

Prediction of precipitation kinetics in recycled aluminum alloy with ClaNG

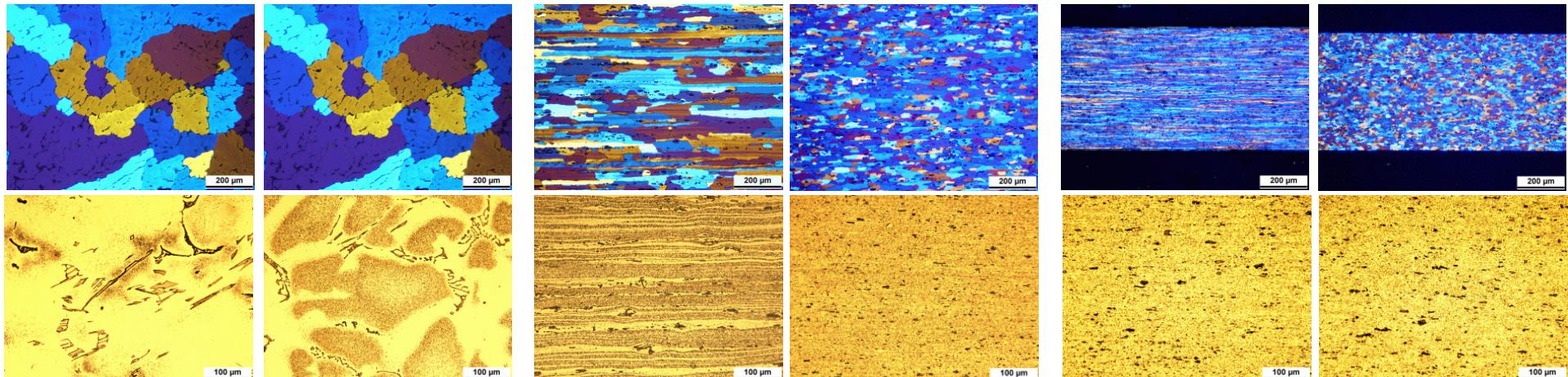
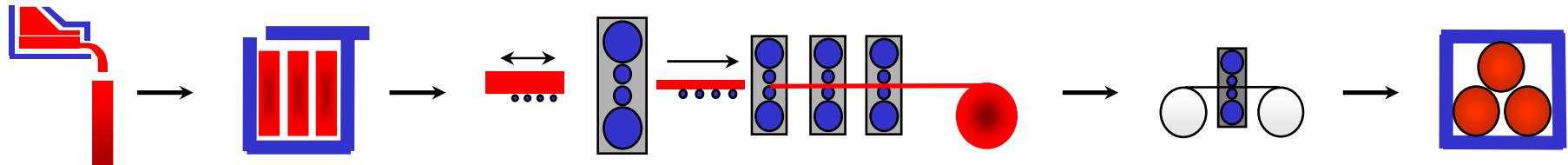
Z. Liu, V. Mohles, G. Gottstein



GTT workshop, Herzogenrath, June 03-05, 2009

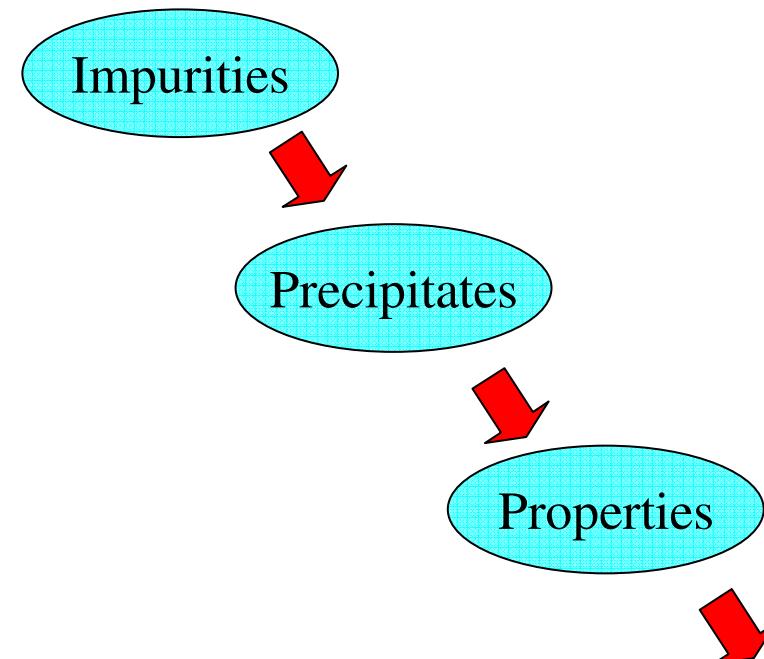
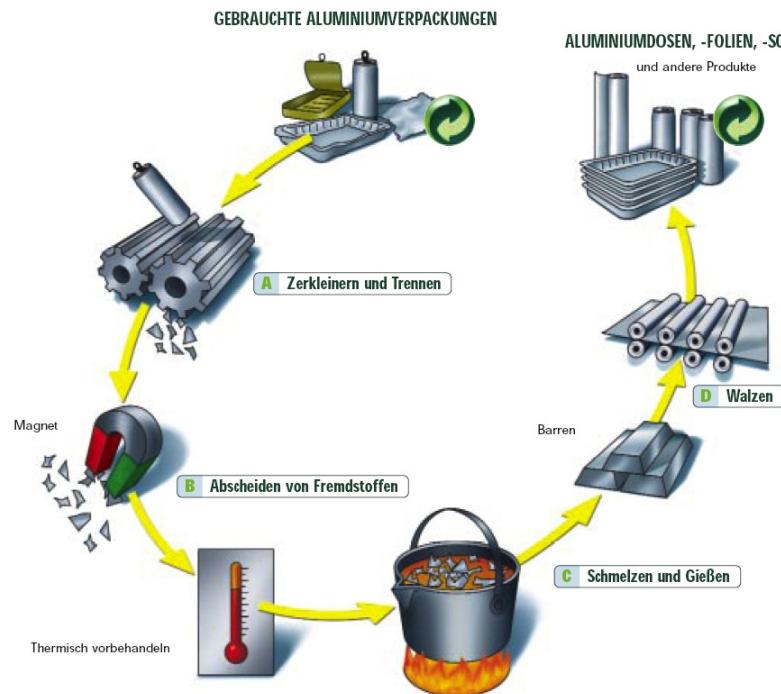
Aluminum sheet production

Casting - Homogenization - Hot Rolling - Cold Rolling - Annealing



- Mechanical and corrosion properties
- Homogenization: Grain size controlling
- Hot/Cold rolling: interaction with RX or recovery

Aluminum sheet production



Project ClaNG⁺

- Goal: determine the precipitation kinetics in Al alloys

→ ClaNG model

- ✓ Collaboration between

- Hydro (Aluminum industry)
- GTT technologies
- IMM: Institut für Metallkunde und Metallphysik



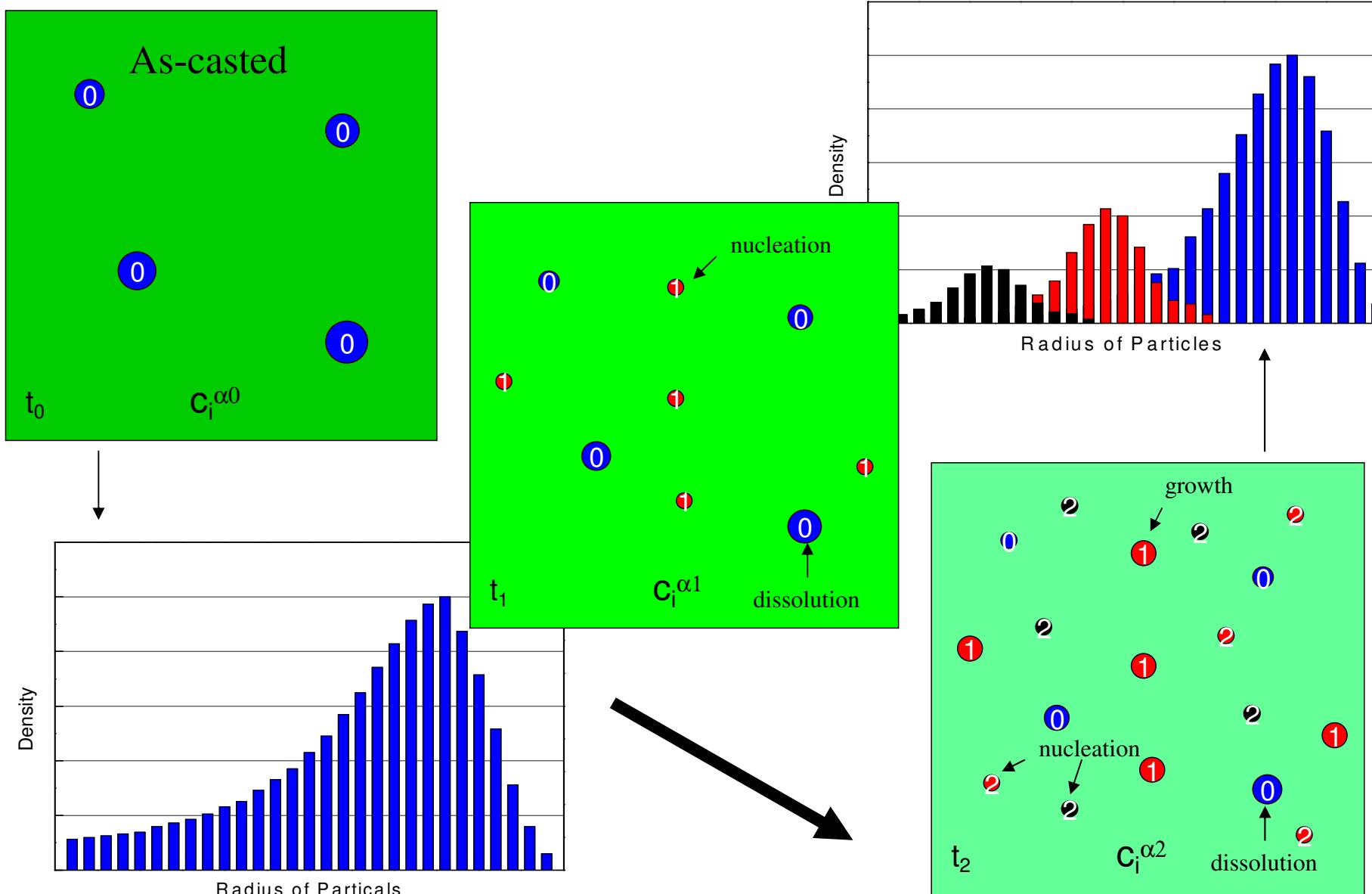
- ♀ Developed by *M. Schneider*:

*„Modellierung und Validierung zeitabhängiger mikrochemischer Prozesse
in Aluminium Knetlegierungen“ RWTH Aachen 2006*

- ♀ Modified by *E. Jannet*

- ♀

ClaNG model overview



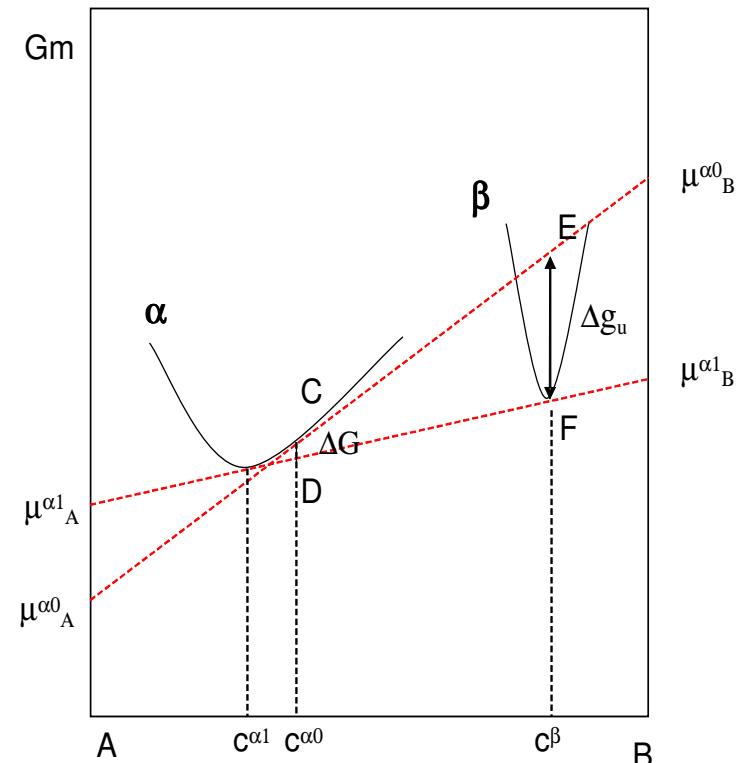
Thermodynamic calculations

□ Use ChemApp (GTT):

- ✓ Set conditions (temperature, concentrations of matrix)
- ✓ Enter 2 phases: matrix + another phase
- ✓ Perform equilibrium
 - Equilibrium concentrations c_i^α in the Al matrix α
 - Equilibrium concentrations c_i^β in the phase β Gm
 - Chemical potentials of the elements

Chemical Driving Force of Nucleation

$$\Delta g_u^\beta = \sum_i c_i^\beta \cdot (\mu_i^{\alpha 0} - \mu_i^{\alpha 1})$$



Nucleation

- When Δg_u is known, the critical radius r_c can be derived
 - User input = interfacial energy γ

$$r_c = \frac{2\gamma \cdot V_m}{\Delta g_u}$$

- The nucleation rate is given by Becker & Döring theory

$$\dot{N} = N^0 \cdot Z \cdot \beta \cdot \exp\left(-\frac{\Delta G(r_c)}{k_B \cdot T}\right) \cdot \exp\left(\frac{-\tau}{t}\right)$$

HOMOGENEOUS NUCL.

$N_{hom}^0 \approx$ number of atoms per volume ($10^{29} m^{-3}$)

Volume Diffusion : D_{hom}

$$\dot{N}_{hom} = N_{hom}^0 \cdot Z_{hom} \cdot \beta_{hom} \cdot \exp\left(-\frac{\Delta G(r_c)}{k_B \cdot T}\right) \cdot \exp\left(\frac{-\tau_{hom}}{t}\right)$$

HETEROGENEOUS NUCL.

$N_{het}^0 =$ funct. (ρ)

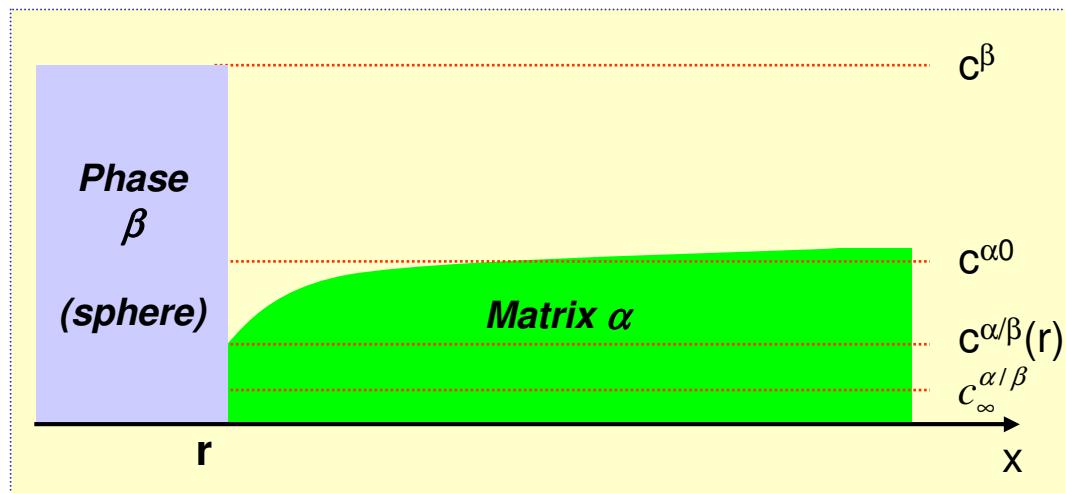
Pipe diffusion : D_{het}

$$\dot{N}_{het} = N_{het}^0 \cdot Z_{het} \cdot \beta_{het} \cdot \exp\left(-f_{het}^{corr} \cdot \frac{\Delta G(r_c)}{k_B \cdot T}\right) \cdot \exp\left(\frac{-\tau_{het}}{t}\right)$$

$$\boxed{\dot{N}_{tot} = \dot{N}_{hom} + \dot{N}_{het}}$$

Growth & Coarsening

- Assumption: precipitate growth is controlled by the diffusion in Al alloys. The slowest diffuser drives the dance(among the elements contained in the phase)
- Growth and coarsening are considered in a single equation
 - A particle above the critical radius grows
 - A particle below the critical radius dissolves
- The growth law used in ClaNG derives from Zener's formulation
 - All the particles are considered spherical

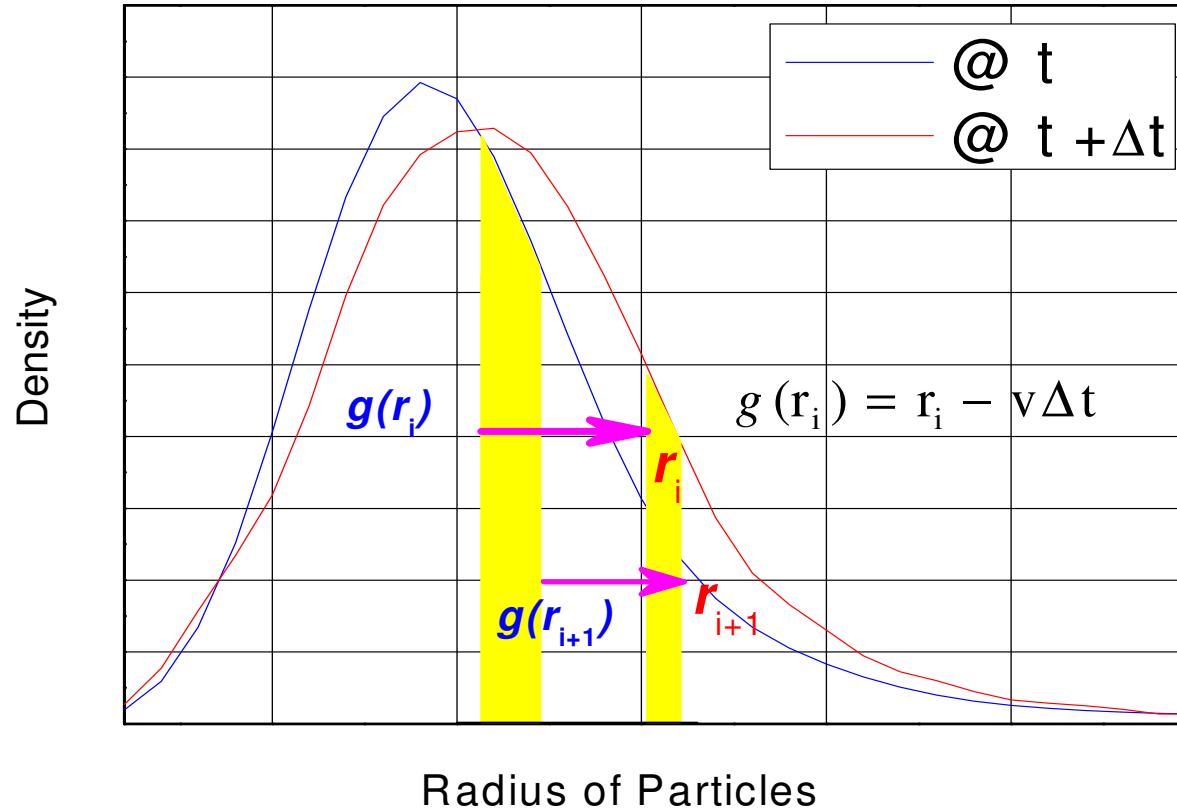


$$v = \frac{dr}{dt} = \frac{c^{\alpha 0} - c^{\alpha/\beta}(r)}{c^\beta - c^{\alpha/\beta}(r)} \cdot \frac{D_{\text{slowest}}}{r}$$

$c^{\alpha/\beta}(r)$ is
Gibbs-Thomson concentration
at the interface

Phase Evolution_Size

□ Discretization in radius classes

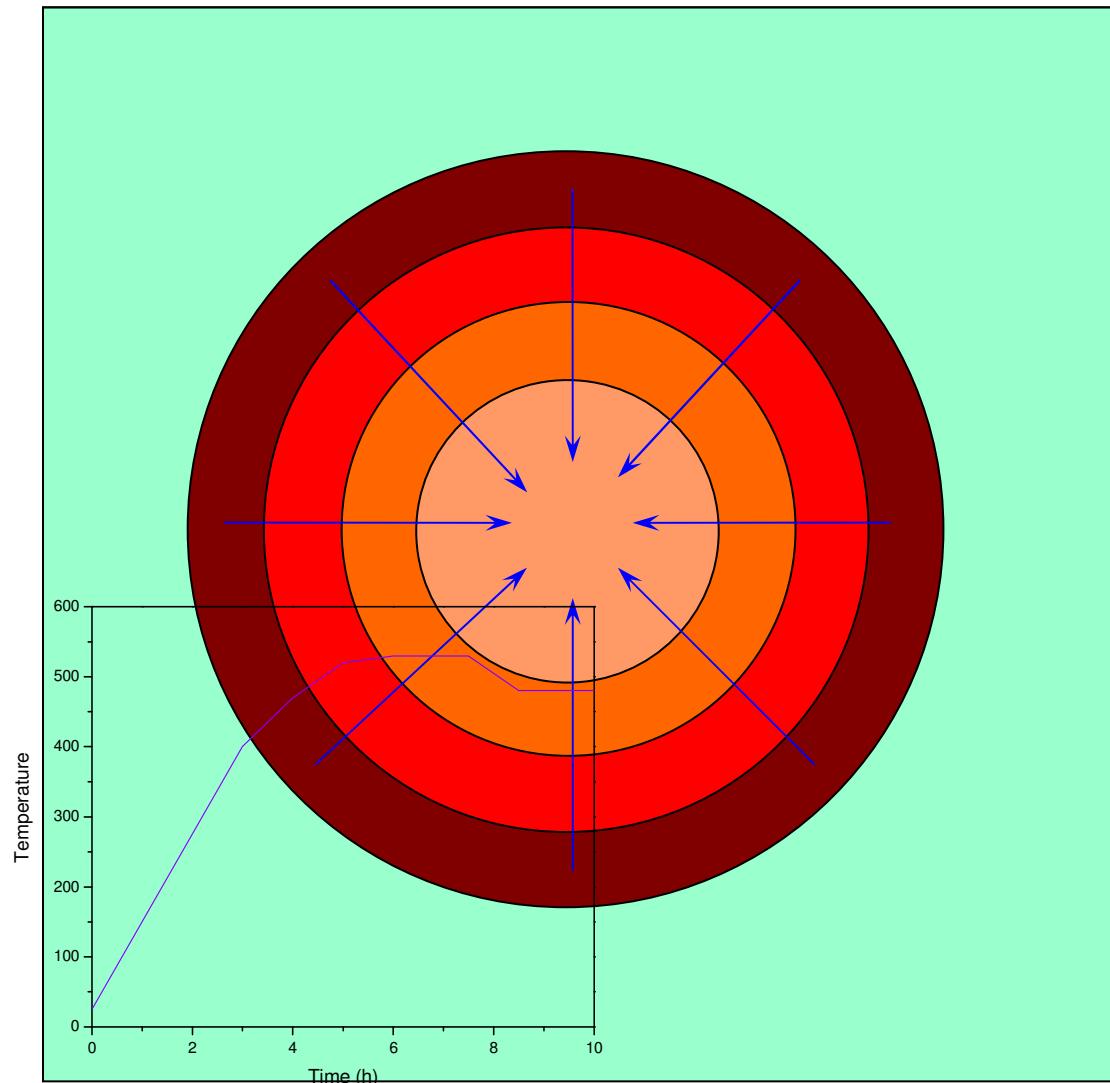


$$N_i(t + \Delta t) = N_i(t) + \int_{g(r_i)}^{r_i} f(r, t) \cdot dr - \int_{g(r_{i+1})}^{r_{i+1}} f(r, t) \cdot dr \quad (+ \dot{N}_i^{\text{nucl}} \Delta t)$$

Runge-Kutta method

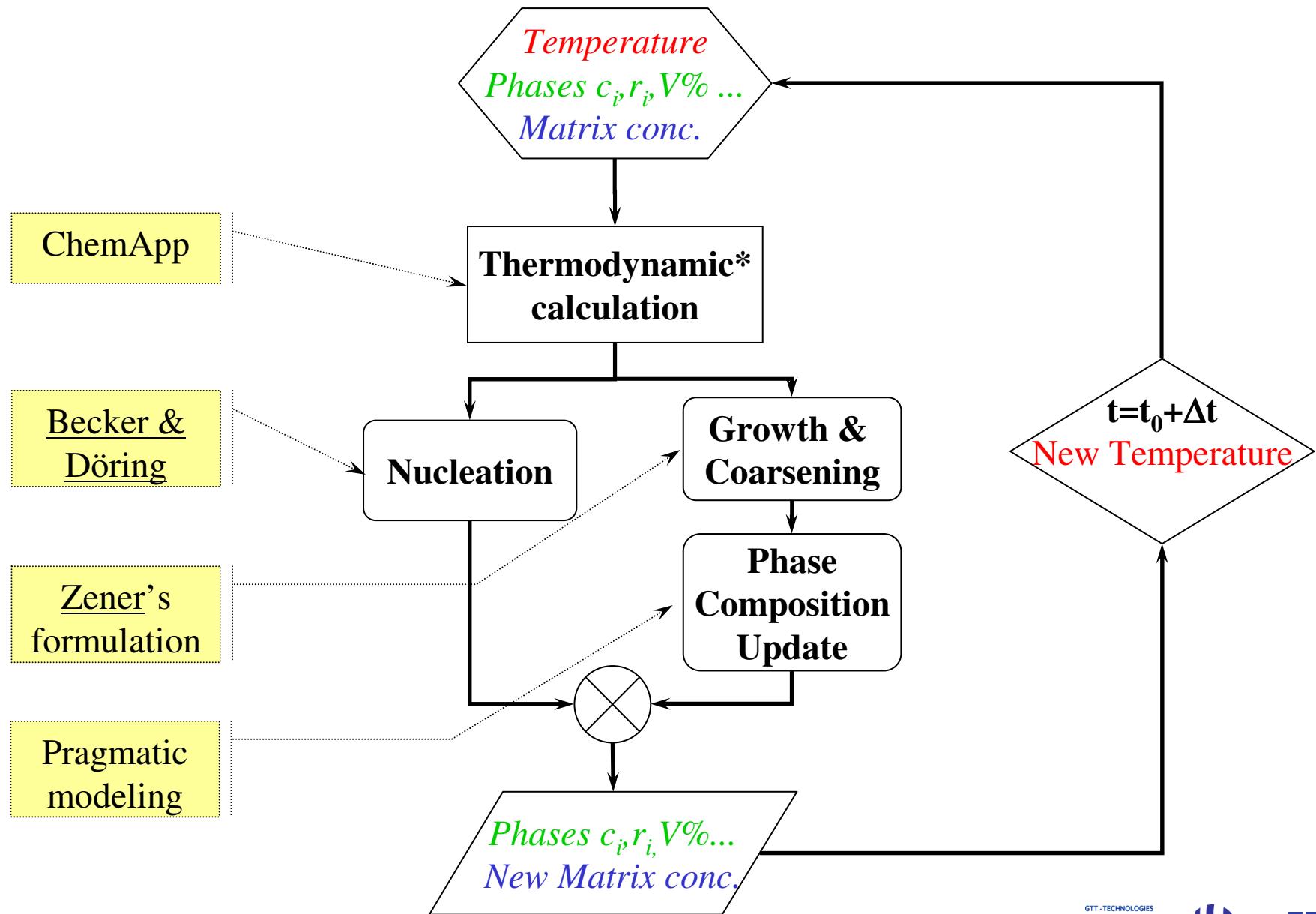


Phase Evolution *Composition*



- ❑ During heat treatment, the phase equilibrium composition changes
- ❑ Diffusion in precipitates does the work
 - :(Diffusion coefficient D_p in the phase is generally unknown
 - ✓ Pragmatic modeling: use of an empirical factor to level diffusion
 - ✓ $D_p = f_{corr} D_{bulk}$

Short Review



Prediction Quality of ClaNG

□ Alloy AA3014

- M.Schneider „*Modellierung und Validierung zeitabhängiger mikrochemischer Prozesse in Aluminium Knetlegierungen*“ RWTH Aachen 2006

□ Alloy AA5182

- A. Knoll „*Untersuchungen zur Entwicklung der Mikrochemie während der Homogenisierung von Aluminiumwalzlegierungen*“

Simulation of AA3014

Composition of alloy AA3014

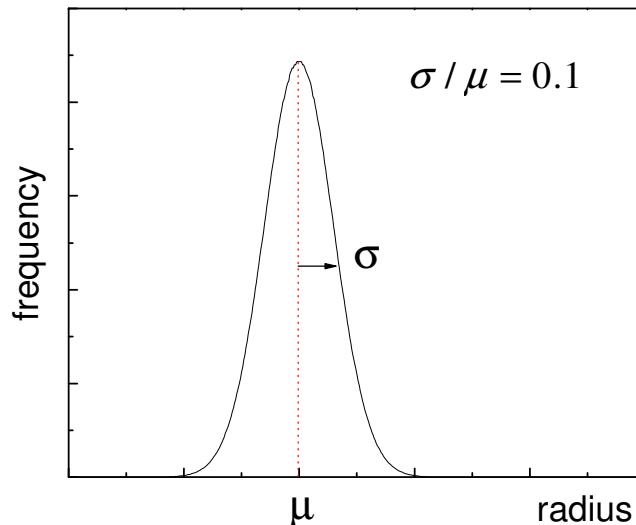
Element	wt%
Mg	1.0964
Mn	0.8425
Fe	0.4691
Si	0.1713
Ti	0.0182
Cu	0.1762
Cr	0.0084
Zn	0.0176
Al	...

➤ AlTT database(8 Elements)

Pre-existed Phases:

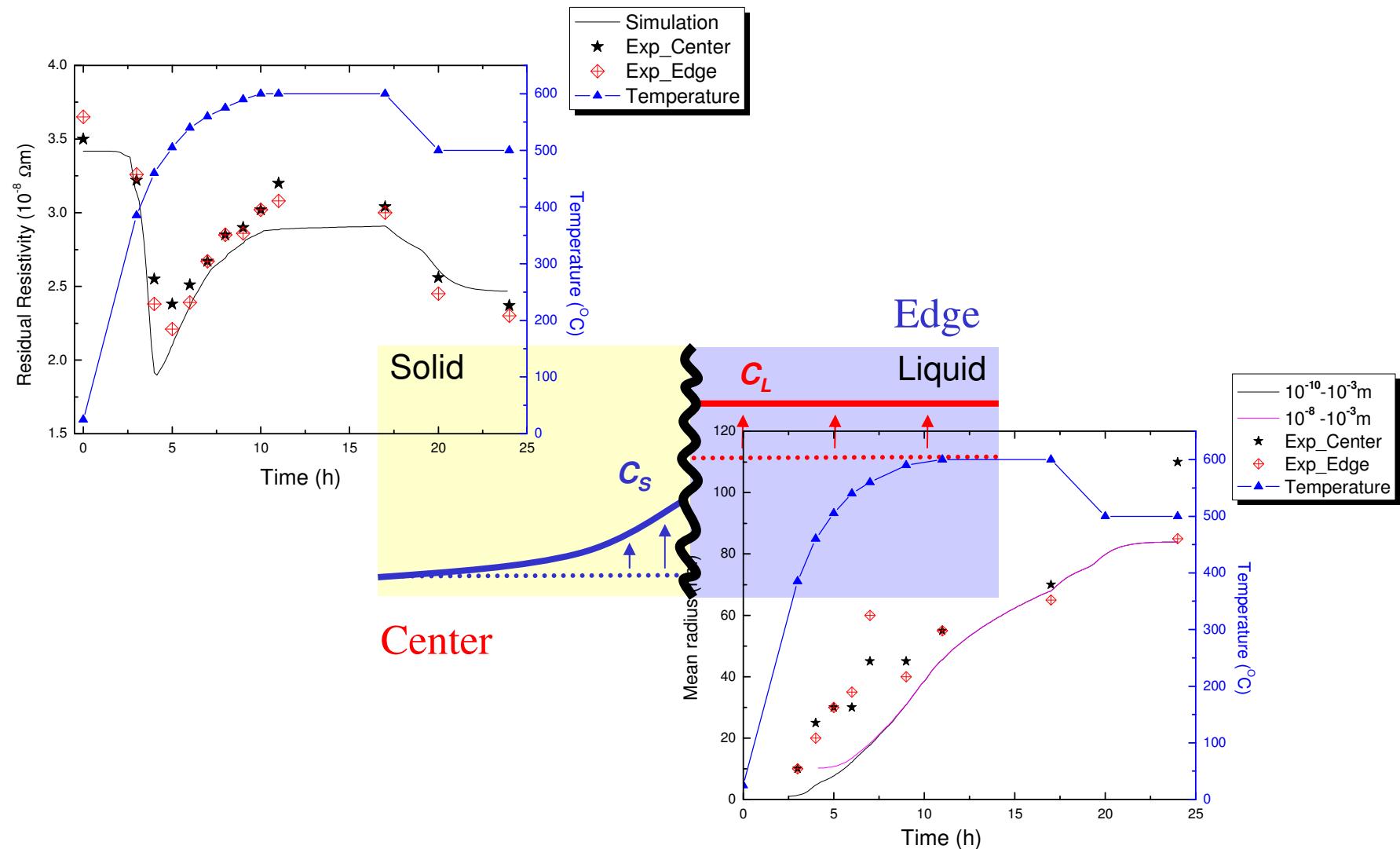
Alpha ($Al_{12}(Mn,Fe)_3Si_2$) $r=2\mu m$
 Mg_2Si $r=30nm$

follow Gauss distribution



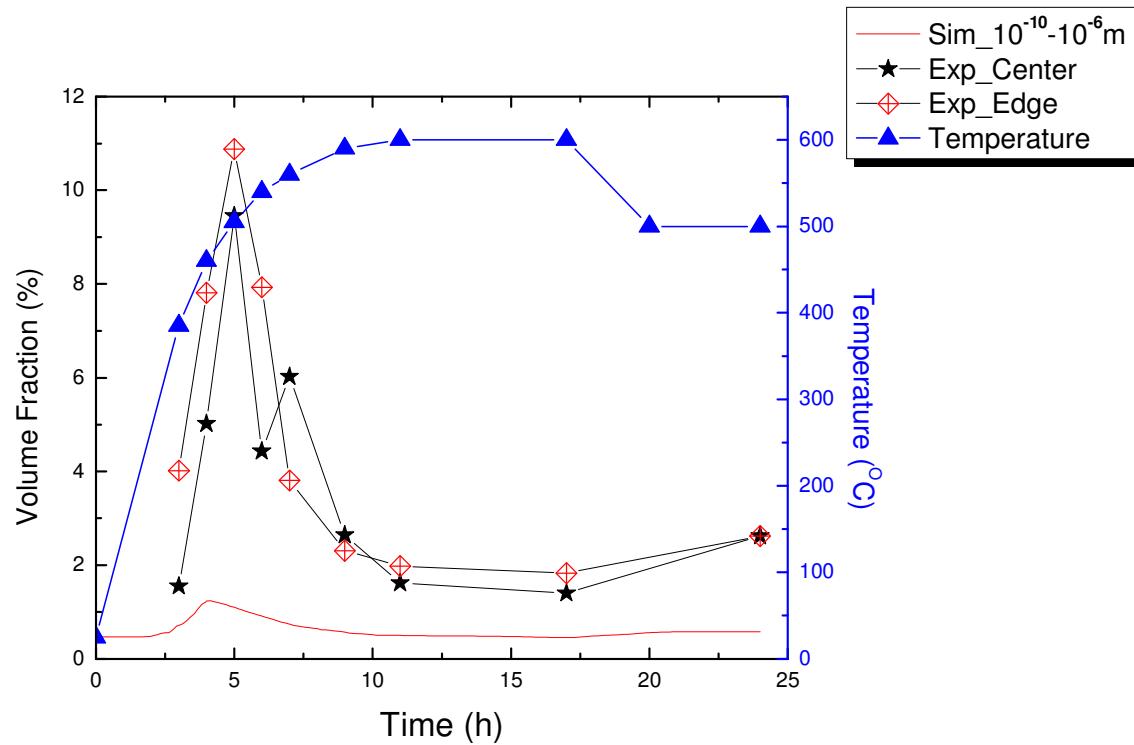
$$f_r = \frac{1}{\sqrt{2\pi\sigma^2}} \exp\left(-\frac{(r-\mu)^2}{2\sigma^2}\right)$$

AA3014_Residual Resist&Mean Radius



AA3014_Volume Fraction

Volume Fraction of Dispersoids



Reason: etching of the dispersoids

Simulation of AA5182

Composition of alloy AA5182

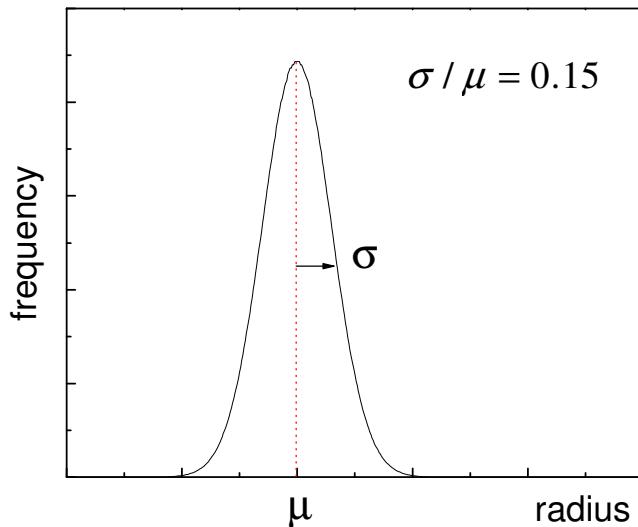
Element	wt%
Mg	4.200
Mn	0.330
Fe	0.250
Si	0.052
Ti	0.021
Cu	0.014
Cr	0.001
Zn	0.003
Al	...

➤ AltT database(8 Elements)

Pre-existed Phases:

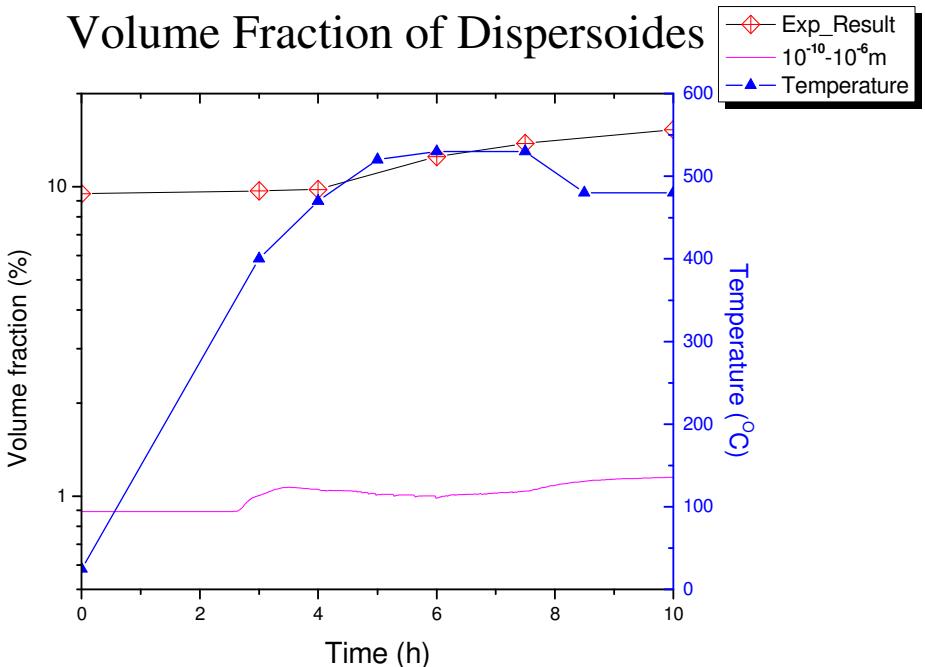
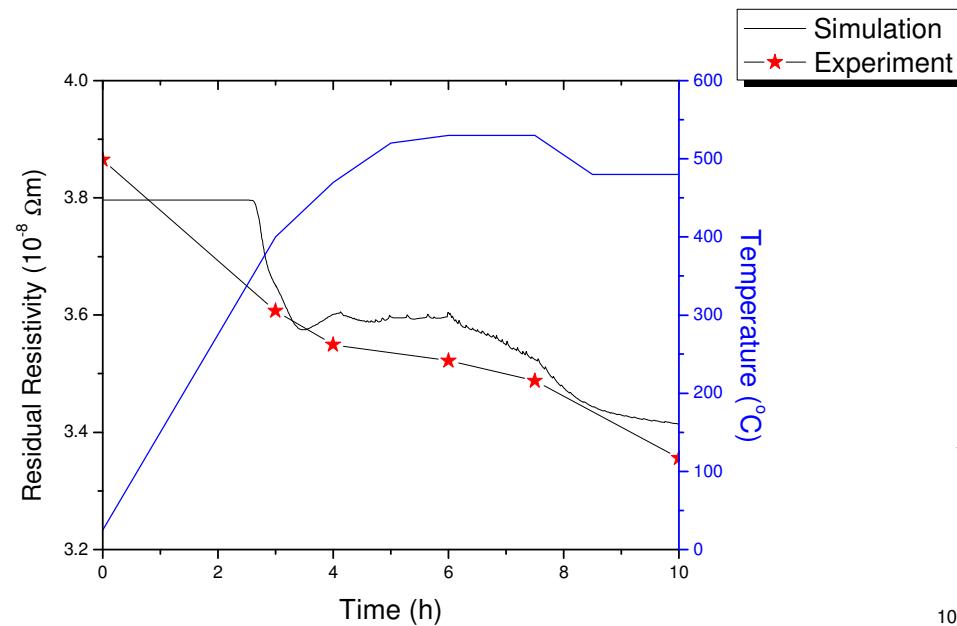


follow Gauss distribution

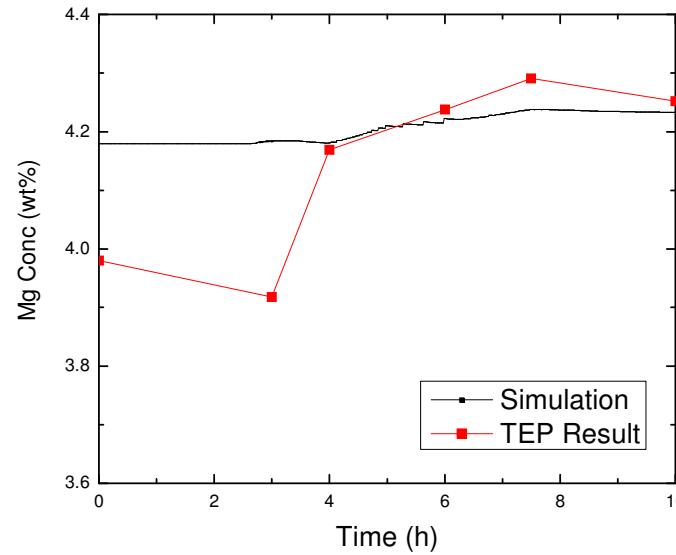
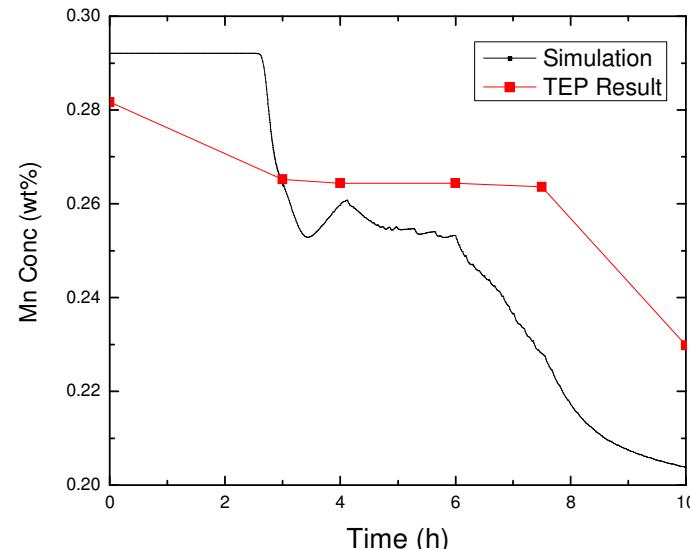
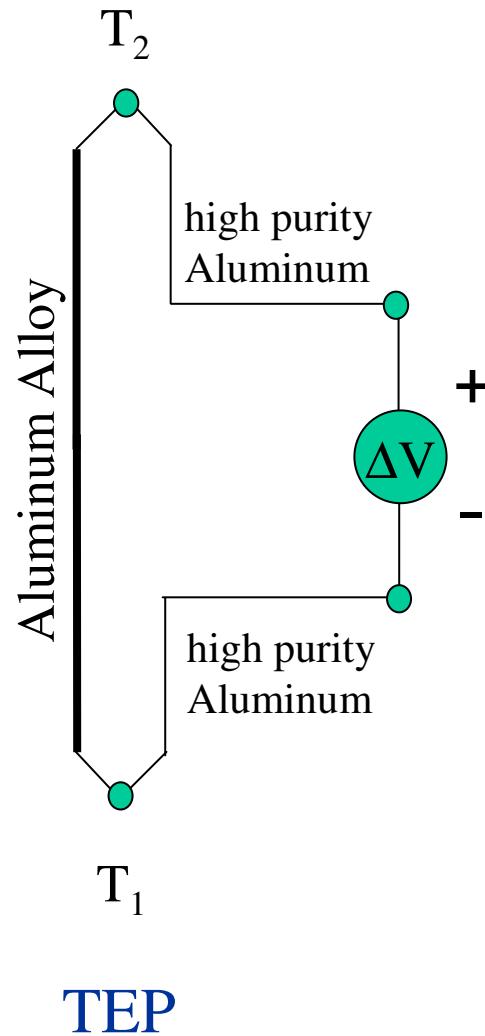


$$f_r = \frac{1}{\sqrt{2\pi\sigma^2}} \exp\left(-\frac{(r-\mu)^2}{2\sigma^2}\right)$$

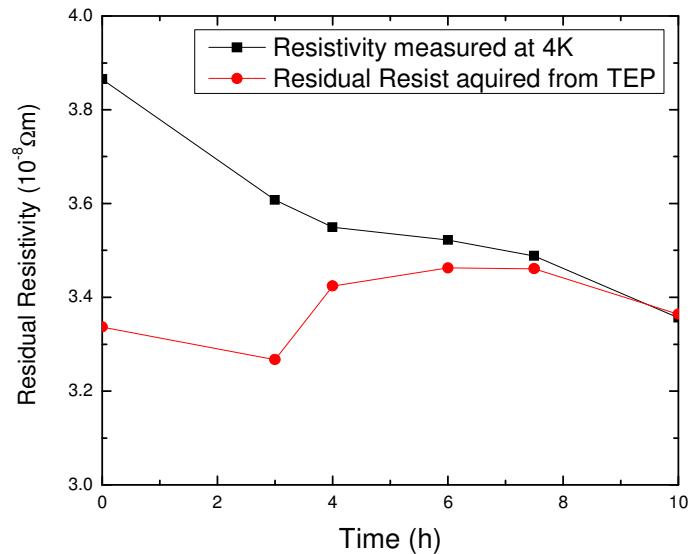
AA5182_Residual Resist&Volume Fraction



AA5182_Thermo Electrical Power

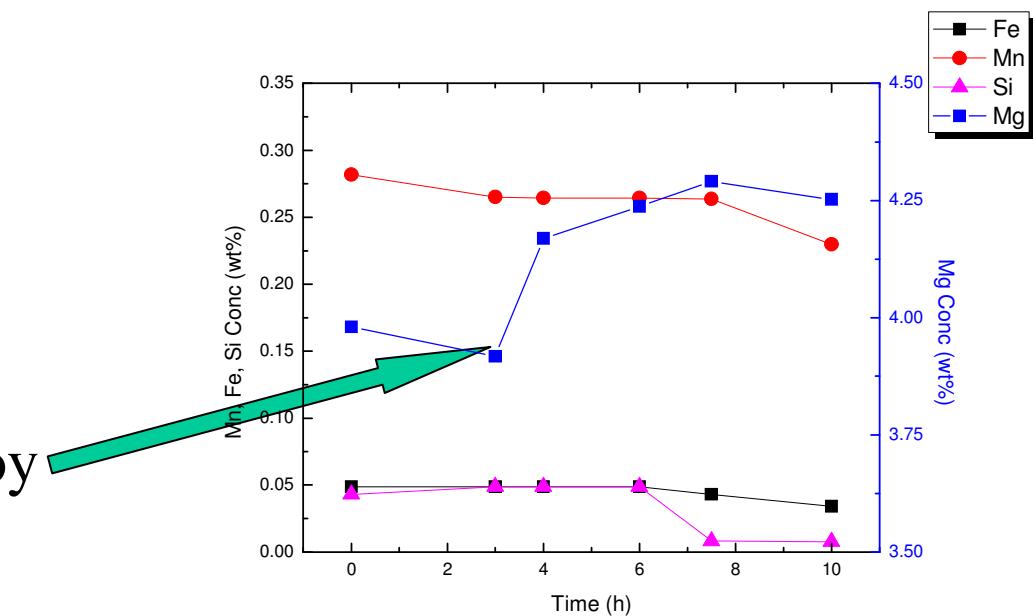


AA5182_Thermo Electrical Power



one of them is incorrect

Inhomogeneity of Alloy



Prediction Quality of ClaNG

- ☺ Residual Resistivity, mean radius, matrix composition (TEP) match very well with each other.
- ☹ Volume fraction of dispersoides measured by experiment is too high,due to experiment method
- ✓ Prediction of ClaNG is correct.

Simulation of AA5182

Composition of alloy AA5182

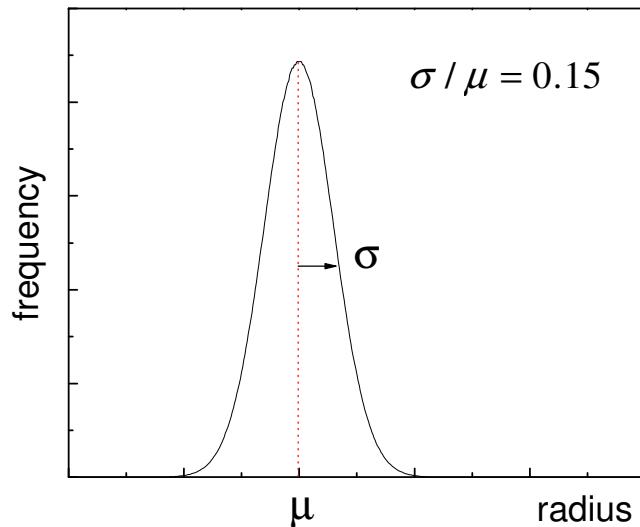
Element	wt%
Mg	4.200
Mn	0.330
Fe	0.250
Si	0.052
Ti	0.021
Cu	0.014
Cr	0.001
Zn	0.003
Al	...

➤ AltT database(8 Elements)

Pre-existed Phases:



follow Gaus distribution



$$f_r = \frac{1}{\sqrt{2\pi\sigma^2}} \exp\left(-\frac{(r-\mu)^2}{2\sigma^2}\right)$$

Database Extension

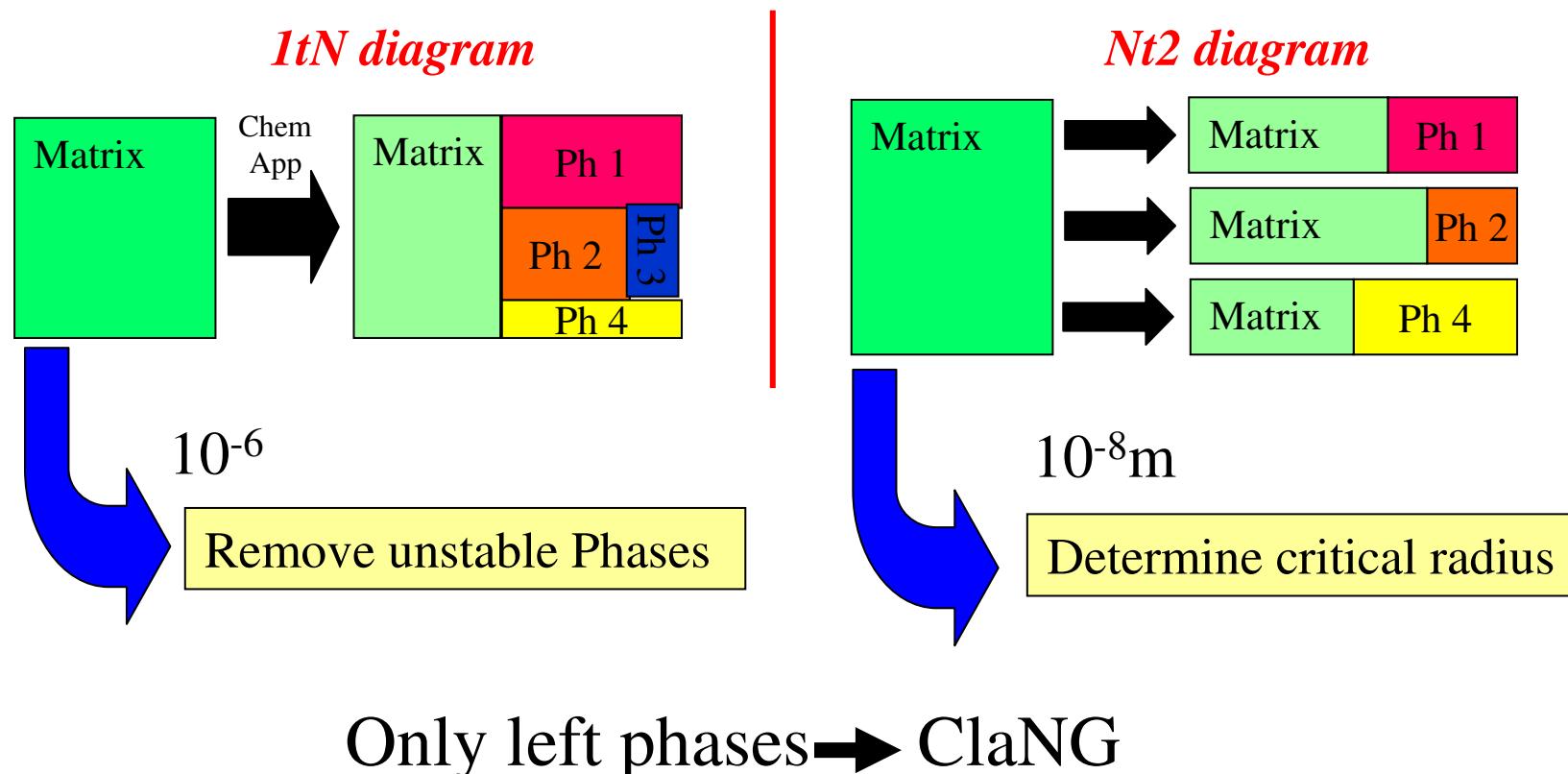


- Original database contains 8 Elts and 27 Phases
- New database contains 15 Elts and 42 Phases

Elt	Al	B	C	Cr	Cu	Fe	Mg	Mn	Ni	Si	Sr	Ti	V	Zn	Zr
TTal15	●	●	●	●	●	●	●	●	●	●	●	●	●	●	●
TTal8	●			●	●	●	●	●		●		●			

Optimization of the computation

- ✓ The phase diagram for the considered alloy is first calculated before starting a ClaNG calculation
- ✓ Two types of phase diagrams calculation



AlTT8 Vs AlTT15

Input composition of Databank AlTT8

Element	wt%
Mg	4.200
Mn	0.330
Fe	0.250
Si	0.052
Ti	0.021
Cu	0.014
Cr	0.001
Zn	0.003
Al	...

Input composition of Databank AlTT15

Element	wt%	Element	wt%
Mg	4.200	B	1e-4
Mn	0.330	C	1e-6
Fe	0.250	Ni	1e-4
Si	0.052	Sr	1e-4
Ti	0.021	V	1e-4
Cu	0.014	Zr	5e-4
Cr	0.001		
Zn	0.003		
Al	...		

Phases in Database A1TT8

2009.05.28 15:32:16	MESSAGE:	1tN Diagram in File Thermodyn_1tN.txt
2009.05.28 15:32:17	MESSAGE:	Nt2 Diagram in File Thermodyn_Nt2.txt
2009.05.28 15:32:20	MESSAGE:	The Phase [2] = [AL2CU] with VolFrac = 0.015 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [3] = [AL3FE] with VolFrac = 1.132 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [5] = [AL3TI] with VolFrac = 0.302 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [6] = [AL6MN] with VolFrac = 2.011 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [7] = [AL7CR] with VolFrac = 2.122 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [8] = [ALMG_BETA] with VolFrac = 11.045 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [9] = [ALPHA] with VolFrac = 1.147 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [10] = [E_ALCRMGMN] with VolFrac = 1.867 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [11] = [FCC_A1] with VolFrac = 100.000 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [12] = [MGZN2#1] with VolFrac = 0.028 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [14] = [T_ALCUMGZN] with VolFrac = 8.488 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [15] = [Al13Cr4Si4] with VolFrac = 0.003 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [16] = [Al20Cu2Mn3] with VolFrac = 0.074 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [17] = [Al3Fe<AL3NI>] with VolFrac = 0.482 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [18] = [Al5Cu2Mg8Si6] with VolFrac = 0.062 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [19] = [Al7Cu2Fe] with VolFrac = 0.030 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [20] = [Al8FeMg3Si6] with VolFrac = 0.150 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [21] = [Al9Fe2<AL9FENI>] with VolFrac = 0.663 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [22] = [Al71Fe19Si1<AFS_ALPHA>] with VolFrac = 0.499 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [23] = [Al692Fe154Si154<AFS_BT>] with VolFrac = 0.324 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [25] = [Mg2Si] with VolFrac = 0.150 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [26] = [Si<SiLICON>] with VolFrac = 0.050 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [27] = [Al2CuMg<S>] with VolFrac = 0.024 Perc. is considered
2009.05.28 15:32:20	MESSAGE:	The Phase [1] = [LIQUID] with VolFrac = 0.175 Perc. is not considered
2009.05.28 15:32:20	MESSAGE:	The Phase [4] = [AL3NI2] has a too low VolFrac = 0.000 Perc.
2009.05.28 15:32:20	MESSAGE:	The Phase [13] = [MGZN2#2] has a too low VolFrac = 0.000 Perc.
2009.05.28 15:32:20	MESSAGE:	The Phase [24] = [Ti<MC>] has a too low VolFrac = 0.000 Perc.
2009.05.28 15:32:20	MESSAGE:	

Considered Phases:23/27

Removed Phases:4 /27

Phases in Database A1TT15

```
2009.05.29 08:52:05 MESSAGE:  
*** Phase Diagram for the given alloy between 25.0 DegC and 530.0 DegC for a min. VolFrac of 1.0E-05 ***  
2009.05.29 08:52:05 MESSAGE: 1tN Diagram in File Thermodyn_1tN.txt  
2009.05.29 08:52:08 MESSAGE: Nt2 Diagram in File Thermodyn_Nt2.txt  
2009.05.29 08:52:22 MESSAGE: The Phase [ 2] = [AL3FE] with VolFrac = 1.156 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [ 3] = [AL3MG2] with VolFrac = 10.983 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [ 4] = [AL3NI] with VolFrac = 0.483 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [ 6] = [AL3TI] with VolFrac = 0.273 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [ 8] = [AL6MN] with VolFrac = 2.011 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [ 9] = [AL7CR] with VolFrac = 2.124 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [10] = [AL9FENI] with VolFrac = 0.664 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [11] = [ALPHA] with VolFrac = 1.662 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [12] = [E_ALCRMGMN] with VolFrac = 1.869 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [13] = [FCC_A1] with VolFrac = 100.000 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [17] = [MGZN2] with VolFrac = 0.029 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [18] = [T_ALCUMGZN] with VolFrac = 8.196 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [19] = [Al13Cr4Si4] with VolFrac = 0.003 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [20] = [Al20Cu2Mn3] with VolFrac = 0.074 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [21] = [Al2Si2Sr] with VolFrac = 0.080 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [23] = [Al4Sr] with VolFrac = 0.080 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [24] = [Al5Cu2Mg8Si6] with VolFrac = 0.062 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [25] = [Al7Cu2Fe] with VolFrac = 0.030 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [27] = [Al8FeMg3Si6] with VolFrac = 2.173 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [29] = [<ALFESI_ALPHA>] with VolFrac = 0.635 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [30] = [<ALFESI_BETA>] with VolFrac = 0.784 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [31] = [Mg2Si] with VolFrac = 2.880 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [33] = [Al2CuMg] with VolFrac = 0.024 Perc. is considered  
2009.05.29 08:52:22 MESSAGE: The Phase [35] = [Si< SILICON >] with VolFrac = 0.960 Perc. is considered
```

Considered Phases: 24/42

Phases in Database A1TT15



Removed Phases: 16/42

iM

Thanks very much for your attention !!!

