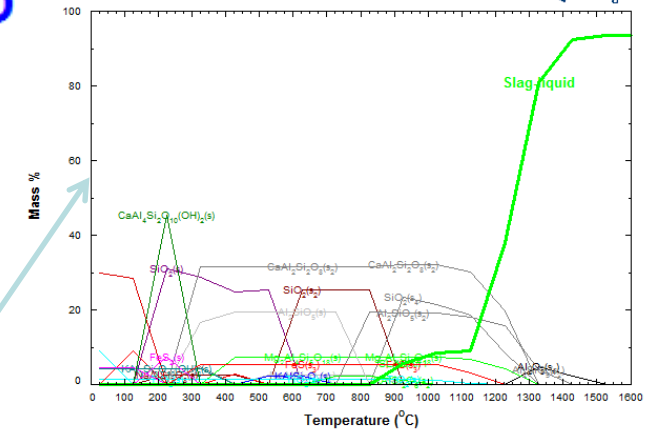
Mineral matter transformation in roof formation of TUCG cavity  
C:\FactSage\Equi0.res 24Apr14 - Dr. JC van Dyk



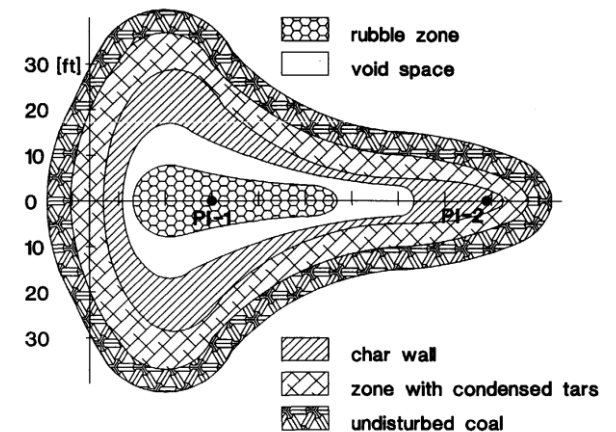
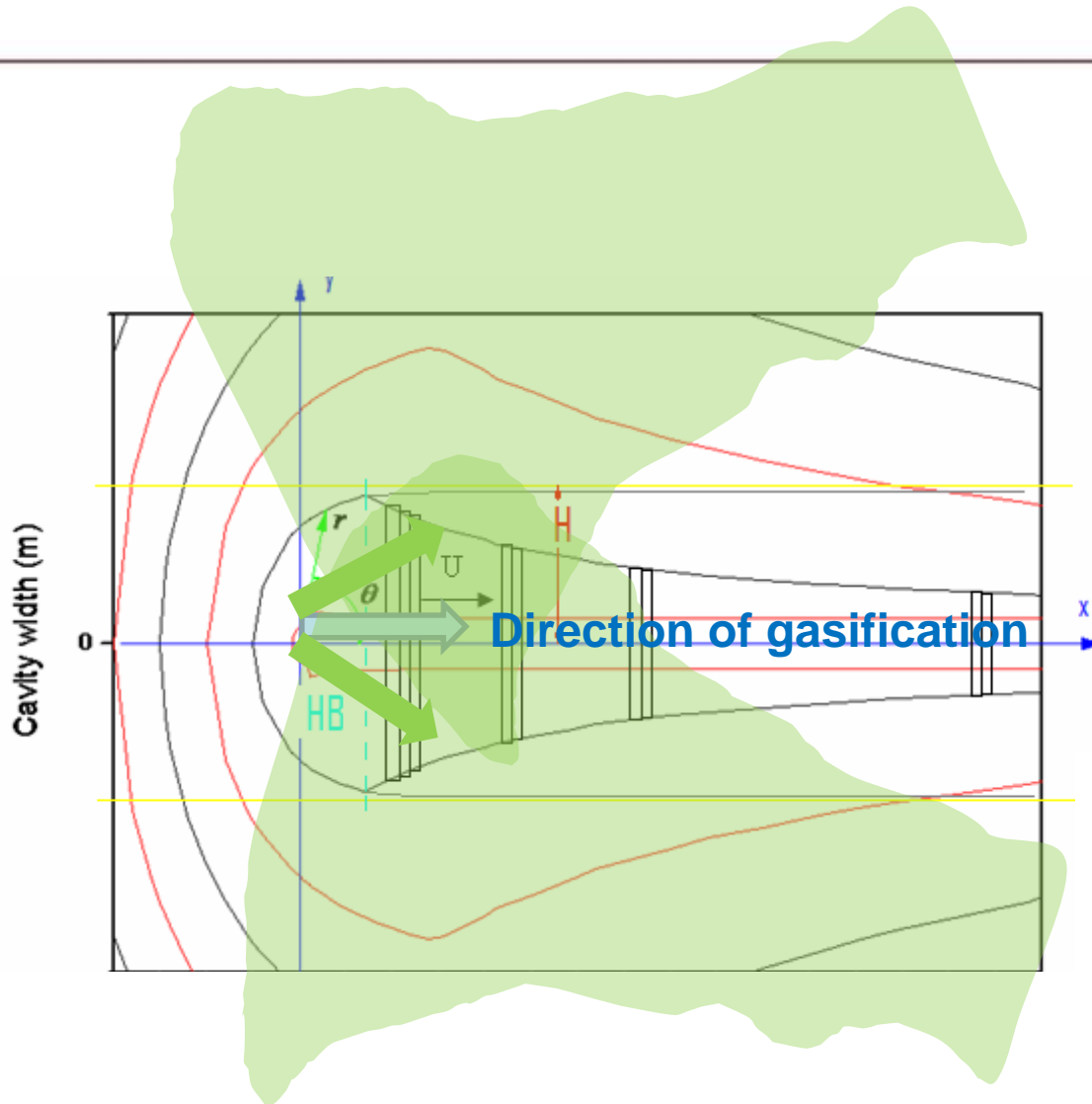
Mineral matter transformation in floor of TUCG cavity  
C:\FactSage\Equi0.res 25Apr14 - Dr. JC van Dyk



Mineral matter transformation in the TUCG cavity  
C:\FactSage\Equi0.res 24Apr14 - Dr. JC van Dyk

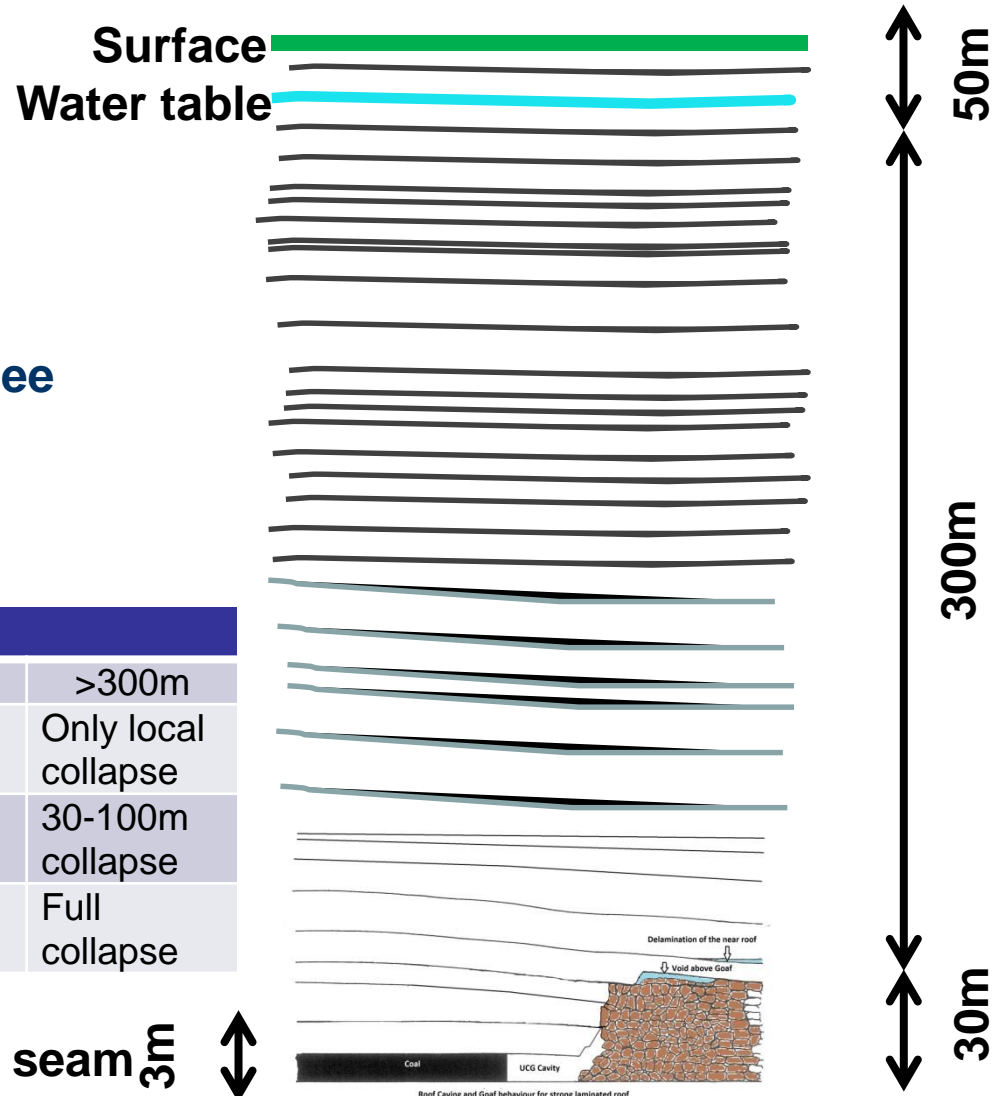


# What next on UCG simulations?



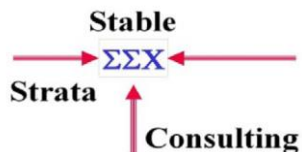
# Geophysical study

**Not to Scale!**  
 (Otherwise you will not be able to see what is happening...)

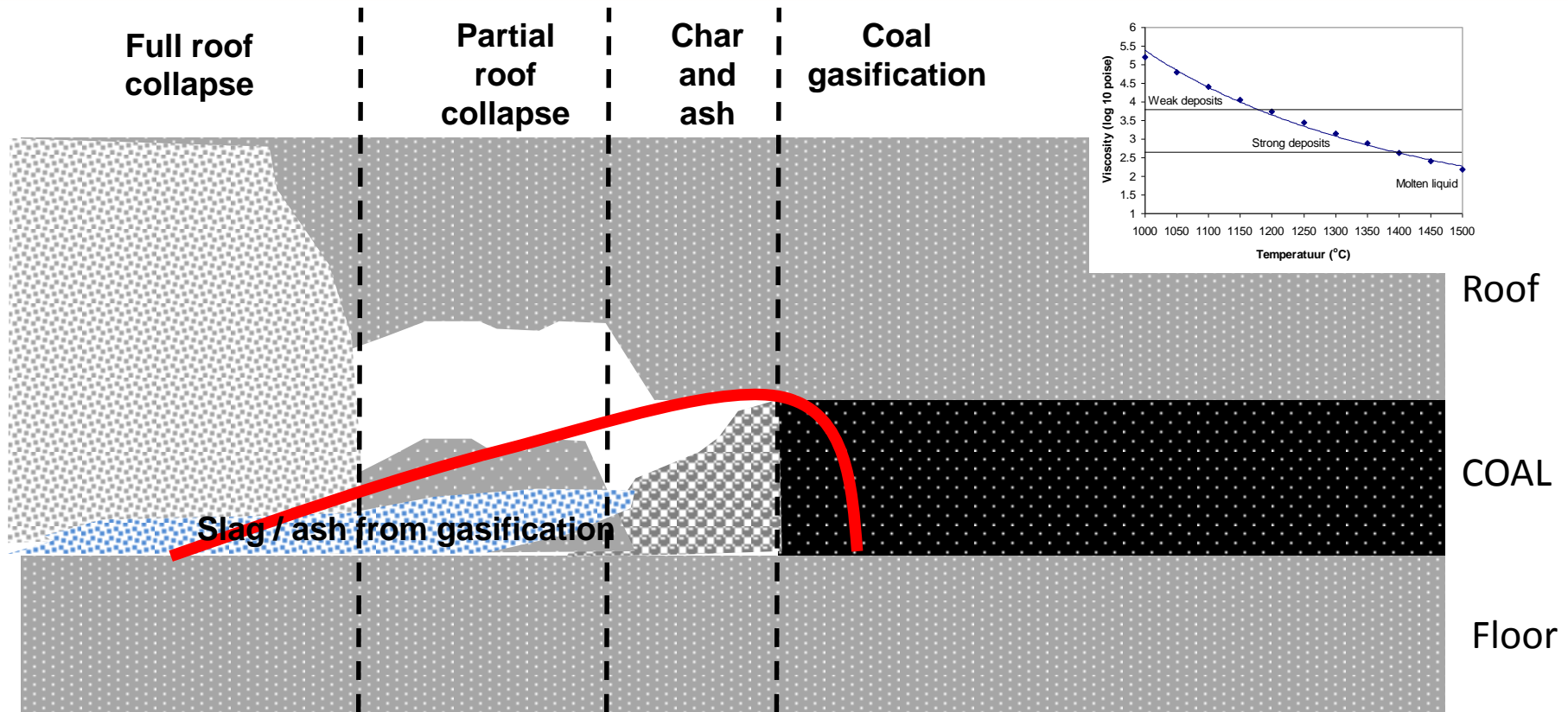


Coal seam 3m

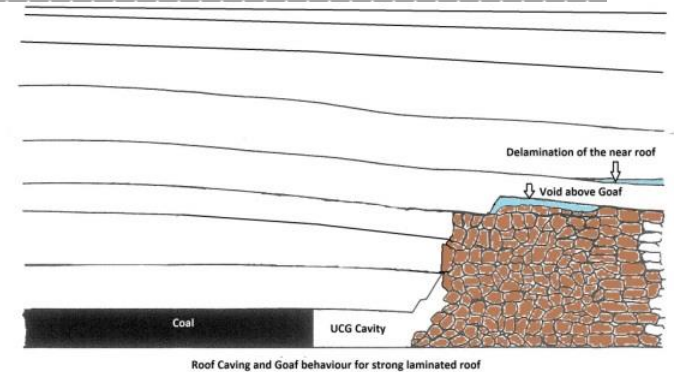
		ADVANCE		
		100m	100-300m	>300m
Cavity Width	0-100m	Only local collapse	Only local collapse	Only local collapse
	100-300m	Only local collapse	30-100m collapse	30-100m collapse
	>300m	Only local collapse	30-100m collapse	Full collapse



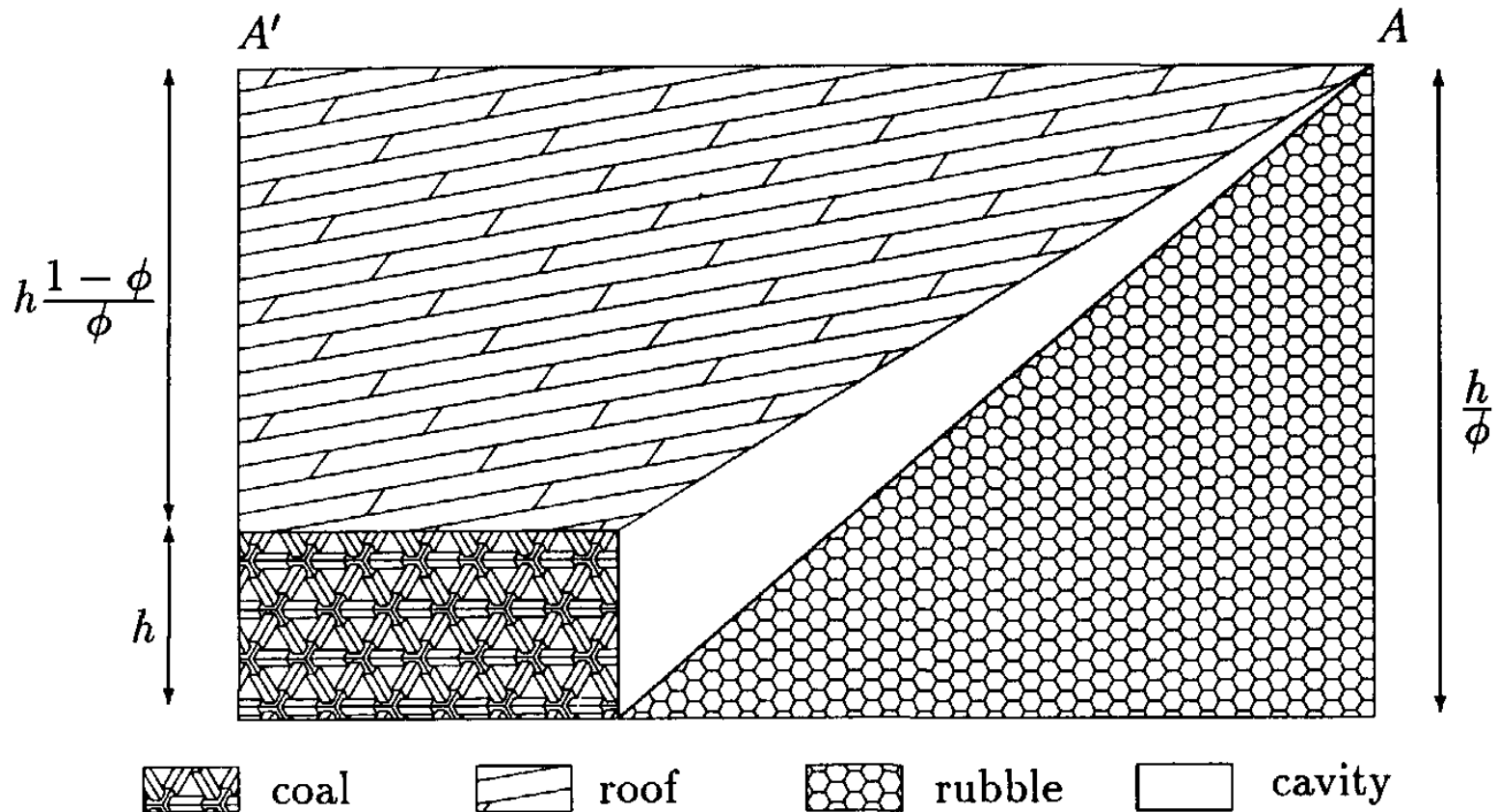
# The approach!



		ADVANCE		
		100m	100-300m	>300m
Cavity Width	0-100m	Only local collapse	Only local collapse	Only local collapse
	100-300m	Only local collapse	30-100m collapse	30-100m collapse
	>300m	Only local collapse	30-100m collapse	Full collapse



# Approach from Batenburg (Delft)



- **Percentage K added according to ash percentage:**

- Similar trend for all the percentage simulations for a coal sample;
- Behavior dependent on the coal composition;

i.e. KFS2U =  $\uparrow$  in slag formation with  $\uparrow$  % K

L#2 =  $\downarrow$  in slag formation with  $\uparrow$  % K

MD =  $\downarrow$  in slag formation with  $\uparrow$  % K

Slag formation temperatures does not vary much due to small increase in % K.

- **High concentrations (10%) K added to the coal:**

- High percentage slag formed,
- Lower slag formation temperatures.

- **Concentrations (1 & 5%) K added to the coal:**

- Different slag percentages, dependent on coal mineral composition,
- Lower slag formation temperatures.

- **Mineral transformation and slag formation is depended on the operating temperatures;**
- **Lower operating temperatures may produce mineral species, which will be leachable from the ash;**
  - Lower temperatures indicate higher percentage of solid mineral species present that at temperatures  $> 1000^{\circ}\text{C}$ .
- **The mineral composition and acid/base ratio for the different coal types influences the slag formation temperatures and mineral transformation.**



# Upcoming International Conference in South-Africa



## SAUCGA - SOUTH AFRICAN UNDERGROUND COAL GASIFICATION ASSOCIATION

*UNDERGROUND COAL GASIFICATION : UNLOCKING SOUTHERN AFRICA'S UNMINEABLE COAL RESERVES*

24 AUGUST 2015



### Thirty Third Annual International PITTSBURGH COAL CONFERENCE

*Conference Announcement and Call for Papers*

Abstracts Due 1 March 2016

8 – 12 August 2016

Cape Town, South Africa

International Convention Centre



**Thank You**