Equations-of-State models in ChemApp/ChemSheet, a Fluent-KilnSimu link, and a metallurgical process simulation tool based on SimuSage

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3 topics

- EOS/GCM models with ChemSheet Karri Penttilä, VTT Process Chemistry
- FKS Fluent-KilnSimu Karri Penttilä, VTT Process Chemistry Eero Immonen, ProcessFlow
- A metallurgical process simulation tool based on SimuSage Stephan Petersen, GTT-Technologies





EOS/GCM Models With ChemSheet

Karri Penttilä, VTT Process Chemistry



Organic systems with ChemApp

- There is a demand from users of ChemApp and other programs built on it (such as ChemSheet) to use data for organic systems on their calculations.
- Enough data is available, but for use with a variety of equation of state (EOS) models.
- Both the EOS models (primarily used for gaseous substances) and the models for activity/fugacity coefficients used for condensed phases have not really been available in ChemApp.
- Work has been done to make such data available to ChemApp, first as "user defined models".



Modelling the Excess Enthalpy for Ethanol-Water with Redlich-Kister



Excess enthalpy of the system ethanol-water at 298,15 and 343.15 K

M. Hack: "Simultane Modellierung thermodynamischer Eigenschaften wäßriger Systeme mit weiten Mischungslücken", Dissertation, RWTH Aachen, Fakultät für Maschinenwesen,1998



Modelling the Excess Enthalpy for Ethanol-Water with Redlich-Kister



Boiling point diagram of the system ethanol-water at 0.1013 MPa

M. Hack: "Simultane Modellierung thermodynamischer Eigenschaften wäßriger Systeme mit weiten Mischungslücken", Dissertation, RWTH Aachen, Fakultät für Maschinenwesen, 1998



Modelling the Excess Enthalpy for Ethanol-Water with Redlich-Kister



Calculated boiling point diagrams of the system ethanol-water at 4 temperatures M. Hack: "Simultane Modellierung thermodynamischer Eigenschaften wäßriger Systeme mit weiten Mischungslücken", Dissertation, RWTH Aachen, Fakultät für Maschinenwesen,1998



Generalized Cubic EOS

VTT Process Chemistry

The model for generalized two parameter cubic EOS is given as:

$$P = \frac{RT}{v - b_m} - \frac{a_m}{(v + c_1 b_m)(v + c_2 b_m)}$$

- *P* pressure [Pa]
- R gas constant [J/mol-K]
- T temperature [K]
- *v* molar volume [m³/mol]
- a attraction parameter [Pa-m^{6/}mol²]
- b covolume parameter [m³/mol]

The classical mixing rules for a_m and b_m parameters are given as:

$$a_{m} = \sum_{i} \sum_{j} x_{i} x_{j} \sqrt{a_{i} a_{j}} (1 - k_{ij}) \quad b_{m} = \sum_{i} x_{i} b_{i} \qquad P_{c} \text{ critical pressure [Pa]} \\ T_{c} \text{ critical temperature [K]} \\ a_{i} = \Omega_{a} \frac{R^{2} T_{ci}^{2}}{P_{ci}} \alpha_{i} (T_{ri}, \omega_{i}) \qquad b_{i} = \Omega_{b} \frac{R T_{ci}}{P_{ci}} \quad \text{from:} \left(\frac{\partial P}{\partial v_{i}}\right)_{T} = \left(\frac{\partial^{2} P}{\partial v_{i}^{2}}\right)_{T} = 0$$

where k_{ij} is emperical interaction parameter ($k_{ii} = 0$).

SRK: $c_1 = 0$ $c_2 = 1$ $W_a = 0.42748$ $W_b = 0.08664$ PR: $c_1 = -\sqrt{2}$ $c_2 = +\sqrt{2}$ $W_a = 0.45724$ $W_b = 0.07780$

GCM Activity Models

In Unifac group contribution model (GCM) the species are represented with molecular groups. The activity coefficient is a sum of the contributions of the groups in each molecular species in the mixture:

$$\ln \gamma_i = \frac{\ln \gamma_i^C}{\text{combinatorial}} + \frac{\ln \gamma_i^R}{\text{residual}}$$

Combinatorial contribution depends on the sizes and shapes of the molecules. It is given as:

$$\ln \gamma_{i}^{C} = 1 - V_{i} + \ln V_{i} - 5q_{i} \left(1 - \frac{V_{i}}{F_{i}} + \ln \frac{V_{i}}{F_{i}} \right)$$

Residual contribution depends on the group areas and group interactions. It is given as:

$$\ln \gamma_i^R = \sum_k v_k^{(i)} \left(\ln \Gamma_k - \ln \Gamma_k^{(i)} \right)$$

•ChemSheet uses ChemApp library for Gibbs energy minimization.

•A special version of ChemApp calls a specific routine in usermod.dll file when it encounters a certain phase model name in ChemSage data-file (USX?). User calculates lng_i (partial excess Gibbs energies) for each species in phase at given temperature, pressure and composition.

SUBROUTINE USERGX[DLLEXPORT](NTXX, ITXX1, ITXX2, ITXX3, ITXX4, WTXX, PXX, TXX, G0XX, V0XX, XXX, MDLXX, NCXX, GINT, GXX)

•••

- **XXX** = MOLE FRACTIONS OF ALL PHASE CONSTITUENTS
- **MDLXX** = **MODEL NAME FOR THE PHASE**
- GXX = PARTIAL DERIVATIVE OF EXCESS INTEGRAL GIBBS ENERGY WITH RESPECT TO THE MOLE FRACTIONS OF ALL PHASE CONSTITUENTS



User Defined Model for ChemApp

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PSRK	RK Ethanol-Water										
3	3	0	2	2	0						
C				н					0		
	12.0)110		1.	008)	15.999	0			
6	1	2	3	4	5	6					
1	1										
GAS											
USXG											
etha	nol										
42	2.	. 0	6.0	1.	0						
150	0.000	0	- 2	25692	4.78	3	31.301	527	-39.120400	74642560E-01	
.70	51434	17E-0	52	28869	6.00)					
3.0	00000	0000		.0	0	. 00000	000	.00	.00000000	.00	
150	1.000	0	- 3	82890	6.32	2	922.89	779	-167.59709	.0000000	
.00	00000	0		0000	0000)					
3.0	00000	000		.0	0	. 00000	000	.00	.00000000	.00	
wate	r										
44	0.	. 0	2.0	1.	0						
110	0.000	0	- 2	25547	5.83	L	-15.282	587	-25.781640	74748581E-02	
.92	05931	L6E-0	71	L3999	.660)					
3 1	107.2	2718		99.0	0	. 00000	000	.00	.00000000	.00	
400	0.000	0	1	15215	2.28	3	164.70	757	-53.145789	80540038E-04	
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3 -83	3128.	276		99.0	0 !	5947.3	700	.50	.00000000	.00	
600	0.000	0	- 4	4694	39.	5	1472.44	424	-155.19083	.0000000	
.00	00000	0		2982	6420)E+09					
3 7	78290).99		99.0	0 - 0	54372.	340	.50	.00000000	.00	
600	1.000	0	- 3	80298	9.70	5	260.622	247	-60.574924	.0000000	
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User Defined Model for ChemApp

VTT Process Chemistry

2							
1 2	1	3	0	INIT			
1	0	41	0.6350	Ethanol	Acentric	factor	[]
2	0	42	63.835	Ethanol	Critical	pressure	[bar]
2 1	0	43	516.20	Ethanol	Critical	temperature	[K]
2 1	0	44	167.00	Ethanol	Critical	volume	[cm3/mol]
2 2	0	41	0.344861	Water	Acentric	factor	[]
2 2	0	42	220.55	Water	Critical	pressure	[bar]
2 2	0	43	645.15	Water	Critical	temperature	[K]
2 2	0	44	55.838	Water	Critical	volume	[cm3/mol]
2	0	40	Δ	Fthanol	λlfa Mati		
2		1)	1				
1 2	T	49	1.3327	Ethanol	MCT		
[] 20	=	н20					
2 0 0	0	21	0	CALC END			



- •Unifac (for only liquid phase)
 - •Original VLE
 - •Original LLE
 - •Dortmund modified
 - •Lyngby modified
 - •Dortmund PSRK (used with PSRK only)
 - •Dortmund VTPR (used with VTPR only)
- •SRK (for only vapour phase or vapour+liquid phases) Soave modification of Redlich-Kwong
- •PR (for only vapour phase or vapour+liquid phases) Peng-Robinson
- •PSRK (for vapour+liquid phases using PSRK Unifac)

Predictive SRK

•VTPR (for vapour+liquid phases using VTPR Unifac) Volume-translated PR





FKS - Fluent-KilnSimu

Karri Penttilä, VTT Process Chemistry

Eero Immonen, ProcessFlow



VTT Process Chemistry

- •KilnSimu contains 1D geometry for the kiln and it calculates 1D-axial temperatures and composition for the bed and the gas sides and temperatures for the kiln inner and outer wall surfaces.
- •Fluent contains 3D geometry for kiln where the kiln inner wall and the bed surface are the boundaries. Fluent calculates the gas side (burner, combustion and gas flow).
- •After the first **KilnSimu** calculation **Fluent** calculates the gas side by using the temperatures for the bed and the inner wall calculated by **KilnSimu**. Also any volatiles formed in the bed (surface) are passed to **Fluent**.
- •Data is passed from **KilnSimu** to **Fluent** by using text files. Data in the files is processed automatically by macros in **Fluent** and mapped from 1D nodes (Kilnsimu) to 3D nodes on the boundary surfaces and then saved into **Fluent** user defined memory.
- •Fluent can use also parallel processing.



Fluent-Kilnsimu Coupling

VTT Process Chemistry



•In *two way coupling* **KilnSimu** uses average gas temperatures calculated by **Fluent** (as **Fluent** is able to calculate flame/gas more accurate) and then **KilnSimu** calculates the bed and the wall temperatures again (without changing the gas temperatures). And after this second **KilnSimu** calculation the gas side can be calculated again in **Fluent**.

Fluent-KilnSimu – Example results

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•3D gas flow temperatures calculated by Fluent
•Kiln inner wall and bed temperatures and formation of volatiles from bed calculated by KilnSimu (as boundary conditions in Fluent).





Bed temperatures calculated by KilnSimu and used as boundary condition in Fluent



Fluent-KilnSimu – Two way coupling in Lime kiln

VTT Process Chemistry

Bed Temperature profiles – Two Way Coupling



•*In Two way coupling* there might be differences in calculated temperatures (between Fluent and KilnSimu) due to using different model parameters like emissivities – so one must be careful.

A metallurgical process simulation tool based on SimuSage

- SimuSage is a flowsheeting tool, used primarily to model processes as networks of interconnected unit operations and streams.
- A number of processes would benefit from a different approach: processes with a single main reactor space submitted to a user-defined, flexible sequence of treatment steps (recipe).



A metallurgical process simulation tool based on SimuSage

- A program was developed which
 - consists of a simple flowsheet (1 reactor plus number of input streams and recycle streams) that is dynamically modified at runtime
 - implements a number of basic treatment steps together with relevant parameters as subroutines/methods
 - provides a GUI that allows the user to flexibly arrange these treatment steps in an arbitrary sequence and set their associated parameters
 - enables the user to store these "recipes" on disk; load, modify, and run them, and view and store results for post-processing



Thank you for your attention!

