



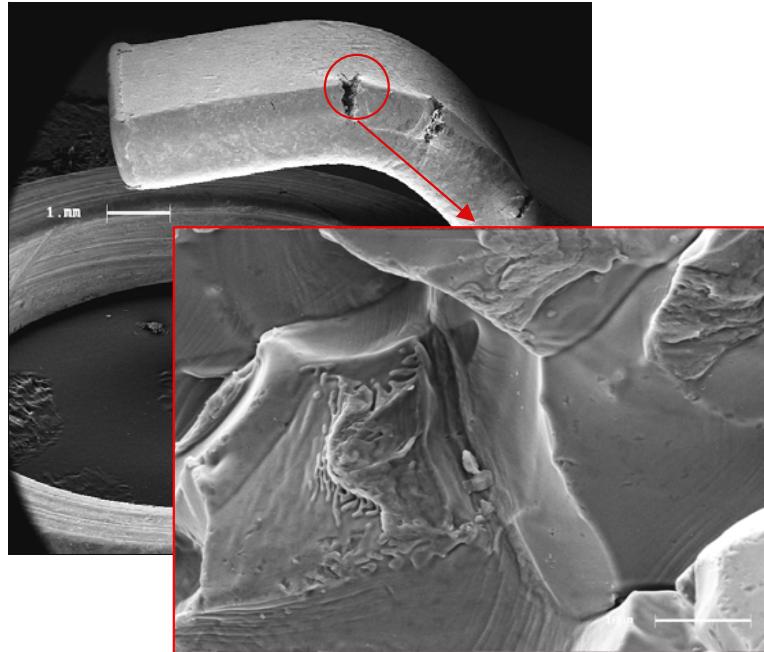
Formation and modification of non-metallic inclusions through slags during ladle treatment

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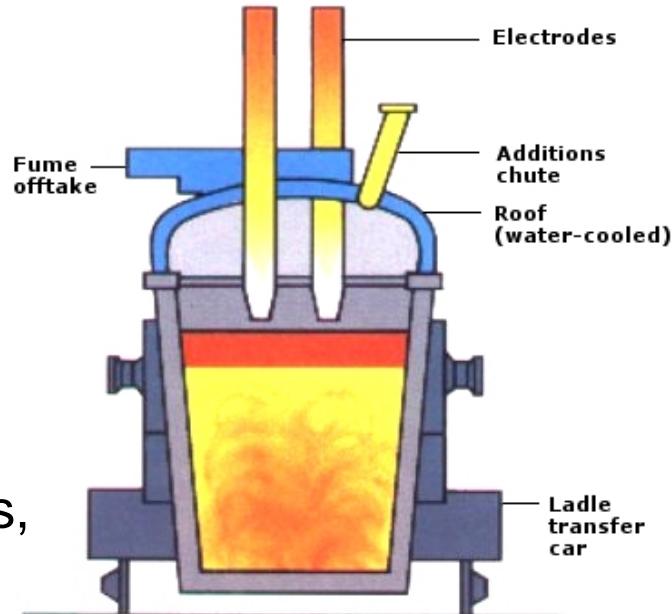
Aim of investigations

- non-metallic inclusions (e.g. alumina, MgO-alumina-spinel) harmful in steels
- complete removal during not possible
 - effect of ladle slag on inclusions?
 - modification to less harmful inclusions possible?
 - low-melting inclusions?
- reoxidation through ladle slag (e.g. FeO, MnO, SiO₂)
- formation of new inclusions
- loss of alloy elements (e.g. Al, Ti)
 - slag with low reoxidation potential?
- optimized ladle treatment strategies for steelmaking industry
- approximate industrial practice and analyze potentials



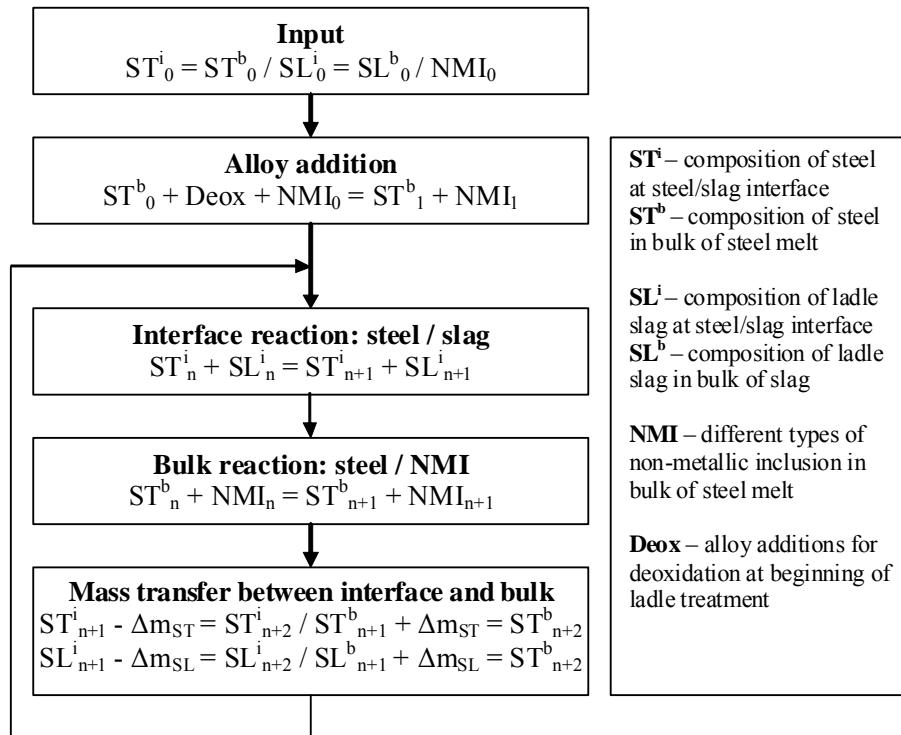
Basic modeling data

- ladle treatment during secondary steelmaking:
deoxidation, alloying, stirring
- equilibrium between steel melt, slag phase,
non-metallic inclusions in steel
→ complex equilibria
- modeling with FactSage using data:
 - steel and slag compositions, alloy additions,
temperature
 - FACT-databases
 - preselection of solution (steel, slag, complex inclusions) and
compound (pure inclusions) phases
- time-dependent behaviour through kinetic approximation
- usage of macro language and Microsoft Excel



Model development

- no overall thermodynamic equilibrium possible
→ reactions only at interfaces
- splitting into two reaction zones
- consecutive equilibrium calculation
- mass transfer according to concentration gradient (diffusion) and stirring (mixing)



Bulk equilibrium: steel melt /
non-metallic inclusions

Interface equilibrium: steel interface /
slag interface

Example - alloyed tool steel

- modeling Al-killed tool steel in secondary steelmaking
- start steel composition [wt%]:

C	Mn	Si	Cr	Ni	Mo	V	Al	heat size
0.4	0.4	1.0	5.3	0.1	1.3	0.9	0.1**	65 t

*[O] assumed to be 0.06 wt% after tapping

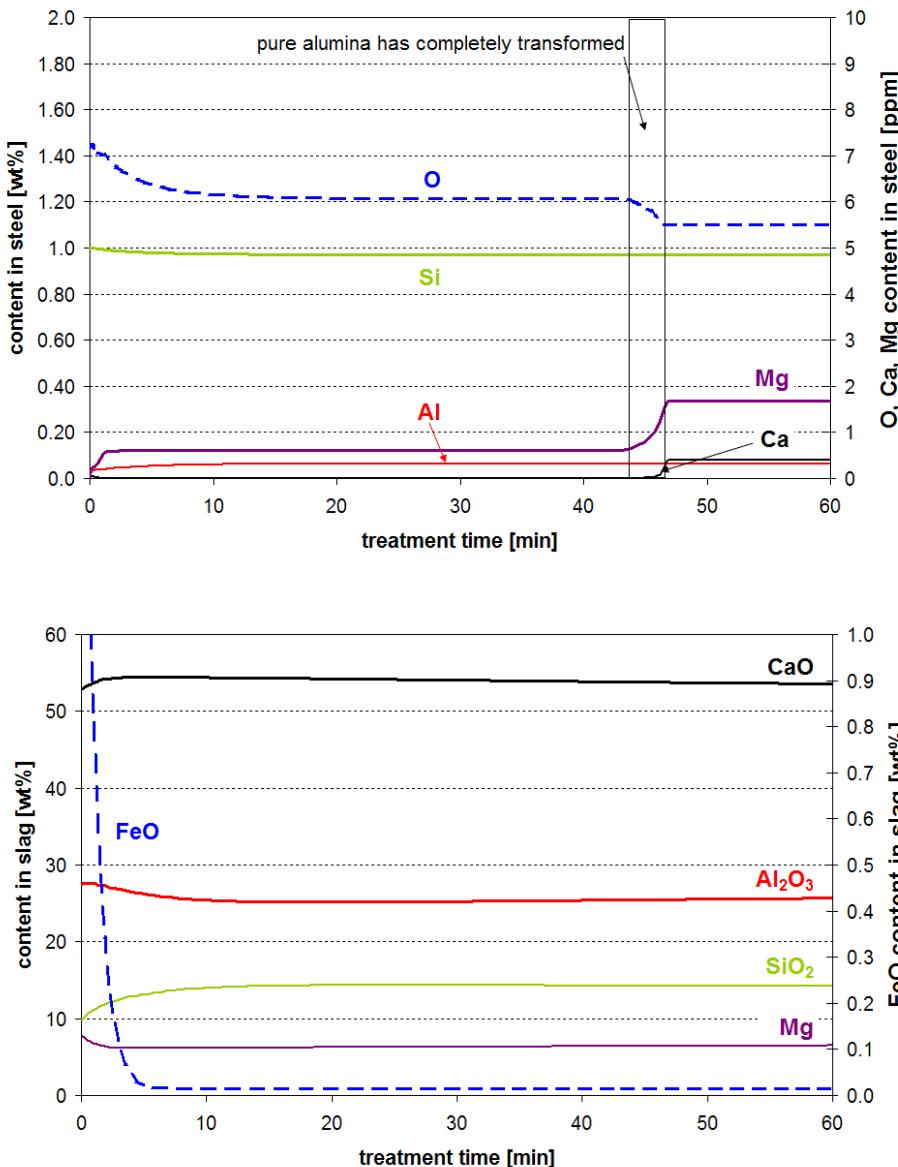
** Al added immediately after tapping

- variation of slag composition [wt%]:

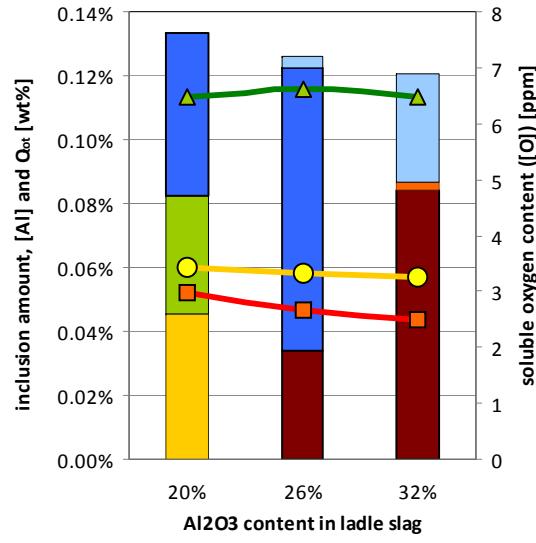
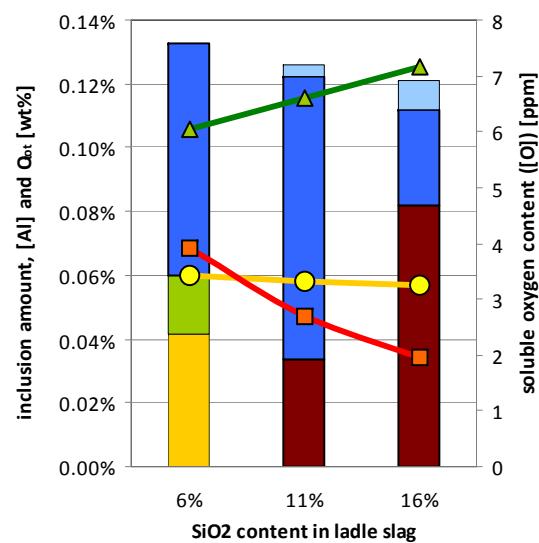
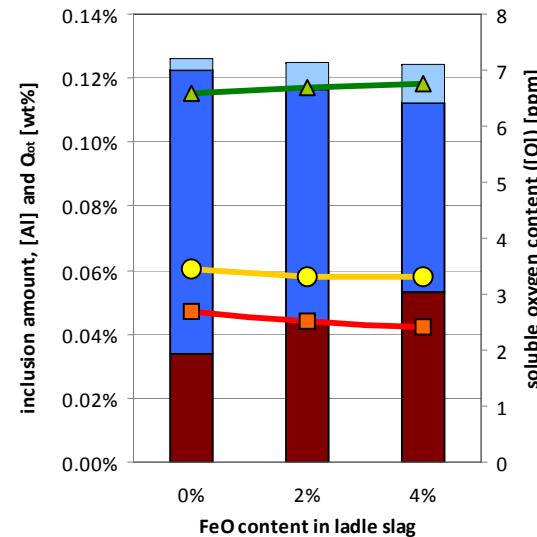
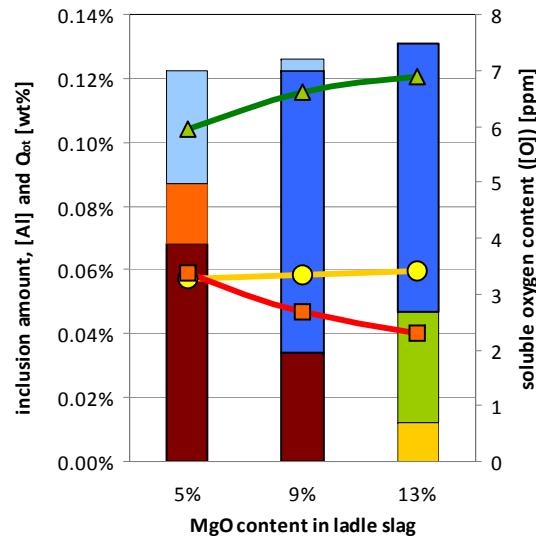
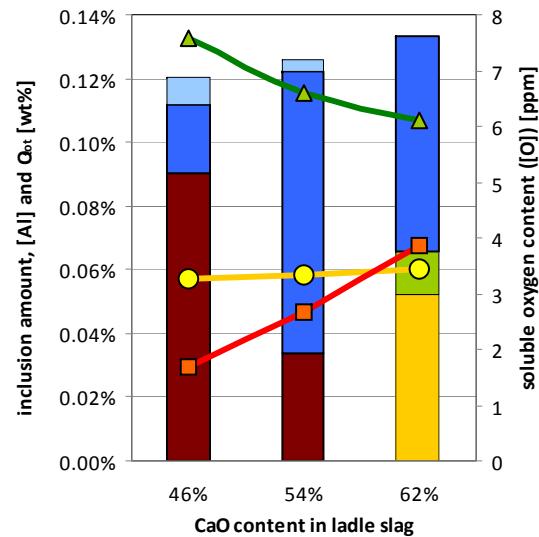
CaO	SiO ₂	Al ₂ O ₃	MgO	FeO	MnO	slag amount
54 (± 8)	11 (± 5)	26 (± 6)	9 (± 4)	0 (+2, +4)	0 (+2, +4)	1 t

Steel and slag chemistry variation

- time-dependent change of:
 - steel composition
 - slag composition
 - type and amount of non-metallic inclusions
- formation of alumina at start
- introduction of Ca, Mg from slag
- loss of silicon
- recovery of aluminum
- steady transformation of alumina into spinel-type and calcium-aluminate inclusions
- decrease of oxygen content



Effect of ladle slag on inclusion types



Legend:

- 2CaO·2MgO·14Al₂O₃ (CMA2)
- CaO·2MgO·8Al₂O₃ (CMA)
- CaO·6Al₂O₃
- MgO·Al₂O₃ (spinell)
- liquid, alumina rich
- alumina
- soluble Al in steel ([Al]) [wt%]
- total O in steel (O_{tot}) [wt%]
- soluble O in steel ($[\text{O}]$) [ppm]

- results after 60 min of ladle treatment / only variation of ladle slag

Main conclusions

- pure alumina only stable at low CaO content of slag
- liquid inclusions only at very high slag basicities (CaO/SiO_2)
- pure MgO-alumina spinels only at high MgO contents

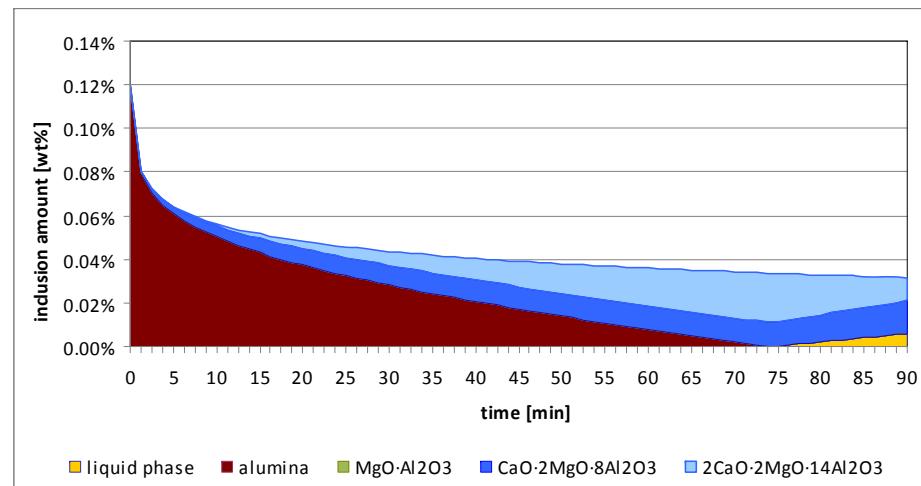
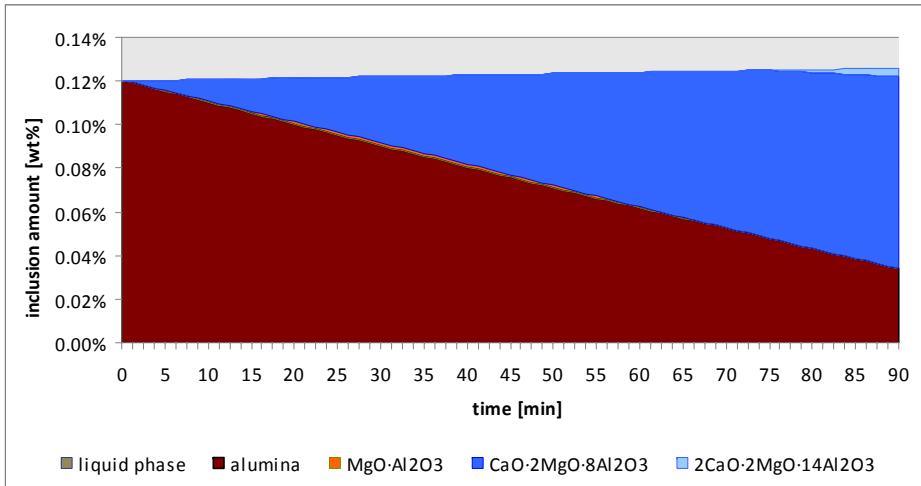
- total oxygen content (amount of inclusions) hardly changeable
- aluminum losses increase strongly with SiO_2 content
- low aluminum losses with liquid inclusions (low activities of components)
- small effect of FeO content at low level

Other effects

- modeling of inclusion settling
 - continuous addition to slag interface
 - remaining inclusions transform faster

- variation of start steel composition
 - modeling of variation of furnace operation

- variation of alloy addition
 - modeling of variation of alloy composition and tramp elements



Future work

- comparison of model results with laboratory and practice experiments
 - inclusion types at certain times, steel composition
- including ladle refractory interaction
 - new reaction zone: steel melt / ladle refractory
- coupling of thermodynamic calculations with fluid flow modeling
 - calculation of equilibria in several parts of the ladle
 - introduction of mixing through fluid flow (steel and slag)

Thank you for
your attention!