

A Fully Consistent Approach for an Associate Solution Model

Dmitry Saulov
GTT Technologies

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Outline

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Part 1

- Theoretical Limits for Configurational Entropy and Entropy Paradox
- Model Description
- Dilute Solutions
- Comparison with Similar Models

Part 2

- Geometric Models
- Power Series Models
- Probabilistic Interpretation



Part 1

Resolving Entropy Paradox



Theoretical Limits for Configurational Entropy

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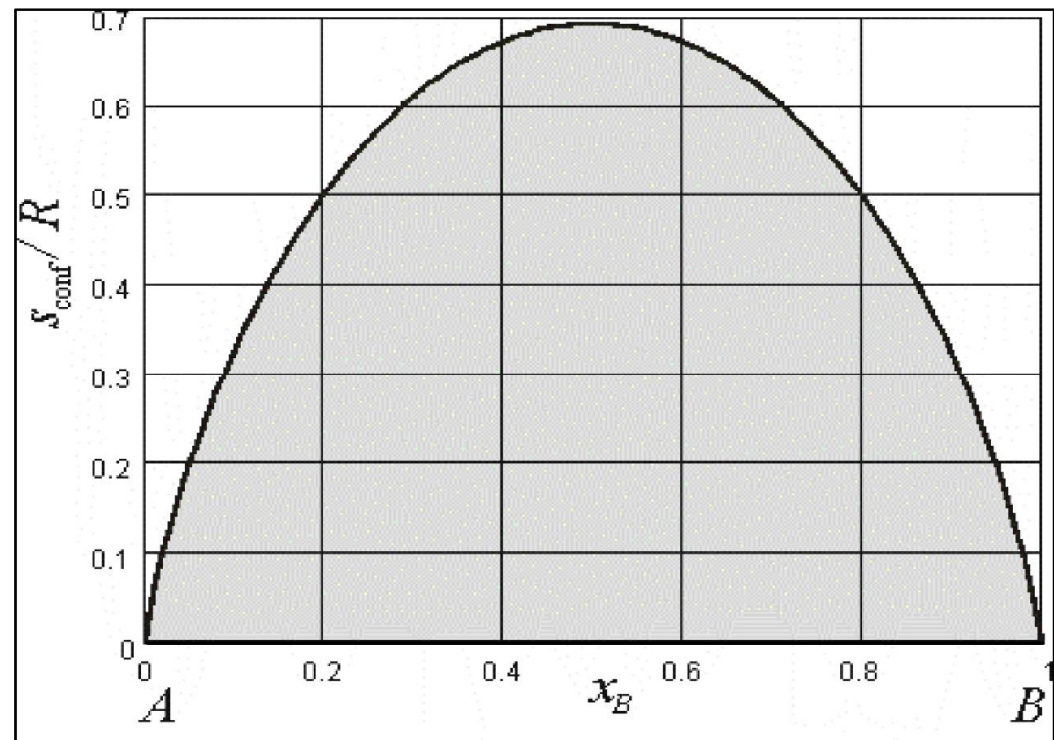
Entropy is a measure of disorder

Ideal solution - maximum disorder

For any solution phase: $S \leq S_{\text{ideal}}$

At the same time

$$S \geq 0$$



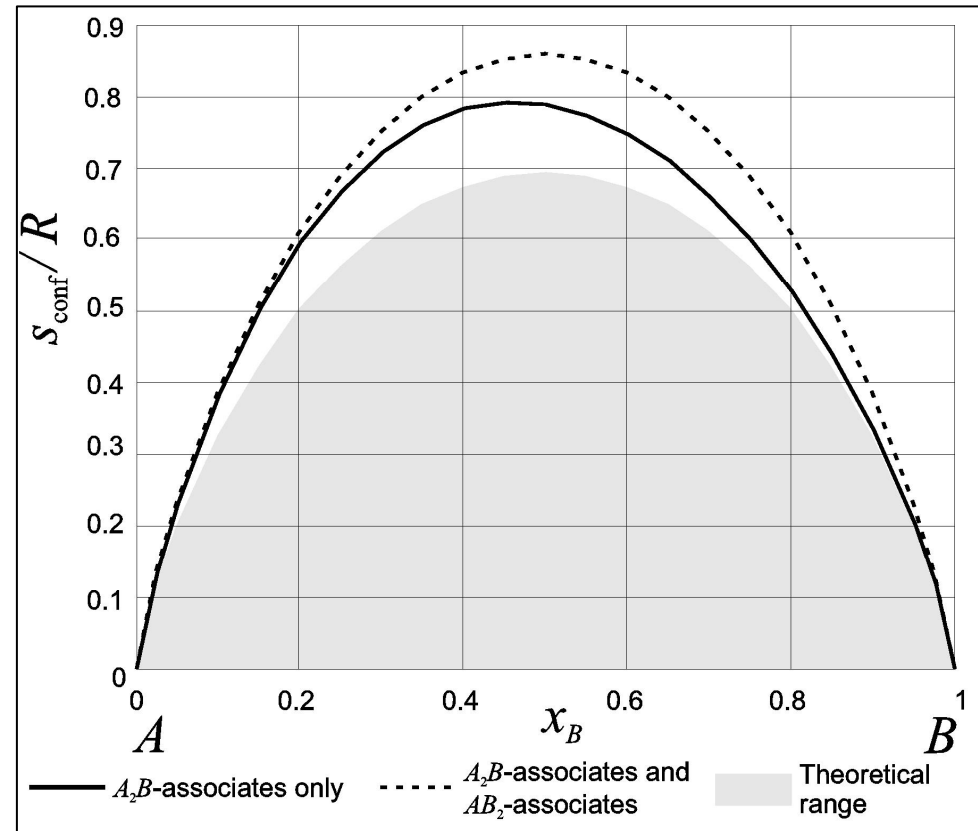
Entropy Paradox

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Luck et al. 1989
High temperature limit

Pelton et al. 2000

$$\Delta g = 0$$



Model Assumptions

- Interactions between mixing particles result in the formation of associates which are at equilibrium with each other.
- Associates are uniformly distributed (ideally mixed) over a lattice.
- All pure solution components and the chemical solution of these components are treated in a **unified way** (associates of the same size).
- Particles of the same type are indistinguishable, while particles of different types and particle sites within an associate are distinguishable.



Model Description

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The system A-B with 3-particle associates

$\begin{array}{c} 1 \ 2 \ 3 \\ [A \ A \ A] \\ [A \ A \ B] \\ [A \ B \ A] \\ [B \ A \ A] \\ [A \ B \ B] \\ [B \ A \ B] \\ [B \ B \ A] \\ [B \ B \ B] \end{array}$

$$\Omega = \frac{\left(N_0 \sum_{i,j,k=A,B} n_{[ijk]} \right)!}{\prod_{i,j,k=A,B} (N_0 n_{[ijk]})!}$$

$$S_{\text{conf}} = k_B \ln(\Omega)$$

$$\frac{2}{3}[AAA] + \frac{1}{3}[BBB] \leftrightarrow [AAB]$$

$$\Delta g_{[AAB]} = g_{[AAB]} - \frac{2}{3} g_{[AAA]} - \frac{1}{3} g_{[BBB]}$$

$$G = \sum_{i,j,k=A,B} n_{[ijk]} \left(g_{[ijk]} + RT \ln(x_{[ijk]}) \right)$$

$$\left(\frac{\partial G}{\partial n_{[ijk]}} \right)_{n_A, n_B, n_{[i'j'k']}} = 0$$



Model Description

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The system A-B with 3-particle associates

$$g_{[AAB]} = g_{[ABA]} = g_{[BAA]} = g_{A_2B}; \quad g_{[BBA]} = g_{[BAB]} = g_{[ABB]} = g_{B_2A}$$

[A A A] - A_3 (pure A)

[A A B]
[A B A]
[B A A]

} A_2B

[A B B]
[B A B]
[B B A]

} AB_2

[B B B] - B_3 (pure B)

$$G = n_{A_3} g_{A_3} + n_{B_3} g_{B_3} + n_{A_2B} g_{A_2B} + n_{AB_2} g_{AB_2} \\ + RT \left(n_{A_3} \ln(x_{A_3}) + n_{B_3} \ln(x_{B_3}) \right) \\ + RT \left(n_{A_2B} \ln\left(\frac{x_{A_2B}}{3}\right) + n_{AB_2} \ln\left(\frac{x_{AB_2}}{3}\right) \right)$$

$$\left(\frac{\partial G}{\partial n_{A_2B}} \right)_{n_A, n_B, n_{AB_2}} = \left(\frac{\partial G}{\partial n_{AB_2}} \right)_{n_A, n_B, n_{A_2B}} = 0$$



Model Description

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The system A-B with 3-particle associates

$$\Delta g_{A_2B} = \Delta g_{B_2A} = 0$$

$$x_{A_3} = x_A^3$$

$$x_{A_2B} = 3x_A^2x_B$$

$$x_{AB_2} = 3x_Ax_B^2$$

$$x_{B_3} = x_B^3$$

***G* expression correctly
reduces to the ideal solution**



Dilute Solutions

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Solution of monparticles A and B and associates A_2B

with $\Delta g_{A_2B} \rightarrow -\infty$

For small x_A $a_B \approx 1 - \frac{x_A}{2}$

Rault's law $a_B \approx 1 - x_A$

$\Delta g_{A_2B} = -\infty$

For any finite Δg_{A_2B}

Rigorously proven for CAM and MAF



Models Comparison

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Models

- CAM
- ASM
- MQM 1986
- MQM 2000
- MAF

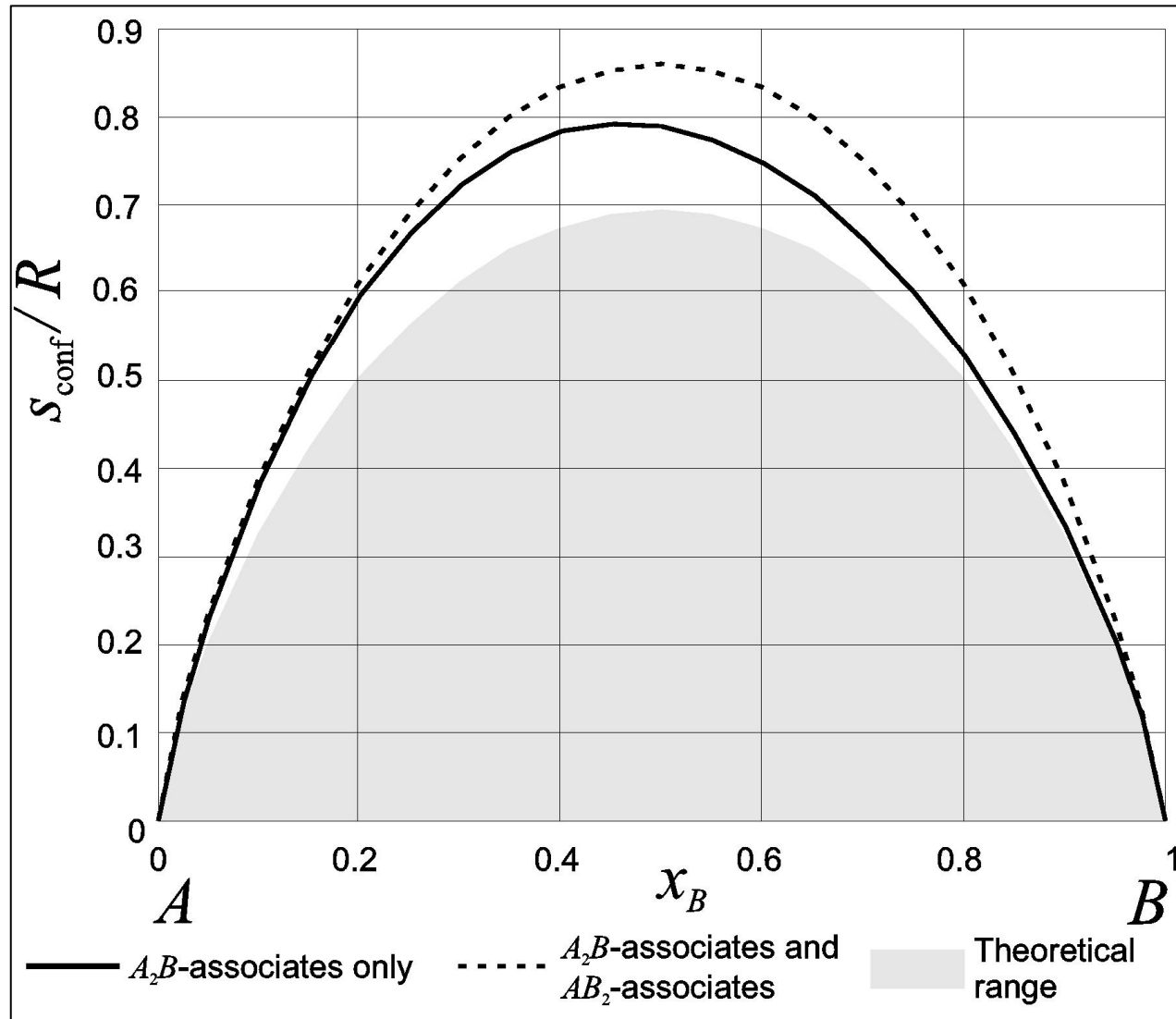
Conf. Entropy

- Ideal solution
- Highly ordered solution
- Immiscible components



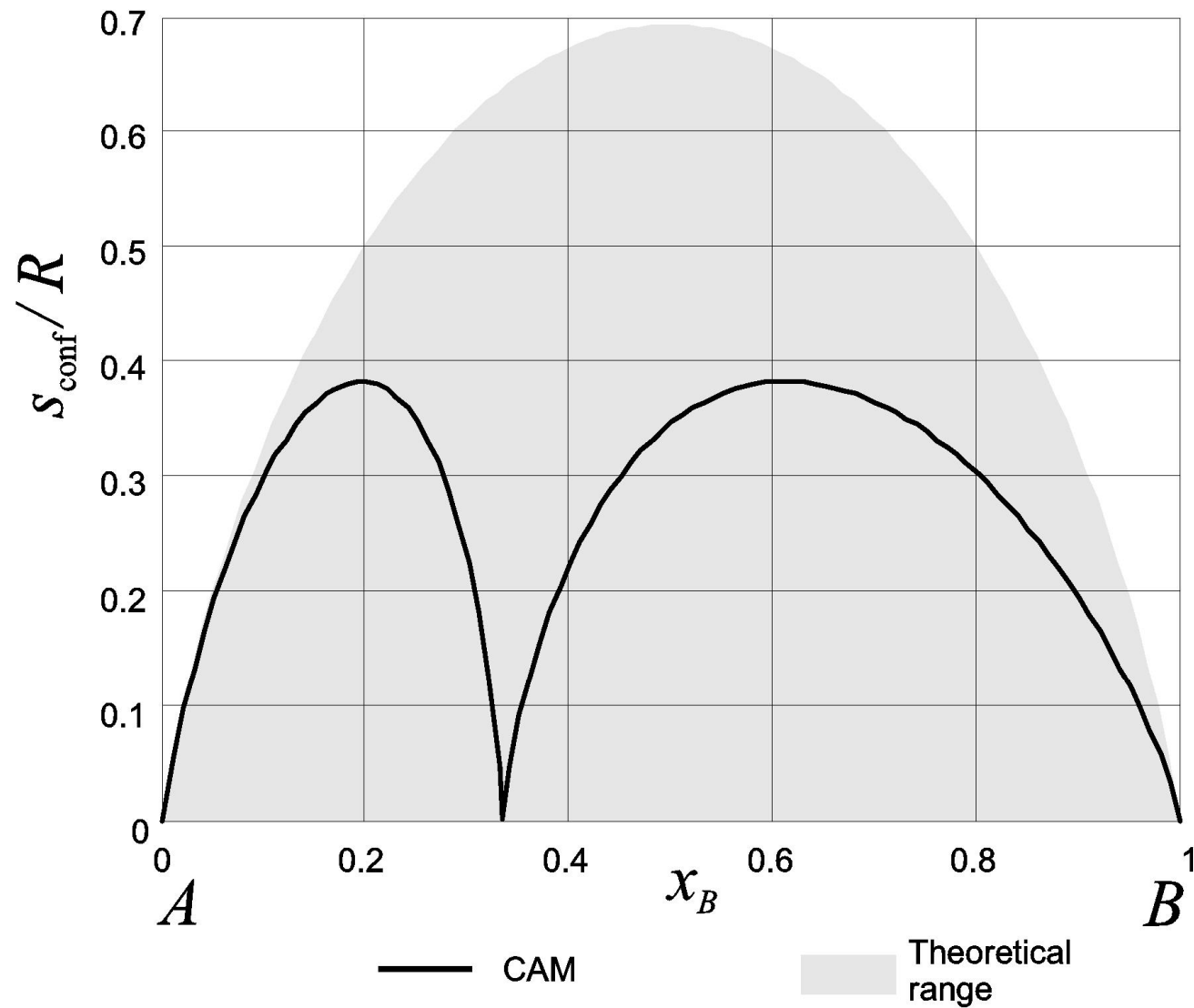
CAM (Ideal)

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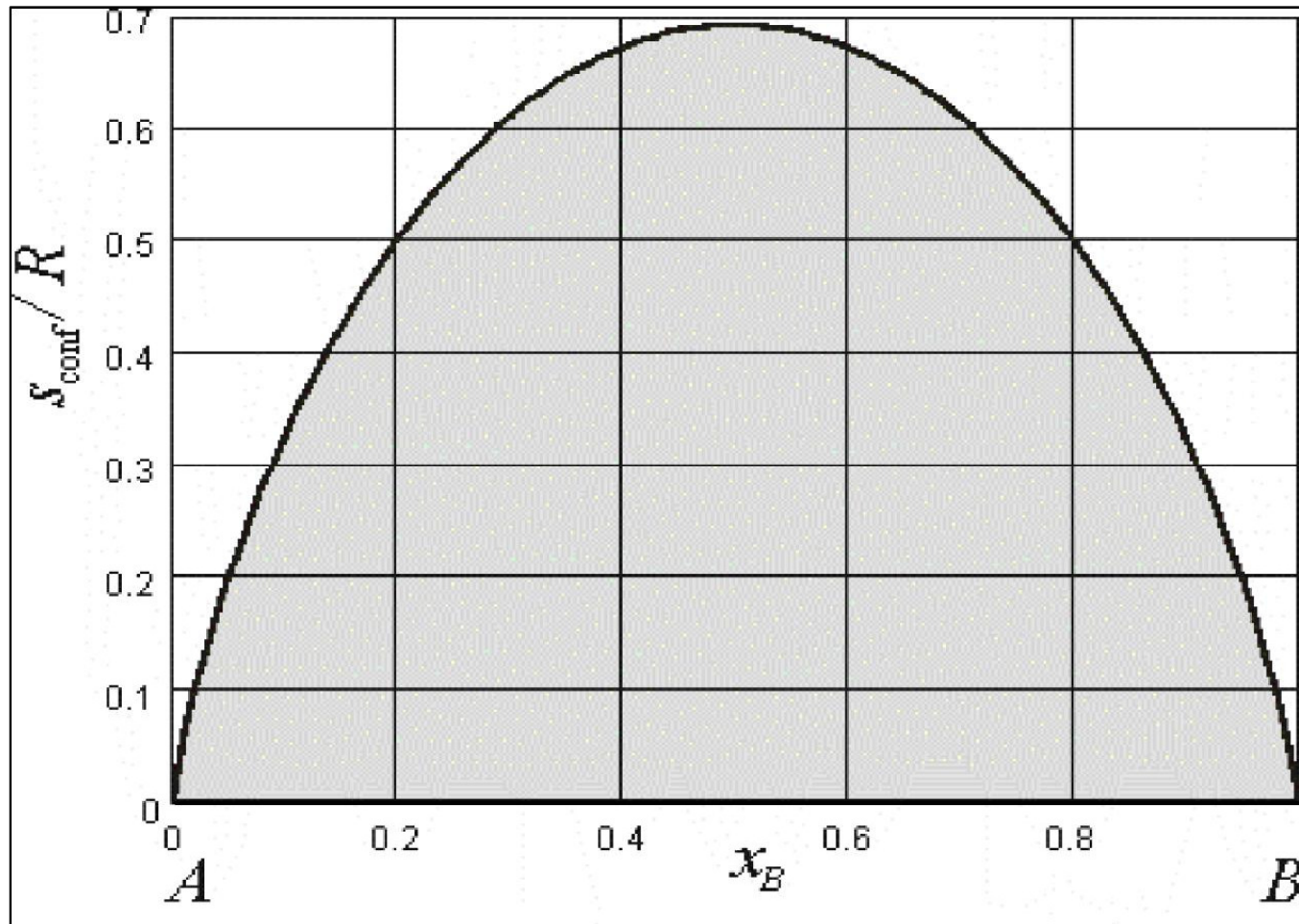
CAM (Ordered)

GTT-Technologies



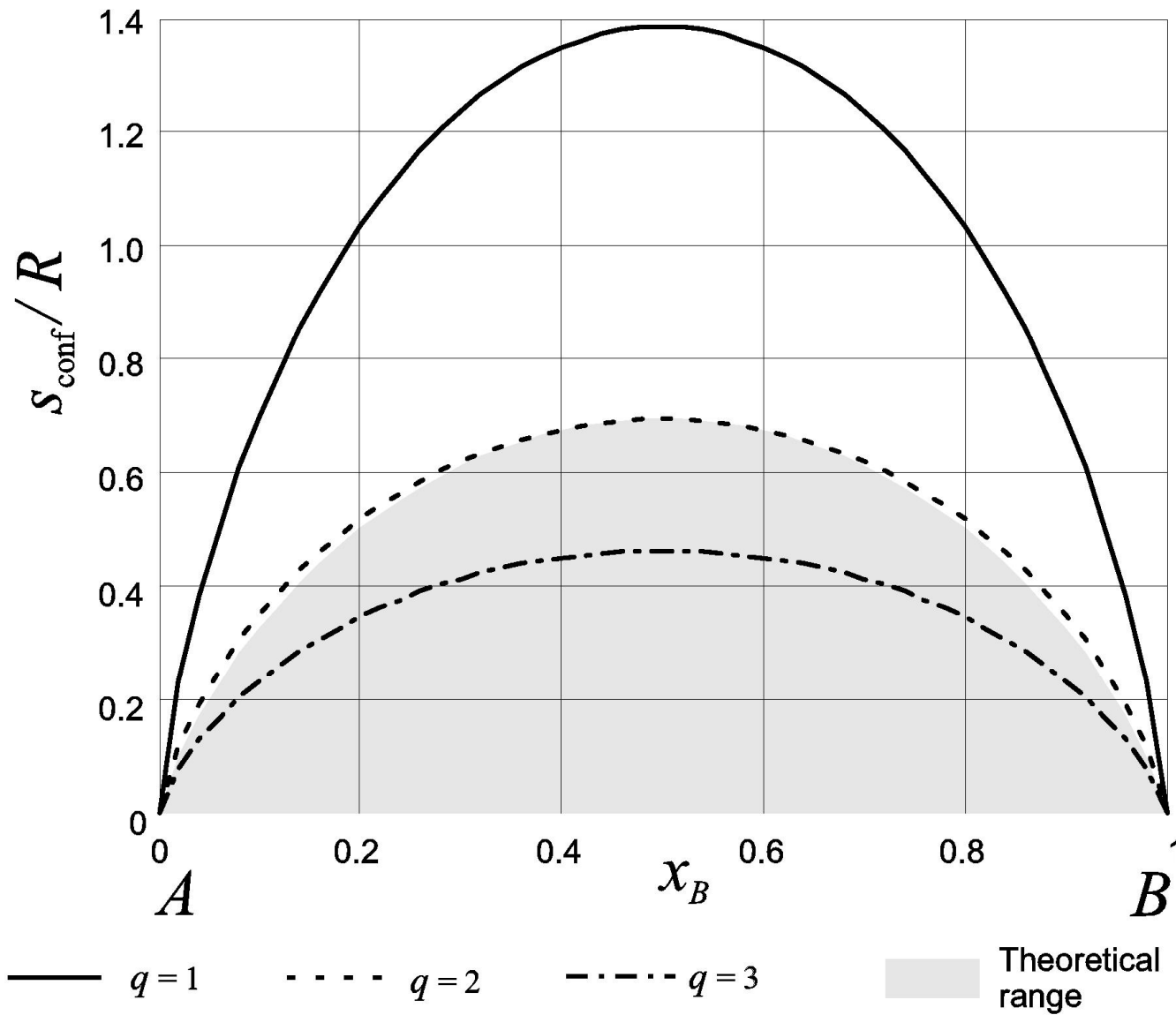
CAM (Immiscible)

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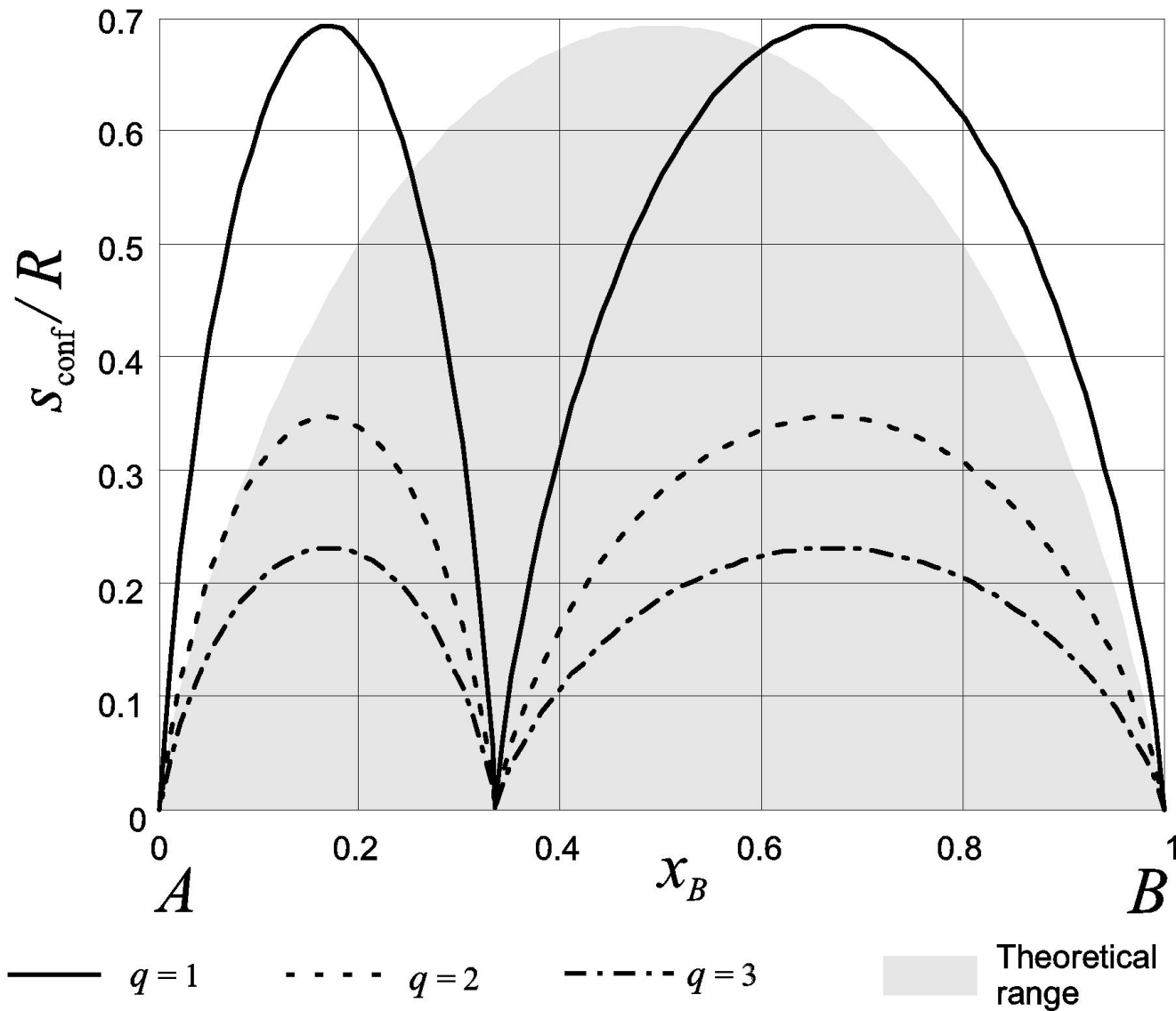
ASM (Ideal)

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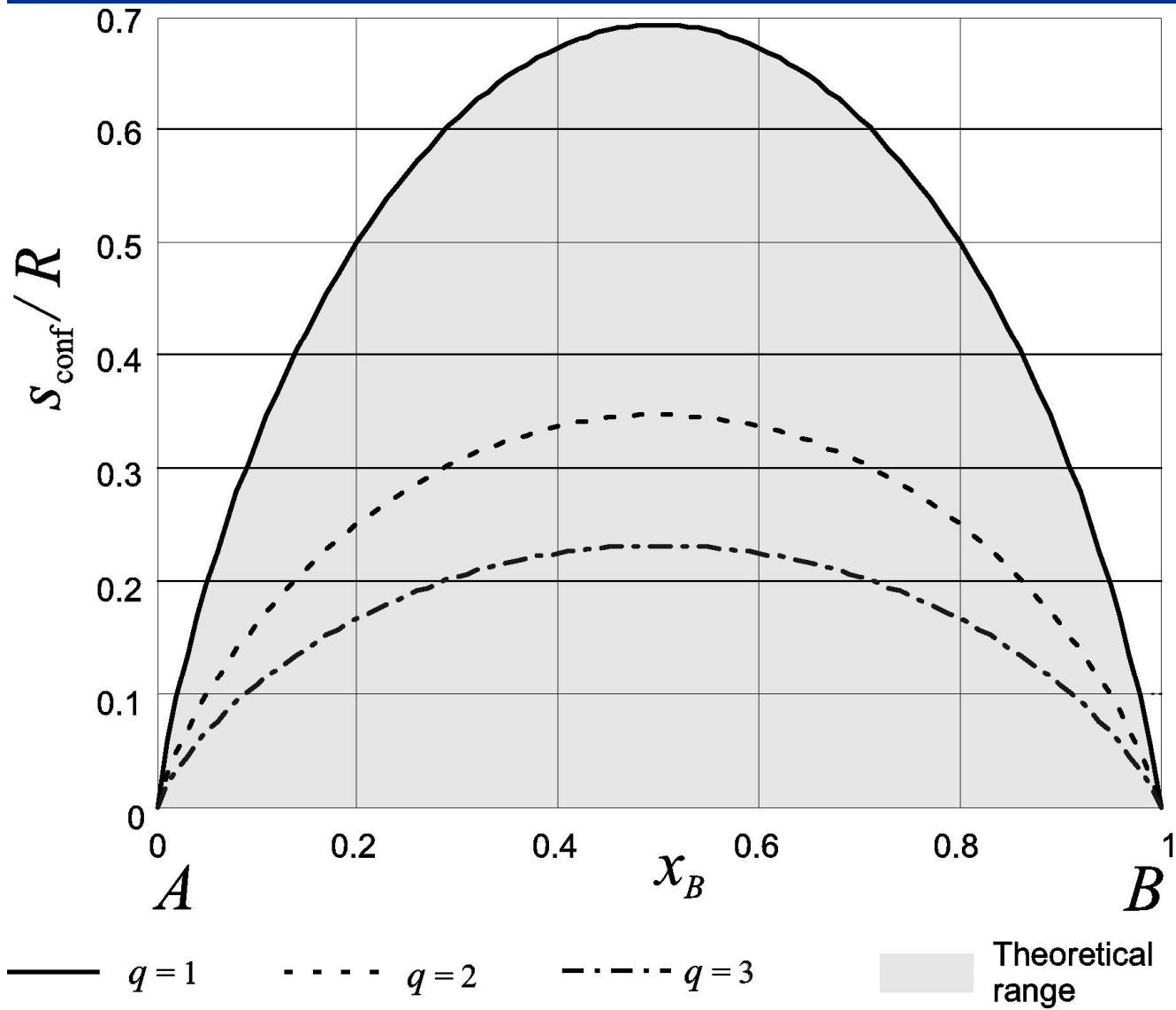
ASM (Ordered)

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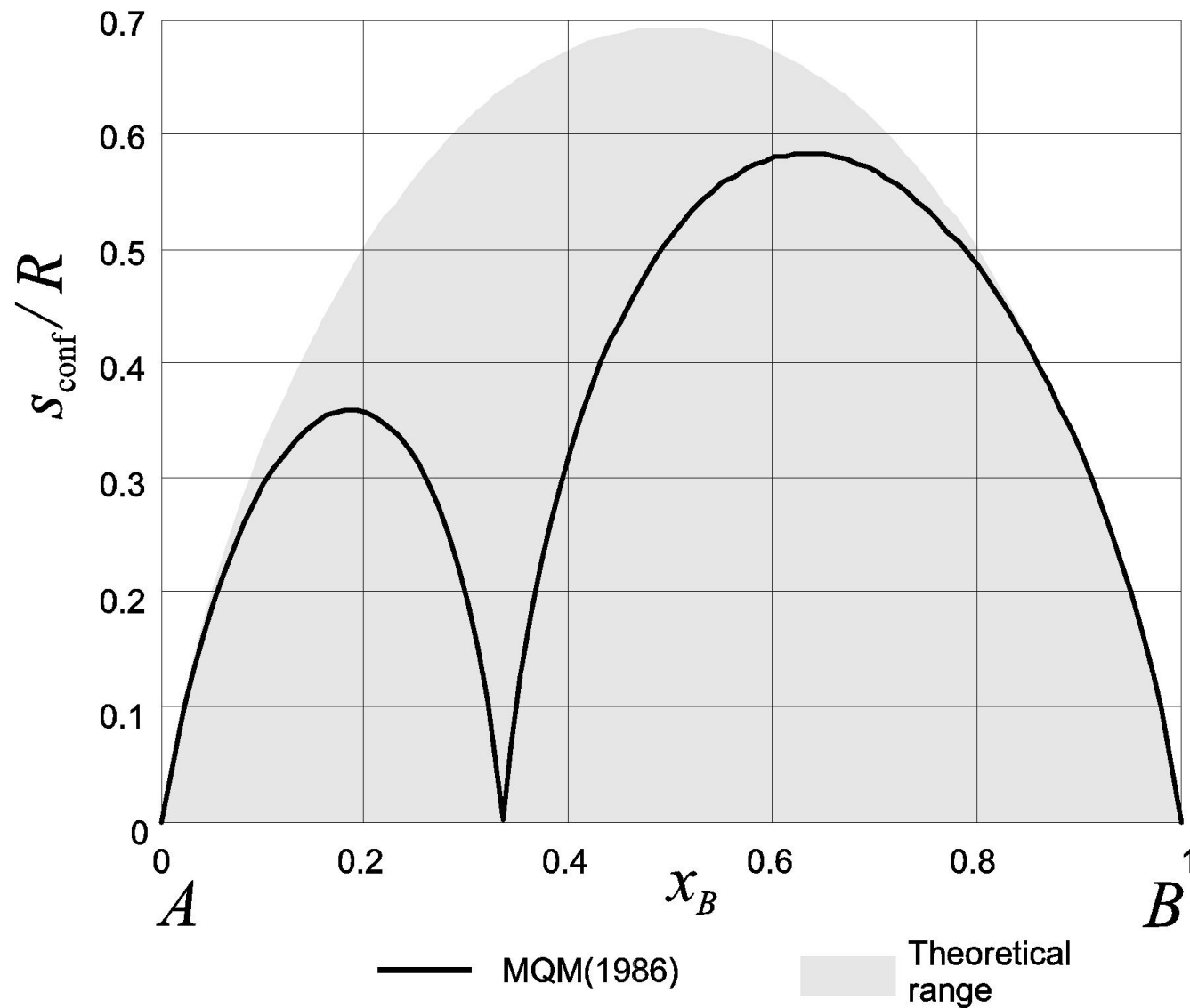
ASM (Immiscible)

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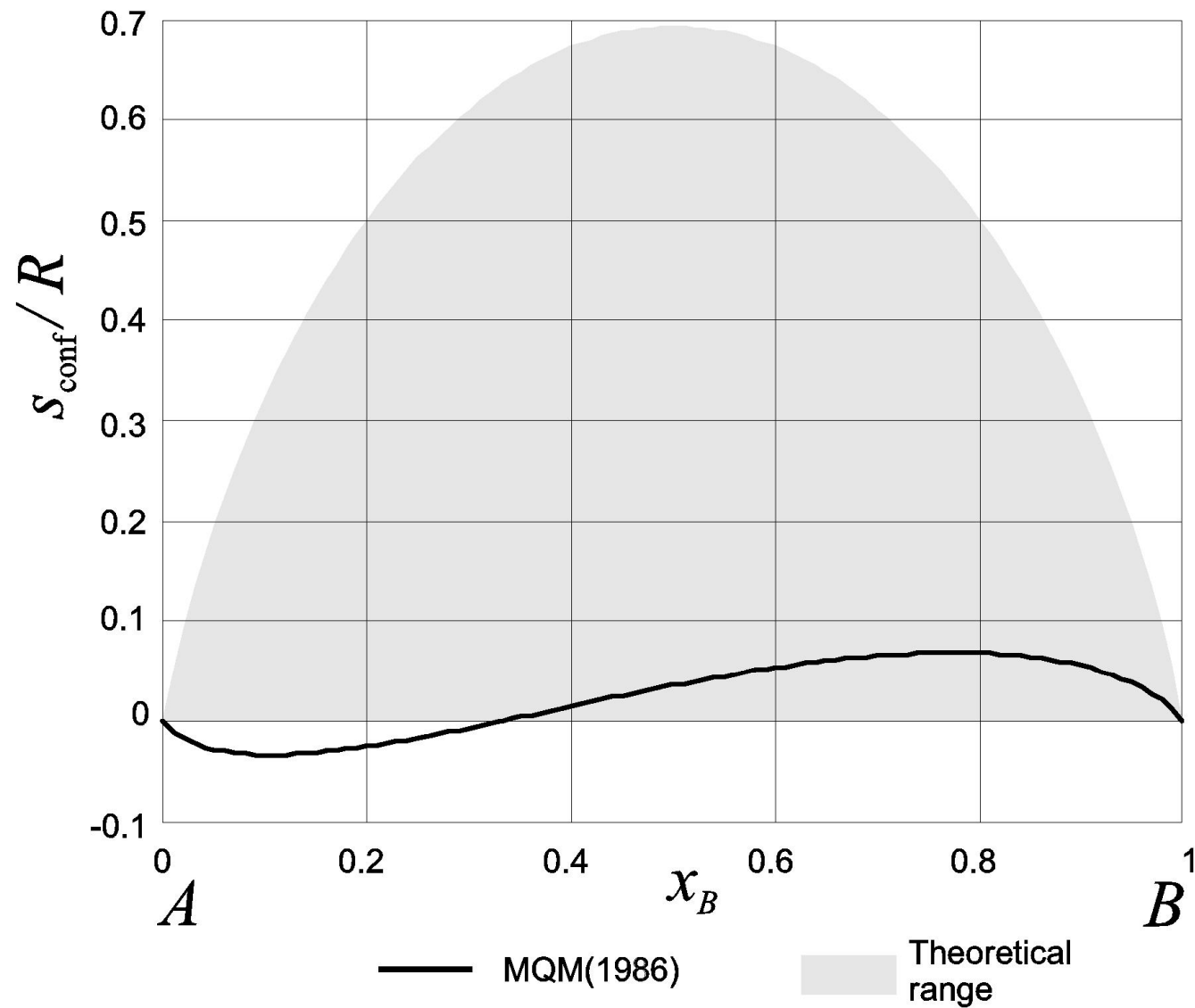
MQM 1986 (Ordered)

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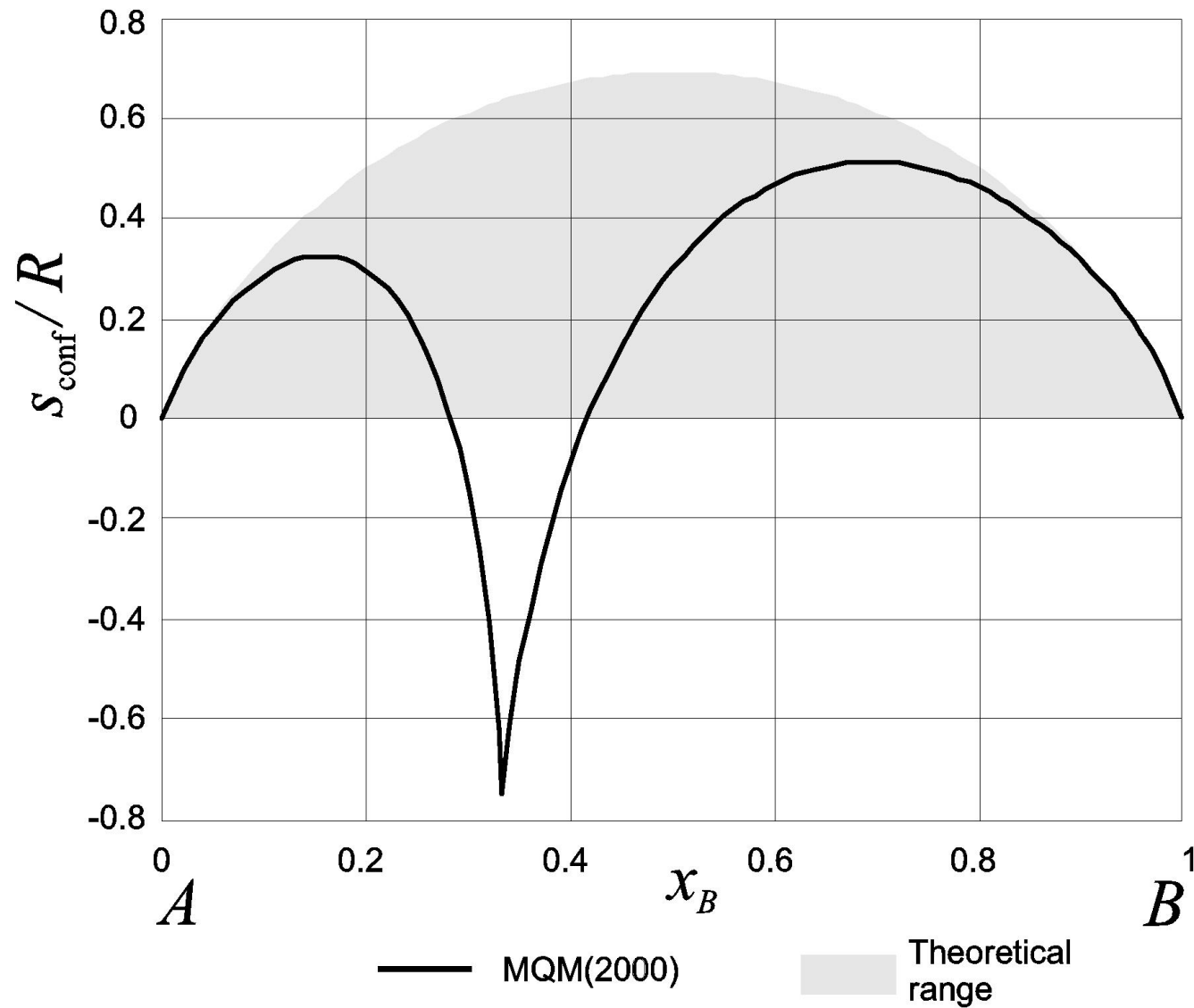
MQM 1986 (Immiscible)

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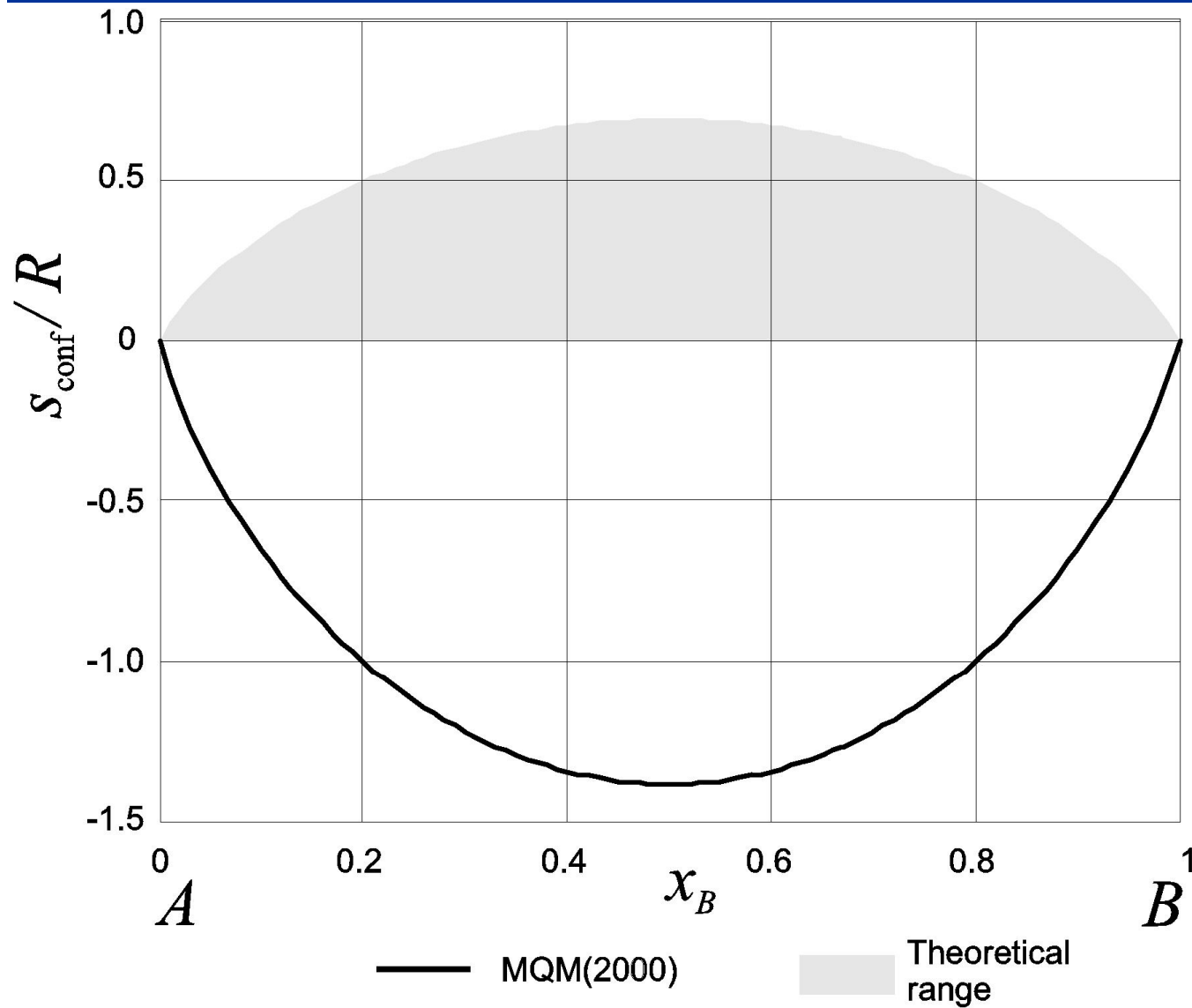
MQM 2000 (Ordered)

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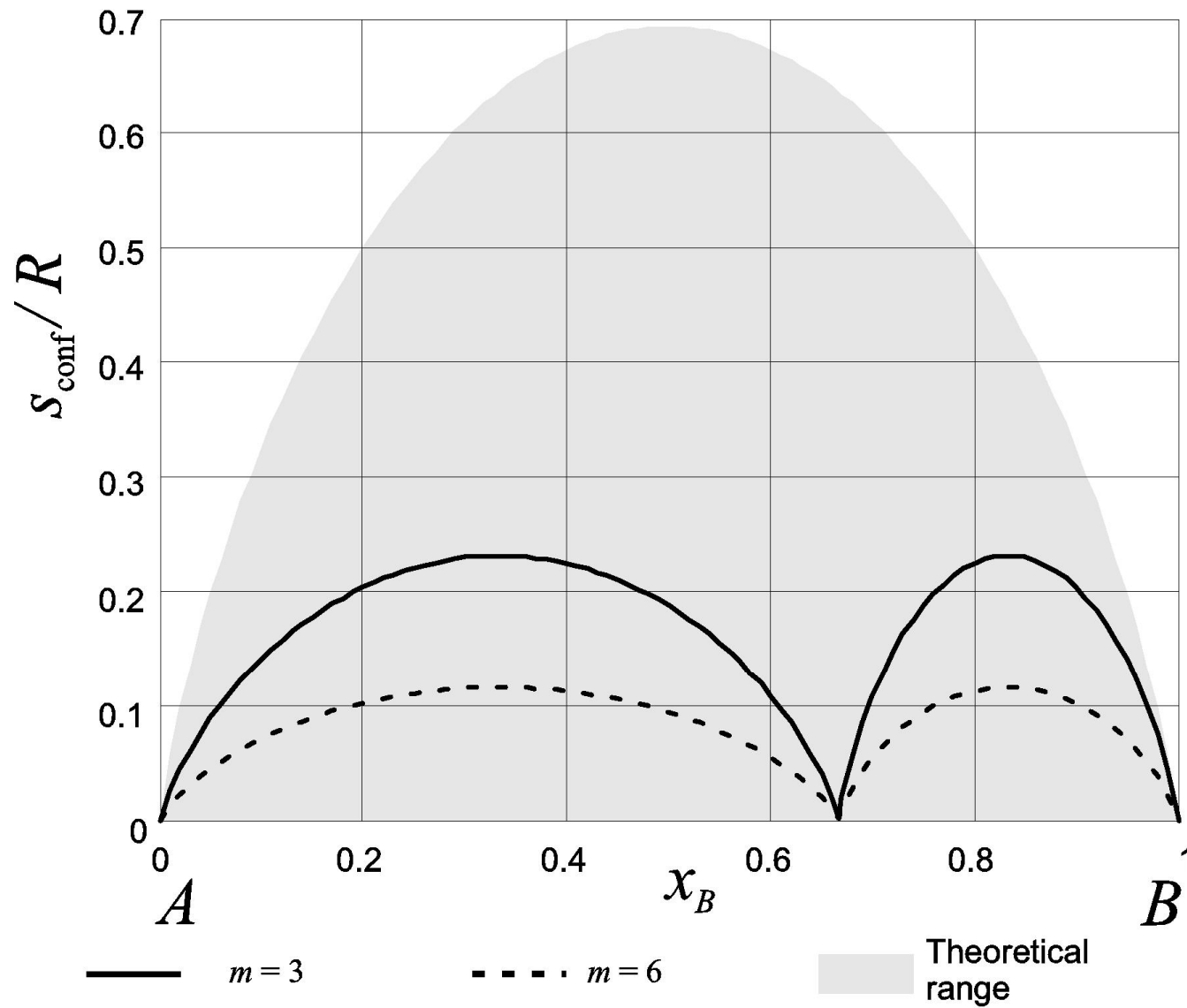
MQM 2000 (Immiscible)

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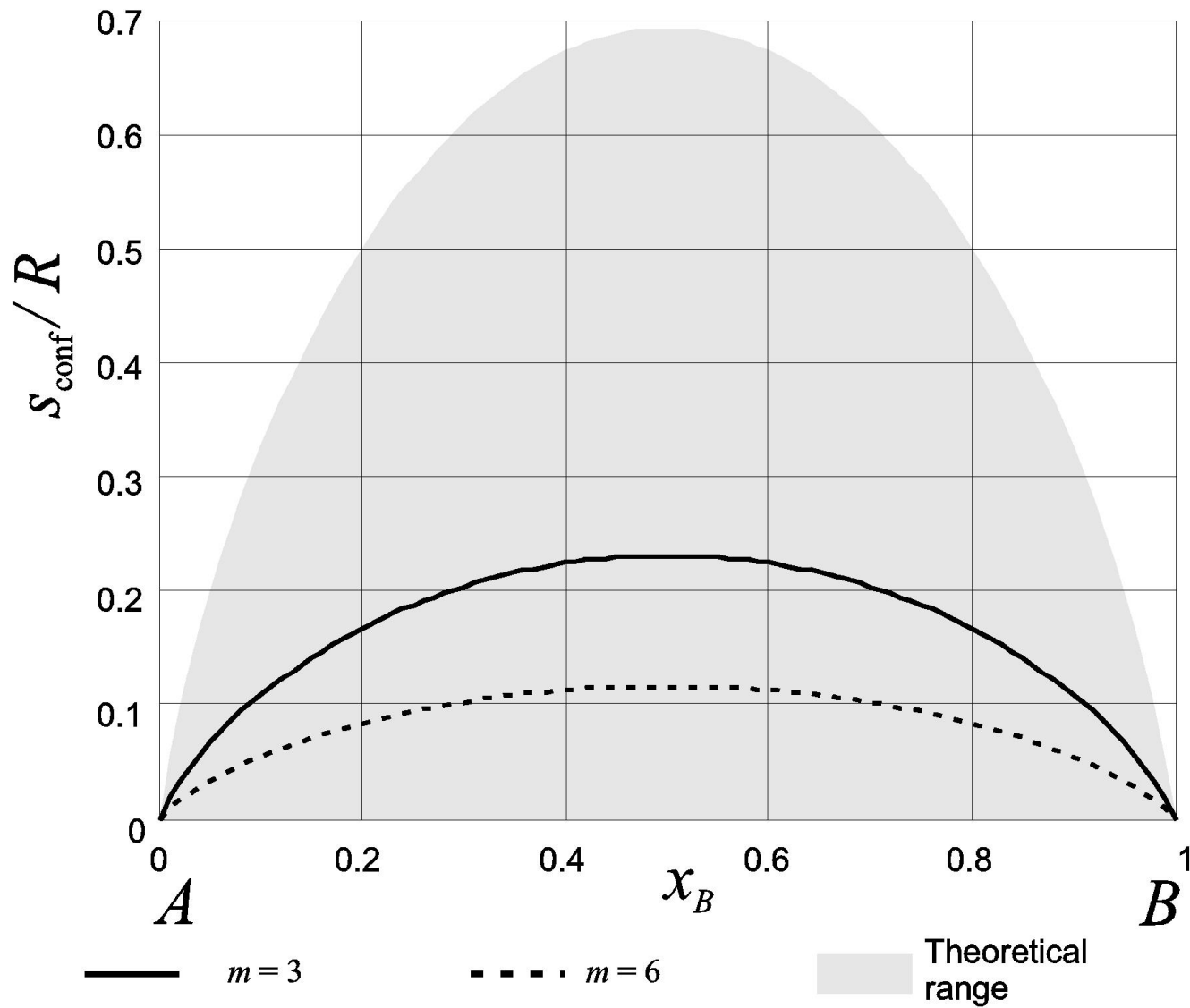
MAF (Ordered)

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MAF (Immiscible)

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Classical QM

Theoretically sound

- Ideal Solution ✓
- Ordered solution ✓
- Immiscible components ($S=0$) ✓

However

Limited applicability



Other Properties

	CAM	ASM	MAF	MQM1	MQM2
Free from entropy paradox	-	-	+	-	-
Raoult's law	+	-+	+	+!	+!
Selection CMO (binary systems)	+	+	+	+ -	+
Selection CMO (multicomponent systems)	+	+	+	-	-
Clear physical meaning of the adjustable parameters	+	+	+	-	-



Correct G-expression MQM 2000

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$$\frac{1}{Z_A} = \frac{1}{Z_{AA}^A} \frac{2n_{AA}}{2n_{AA} + n_{AB}} + \frac{1}{Z_{AB}^A} \frac{n_{AB}}{2n_{AA} + n_{AB}}$$
$$\frac{1}{Z_B} = \frac{1}{Z_{BB}^B} \frac{2n_{BB}}{2n_{BB} + n_{AB}} + \frac{1}{Z_{BA}^B} \frac{n_{AB}}{2n_{AA} + n_{AB}}$$

$$G = n_A g_A^o + n_B g_B^o - T\Delta S^{\text{conf}} + \frac{n_{AB}}{2} \Delta g_{AB}$$

In error

Correct expression

$$G = \frac{Z_A}{Z_{AA}^A} n_A g_A^o + \frac{Z_B}{Z_{BB}^B} n_B g_B^o - T\Delta S^{\text{conf}} + \frac{n_{AB}}{2} \Delta g_{AB}$$



Part 2

Treatment of polynomial excess functions



Model Used

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Cheng and Ganguly:

“a major problem in solution thermodynamics”

Power Series

- Margules
- Wohl
- Helffrich and Wood
- others

Empirical

Geometric

$$\alpha_{12}x_1x_2$$

- Kohler
- Muggianu (Hillert)
- Kohler / Toop
- Muggianu / Toop

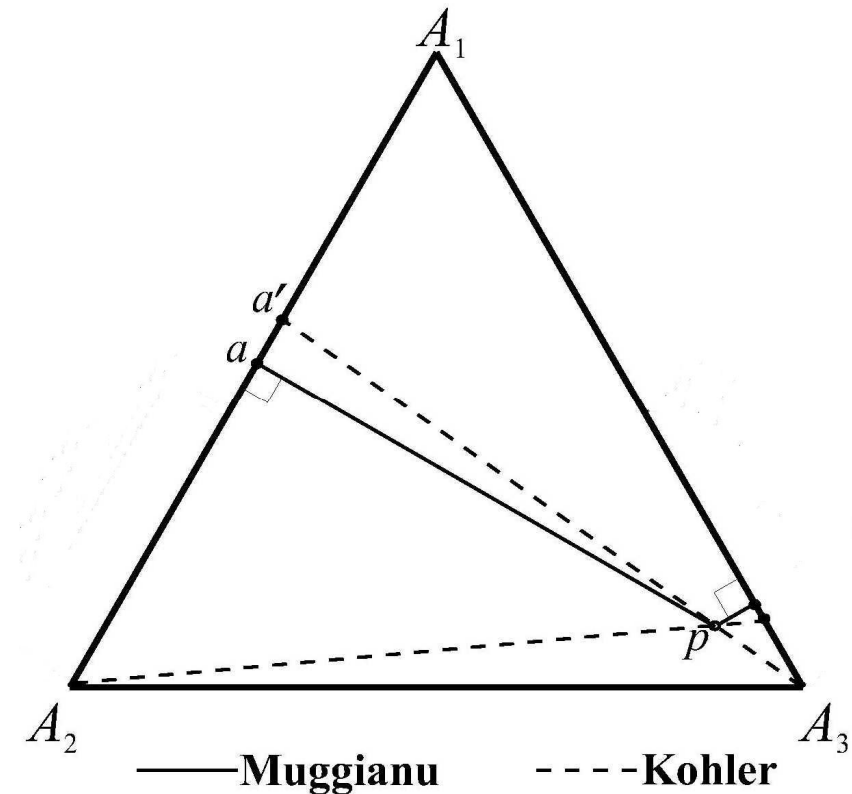
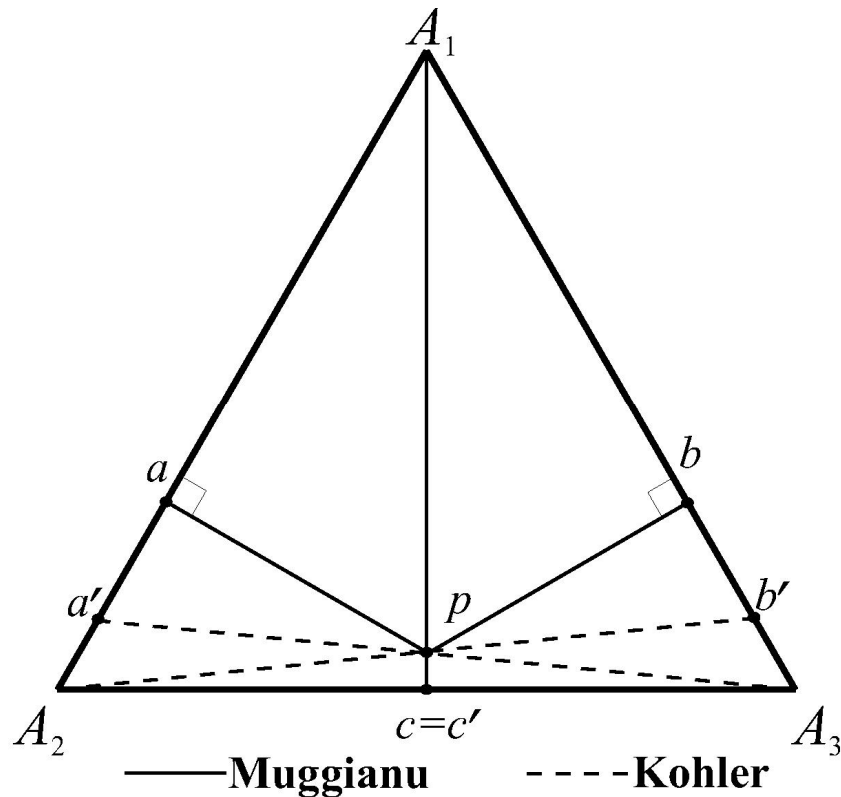
Inline with Calphad approach



Kohler vs Muggianu

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Dilute solution (Chartrand and Pelton)

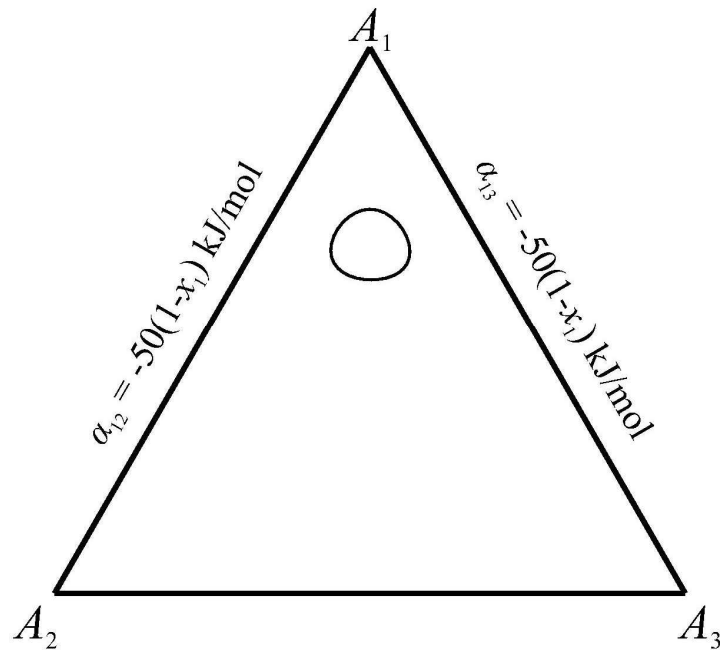


**Kohler: singularities at terminal composition (Brynstad)
should not be used (Howald and Row)**

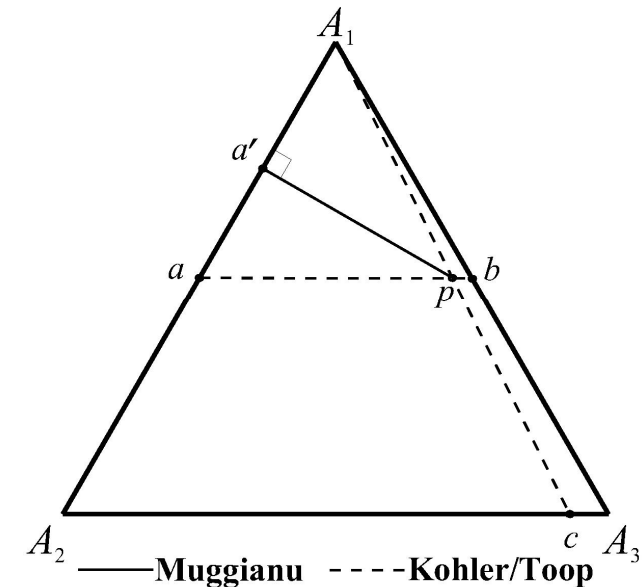


Symmetric vs Asymmetric

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Dilute solution



- No quantitative criteria
- **Chemically** dissimilar components are treated in a **thermodynamically** different way
- Muggianu with the specially selected ternary term resolves the issue



Power Series vs Geometric

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Helfrich and Wood:

- Subregular model - 3-particle interactions
- Ternary terms exist independently of the properties
- of the bounding binaries
- Ternary terms could be required even for Regular model

Power Series

integral part

Geometric

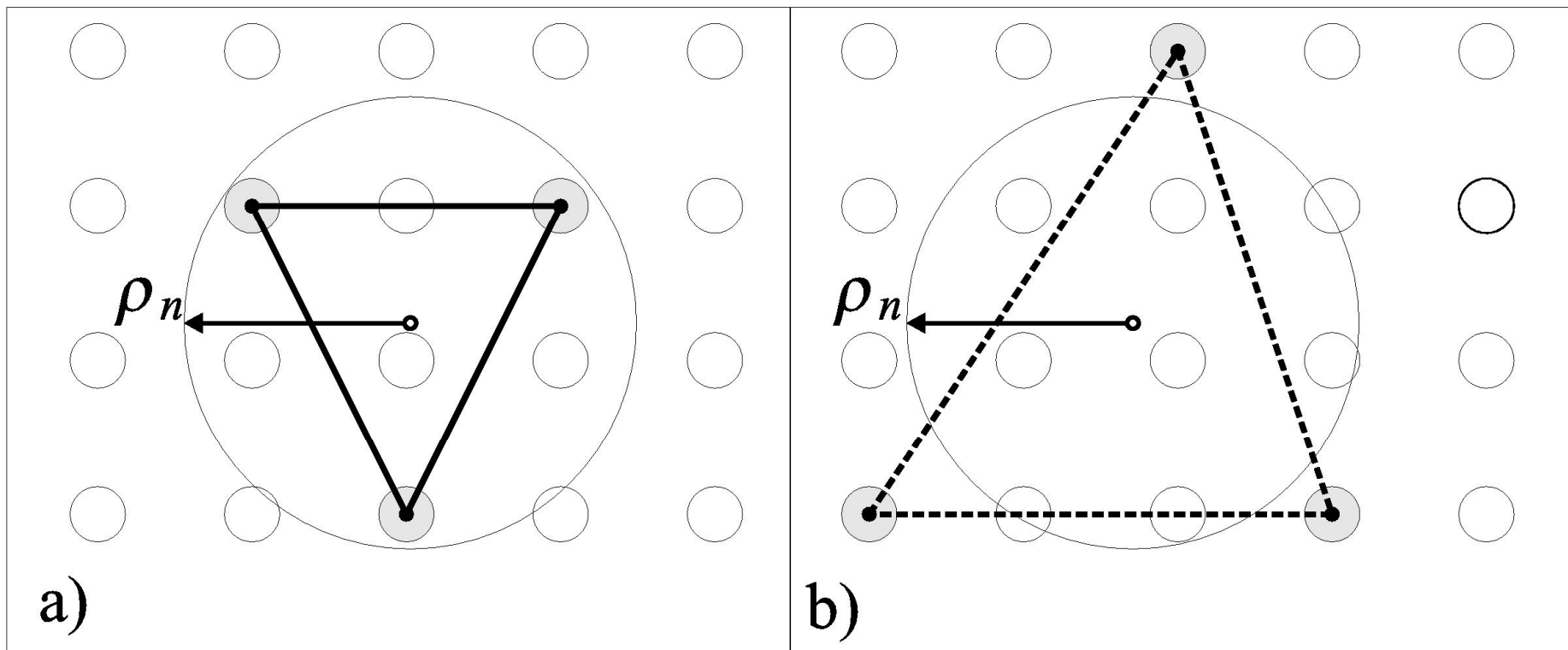
empirical



Probabilistic Interpretation

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- The Bragg-Williams approximation
- Short-range interaction ρ_n
- Uniform lattice z_n



Probabilistic Interpretation

$$g = \sum_{i=1}^r x_i (g_i + RT \ln(x_i)) + \sum_{k_1 + \dots + k_r = n} c_{n; k_1, \dots, k_r} x_1^{k_1}, \dots, x_r^{k_r}$$

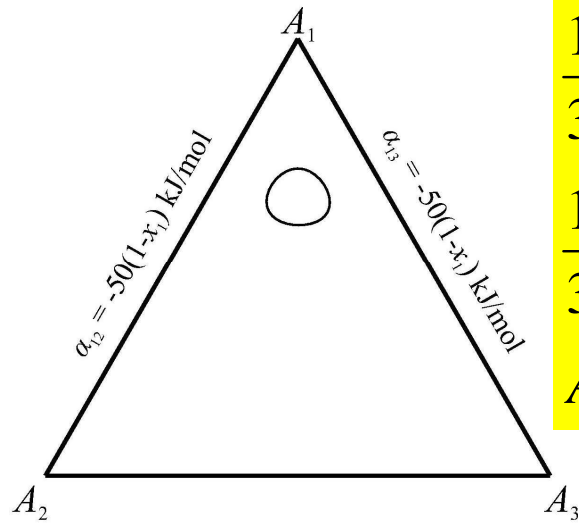
$$c_{n; k_1, \dots, k_r} = \frac{n!}{k_1! \dots k_r!} \frac{z_n N_0}{n} \Delta g_{n; k_1, \dots, k_r}$$

$$\frac{k_1}{n} A_1 \dots A_1 + \dots + \frac{k_r}{n} A_r \dots A_r \leftrightarrow A_1 \dots A_1 \dots A_r \dots A_r$$



Probabilistic Interpretation

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$$\frac{1}{3} A_1 A_1 A_1 + \frac{2}{3} A_2 A_2 A_2 \leftrightarrow A_1 A_2 A_2; \quad \Delta g = -\frac{50}{z_3 N_0} \text{ kJ/mol}$$

$$\frac{1}{3} A_1 A_1 A_1 + \frac{2}{3} A_3 A_3 A_3 \leftrightarrow A_1 A_3 A_3; \quad \Delta g = -\frac{50}{z_3 N_0} \text{ kJ/mol}$$

$$A_2 A_2 + A_3 A_3 \leftrightarrow 2 A_2 A_3; \quad \Delta g = 0$$

$$A_1 (A_2 A_2) + A_1 (A_3 A_3) \leftrightarrow 2 A_1 (A_2 A_3); \quad \Delta g = 0$$

$$\frac{1}{3} A_1 A_1 A_1 + \frac{1}{3} A_2 A_2 A_2 + \frac{1}{3} A_3 A_3 A_3 \leftrightarrow A_1 A_2 A_3; \quad \Delta g = -\frac{50}{z_3 N_0} \text{ kJ/mol}$$

$$^{\text{Ex}} g = -50 x_1 (1 - x_1)^2 \text{ kJ}$$

