

Modelling and Experiments for Stainless Steel Slag Valorisation

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World steel production: $\sim 1.25 \cdot 10^9$ tons

Slag/steel ratio: $1/3 \sim 1/4$

World steel slag production: $\sim 350 \cdot 10^6$ tons

- 50 pyramids of Gizeh every year

- WASTE?

- Large dumping fees
- Ecological implications

- Or... VALORISATION

- Small revenue for the “secondary resource”
- Sustainability & Zero waste



Secondary resource



Primary resources

slag



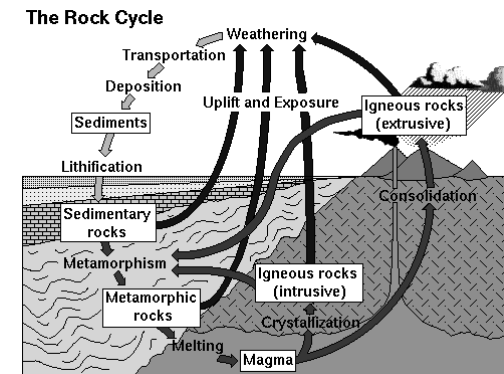
sand, gravel, natural rocks

When balancing the choice of resource:

Cost, availability, grain size distribution, strength, abrasion resistance, polishing resistance, volume stability, hydraulic properties, heavy metal leaching, porosity...



Human time-scale vs. Geological time-scale



Can the slag be used?

Properties

Process

Structure

- Composition
- Cooling path

Thermochemistry!

- Microstructure
- Crystallography

Industrial project

- Research group: Thermodynamics in metallurgy
 - High-T centre
 - IWT-project on stainless steel slag valorisation
- Use FactSage data for
 - Process modelling
 - Previous project: VOD, EAF
 - Current project: de-sulphurisation, slag additions
 - Solidification modelling
- Also a need for data valorisation

Content

- Experimental verification of liquidus
- Slag solidification modelling
- Process model for additions

Experimental liquidus verification

- Goal:

- Verify the applicability of FactSage for multicomponent slags at high temperature
- Simplify stainless steel slag system to:
 $\text{CaO-SiO}_2\text{-MgO-Al}_2\text{O}_3\text{-Cr}_2\text{O}_3$ (CSMAK)

Literature data on liquidus

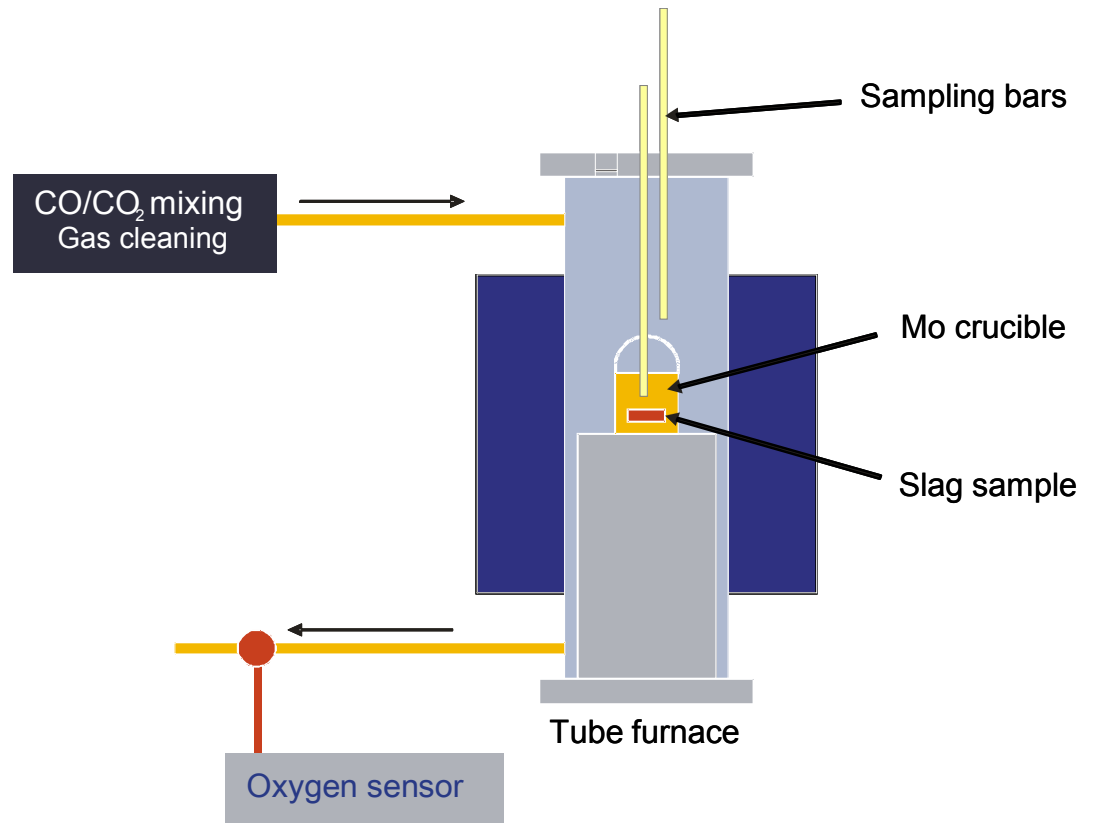
- Experimental
 - Mostly ternary
 - Up to ternary already used in optimizations
 - Some quaternary (CSAK...)
- Optimizations
 - 7 out of 10 ternaries
 - Other 3 have high liquidus temperatures (CMK, CAK, MSK)

Optimized data

- FactSage 5.4: database info
 - Slag
 - In the absence of SiO_2 : extensively optimized for oxides of Al, Ca, Co, Cr^{II} , Cr^{III} , Fe^{II} , Fe^{III} , Mg, Ni, Zn, some regions.
 - In the absence of Cr: fully evaluated and optimized for oxides of Al, Ca, Fe^{II} , Fe^{III} , Mg, Si, all regions.
 - Fully optimized for Al_2O_3 -CaO-CrO- Cr_2O_3 - SiO_2 solutions
 - Roughly optimized for $\text{CrO-}\text{Cr}_2\text{O}_3$ -MgO- SiO_2 solutions.
 - Spinel-solution
 - Evaluated and optimized over all compositions (but... pure MK is formed in calculations)
- CRCT: Completely optimized

Experimental method

- Mixing powders
- Equilibration
- Sampling
- EPMA-WDS

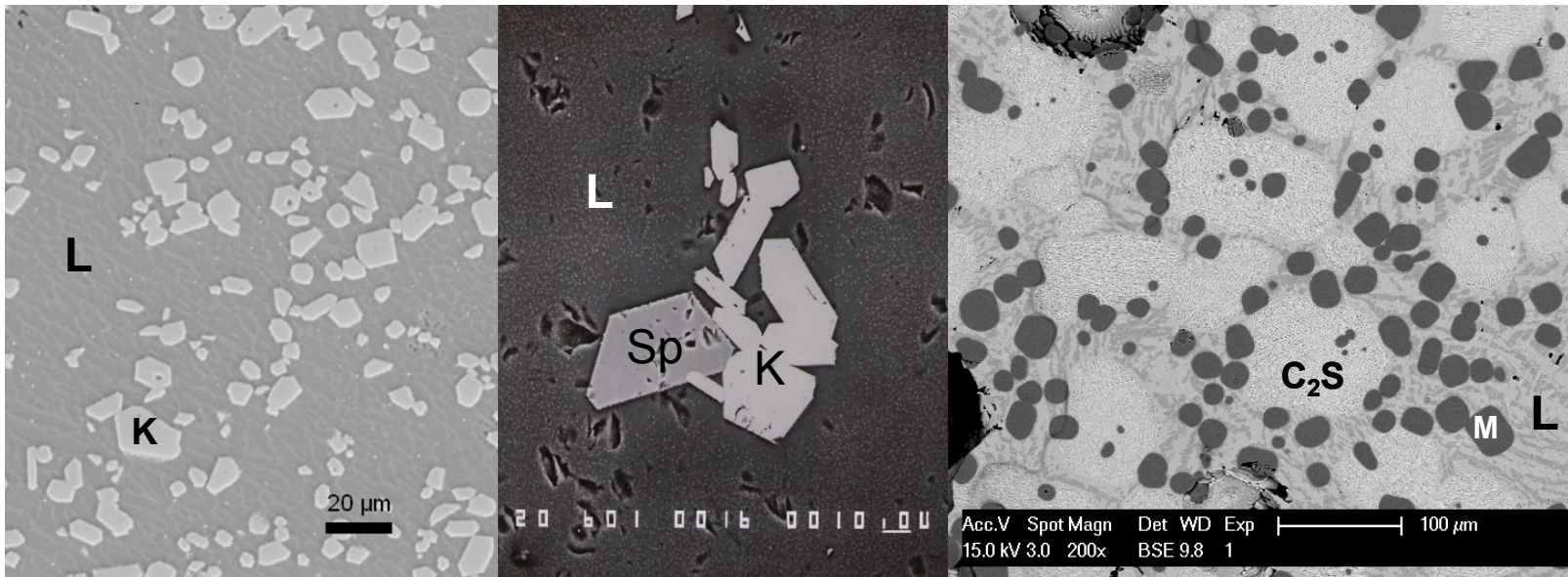


Parameters

Temperature	1500°C and 1600°C
CO/CO ₂ ⇒ p _{O2}	20 and 50 10 ^{-10.16} and 10 ^{-9.36} atm
Basicity C/S	0.5 and 1.2
Al ₂ O ₃	0, 10, 20, 30 wt%
MgO Cr ₂ O ₃	chosen in 2 or 3-phase area

Results: microstructures

- Phases correspond with the expected
- Amount of precipitates mostly higher than calculated



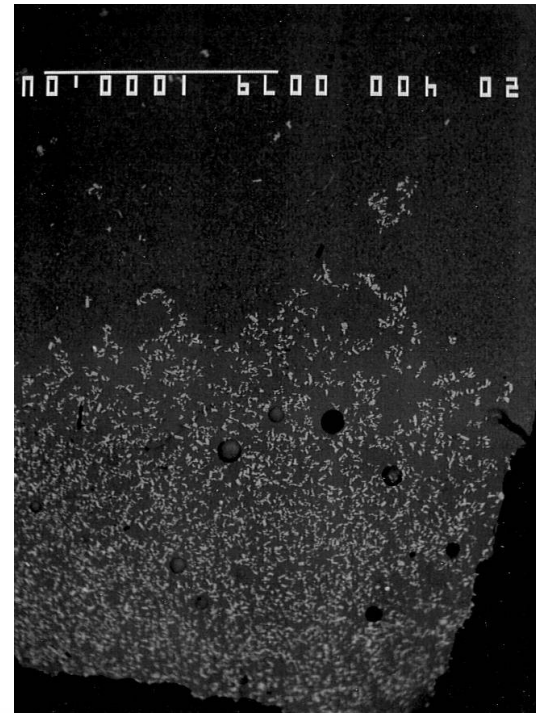
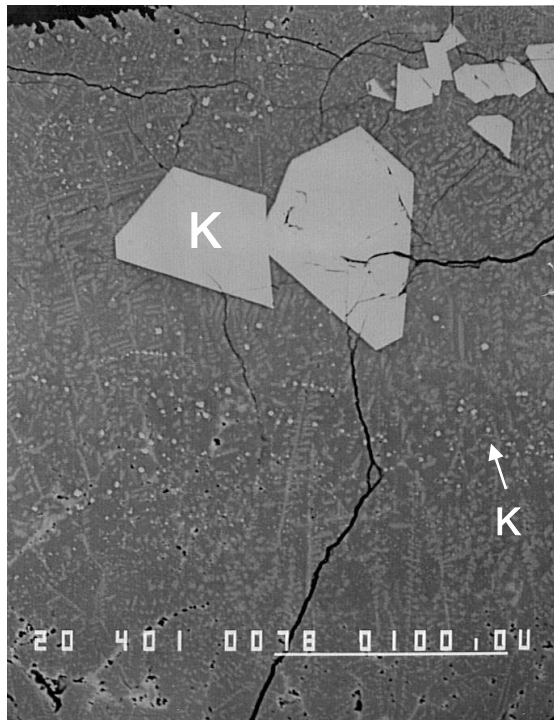
Slag + Cr_2O_3

Slag + Cr_2O_3 + spinel

Slag + C_2S + MgO

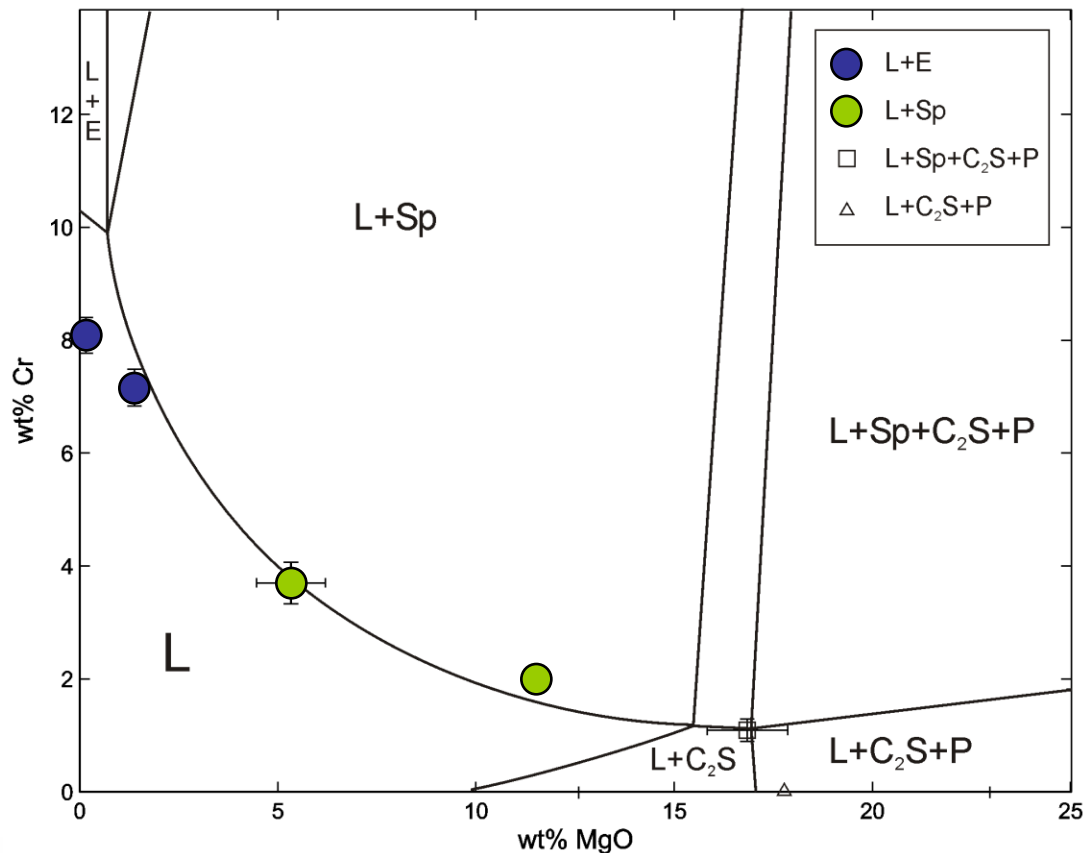
Results: microstructures

- Cooling speed is finite
- Matrix crystallisation occurs
- Interaction with precipitates
- Sedimentation $\Rightarrow$ zones
- Wetting $\Rightarrow$ loss of liquid



Results: liquidus

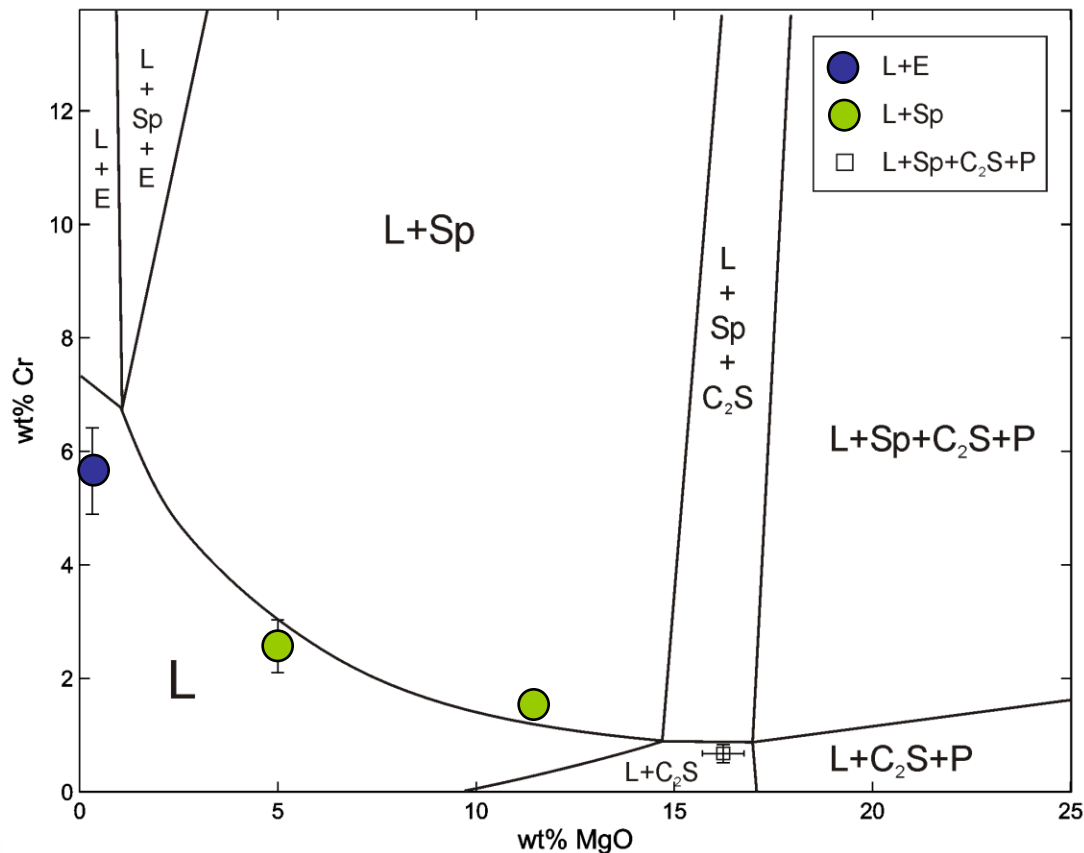
- Tests in different p_{O_2} , no Al_2O_3



$T=1600^{\circ}\text{C}$
 $p_{O_2}=10^{-10.16}$ atm
 $B=1.2$

Results: liquidus

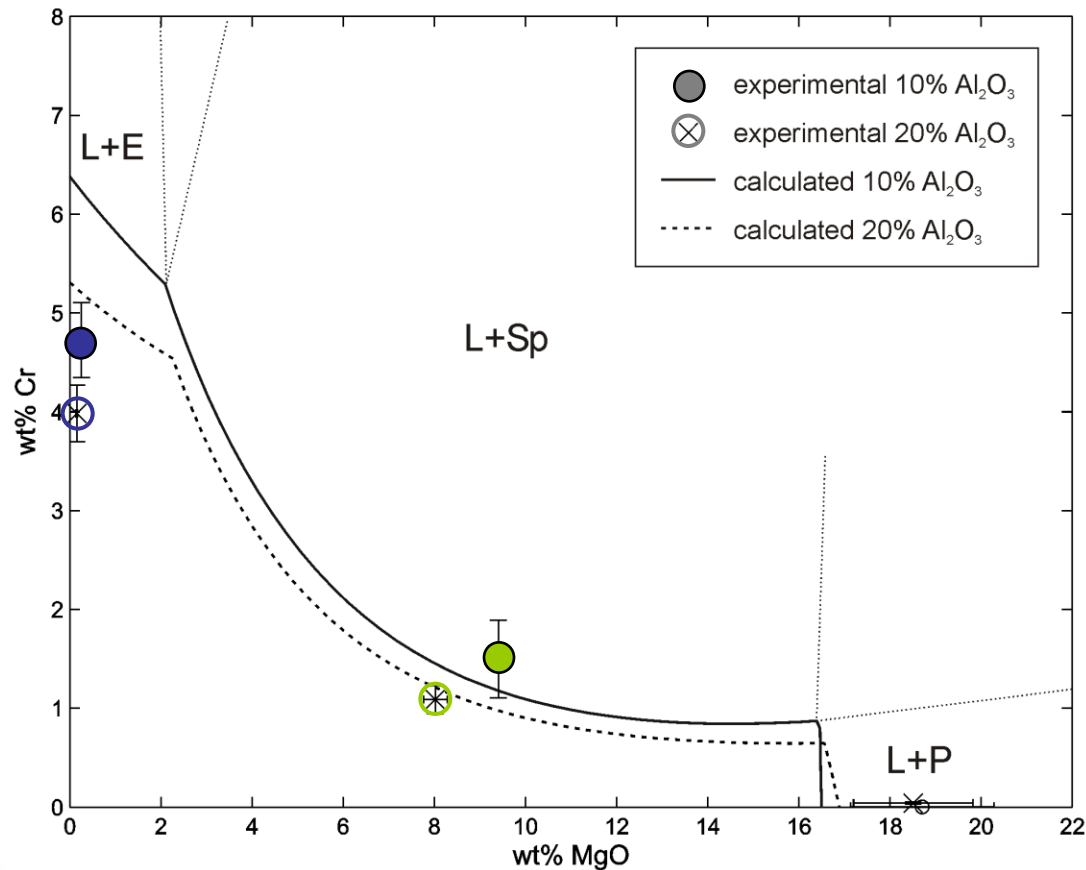
- Tests in different p_{O_2} , no Al_2O_3



$T=1600^{\circ}\text{C}$
 $p_{O_2}=10^{-9.36}\text{ atm}$
 $B=1.2$

Results: liquidus

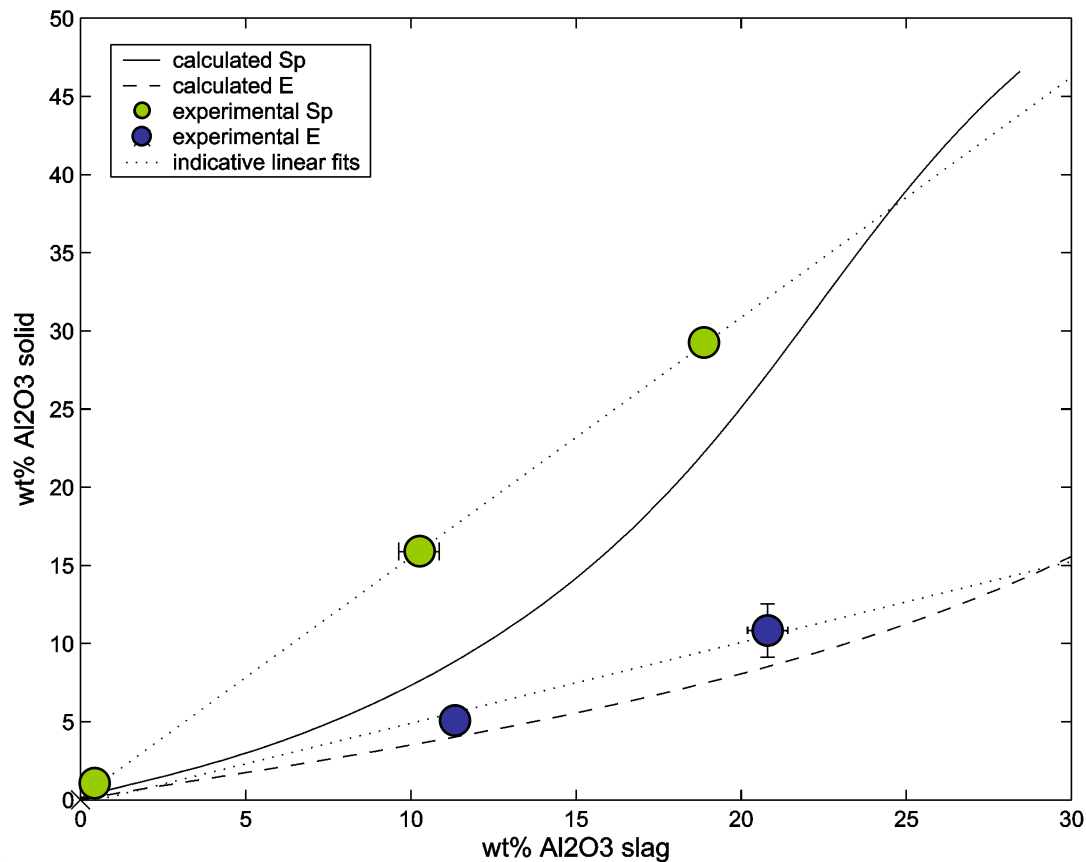
- Tests in known p_{O_2} , changing Al_2O_3



$T=1600^{\circ}C$
 $p_{O_2}=10^{-9.36}$ atm
 $B=1.2$

Results: solidus

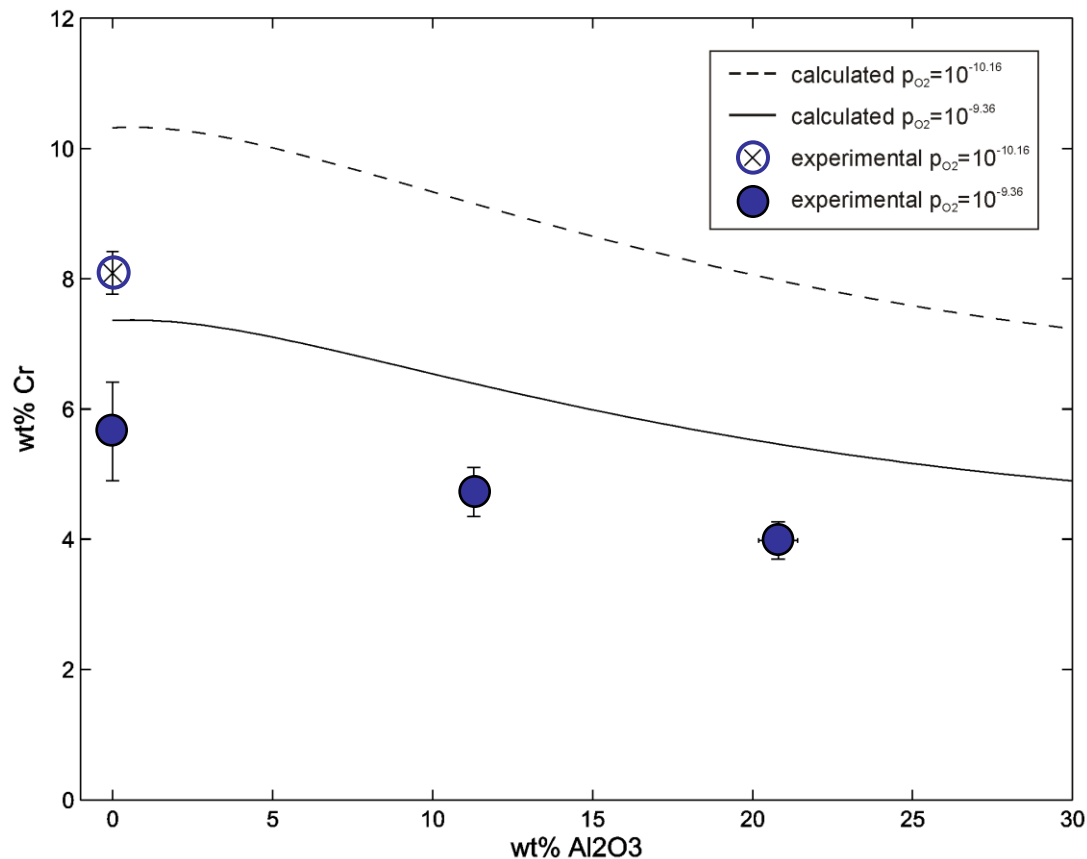
■ Relation between Al_2O_3 in slag and in solid



$T=1600^\circ\text{C}$
 $p_{\text{O}_2}=10^{-9.36}$ atm
 $B=1.2$
 (E: 0% MgO) ●
 (Sp: 9% MgO) ●

Results: eskolaite liquidus

■ Relation between Al_2O_3 and Cr_2O_3 in slag

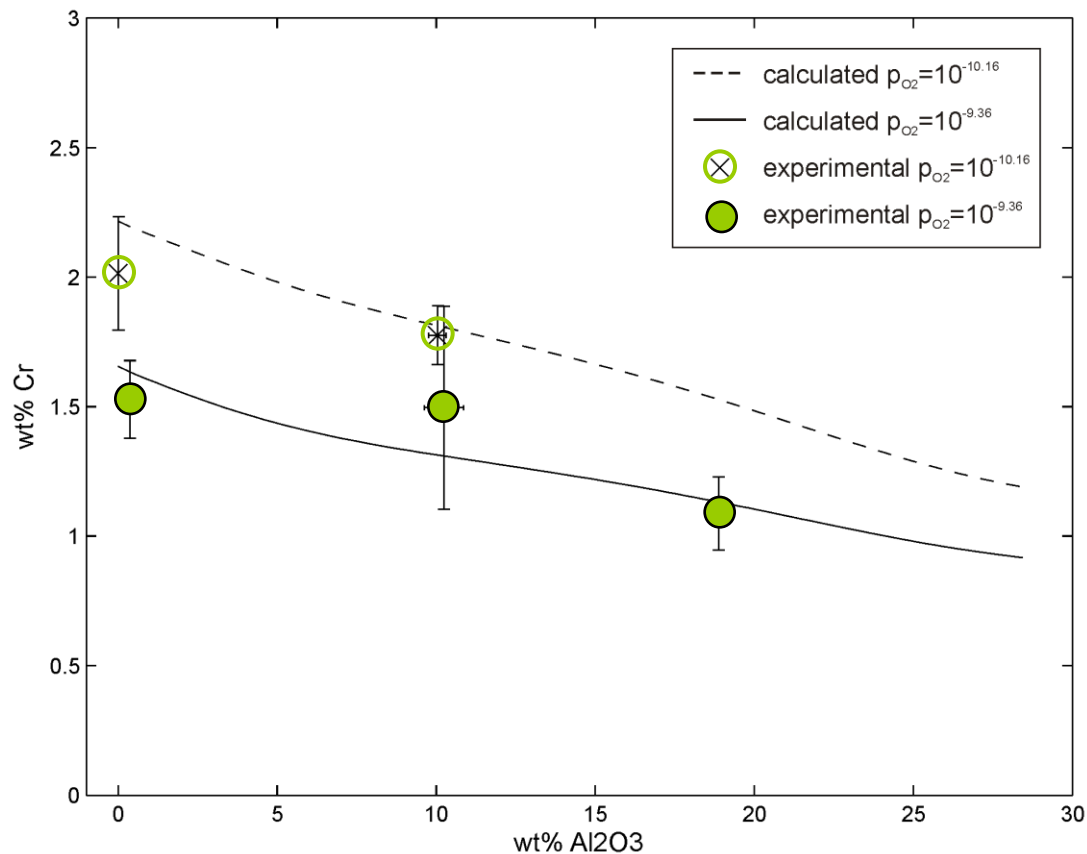


$T = 1600^\circ\text{C}$
 $p_{\text{O}_2} = 10^{-9.36}$ atm
 $B = 1.2$

E: 0% MgO

Results: spinel liquidus

■ Relation between Al_2O_3 and Cr_2O_3 in slag



$T=1600^\circ\text{C}$
 $p_{\text{O}_2}=10^{-9.36}$ atm
 $B=1.2$

Sp: 9% MgO

Discussion

- Sampling technique with nice results
- Good match with FactSage
- Improvement possible for
 - Eskolaite liquidus
 - Difference originates in ternary system
 - Solid compositions

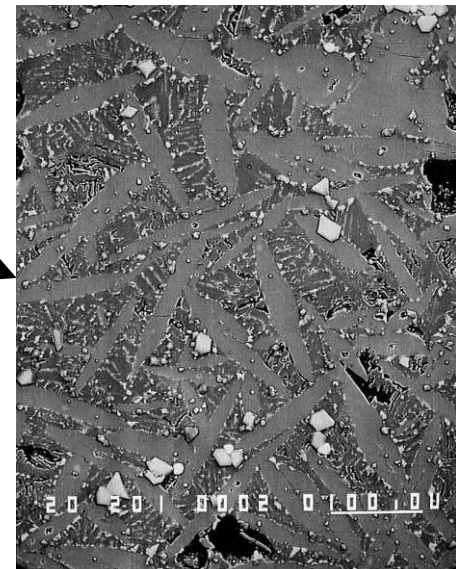
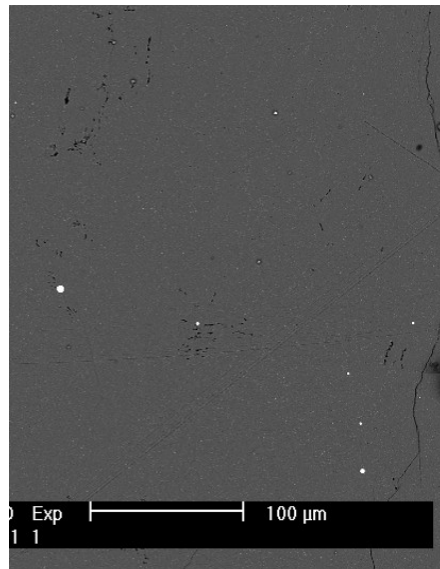
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Industrial slag solidification

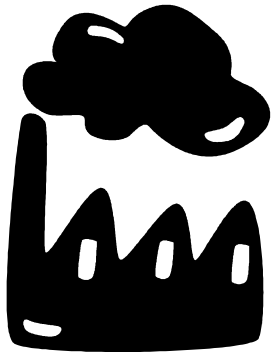


24 h



Research methodology

Industrial observation

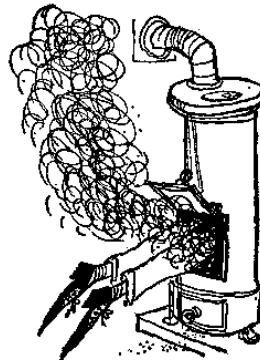


Complex system & cooling path

Relevant data:

- Microstructure after complete cooling

Laboratory experiment



Simple system & cooling path

Relevant data:

- Microstructure after complete cooling
- Microstructure during cooling

Simulation



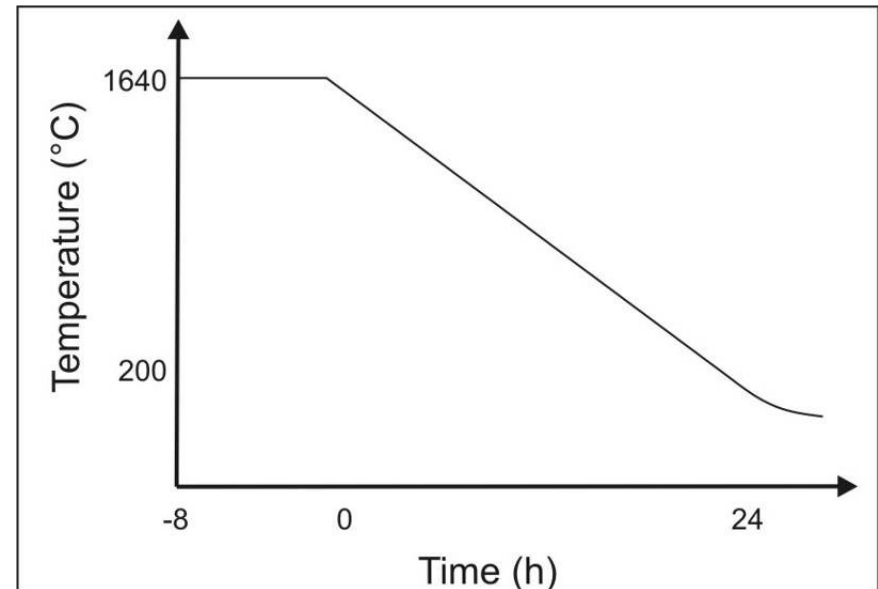
Known kinetic restrictions

Relevant data:

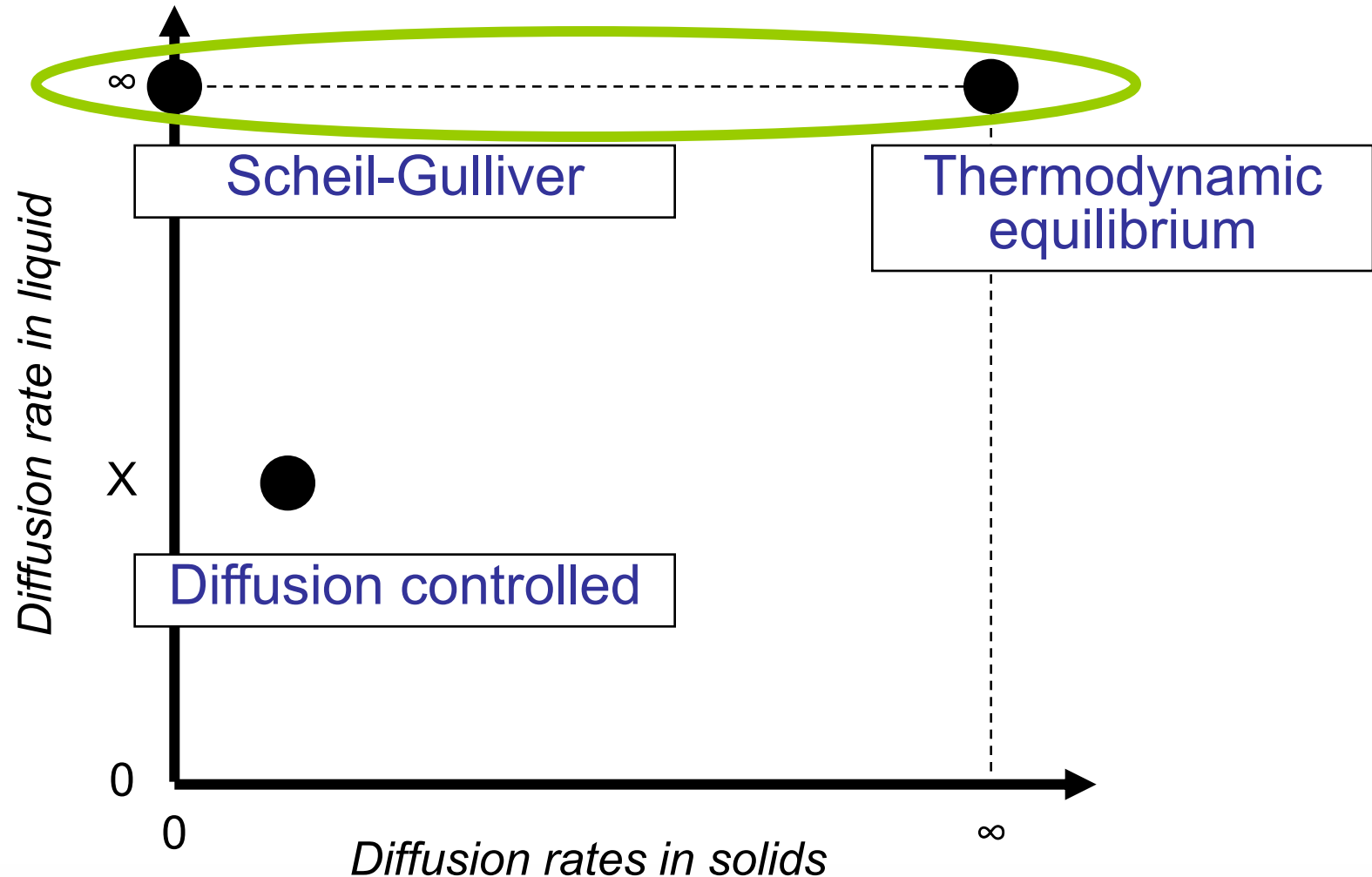
- Microstructure after complete cooling
- Microstructure during cooling

Laboratory experiment

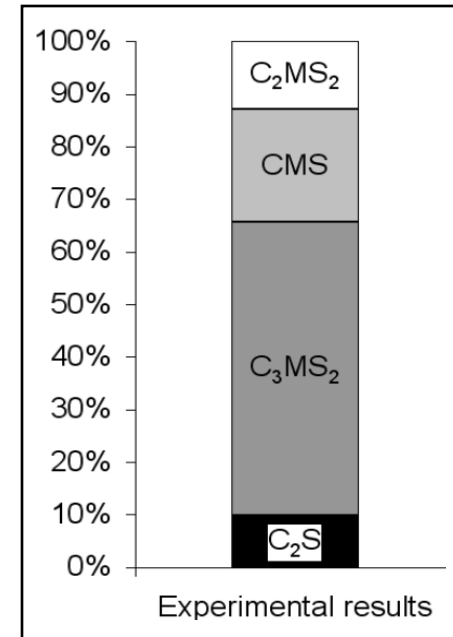
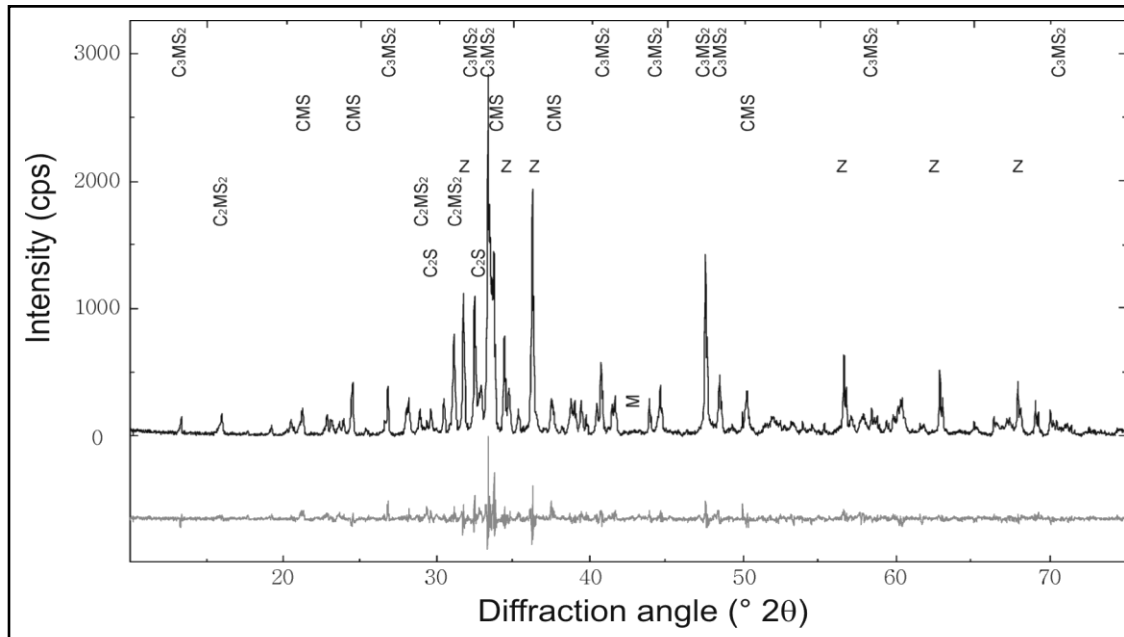
- Synthetic CaO-MgO-SiO₂ slag
- Thermal path
- Analysis techniques
 - Quantitative XRD (Rietveld analysis)
 - EPMA-EDS



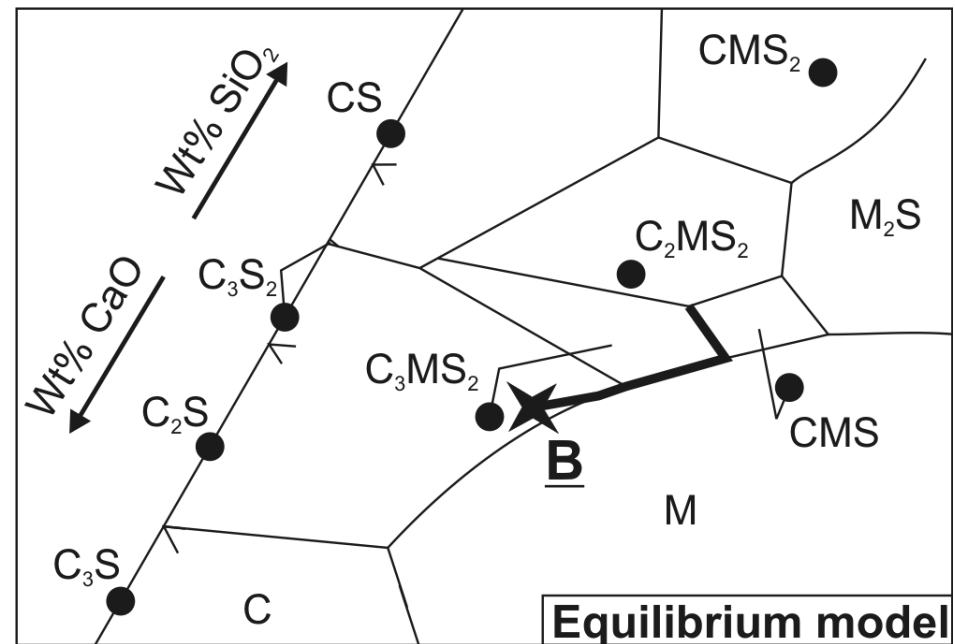
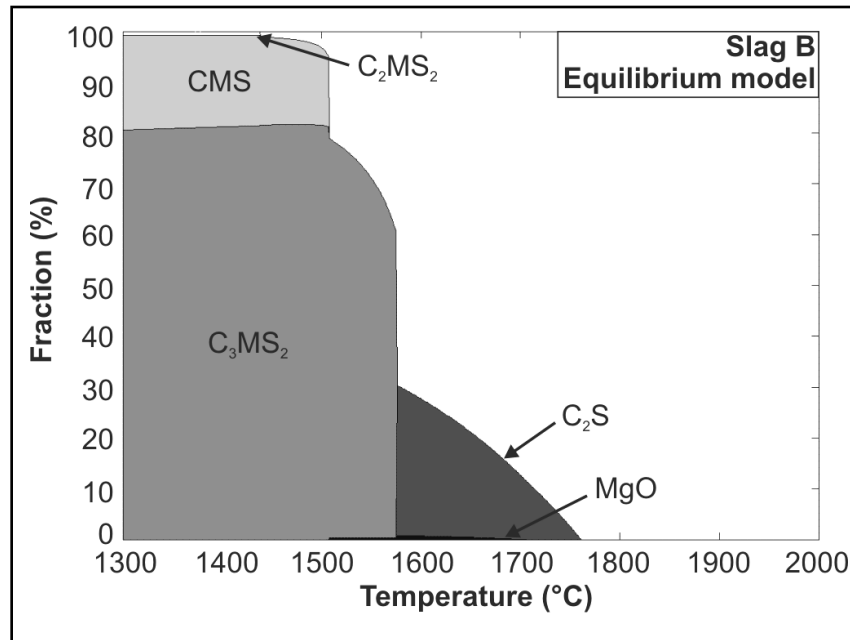
Common approaches to solidification modelling



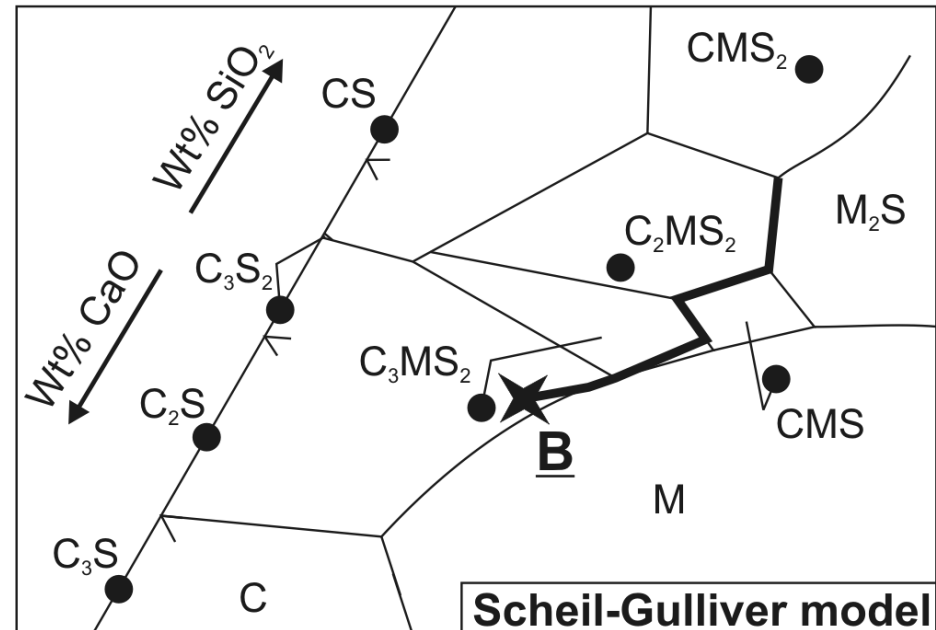
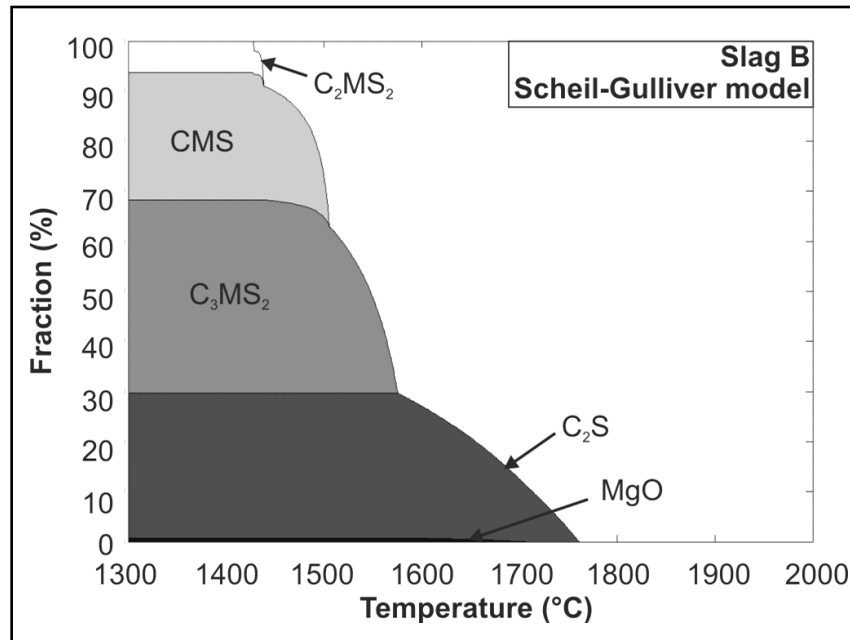
Mineralogy: Experiment



Mineralogy: Equilibrium

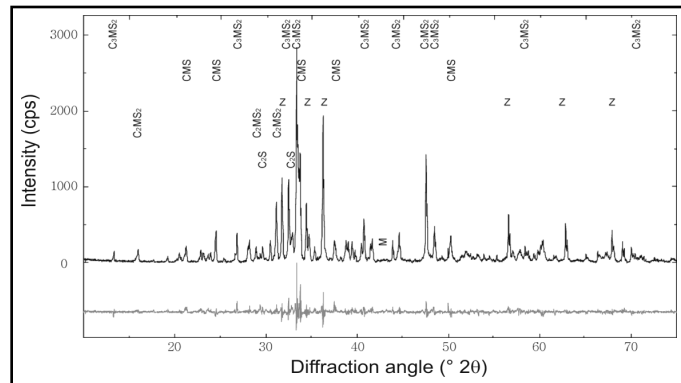


Mineralogy: Scheil-Gulliver

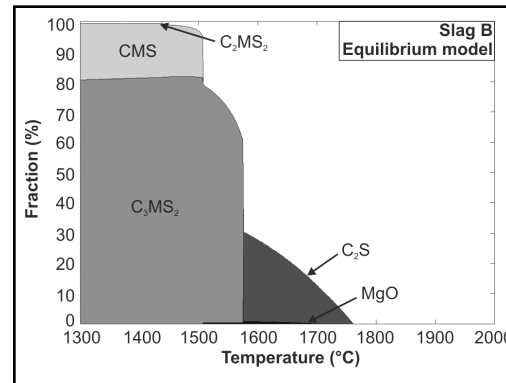


Mineralogy

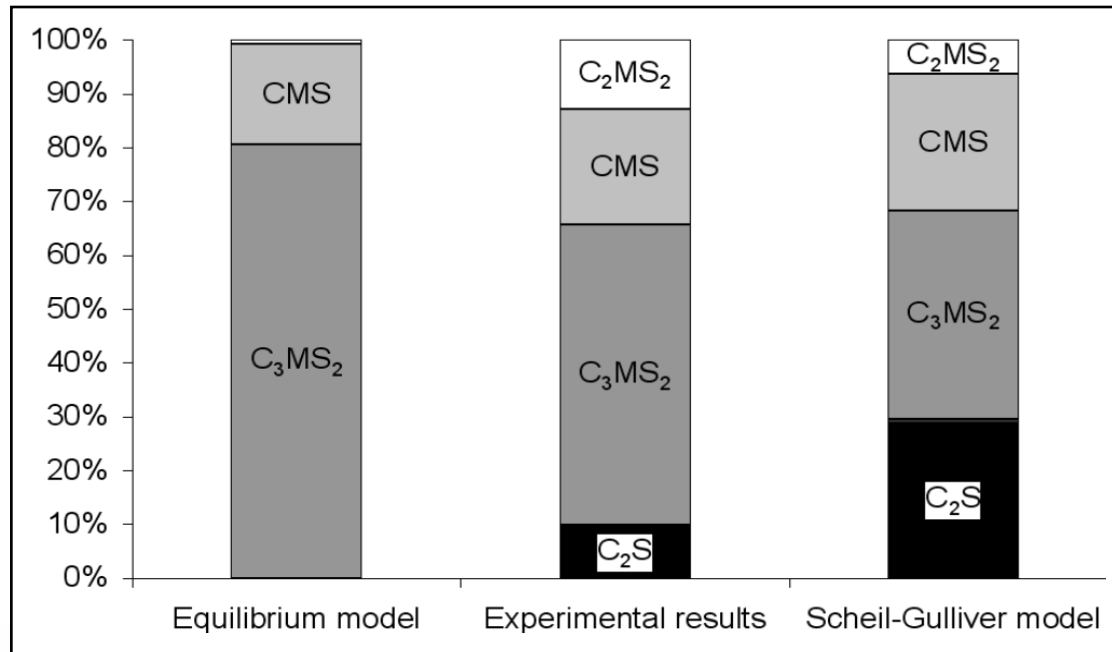
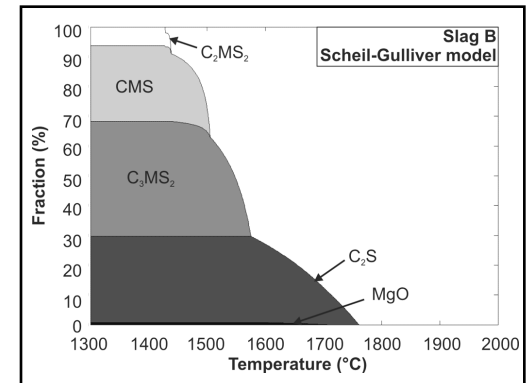
Experiment



Equilibrium



Scheil-Gulliver

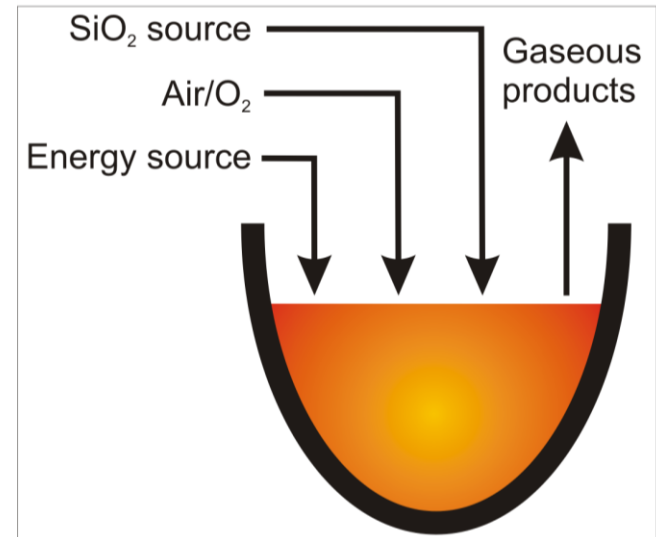


Content

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Slag post-treatment

- Idea:
 - Lower slag basicity
 - Avoid C_2S formation
 - Avoid subsequent slag dusting
- Main problem:
 - Limited heat capacity and conductivity
 - Limited potential to dissolve SiO_2 .
- Possible solution:
 - Co-inject energy source (metal + O_2)



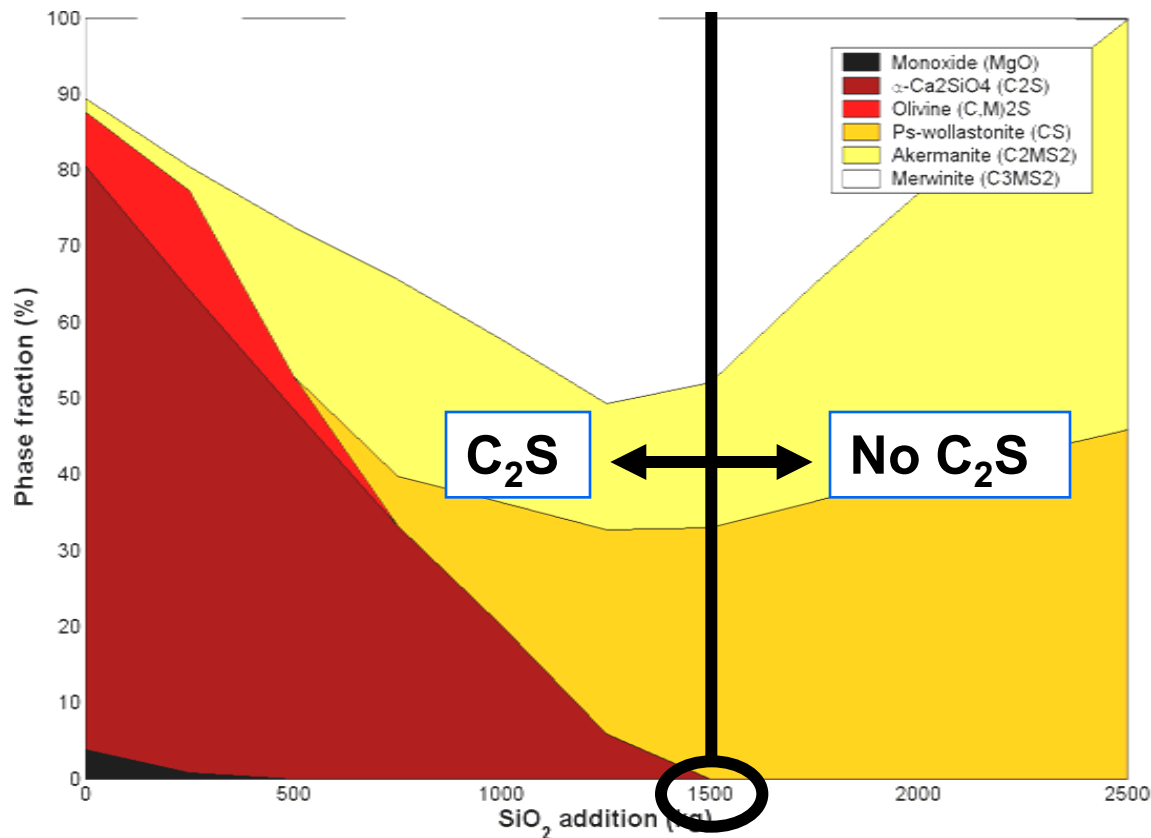
Slag post-treatment model

- Developed based on:
 - Solidification model (composition → microstructure)
 - Process model (additions → composition & temperature)

- Guidelines concerning:
 - Required SiO_2 addition
 - Al or FeSi as energy source
 - Air or oxygen as gas (stirring + oxidation)

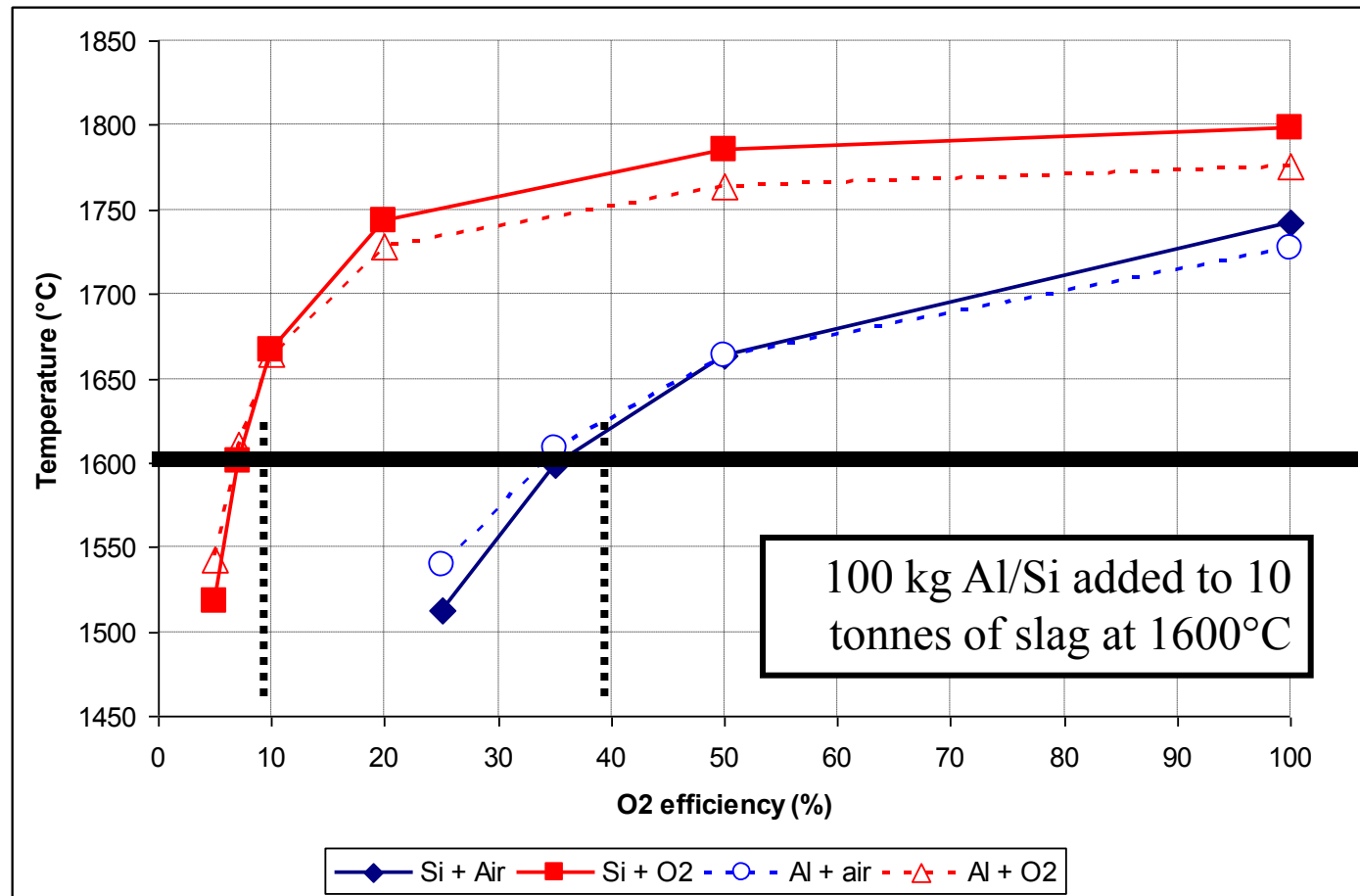
Slag post-treatment model

■ Required SiO₂ addition?



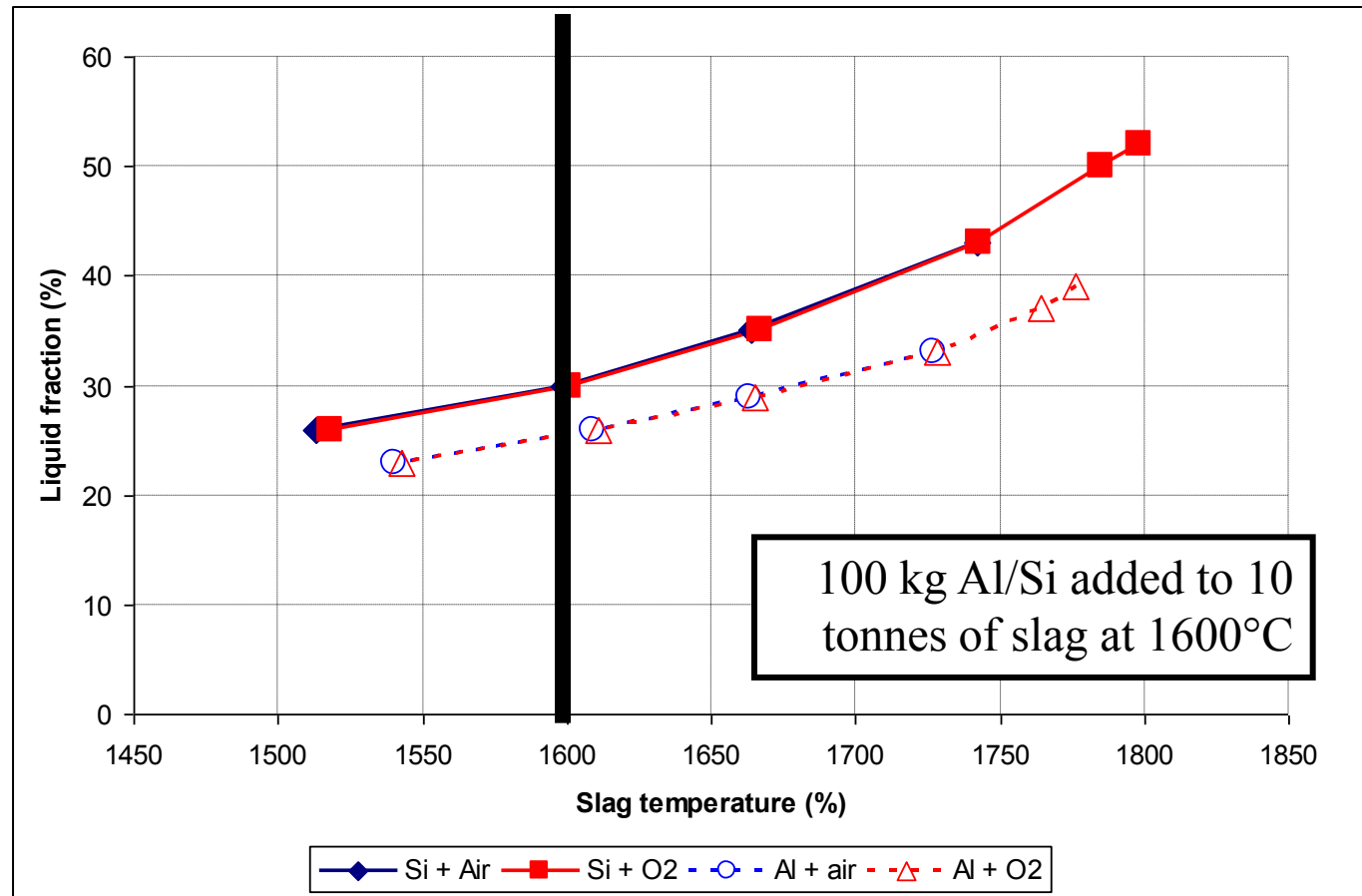
Enthalpy source

- Al or FeSi? → Heating effect



Enthalpy source

- Al or FeSi? → Slag liquidity



Guidelines from process model

■ Guidelines

Oxygen injection

- 100 kg of Si or 130kg of FeSi
- 340 kg or 250 Nm³ of O₂
- 1300 kg of SiO₂

- Practical implementation ongoing
- Macroscopic properties should not deteriorate

Conclusions

- Liquidus measurements:
 - Data at high temperature match well
- Cooling microstructure:
 - Easy assumptions give good approximation
- Process model:
 - Fast guidelines